1 Positively selected modifications in the pore of TbAQP2 allow pentamidine to enter

2 Trypanosoma brucei

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31 Abstract

Mutations in the Trypanosoma brucei aquaporin AQP2 are associated with resistance to 32 33 pentamidine and melarsoprol. We show that TbAQP2 but not TbAQP3 was positively selected for increased pore size from a common ancestor aquaporin. We demonstrate that 34 35 TbAOP2's unique architecture permits pentamidine permeation through its central pore and 36 show how specific mutations in highly conserved motifs affect drug permeation. Introduction 37 of key TbAQP2 amino acids into TbAQP3 renders the latter permeable to pentamidine. 38 Molecular dynamics demonstrates that permeation by dicationic pentamidine is energetically 39 favourable in TbAQP2, driven by the membrane potential, although aquaporins are normally 40 strictly impermeable for ionic species. We also identify the structural determinants that make 41 pentamidine a permeant but exclude most other diamidine drugs. Our results have wide-42 ranging implications for optimising antitrypanosomal drugs and averting cross-resistance. 43 Moreover, these new insights in aquaporin permeation may allow the pharmacological 44 exploitation of other members of this ubiquitous gene family.

45

46 Keywords:

47 Drug transport / Aquaporin / evolution of membrane proteins / Trypanosoma brucei /

48 pentamidine-melarsoprol cross-resistance

49 Introduction

50 The Trypanosoma brucei-group species are protozoan parasites that cause severe and fatal 51 infections in humans (sleeping sickness) and animals (nagana, surra, dourine) (Giordani et al, 52 2016; Büscher *et al*, 2017). The treatment is dependent on the sub-species of trypanosome, on 53 the host, and on the stage of the disease (Giordani et al, 2016; De Koning, 2020). Many antiprotozoal drugs are inherently cytotoxic but derive their selectivity from preferential uptake 54 55 by the pathogen rather than the host cell (Munday et al, 2015a; De Koning, 2020). Conversely, loss of the specific drug transporters is a main cause for drug resistance (Barrett 56 57 et al, 2011; Baker et al, 2013; Munday et al, 2015a; De Koning, 2020). This is the case for 58 almost all clinically used trypanocides, including diamidines such as pentamidine and 59 diminazene (Carter et al, 1995; De Koning, 2001a; De Koning et al, 2004; Bridges et al, 60 2007), melaminophenyl arsenicals such as melarsoprol and cymelarsan for cerebral stage 61 human and animal trypanosomiasis, respectively (Carter & Fairlamb, 1993; Bridges et al, 62 2007), and the fluorinated amino acid analogue effornithine for human cerebral 63 trypanosomiasis (Vincent et al, 2010). The study of transporters is thus important for anti-64 protozoal drug discovery programmes as well as for the study of drug resistance (Lüscher et 65 al, 2007; Munday et al, 2015a).

66 In Trypanosoma brucei, the phenomenon of melarsoprol-pentamidine cross-resistance 67 (MPXR) was first described shortly after their introduction (Rollo & Williamson, 1951), and was linked to reduced uptake rather than shared intracellular target(s) (Frommel & Balber, 68 69 1987). The first transporter to be implicated in MPXR was the aminopurine transporter TbAT1/P2 (Carter & Fairlamb, 1993; Mäser et al, 1999; Munday et al, 2015b) but two 70 71 additional transport entities, named High Affinity Pentamidine Transporter (HAPT1) and Low Affinity Pentamidine Transporter (LAPT1), have been described (De Koning, 2001a; 72 73 De Koning & Jarvis, 2001; Bridges et al, 2007). HAPT1 was identified as Aquaglyceroporin

74 2 (TbAQP2) via an RNAi library screen, and found to be the main determinant of MPXR 75 (Baker et al, 2012, 2013; Munday et al, 2014). The apparent permissibility for high molecular 76 weight substrates by TbAQP2 was attributed to the highly unusual selectivity filter of 77 TbAQP2, which lacks the canonical aromatic/arginine (ar/R) and full NPA/NPA motifs, 78 resulting in a much wider pore (Baker et al, 2012; Munday et al, 2014, 2015a). Importantly, 79 the introduction of TbAOP2 into *Leishmania* promastigotes greatly sensitised these cells to pentamidine and melarsen oxide (Munday et al, 2014). Moreover, in several MPXR 80 81 laboratory strains of T. brucei the AQP2 gene was either deleted or chimeric after cross-over 82 with the adjacent TbAOP3 gene, which, unlike AQP2, contains the full, classical ar/R and 83 NPA/NPA selectivity filter motifs and is unable to transport either pentamidine or 84 melaminophenyl arsenicals (Munday et al, 2014). Similar chimeric genes and deletions were 85 subsequently isolated from sleeping sickness patients unresponsive to melarsoprol treatment 86 (Graf et al, 2013; Pyana Pati et al, 2014) and failed to confer pentamidine sensitivity when 87 expressed in a tbaqp2-tbaqp3 null T. brucei cell line whereas wild-type TbAQP2 did 88 (Munday et al, 2014; Graf et al, 2015).

The model of drug uptake through a uniquely permissive aquaglyceroporin (Munday 89 90 et al, 2015a) was challenged by a study arguing that instead of traversing the TbAQP2 pore, 91 pentamidine merely binds to an aspartate residue (Asp265) near the extracellular end of the 92 pore, above the selectivity filter, followed by endocytosis (Song et al, 2016). This alternative, 93 'porin-receptor' hypothesis deserves careful consideration given that (i) it is an exceptional 94 assertion that drug-like molecules with molecular weights grossly exceeding those of the 95 natural substrates, can be taken up by an aquaglyceroporin and (ii) the fact that bloodstream 96 form trypanosomes do have, in fact, a remarkably high rate of endocytosis (Field & 97 Carrington, 2009; Zoltner et al, 2016). The question is also important because aquaporins are 98 found in almost all cell types and the mechanism by which they convey therapeutic agents

and/or toxins into cells is of high pharmacological and toxicological interest. While TbAQP2
is the first aquaporin described with the potential to transport drug-like molecules, this ability
might not be unique, and the mechanism by which the transport occurs should be carefully
investigated.

103 We therefore conducted a mutational analysis was undertaken, swapping TbAQP2 and TbAOP3 selectivity filter residues and altering pore width at its cytoplasmic end. This 104 105 was complemented with a thorough structure-activity relationship study of the interactions 106 between pentamidine and TbAQP2, using numerous chemical analogues for which inhibition 107 constants were determined and interaction energy calculated. The pentamidine-TbAQP2 108 interactions were further modelled by running a molecular dynamics simulation on a protein-109 ligand complex, and In addition, we investigated a potential correlation between the T. brucei 110 endocytosis rate and the rate of pentamidine uptake. Our results unequivocally show that 111 pentamidine permeates directly through the central pore of TbAQP2 and that uptake is 112 dependent on the microbial membrane potential. Having identified the essential 113 characteristics that allow the transport of large, flexible molecules through TbAQP2, this 114 should now allow the evaluation of aquaporins in other species for similar adaptations.

115 **Results**

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117 12. Investigation of the structural determinants of AQP2 for pentamidine transport

118 *1.1. Positive selection for pore size*

119 In T. brucei, the AQP2 and AQP3 genes are arranged as a tandem pair on chromosome 10 120 and have 74% amino acid identity. Whereas TbAQP2 clearly mediates pentamidine uptake, 121 TbAQP3 does not (Baker et al, 2012; Munday et al, 2014), nor do various chimeric AQP2/3 122 rearrangements that give rise to pentamidine resistance (Munday et al, 2014; Graf et al, 123 2015). To investigate the origin of the AQP2 gene, a phylogenetic analysis of AQPs in 124 African trypanosomes was performed. The number of aquaporin genes varies: there is a 125 single aquaporin in T. vivax and T. congolense, two in T. suis and three in T. brucei and its 126 derivatives (Supplemental Fig. 1A). The most probable tree (Supplemental Fig. 1B) is 127 consistent with the evolutionary history of the four species (Hutchinson & Gibson, 2015) and 128 indicates AQP1 as the ancestral AQP present in all trypanosome species. A duplication 129 occurred in the common ancestor of T. suis and T. brucei after divergence from T. congolense and a further duplication, to form AQP2 and AQP3, in the ancestor of T. 130 131 brucei after divergence from T. suis. Multiple alignment (Supplemental Fig. 1A) shows that 132 the classical NPA/NPA and ar/R AQP selectivity filter elements are present in all AQPs 133 except T. brucei AQP2. The divergence of T. brucei AQP2 and 3 was investigated by 134 calculating the non-synonymous/synonymous codon ratio (dN/dS) for different AQPs 135 (Supplemental Fig. 1C). For T. brucei aligned with T. suis AQP1, dN/dS is 0.21 and for 136 AQP3, dN/dS is 0.30 indicating purifying selection. However, comparing T. brucei AQP2 137 with T. brucei AQP3, dN/dS is 2.0 indicating strong selection pressure for divergence on 138 AQP2 towards an aquaporin with increased pore size. In order to verify any role of amino

acids along the TbAQP2 pore in facilitating pentamidine sensitivity and/or uptake, weperformed a mutational analysis.

141

142 1.2 Introduction of AQP3 residues into the AQP2 selectivity filter

143 One highly conserved motif of aquaporins, believed to be essential for permeant selectivity, is NPA/NPA which is present in TbAOP3 but not in TbAOP2, where, uniquely, it is 144 NS¹³¹A/NPS²⁶³ instead. We therefore constructed a TbAQP2 variant with the classical 145 NPA/NPA motif (TbAQP2^{S131P/S263A}) and expressed it in the *aqp2/aqp3* null cell line (Baker 146 147 et al, 2012; Munday et al, 2014). In this cell line, uptake of 30 nM [³H]-pentamidine was 148 reduced to $4.40 \pm 0.71\%$ (n=4) of the rate in the control line expressing TbAQP2WT (P<0.05, 149 Student's unpaired t-test), as well as significantly different from the rate measured in parallel 150 in the tbaqp2/tbaqp3 null cells (P<0.01) (Fig. 1A). The remaining pentamidine uptake was sufficient to strongly sensitise the TbAQP2^{S131P/S263A} cells to pentamidine in a standard 151 152 protocol of 48 h incubation with the drug followed by a further 24 h in the presence of the resazurin indicator dye (P<0.0001 vs *tbaqp2/tbaqp3* null) but the EC₅₀ was still significantly 153 higher than the TbAOP2WT control (P<0.05) (Fig. 1B). A similar effect was observed for the 154 155 melaminophenyl arsenical drug cymelarsan, but there was no change in sensitivity to diminazene or the control drug phenylarsine oxide (PAO), which is believed to diffuse 156 157 directly across the membrane (Fairlamb et al, 1992) (Fig. 1B).

The mutant L258Y, which has the AQP3 Tyr-250 half of the highly conserved aromatic/arginine (ar/R) motif, responsible for pore restriction and proton exclusion (Wu *et al*, 2009), introduced into the TbAQP2 pore, yielded a drug transport phenotype similar to TbAQP2^{S131P/S263A}. The [³H]-pentamidine transport rate was reduced to $6.6 \pm 1.4\%$ of TbAQP2WT (P<0.05) but remained above the rate in the *tbaqp2/tbaqp3* null cells (P<0.01)

163 (Fig. 1A). Pentamidine and cymelarsan EC_{50} values were also significantly different from 164 both the TbAQP2WT and the *tbaqp2/tbaqp3* null controls (Fig. 1C).

The ar/R motif is part of the larger selectivity filter, usually WGYR, present in both 165 TbAQP1 and TbAQP3 but uniquely consisting of I¹¹⁰VL²⁵⁸L²⁶⁴ in TbAQP2 (Baker et al, 166 2013), all non-polar, open chained residues. Cell lines expressing mutations AQP2^{I110W} and 167 AOP2^{L264R}, either alone or in combination, displayed pentamidine transport rates, and 168 169 pentamidine and cymelarsan EC₅₀ values that were not significantly different from the 170 tbaqp2/tbaqp3 null controls but highly significantly different from the TbAQP2WT drug-171 sensitive controls, showing that their capacity for pentamidine and cymelarsan uptake had 172 been reduced to zero (Fig. 1A, D-F).

We conclude that the unique TbAQP2 replacement of the NPA/NPA motif and all of the WGYR selectivity filter mutations are necessary for the observed pentamidine and melaminophenyl arsenical sensitivity observed in cells expressing wild-type TbAQP2.

176

177 1.3 Introduction of TbAQP2 selectivity filter residues into the AQP3 pore enables
178 pentamidine transport

179 An interesting question was whether the introduction of (some of) the critical TbAQP2 residues in TbAQP3 might give the latter the capacity to take up pentamidine. We 180 therefore constructed TbAQP3^{W102I/R256L} and TbAQP3^{W102I/R256L/Y250L} and tested whether 181 *tbaqp2/tbaqp3* null cells transfected with these mutant aquaporins were able to take up 25 nM 182 ³H]-pentamidine in the presence of 1 mM adenosine (which blocks uptake via TbAT1/P2). 183 184 Pentamidine uptake in the tested cell lines was very low compared to the same cells expressing TbAQP2WT (Fig. 1G). However, by measuring $[^{3}H]$ -pentamidine uptake over 30 185 min it was possible to reliably and reproducibly measure radiolabel accumulation in each cell 186 line. This showed that while uptake in TbAQP3^{W102I/R256L} only trended slightly upwards 187

188	(P>0.05), the mutant AQP3 with all three AQP2 WGYR residues (W102I, R256L and
189	Y250L) accumulated significantly more $[^{3}H]$ -pentamidine than the <i>tbaqp2/tbaqp3</i> null cells
190	(P<0.01) or the null cells expressing TbAQP3WT (P=0.011). This is further corroborated by
191	comparing the pentamidine sensitivity profile of these cell lines: only TbAQP3 ^{W102I/R256L/Y250L}
192	conveyed significant sensitisation to tbaqp2/tbaqp3 null cells (P<0.0001; Fig. 1H). Thus,
193	TbAQP3 is converted into a pentamidine transporter by the insertion of the AQP2 WGYR
194	residues, although this does not convey as high a rate of pentamidine uptake as TbAQP2.
195	

196 *1.4 Mutations of amino acids modelled to potentially bind pentamidine or melarsoprol*197 *dramatically reduce pentamidine transport*

Our previous attempts at modelling the binding of pentamidine and melarsoprol into the pore of TbAQP2 tentatively identified several residues that could be involved in this process (Munday *et al*, 2015a), from which we selected two residues, Ile190 and Trp192, at the extracellular end of the channel (position shown in Fig. 6), to swap with the corresponding residues of TbAQP3, creating TbAQP2^{I190T} and TbAQP2^{W192G}. Both residues were predicted to interact with the substrate(s) via main-chain carbonyl oxygen atoms, but the side chains could nonetheless affect the interactions.

TbAQP2^{I190T} displayed dramatically reduced [³H]-pentamidine uptake, at $2.7 \pm 0.7\%$ 205 206 (P < 0.01, n=4) of the TbAQP2WT control, although significantly higher than the rate of the tbaqp2/tbaqp3 null negative control (P<0.05) (Fig. 2A). The reduced rate was the result of a 207 reduced V_{max} of the high affinity [³H]-pentamidine uptake, rather than a change in Km; the 208 LAPT1 V_{max} and K_m were unchanged in cells expressing TbAQP2^{I190T} or TbAQP2WT 209 (Supplemental Fig. S2). TbAQP2^{1190T} still conferred some increased pentamidine sensitivity 210 211 in the standard resazurin test (P<0.0001), although highly significantly less sensitizing than 212 TbAQP2WT (P<0.001); an intermediate sensitivity was also observed for cymelarsan (Fig.

213 2B). Substitution W192G also produced intermediate sensitivity to both drugs (Fig. 2C) but
214 the double substitution TbAQP2^{I190T/W192G} displayed no significant pentamidine uptake above
215 *tbaqp2/tbaqp3* null (Fig. 3A) and did not sensitise to pentamidine or cymelarsan (Fig. 2D).
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217 *1.5. The effect of large amino acids at the cytoplasmic end of the pore.*

218 To test whether restrictions at the cytoplasmic end of TbAOP2 would impact on pentamidine 219 transport, we selected three leucine residues and exchanged each with tryptophan, creating 220 L84W, L118W and L218W (positions indicated in Fig. 6). Expressing each of the L-to-W 221 mutants in *tbaqp2/tbaqp3* null cells revealed significantly reduced pentamidine sensitivity 222 compared to the same cells expressing TbAQP2WT (Fig. 3A), while also exhibiting 223 dramatically reduced rates of [³H]-pentamidine transport (Fig. 3B). This effect was additive, with TbAQP2^{L84W/L118W} not significantly sensitising for pentamidine and displaying no 224 detectable increase in [³H]-pentamidine transport relative to *tbagp2/tbagp3* null cells (Fig. 225 226 3A,B). None of these L-to-W mutants sensitised the cells to cymelarsan, diminazene or PAO 227 (Supplemental Fig. S3). When the same leucine residues were replaced with methionine 228 instead of tryptophan, variants L84M and L218M were not or barely different from 229 TbAQP2WT with respect to pentamidine sensitisation (Fig. 3A) or transport (but highly significantly different from their respective tryptophan variants). For position 118 the Met 230 231 replacement had similar effects as the Trp variant had, albeit with a significantly higher rate of pentamidine transport (1.88 \pm 0.20 (n=6) versus 9.38 \pm 0.63% (n=3) of TbAQP2WT, 232 233 P < 0.001; Fig. 3A,B). The L84M and L218M mutants also sensitised to cymelarsan (P < 0.01) 234 and, surprisingly, the L218 W and M mutants also sensitised slightly to diminazene (~2-fold, 235 P<0.05) (Supplemental Fig. S3).

236 These results strongly argue that the introduction of large amino acids at the cytosolic237 end significantly blocks the transport of pentamidine, whereas the change to Leu→Met

238 mutants were more permissive for pentamidine, but not cymelarsan. In order to check 239 whether these variants were still functional aquaglyceroporins, we used the observation of Jeacock et al, (2017) that T. brucei cells lacking all three AQPs are sensitised to the 240 Trypanosome Alternative Oxidase inhibitor SHAM, as a result of cellular glycerol 241 242 accumulation. By this measure, all of the position 84, 118 and 218 Trp and Met mutants were 243 able to transport glycerol, as each displayed SHAM EC_{50} values significantly different from 244 the *tbaqp1-2-3* null cells (Fig. 3C,D); several variants displayed an intermediate SHAM EC₅₀, being also significantly different from TbAQP2WT, indicating some attenuation of glycerol 245 efflux capacity for those mutants. Indeed, uptake of $[^{3}H]$ -glycerol closely mirrored the 246 247 SHAM observations (Fig. 3E).

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249 1.6. Overall correlation between $[{}^{3}H]$ -pentamidine transport rate and pentamidine EC_{50}

The results presented in Figures 2-4 consistently show that even TbAQP2 mutants that 250 251 display a large reduction in $[^{3}H]$ -pentamidine uptake rate results can show intermediate 252 pentamidine sensitivity phenotypes (EC₅₀s), due to the nature of the standard drug sensitivity 253 test employed, which involves a 48-h incubation with the drug prior to a further 24-h 254 incubation with resazurin: even a much-reduced transport rate will be sufficient to accumulate significant amounts of intracellular pentamidine over 3 days. A plot of [³H]-255 256 pentamidine transport rates versus pentamidine EC_{50} , using the data for all 19 TbAQP2 and 257 TbAQP3 mutants for which the transport rates were determined, shows that relatively small 258 changes in EC₅₀ occur, even with up to approximately 95% reduction in transport rates; at >95% reduction large EC₅₀ increases become apparent (Supplemental Fig. S4). 259

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261 2. Partially blocking endocytosis does not alter the rate of pentamidine transport

262 The knockdown of the CRK12 kinase in T. brucei causes a highly reproducible defect in 263 endocytosis that affects an estimated one third of cells 12 h after RNAi induction and is 264 ultimately lethal (Monnerat et al, 2013). We utilized this system to investigate whether a link 265 between endocytosis and pentamidine transport exists. At 12 h of CRK12 RNAi induction 266 with tetracycline, CRK12 mRNA levels were reduced by 42% (P<0.001) relative to uninduced controls as determined by qRT-PCR (Fig. 4A). Samples from the culture taken at 267 268 this time point showed an increased abundance of cells with swelling characteristic of endocytosis defects, although this was hard to quantify as a minority of cells were affected, 269 270 and to various degrees, as the 12 h time point was deliberately taken as an early point that 271 would not yet affect cell viability (Supplemental Fig. S5) or cause excessive cellular 272 pathology. We thus performed parallel uptake experiments with [³H]-pentamidine and [³H]-273 suramin, with suramin acting as positive control as it is known to enter T. brucei bloodstream 274 forms through endocytosis after binding to surface protein ISG75 (Zoltner et al, 2016). After 275 12 h of CRK12 RNAi induction, pentamidine uptake was not significantly less than in the T. *brucei* 2T1 parental cells, whereas uptake of $[^{3}H]$ -suramin was (P=0.019, n=5; Fig. 4B,C). 276

277

3. The protonmotive force drives AQP2-mediated pentamidine uptake in bloodstream forms
of T. brucei

It has been reported that knock-down of the HA1–3 plasma membrane proton pumps of *T*. *brucei* (which are essential for maintaining the plasma membrane potential), confers pentamidine resistance (Alsford *et al*, 2012; Baker *et al*, 2013). Interestingly, this locus only conferred resistance to (dicationic) pentamidine, not to the (neutral) melaminophenyl arsenicals, unlike knockdown of the TbAQP2/TbAQP3 locus (Alsford *et al*, 2012). We have previously reported that the HAPT-mediated pentamidine uptake in *T. brucei* procyclics correlates strongly with the proton-motive force (PMF) (De Koning, 2001a). However, it is

287 not clear whether this dependency indicates that pentamidine uptake is mediated by a proton 288 symporter, as known for many T. brucei nutrient transporters (De Koning & Jarvis, 1997a,b, 289 1998; De Koning et al, 1998), or reflects the energetics of uptake of cationic pentamidine being driven by the strong inside-negative membrane potential V_m. The absence of an effect 290 291 of HA1–3 knockdown on sensitivity to the neutral melaminophenyl arsenicals strongly argues 292 against a mechanism of proton symport for HAPT1/AQP2 but a (partial) dependency of 293 HAPT1/AQP2-mediated uptake of dicationic pentamidine on PMF or V_m would be expected 294 if the substrate traverses the channel, as opposed to binding a single Asp residue on the 295 extracellular side of the protein, as suggested in the endocytosis model (Song et al. 2016). 296 Here we show that the same ionophores that inhibit HAPT1-mediated pentamidine transport 297 in procyclic cells, and inhibit hypoxanthine uptake in both bloodstream form (BSF) (De 298 Koning & Jarvis, 1997b) and procyclic (De Koning & Jarvis, 1997a) T. brucei, also dose-299 dependently inhibit [³H]-pentamidine uptake in BSF (Fig. 5A). This confirms that 300 pentamidine needs the membrane potential for rapid uptake, as predicted by the dependence on the HA1-3 proton pumps. Using $[{}^{3}H]$ -suramin as an endocytosed substrate (Zoltner *et al*, 301 2016), we found that 20 µM CCCP also inhibits endocytosis in T. brucei, by 32.6% 302 303 (P=0.029; pre-incubation 3 min, plus suramin accumulation over 10 minutes) (Fig. 5B). 304 While that means that the ionophore experiments do not perfectly discriminate between 305 endocytosis and trans-channel transport for di-cationic pentamidine, they do for neutral 306 melaminophenyl arsenicals: the non-dependence of these neutral TbAQP2 substrates on the 307 proton gradient (Alsford et al, 2012) indicates that, unlike suramin, they are not endocytosed. 308 Although there is a good correlation between the proton-motive force and TbAQP2-309 mediated pentamidine transport (Fig. 5C), the effect of CCCP was stronger than expected, 310 and stronger than previously observed for $[^{3}H]$ -hypoxanthine uptake in *T. brucei* bloodstream

311 forms (De Koning & Jarvis 1997b) and we thus investigated whether CCCP might have a

direct effect on TbAQP2. Indeed, CCCP inhibited uptake of (neutral) [3 H]-glycerol in *tbaqp1-2-3* null cells expressing TbAQP2-WT, with an IC₅₀ of 20.7 ± 2.6 μ M (n=3) and inhibited [3 H]-pentamidine uptake in the same cells with a similar IC₅₀ (Fig. 5D,E), showing CCCP to inhibit TbAQP2 directly, irrespective of effects on the membrane potential.

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317 4. Molecular dynamics modelling of pentamidine interactions with TbAQP2

318 To further investigate pentamidine binding and permeation in TbAQP2, we used the 319 coordinates of the TbAQP2-pentamidine complex that was modelled in our previous study 320 (Munday et al, 2015a). The stability of the protein model was first confirmed by unbiased 321 atomistic molecular dynamics simulations (Supplemental Fig. S6). We then conducted force-322 probe simulations, in which a moving spring potential was used to enforce unbinding of 323 pentamidine from its docked binding position and subsequently reconstructed the free-energy 324 profile of pentamidine association-dissociation along the pore axis by employing Jarzynski's 325 equality (Park et al, 2003).

326 Figure 6 shows that the docked position of pentamidine correctly identified its 327 minimum free-energy binding site inside the TbAQP2 pore. Pentamidine adopts an extended 328 state inside the TbAQP2 pore, adapting its molecular shape to the narrow permeation channel; pentamidine binding poses display inter-amidine lengths in the range 16.5 - 17 Å. 329 330 Importantly, our steered simulations reveal that pentamidine can exit the channel in either 331 direction, and that unbinding on the route towards the cytoplasm occurs on a free-energy 332 surface roughly symmetric to that towards the extracellular side. Apart from overcoming the 333 strongly attractive binding interaction in the centre, there are no major further free-energy 334 barriers in either direction. The computed free-energy profile of pentamidine binding to the 335 TbAQP2 structural model slightly overestimates its experimentally recorded binding affinity. 336 However, the pentamidine conformation binding the narrow pore may not be the lowest-

energy internal conformation of the small molecule, a factor that may be underrepresented in
the profile as simulations were started from the protein-bound state. A further source of
uncertainty stems from the protein model, which is expected to be somewhat less accurate
than a crystal structure.

341 Due to the dicationic character of pentamidine, the free-energy profile of the molecule within TbAOP2 strongly depends on the membrane voltage. The voltage drop of -125 mV 342 343 across the cytoplasmic membrane of T. b. brucei (De Koning & Jarvis, 1997b), with a negative potential inside the cell, results in an overall inward attraction of ~ 22 kJ/mol (Fig. 6, 344 345 arrow), i.e. exit from TbAQP2 into the cytoplasm is substantially more favourable for 346 pentamidine than towards the extracellular side. Taken together, the free-energy profile under 347 membrane voltage explains the strong coupling between pentamidine uptake and V_m 348 observed in the experiments. The high affinity of the binding interaction leads to slow offrates and a relatively low V_{max} (0.0044 ± 0.0004 pmol(10⁷ cells)⁻¹s⁻¹) (De Koning, 2001a). 349

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351 5. SAR of the pentamidine-AQP2 interaction

352 In order to study substrate binding and selectivity by the T. b. brucei High Affinity 353 Pentamidine Transporter (HAPT1/TbAQP2), competition assays were performed with a series of pentamidine analogues and other potential inhibitors, in the presence of 1 mM 354 355 unlabelled adenosine to block diamidine uptake by the TbAT1 aminopurine transporter (De Koning, 2001a; Bridges *et al*, 2007). High specific activity [³H]-pentamidine was used at 30 356 nM, below the K_m value (De Koning, 2001a). Uptake was linear for at least 3 min (De 357 358 Koning, 2001a) and we utilized 60 s incubations for the determination of inhibition constants 359 (K_i). At 30 nM $[^{3}H]$ -pentamidine there is virtually no uptake through LAPT1 (Bridges *et al*, 360 2007) (K_m value ~1000-fold higher than HAPT1) (De Koning, 2001a). The full dataset of 71 361 compounds is presented in Supplemental Table S1, featuring K_is spanning 5 log units.

362

363 5.1. The linker length and composition is a strong determinant for high affinity binding of364 pentamidine

We determined the K_i values for analogues with a 2–8 methylene unit linker (Fig. 7A, Table 365 1). Pentamidine analogues featuring 5-7 units displayed submicromolar binding affinities (5 366 > 6 > 7), while fewer (3-4) or more (8) only conveyed low micromolar binding affinity, 367 equivalent to a decrease in Gibbs free energy of binding (ΔG^0) from 10.2 to 13.0 kJ/mol 368 (Table 1). Energy minimalization using Gaussian16 yielded an elongated conformation for 369 370 pentamidine, with an inter-amidine length of 17.8 Å (Fig. 7B). Replacement of the ether oxygens with S or NH, analogues RT-48 and RT-50, respectively (Table 1) resulted in 371 372 $\delta(\Delta G^0)$ of 10.0 and 12.9 kJ/mol, respectively, indicating that the ether oxygens potentially act 373 as H-bond acceptors: the NH group serves only as an H-bond donor, as its lone pair is 374 conjugated with the aromatic system, and the sulphur mimics an aromatic NH (Beno et al, 375 2015). The sulfone analogue (RT-49), which introduces a dihedral angle of 180° between the 376 benzamidine and the linker (Brameld et al, 2008), displayed no binding affinity. We propose that a near-planar conformation of the Phe-O-CH₂ segment is required for efficient 377 engagement of the binding site. This is supported by examining the binding affinities found 378 379 for the analogous benzofuramidine series (e.g. RT-14, Fig. 7B), which has a conformationally predefined ether-methylene bond orientation. Replacement of the middle methylene unit of 380 381 pentamidine with an isosteric oxygen (ethylene glycol derivative DB1699, Table 1) results in a less flexible linker and a remarkable drop in binding affinity ($\delta(\Delta G^0) = 15.3 \text{ kJ/mol}$). 382

383

5.2. Two amidine groups are required for high affinity binding

385 Matched-molecular pair analysis of non-symmetric analogues identified that both amidines 386 contribute to high affinity binding (compare pairs pentamidine/RT-36 and

387 pentamidine/CHI/1/72/1; Table 1). Removal of an H-bond donor (as in CHI/1/72/1) leads to a loss in $\Delta G^0 > 10$ kJ/mol. The aniline derivative RT-36 can still act as an H-bond donor, albeit 388 with significantly reduced basicity (and thus H-bond acceptor propensity), and accordingly 389 displayed intermediate affinity ($\delta(\Delta G^0) = 6.2 \text{ kJ/mol}$). Interestingly, the removal of one 390 391 amidine (compare butamidine and CHI/1/69/1) did not produce a significant effect on the binding affinity ($K_i = 3.87 \mu M$ and $K_i = 2.33 \mu M$, respectively), indicating that the low 392 affinity of butamidine (compare 36 nM for pentamidine) is due to an inability to attain a 393 394 productive interaction with the second amidine. Capping of the amidine group, resulting in 395 imidazoline analogue RT-32, or methylation (analogue RT-30) reduced binding to HAPT1, 396 probably due to increased steric crowding at the interaction site, impairing H-bonding. 397 Reducing pentamidine to just 4-hydroxybenzamidine removed essentially all affinity (K_i = 2.9 mM; $\delta(\Delta G^0) = 28.1$ kJ/mol), and the replacement of one amidine with a carboxylic group 398 399 (compare propamidine, RT-38) was highly deleterious for engagement with the binding site. 400 Finally, the orientation of the amidine group is crucial as shown by a meta to para change on 401 the phenyl ring (*meta*-pentamidine, Table 1). We conclude that for high affinity both amidine 402 groups must be able to interact unimpeded with the transporter, and in the linear (para) 403 conformation.

404

405 5.3. Fully conjugated linking units

Stilbamidine and the short-linker analogues FR39 and CRMI8 (Ríos Martínez *et al*, 2015) displayed low binding affinity (Table 1). Diminazene also displayed similar low affinity ($K_i = 63 \mu M$), and [³H]-diminazene uptake can only just be detected in procyclic *T. b. brucei*, *i.e.* in the absence of the TbAT1/P2 transporter (Teka *et al*, 2011), potentially indicating a minimal uptake via HAPT1. Stilbamidine and diminazene feature a similar interamidine distance, much shorter than pentamidine (12.35 and 12.25 Å, respectively). DB75

412 (furamidine) likewise displayed low affinity (Table 2) and is only internalised by TbAT1/P2 413 (Ward et al, 2011). The 2,5-furan linker imposes a fixed, inflexible angle of 136° on the benzamidine moieties and the phenyl rings will adopt a planar orientation with respect to the 414 415 furan plane. This appears to allow only one benzamidine end to interact with the transporter, 416 as (unlike the flexible linker of aliphatic diamidines, vide supra) the replacement of one amidine group actually increases the binding affinity, presumably by allowing an improved 417 418 bonding orientation of the remaining amidine. Thus, DB607 (methoxy for amidine) and 419 DB960 (N-methyl benzimidazole for benzamidine) display a somewhat higher affinity than 420 DB75, although the fixed angle was unchanged. Introduction of a pyridine-N in the ortho-421 position with respect to the amidine functionality (DB994), dramatically reduces the pK_a of 422 the amidine moiety (Wang *et al*, 2010), resulting in a complete loss of binding affinity ($K_i =$ 423 $167 \pm 20 \mu$ M), while this was not observed for the corresponding *meta*-pyridine derivative 424 (DB829). The unfavourable furan bond angle is further demonstrated by the distally 425 elongated analogues DB1061 and DB1062 that approximate the inter-amidine distance of 426 pentamidine but showed no improvement in binding affinity (Table 2). Replacement of furan 427 with thiazole (ER1004) or methylpyrrole (DB320), which feature a similar bond angle, also 428 revealed comparable binding affinities. In contrast, a 2,5-substituted thiophene (DB686 and DB1063) or 2,5-substituted selenophene (DB1213) as a bio-isosteric replacement for the 429 430 furan linker resulted in significantly higher binding affinities when compared to their matched pair analogue (DB1063/DB1061 and DB1213/DB75), which we attribute to a larger 431 432 benzamidine-benzamidine angle. This is corroborated by the much weaker binding of the 2,4-433 thiophene derivative DB1077. A terminal amidine cap (imidazoline) reduced affinity as it did 434 for pentamidine (compare DB1061/DB1062 and DB1063/DB1064 (Table 2)). A difuran 435 spacer (DB914) resulted in a high affinity binder ($K_i = 0.073 \mu M$) because the two furans 436 orient themselves in a *trans* conformation, resulting in a near-linear molecule.

437

438 5.4. Modifications to the phenyl rings of pentamidine

Substituents in the ortho-position (relative to the alkyloxy substituent) of pentamidine were 439 poorly tolerated, including chloride or iodide (RT-43, iodopentamidine; Table 1); the amide 440 441 analogue displayed no affinity at all (RT-46). Such substituents will cause an out-of-plane conformation of the alkoxy-group to avoid clashing with the *ortho*-substituent; high-affinity 442 443 pentamidine binding appears to require a coplanar arrangement of the first methylene bound to the oxygen. Similarly, the introduction of an *ortho*-pyridine N (RT-52) led to a $\delta(\Delta G^0)$ of 444 445 13.7 kJ/mol. This derivative exhibits a conformational bias towards an *anti*-orientation of the 446 ether oxygen and pyridine nitrogen (Fig. 7C) (Chein & Corey, 2010). The regio-isomeric 447 *meta*-pyridine (RT-53) was completely inactive, reflecting the need for a positively charged 448 amidine, as this analogue has a significantly reduced pKa (Wang et al, 2010) (see furan-449 spaced analogue DB994, supra).

450

451 5.5. Non-diamidine trypanocides

The important veterinary trypanocide isometamidium, a hybrid of the phenanthridine 452 ethidium and the diamidine diminazene, inhibited HAPT1-mediated [³H]-pentamidine uptake 453 454 with a K_i of only 3.5 μ M (Supplemental Table S1), most probably through an interaction with its benzamidine moiety, as ethidium displayed virtually no affinity ($K_i = 97 \mu M$). However, 455 we found no evidence that HAPT1/AQP2 is able to transport the bulky isometamidium 456 457 molecule. For instance, the 2T1, tbaqp2 null, TbAQP2 expressed in tbaqp2 null, and the tbaqp2/tbaqp3 null strains displayed statistically identical EC₅₀ values for isometamidium 458 459 $(112 \pm 12 \text{ nM}, 103 \pm 14 \text{ nM}, 98 \pm 24 \text{ nM} \text{ and } 95 \pm 12 \text{ nM}, \text{ respectively; } P>0.05, \text{ Student's}$ 460 unpaired t-test), and the EC_{50} values for ethidium were also identical for each of these strains 461 $(1.32 \pm 0.07 \ \mu\text{M}, 1.39 \pm 0.08 \ \mu\text{M}, 1.35 \pm 0.11 \ \mu\text{M}$ and $1.38 \pm 0.14 \ \mu\text{M}$, respectively). It is

462	thus likely that isometamidium acts as an extracellular inhibitor rather than a substrate for
463	HAPT/AQP2, as it does for the TbAT1/P2 transporter (De Koning, 2001b). The nitro-
464	heterocyclic trypanocide megazol (Carvalho et al, 2014), curcumin and its trypanocidal
465	analogue AS-HK14 (Alkhaldi et al, 2015) failed to inhibit HAPT1. Two trypanocidal bis-
466	phosphonium compounds, CD38 (Taladriz et al, 2012) and AHI43 (Alkhaldi et al, 2016) did
467	inhibit pentamidine uptake (K _i 5-10 μ M), whereas two related compounds, CDIV31 and
468	AHI15 (Taladriz et al, 2012), did not. Phloretin, which inhibits human AQP9 and AQP3
469	(Geng <i>et al</i> , 2017), ⁴⁷ displayed a K_i of 1.76 μ M for HAPT1/TbAQP2.

470

471 5.6. Are all the HAPT1/AQP2 inhibitors transported?

472 In an uptake-by-endocytosis model it would be expected that TbAQP2 binding energy 473 correlates well with TbAQP2-mediated uptake rates for each analogue. We were unable to 474 ascertain the existence of such a correlation directly, for lack of radiolabelled substrates other 475 than pentamidine and diminazene and thus used the Resistance Factor (RF; $EC_{50}(agp2/3$ null/ EC_{50} (TbAQP2-WT)) as a proxy: clearly, a compound with a significant RF is 476 internalized by TbAQP2. We observed a poor correlation between HAPT1 binding affinity 477 and the level of resistance in the *tbagp2/tbagp3* null strain ($r^2=0.039$, Supplemental Fig. S7; 478 n=30, with many inhibitors, even those with high affinity, not displaying any significant 479 480 resistance in the null line. This indicates that many of these inhibitors inhibit HAPT1/TbAQP2 but are not transported by it. This is not compatible with a model in which 481 482 pentamidine binds and is then internalized by endocytosis: the inhibitors do not show resistance in the *tbaqp2/tbaqp3* null line, whereas substrates do. The caveat inherent to using 483 484 the RF instead of rate of transport is that it cannot be excluded that some of the test 485 compounds are AQP2 substrates yet predominantly taken up by transporters other than 486 TbAQP2, and hence show a low RF.

487

488 *5.7. SAR summary*

Figure 7D summarises the structure-activity relationship of pentamidine interactions with 489 490 HAPT1/TbAQP2. No modification in any part of the molecule improved affinity for 491 TbAQP2, but virtually every modification resulted in a significant loss of binding activity. 492 The results clearly demonstrate that at least both amidine groups and one or both ether 493 oxygens are involved in interactions with AQP2, the sum of which adds up to the unusually high binding energy for this substrate-transporter pair ($\Delta G^0 = -42.6 \text{ kJ/mol}$). These results, 494 495 unambiguously demonstrating pentamidine binding in an elongated orientation, are in 496 complete agreement with the modelling and molecular dynamics, and the mutational analysis 497 presented above, strengthening those conclusions using a completely different approach. 498

499 Discussion and conclusion

500

501 There is overwhelming consensus that expression of TbAQP2 is associated with the 502 extraordinary sensitivity of T. brucei to pentamidine and melaminophenyl arsenicals, and that 503 mutations and deletions in this locus cause resistance (Baker et al. 2012, 2013; Graf et al. 504 2013, 2015, 2016; Pyana Pati et al, 2014; Munday et al, 2014, 2015a; Unciti-Broceta et al, 505 2015). What has remained however unclear is the mechanism underpinning these phenomena 506 - there are currently no documented other examples of aquaporins transporting such large 507 molecules.. Yet, considering how ubiquitous aquaporins are to almost all cell types, this 508 question is of wide pharmacological importance: if large cationic and neutral drugs 509 (pentamidine and melarsoprol, respectively) can be taken up via an aquaglyceroporin of T. 510 *brucei*, what other pharmacological or toxicological roles may these channels be capable of in 511 other cell types? This manuscript shows clearly that changes in the TbAQP2 WGYR and 512 NPA/NPA motifs, which collectively enlarge the pore and remove the cation filter, allow the 513 passage of these drugs into the cell, and thereby underpin the very high sensitivity of the 514 parasite to these drugs.

515 TbAQP2 has evolved, apparently by positive selection given the high dN/dS ratio, to 516 remove all main constriction points, including the aromatic amino acids and the cationic 517 arginine of the selectivity filter, and the NPA/NPA motif, resulting in an unprecedentedly 518 enlarged pore size. Whereas the advantage of this to T. b. brucei is yet unknown, the 519 adaptation is stable within the *brucei* group of trypanosomes, and found in T. b. rhodesiense 520 (Munday et al, 2014; Graf et al, 2016), T. b. gambiense (Graf et al, 2013, 2015; Munday et 521 al, 2014; Pyana Pati et al, 2014), T. equiperdum and T. evansi (Philippe Büscher and Nick Van Reet, unpublished). As such, it is not inappropriate to speculate that the wider pore of 522 523 TbAQP2 (i) allows the passage of something not transported by TbAQP1 and TbAQP3; (ii)

that this confers an a yet unknown advantage to the cell; and (iii) that uptake of pentamidineis a by-product of this adaptation.

It is difficult to reconcile the literature on pentamidine transport/resistance with 526 527 uptake via endocytosis. For instance, the rate of endocytosis in bloodstream trypanosomes is 528 much higher than in the procyclic lifecycle forms (Langreth & Balber, 1975; Zoltner et al, 2016), yet the rate of HAPT-mediated $[^{3}H]$ -pentamidine uptake in procyclics is ~10-fold 529 530 higher than in bloodstream forms (De Koning, 2001a; Teka et al, 2011), despite the level of 531 TbAQP2 expression being similar in both cases (Siegel et al, 2010; Jensen et al, 2014). 532 Moreover, in procyclic cells TbAQP2 is spread out over the cell surface (Baker *et al*, 2012) 533 but endocytosis happens exclusively in the flagellar pocket (Field & Carrington, 2009) 534 (which is 3-fold smaller in procyclic than in bloodstream forms (Demmel et al, 2014)), as the 535 pellicular microtubule networks below the plasma membrane prevent endocytosis (Zoltner et 536 al, 2016). Similarly, the expression of TbAQP2 in Leishmania mexicana promastigotes produced a rate of $[^{3}H]$ -pentamidine uptake more than 10-fold higher than observed in T. 537 brucei BSF (Munday et al, 2014), despite these cells also having a low endocytosis rate 538 (Langreth & Balber, 1975). The K_m and inhibitor profile of the TbAQP2-mediated 539 540 pentamidine transport in these promastigotes was indistinguishable from HAPT in procyclic or bloodstream form T. brucei (De Koning, 2001a). Moreover, the experimental V_{max} for 541 HAPT-mediated pentamidine uptake in T. brucei BSF and procyclics (Baker et al, 2013) can 542 be expressed as 9.5×10^5 and 8.5×10^6 molecules/cell/h, respectively; given a 1:1 stoichiometry 543 for AQP2:pentamidine the endocytosis model would require the internalisation and recycling 544 545 of as many units of TbAQP2 and this seems unlikely, especially in procyclic cells. These 546 observations are all inconsistent with the contention that pentamidine uptake by 547 trypanosomes is dependent on endocytosis.

Furthermore, the Gibbs free energy of -42 kJ/mol for the pentamidine/AQP2 548 interaction (De Koning, 2001a; Zoltner et al, 2016) is highly unlikely to be the result of the 549 one interaction between one terminal amidine and Asp265 as required in the endocytosis 550 551 model (Song *et al*, 2016). For the TbAT1 transporter, a double H-bond interaction of Asp140 552 with the $N1(H)/C(6)NH_2$ motif of adenosine or with one amidine of pentamidine (Munday et al. 2015b) is estimated to contribute only ~16 kJ/mol to the total ΔG^0 of -34.5 kJ/mol for 553 554 adenosine (-36.7 kJ/mol, pentamidine) (De Koning & Jarvis, 1999). The endocytosis model also does not address the internalisation of melaminophenyl arsenicals, which presumably 555 556 would equally need access to Asp265.

557 Here we systematically mapped the interactions between the aquaporin and pentamidine (ΔG^0 for 71 compounds), yielding a completely consistent SAR with multiple 558 559 substrate-transporter interactions, summarised in Fig. 7D. The evidence overwhelmingly 560 supports the notion that pentamidine engages TbAQP2 with both benzamidine groups and most probably with at least one of the linker oxygens, and that its flexibility and small width 561 562 are both required to optimally interact with the protein. This is completely corroborated by 563 molecular dynamics modelling, which shows minimal energy to be associated with a near-564 elongated pentamidine centrally in the TbAQP2 pore, without major energy barriers to exiting in either direction, but driven to the cytoplasmic side by the membrane potential. This 565 566 contrasts with the contention (Song et al, 2016) that pentamidine could not be a permeant for 567 TbAQP2 because it did not transport some small cations and that this proves that the larger 568 pentamidine cannot be a substrate either. There is scant rationale for that assertion: out of 569 many possible examples: there are 5 orders of magnitude difference in affinity for 570 pentamidine and *para*-hydroxybenzamidine (35 nM vs 2.9 mM; Supplemental Table S1); 571 adenine is not a substrate for the T. brucei P1 adenosine transporter (De Koning & Jarvis, 572 1999), the SLC1A4 and SLC1A5 neutral amino acid transporters transport Ala, Ser, Cys and

573 Thr but not Gly (Kania *et al*, 2013), and Na⁺ is not a permeant of K⁺ channels (Zhorov & Tikhonov, 2013).

The endocytosis model identifies only two key residues for pentamidine access 575 576 (Leu264) and binding (Asp265) in TbAQP2 (Song et al, 2016). Yet, multiple clinical isolates 577 and laboratory strains contain chimeric AQP2/3 genes associated with resistance and/or noncure that have retained those residues and should thus allow binding and internalisation of 578 579 pentamidine (Graf et al, 2013; Pyana Pati, 2014; Unciti Broceta et al, 2015; Munday et al, 2014). Although we find that introduction of the AQP3 Arg residue in position 264 580 (TbAOP2^{L264R}) disables pentamidine transport, this is because the positively charged 581 582 arginine, in the middle of the pore, is blocking the traversing of all cations through the pore, 583 as is its common function in aquaporins (Beitz et al, 2006; Wu et al, 2009). Indeed, the 584 W(G)YR filter residues appear to be key determinants for pentamidine transport by AQPs and the introduction of all three TbAOP2 residues into TbAOP3 (AOP3^{W102I/R256L/Y250L}) was 585 586 required to create an AQP3 that at least mildly sensitised to pentamidine, and facilitated a detectable level of pentamidine uptake. Conversely, any one of the mutations I110W, L258Y 587 588 or L264R was sufficient to all but abolish pentamidine transport by TbAOP2. Similarly, the 589 conserved NPA/NPA motif, and particularly the Asp residues, present in TbAQP3 but NSA/NPS in TbAQP2, is also associated with blocking the passage of cations (Wree et al, 590 591 2011). The unique serine residues in this TbAQP2 motif, halfway down the pore, might be 592 able to make hydrogen bonds with pentamidine. Reinstating the NPA/NPA motif resulted in a TbAQP2 variant with a 93.5% reduced rate of $[^{3}H]$ -pentamidine transport. 593

Tryptophan residues were introduced towards the cytoplasmic end of the TbAQP2 pore (L84W, L118W, L218W) to test the hypothesis that introducing bulky amino acids in that position would block the passage of pentamidine. Each of these mutants was associated with reduced sensitivity to pentamidine and cymelarsan and a >90% reduction in $[{}^{3}H]$ -

pentamidine uptake. This effect was size-dependent as the pentamidine transport rate of L84M and L218M was statistically identical to that of control TbAQP2 cells, and L118M also displayed a higher transport rate than L118W (P<0.0001). These mutant AQPs were still functional aquaglyceroporins as their expression in *tbaqp1-2-3* null cells made those cells less sensitive to the TAO inhibitor SHAM, and increased glycerol uptake.

603 Independence from endocytosis was demonstrated by employing the tetracycline-604 inducible CRK12 RNAi cell line previously described to give a highly reproducible and 605 progressive endocytosis defect in T. brucei (Monnerat et al, 2013), and unambiguously 606 distinguishes between uptake via endocytosis and transporters. Twelve hours after CRK12 RNAi induction pentamidine transport was not significantly reduced although uptake of [³H]-607 608 suramin, which is accumulated by endocytosis through the *T. brucei* flagellar pocket (Zoltner 609 et al, 2016), was reduced by 33% (P=0.0027), indicating successful timing of the experiment 610 to the early stage of endocytosis slow-down. Although several ionophores, including CCCP, 611 nigericin and gramicidin strongly inhibited pentamidine uptake, similar to what has been 612 previously reported for transport processes in T. brucei that are linked to the protonmotive force (De Koning & Jarvis, 1997a,b, 1998; De Koning et al, 1998), this is probably due to the 613 614 inside-negative membrane potential of -125 mV (De Koning & Jarvis, 1997b) attracting the 615 dicationic pentamidine. This is consistent with the prediction of the molecular dynamics 616 modelling, and the reported role of the HA1-3 proton pumps in pentamidine but not 617 melarsoprol resistance (Alsford et al, 2012; Baker et al, 2013).

Altogether, we conclude that the primary entry of the sleeping sickness drugs pentamidine and melarsoprol into *T. brucei* spp. is through the unusually large pore of TbAQP2, rendering the parasite extraordinarily sensitive to the drugs (compare *Leishmania mexicana* (Munday *et al*, 2014)). This is the first report providing detailed mechanistic evidence of the uptake of organic drugs (of MW 340 and 398, respectively) by an aquaporin.

We show that this porin has evolved through positive selection and identify the adaptations in the constriction motifs that enabled it. We consider that other pore-opening adaptations may have evolved in other organisms, including pathogens, which could initiate the pharmacological exploitation of aquaporins and lead to the design of new drug delivery strategies.

628

630 Materials and Methods

631

632 *Trypanosome strains and cultures.*

The drug-sensitive clonal T. b. brucei strain 427 (MiTat 1.2/BS221) (De Koning et al, 2000) 633 634 was used for all the work on the SAR of pentamidine transport. The *tbaqp2/tbaqp3* null cells 635 (Baker et al, 2012) and tbagp1-2-3 null cells (Jeacock et al, 2017) (both obtained from David Horn, University of Dundee, UK) are derived from the 2T1 strain of T. b. brucei (Alsford & 636 Horn, 2008). The CRK12 RNAi cell line²⁸ was obtained from Dr Tansy Hammarton 637 638 (University of Glasgow, UK) and is also based on the 2T1 cell line; RNAi expression was 639 induced with 1 μ g/ml tetracycline in the medium. All experiments were performed with 640 bloodstream form trypanosomes grown in vitro in HMI-11 medium as described (Wallace et 641 al. 2002) at 37 °C in a 5% CO₂ atmosphere. Cultures were routinely maintained in 10 ml of this medium, being seeded at 5×10^4 cells/ml and passed to fresh medium at reaching 642 approximately 3×10⁶ cells/ml after 48 h. For transport experiments 150 or 200 ml of culture 643 644 was seeded at the same density in large flasks and incubated until the culture reached late-log 645 phase.

646

647 Materials

A complete list of diamidine analogues and other chemicals used for the SAR study is given as a table in the supplemental materials with their sources (Supplemental Table S1). Ionophores and uncouplers nigericin, gramicidin, carbonyl cyanide m-chlorophenyl hydrazone (CCCP) and valinomycin, as well as the *T. brucei* proton pump inhibitor *N*ethylmaleimide (NEM) were all purchased from Sigma-Aldrich. New compounds synthesised for this study are listed and described in the Supplemental Materials.

654

Transport of $[{}^{3}H]$ -pentamidine - Transport of $[{}^{3}H]$ -pentamidine was performed exactly as 655 656 previously described for various permeants (Wallace et al, 2002; Bridges et al, 2007; Teka et al, 2011) in a defined assay buffer (AB; 33 mM HEPES, 98 mM NaCl, 4.6 mM KCl, 0.55 657 mM CaCl₂, 0.07 mM MgSO₄, 5.8 mM NaH₂PO₄, 0.3 mM MgCl₂, 23 mM NaHCO₃, 14 mM 658 glucose, pH 7.3). [³H]-pentamidine was custom-made by GE Healthcare Life Sciences 659 (Cardiff, UK) with a specific activity of 88 Ci/mmol. Incubations of bloodstream form 660 661 trypanosomes with 30 nM of this label (unless otherwise indicated) were performed in AB at room temperature for 60 s (unless otherwise indicated) and terminated by addition of 1 ml 662 663 ice-cold 'stop' solution (1 mM unlabelled pentamidine (Sigma) in AB) and immediate 664 centrifugation through oil (7:1 dibutylphthalate:mineral oil v/v (both from Sigma)). Transport 665 was assessed in the presence of 1 mM adenosine to block uptake through the P2 aminopurine 666 transporter; adenosine does not affect HAPT1-mediated transport (De Koning, 2001a; Bridges et al, 2007). Inhibition assays were performed routinely with 6 - 10 different 667 668 concentrations of inhibitor over the relevant range, diluting stepwise by one third each time, in order to obtain a well-defined and accurate sigmoid plot and IC50 value (inhibitor 669 concentration giving 50% inhibition of pentamidine transport; calculated by non-linear 670 671 regression using Prism 6.0 (GraphPad), using the equation for a sigmoid curve with variable 672 slope). Highest concentration was usually 1 mM unless this was shown to be insufficient for 673 good inhibition, or when limited by solubility. K_i values were obtained from IC₅₀ values 674 using

675
$$K_{\rm i} = \mathrm{IC}_{50} / [1 + (L + K_{\rm m})]$$
 (Eq. 1)

in which L is the [³H]-pentamidine concentration and K_m the Michaelis-Menten constant for pentamidine uptake by HAPT1 (Wallace *et al*, 2002). The Gibbs Free energy of interaction ΔG^0 was calculated from

$$\Delta G^0 = -RT \ln K_i \tag{Eq. 2}$$

680 in which R is the gas constant and T is the absolute temperature (Wallace *et al*, 2002).

681

682 Construction of AQP mutants and transfection

683 All mutations in the TbAQP2 and TbAQP3 genes were introduced to the relevant backbone WT vector, either pRPa^{GFP-AQP2} or pRPa^{GFP-AQP3} (Baker et al, 2012), by site-directed 684 mutagenesis. For mutations S131P, S263A, I110W, L264R, L258Y, I190T and W192G in 685 686 AQP2, and W102I, R256L and Y250L in AQP3 mutations were inserted using the QuikChange II kit (Agilent, Santa Clara, CA, USA), following the manufacturer's 687 688 instructions. For mutations L84W, L118W, L218W, L84M, L118M and L218M were 689 introduced using the Q5 Site-Directed Mutagenesis Kit (E0554S), (New England BioLabs) 690 according to manufacturer's instructions.

691 The following primer pairs (itemised in Supplemental Table S2) were used to insert 692 the named TbAQP2 mutations: for S131P, primers HDK1062 and HDK1063; in combination with mutation S263A, using primers HDK1064 and HDK1065 to produce plasmid 693 694 pHDK166. For I110W, primers HDK607 and HDK608, to produce pHDK84; for L264R, primers HDK609 and HDK610 to produce pHDK167; the combination I110W/L264R was 695 696 produced using primers HDK609 and HDK610 on plasmid pHDK84 to give plasmid 697 pHDK78; for L258Y, primers HDK1109 and HDK1110 to produce pHDK168; for I190T, primers HDK1056 and HDK1057 to produce pHDK163; for W192G, primers HDK1058 and 698 HDK1059 to produce pHDK164; the combination I190T/W192G was produced using 699 700 primers HDK1060 and HDK1061 on plasmid pHDK163 to give plasmid pHDK165; for 701 L84W, primers HDK1276 and HDK1277, producing pHDK210; for L118W, primers 702 HDK1274 and HDK1275, producing pHDK208; for L218W, primers HDK1272 and 703 HDK1273, producing pHDK209; for the combination L84W/L118W, primers HDK1276 and 704 HDK1277 on template pHDK208, producing pHDK227; for L84M, primers HDK1364 and

705 HDK1367, producing pHDK234; for L118M, primers HDK1365 and HDK1367, producing 706 pHDK235; and for L218M, primers HDK1366 and HDK1367, producing pHDK236. To 707 insert the named mutations into TbAQP3, the following primers were used (Supplemental 708 Table S3): for W102I, primers HDK511 and HDK512, in combination with mutation R256L, 709 with primers HDK513 and HDK514, to produce plasmid pHDK71; and to add mutation Y250L to this combination, primers HDK795 and HDK796, to produce pHDK121. All 710 711 plasmids were checked by Sanger Sequencing (Source BioScience, Nottingham, UK) for the 712 presence of the correct mutation(s) and the cassette for integration digested out with AscI 713 (NEB, Hitchin, UK) prior to transfection.

For transfection, 10 μ g of digested plasmid and $1-2 \times 10^7$ parasites of the desired cell line (either *aqp2/aqp3* null or *aqp1/aqp2/aqp3* null) were resuspended in transfection buffer and transfected using an Amaxa Nucleofector, with program X-001. After a recovery period (8-16 h) in HMI-11 at 37 °C and 5% CO₂, the parasites were cloned out by limiting dilution with the selection antibiotic (2.5 μ g/ml hygromycin). In all cases the correct integration of the expression cassettes was analysed by PCR.

720

721 Drug sensitivity assays

Drug sensitivity assays for *T. b. brucei* bloodstream forms used the cell viability dye resazurin (Sigma) and were performed exactly as described (Wallace *et al*, 2002; Bridges *et al*, 2007) in 96-well plates with doubling dilutions of test compound, starting at 100 μ M, over 2 rows of the plate (23 dilutions plus no-drug control). Incubation time with test compound was 48 h (37 °C/5% CO₂), followed by an additional 24 h in the presence of the dye.

727

728 Molecular dynamics

729 Molecular dynamics simulations were performed using the GROMACS software package, 730 version 5.1.1 (Abraham et al, 2015). We used the coordinates from the homology model of TbAQP2 published in Figure 2A in Munday et al (2015a), which was inserted into 731 732 POPC/POPE (4:1) membranes, approximately reflecting the membrane composition of T. b. 733 brucei (Smith & Bütikofer, 2010). The membrane models were constructed using the 734 CHARMM-GUI webserver (Jo et al, 2008). Subsequently, extended stability tests of the 735 modelled structure and the bound pentamidine were carried out using unbiased simulations of 736 100 ns length. The root-mean-square deviation (RMSD) of the protein remained relatively 737 low with a backbone RMSD converging to ~ 3 Å after 100 ns simulated time (Supplemental 738 Fig. S6) bound to the binding site defined previously using molecular docking (Fig. 6) 739 (Munday et al, 2015a). For these and all following simulations, we used the CHARMM36 740 force field (Klauda et al, 2010); pentamidine was parameterised using the CHARMM 741 generalized force field approach (CHGenFF (Vanommeslaeghe et al, 2010)). All simulations 742 employed a time step of 2 fs under an NPT ensemble at p = 1 bar and T = 310 K. To obtain 743 non-equilibrium work values for removing pentamidine from the internal AQP2 binding site, we then conducted steered MD simulations with a probe speed of 0.005 nm/ns and a 744 harmonic force constant of 300 kJ/mol nm², pulling pentamidine in both directions along the 745 746 pore axis. The free energy profile of pentamidine binding to the AQP2 pore was 747 reconstructed by using the Jarzynski equality (Jarzynski, 1997).

748

749 *Statistical analysis*

All transport experiments were performed in triplicate and all values such as rate of uptake, percent inhibition, K_i , K_m , V_{max} etc were performed at least three times completely independently. For drug sensitivity tests, all EC₅₀ values were based on serial dilutions over two rows of a 96-well plate (23 doubling dilutions plus no-drug control), which were

754	obtained independently at least three times. EC_{50} and IC_{50} values were determined by non-
755	linear regression using the equation for a sigmoid curve with variable slope and are presented
756	as average \pm SEM. Statistical significance between any two data points was determined using
757	Student's t-test (unpaired, two-tailed).
758	
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767	cell line.
768	Author contributions
769	Performed experiments: AHA, JCM, GDC, MEMA, PM, LW, DP, AD, JW, GS, LFA,
770	SPYK, HMSI, MIAS, AAE, IAT, SG, HPDK

- 771 Chemical synthesis and SAR: CEO, AK, FH, CD
- 772 Computational analysis and modelling: DG, FS, LS, CMW, MC
- 773 Supervision: RRT, DWB, PMON, UZ, HPDK
- 774 Writing manuscript: FH, UZ, HPDK

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776 **Conflict of Interest**

The authors declare that they have no conflict of interest.

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1021 Figures and Legends



Fig. 1. The selectivity filter differences between TbAQP2 and TbAQP3 are largely
responsible for their differences in pentamidine sensitivity and transport rates.

- 1025 (A) Transport of 30 nM [3 H]-pentamidine by *tbaqp2/aqp3* null cells expressing TbAQP2-WT
- 1026 or one of the TbAQP2 mutants as indicated (blue bars). The corresponding brown bars are

1027 pentamidine transport in the control tbaqp2/aqp3 null cells assessed in parallel in each 1028 experiment. Transport was determined in the presence of 1 mM adenosine to block the TbAT1/P2 transporter. Bars represent the average and SEM of at least three independent 1029 1030 experiments, each performed in triplicate. Blue stars: statistical significance comparison, by 1031 two-tailed unpaired Student's tests, between the cells expressing TbAQP2WT and mutants; red stars: statistical comparison between the AQP2-expressing cells and control cells; NS, not 1032 1033 significant. (B-F) EC₅₀ values indicated test drugs, expressed as a percentage of the resistant control (tbaqp2/tbaqp3 null), against cell lines either expressing the indicated TbAQP2 1034 1035 mutant or TbAQP2WT (sensitive control). Red stars and green stars: comparison with 1036 tbaqp2/aqp3 null or TbAQP2WT-expressing cells, respectively, which were always assessed in parallel in each experiment. (G) Transport of 30 nM [³H]-pentamidine by *tbaqp2/aqp3* null 1037 1038 cells expressing TbAQP3 or an AQP3 mutant as indicated. (H) EC₅₀ values of the indicated 1039 drugs against tbaqp2/aqp3 null cells expressing either TbAQP3 or a mutant thereof, 1040 expressed as percentage of *tbaqp2/aqp3* null. 1041 All experiments are the average and SEM of at least 3 independent experiments. *, P<0.05; **, P<0.01; ***, P<0.001, ****, P<0.0001 by unpaired Student's t-test, two-tailed. 1042





Fig. 2. Mutational analysis of TbAOP2 residues I190 and W192. (A) Transport of 30 nM 1045 ³H]-pentamidine by *tbaqp2/tbaqp3* null cells or TbAQP2 variants expressed therein. 1046 1047 Transport was expressed as a percentage of the rate of the AQP2WT control, performed in parallel. Blue stars are comparison with TbAQP2WT, red stars, comparison with the 1048 *tbaqp2/tbaqp3* null control. NS, not significant. (B) EC_{50} values for the indicated drugs 1049 against *tbaqp2/tbaqp3* null cells, and against TbAQP2WT and TbAQP2^{I190T} expressed 1050 therein; values were expressed as % of the *tbaqp2/tbaqp3* null (resistant) control. Red stars, 1051 comparison with the resistant control; green stars, comparison with the internal sensitive 1052 control (TbAQP2WT). The assays for all three strains and all three drugs were done 1053 simultaneously on at least 3 different occasions. (C) As B but for TbAQP2^{W192G}. (D) As B 1054 but for TbAOP2^{I190T/W192G}. 1055

1056 *, P<0.05; **, P<0.01; ***, P<0.001, ****, P<0.0001 by unpaired Student's t-test.



1057

Fig. 3. Analysis of TbAQP2 variants with a leucine-to-tryptophan or leucine-to-1058 1059 methionine substitution near the cytoplasmic end of the pore. (A) Pentamidine EC_{50} values (nM) for mutant and WT TbAQP2 expressed in *tbaqp2/tbaqp3* cells (aqp2-3 null). The 1060 1061 mutants are either a Trp (dark blue bars) or Met (orange bars) substitution at the indicated positions. The resistant control (aqp2-3 null) and sensitive control (AQP2WT) for the 1062 separate datasets (Trp or Met) are indicated as hatched bars in the same colours. (B) As (A) 1063 but showing transport of 30 nM [³H]-pentamidine by the same cell lines, expressed as 1064 percentage of the transport rate in the TbAQP2 control cells. (C) Pentamidine EC₅₀ values for 1065 1066 the same mutants as in (A) but expressed in the *tbaqp1-2-3* null cells, performed in parallel with the determination of EC₅₀ values for SHAM, shown in (D). As all cell lines were done 1067

simultaneously, the resistant and sensitive strain control values are identical for the Trp and

1069 Met mutants in this series.

1070 All bars represent the average and SEM of at least three independent replicates. *, P < 0.05;

1071 **, P<0.01; ***, P<0.001, ****, P<0.0001 by unpaired Student's t-test; ns, not significant;

1072 nd, not determined.

1073



1074

Fig. 4. Disabling endocytosis does not reduce uptake of pentamidine. (A) qRT-PCR of 1075 CRK12, normalised to housekeeping gene GPI-8 (n=3). (B) Transport of 0.025 µM [³H]-1076 Pentamidine measured in control (non-induced) and CRK12 cell after exactly 12 h of 1077 tetracycline induction; incubation time with label was 30 s. Bar is average and SEM of 5 1078 1079 independent determinations, each performed in triplicate. NS, not significant by unpaired Student's t-test. (C) As frame B but uptake of 0.25 µM [³H]-suramin over 15 min; average 1080 and SEM of 5 independent determinations, each in quadruplicate. **, P=0.0027 by Student's 1081 unpaired, two-tailed t-test. 1082





Fig. 5. High affinity pentamidine uptake in *T. b. brucei* is sensitive to ionophores. (A) Uptake of 25 nM [³H]-pentamidine in s427WT bloodstream forms was measured in the presence of 1 mM adenosine to block the P2 transporter, and in the further presence of various ionophores at the indicated concentrations in μ M. Incubation with radiolabel was 5 min after a 3 min pre-incubation with ionophore. Accumulation of radiolabel was expressed as a percentage of the control, being a parallel incubation in the absence of any ionophore.

Bars represent the average of 3 - 5 independent determinations (each performed in 1092 quadruplicate) and SEM. (B) Uptake of 0.25 μ M [³H]-suramin by T. b. brucei s427WT cells 1093 over 10 minutes. Cells were incubated in parallel, with or without the presence of 20 µM 1094 CCCP (plus 3-minute pre-incubation). Saturation of the suramin-receptor interaction was 1095 1096 demonstrated by including 100 µM unlabelled suramin (blue bars). Bars represent average and SEM or three independent experiments, each performed in quadruplicate. (C) Correlation 1097 plot of pentamidine transport rate versus protonmotive force (PMF), $r^2 = 0.93$, P<0.05 by F-1098 test. Concentrations in µM are indicated in the frame. CCCP is shown in red and not included 1099 in the regression analysis. Each data point is the average of 4 or more independent repeats 1100 1101 performed in quadruplicate. The values for PMF were taken from (De Koning and Jarvis, 1997b). (D) Uptake of 0.25μ M [³H]-glycerol by aqp1/aqp2/aqp3 null cells expressing 1102 1103 TbAQP2-WT. Dose response with CCCP and pentamidine (PMD), using an incubation time of 1 min. The graph shown was performed in triplicate and representative of three 1104 independent repeats. (E) As C but using 0.025 µM [³H]-pentamidine and 30 s incubations. 1105 1106 Representative graph in triplicate from 3 independent repeats. *, P<0.05; **, P<0.01; ***, P<0.001 by Student's unpaired t-test. 1107



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1110

Fig. 6. Pentamidine binding in TbAQP2 and free-energy profile of permeation (Left). 1111 Docked conformation of pentamidine (blue) bound to the TbAQP2 (wheat). The protein and 1112 the ligand were modelled as described.⁴ The protein pore is shown in grey mesh, and the 1113 mutated positions described in the text are in magenta. (Right) Free-energy profile G(L)1114 (solid blue line) along the pore axis of TbAQP2 (L). The membrane voltage of T. b. brucei 1115 1116 gives rise to a voltage drop across the membrane (gray dotted line), which alters the free-1117 energy profile (dashed blue line includes V_m effect) and reduces the free-energy of 1118 pentamidine exit into the intracellular bulk by ~22 kJ/mol as compared to the extracellular 1119 side (black arrow).



1123 Fig. 7. Correlation between linker chain length and affinity to HAPT1.

1124 (A) A series of pentamidine analogues with different methylene linker length was tested for inhibition of TbAQP2/HAPT1-mediated 25 nM [³H]-pentamidine transport (i.e. in the 1125 presence of adenosine to block the TbAT1/P2 transporter). The K_i values are listed in Table 1126 1. All K_i values are shown as average and SEM of 3 or more independent experiments, each 1127 performed in triplicate. (B) The distance between the amidine carbon atoms in the lowest-1128 1129 energy conformation was calculated using density functional theory as implemented in Spartan '16 v2.0.7. Geometry optimisations were performed with the wB97XD functional 1130 and the 6-31G* basis set at the ground state in gas phase. Structures and distances shown 1131 1132 represent the dication state that is overwhelmingly prevalent in aqueous solution at neutral 1133 pH. The numbered red data points correspond to the propamidine - octamidine series in frame 1134 A. (C) Repulsion between free electron pairs (double dots), indicated by curved blue lines for 1135 RT-52 in the *cis*-conformation, causing it to exist overwhelmingly in the *anti*-conformation. (D) Overview of SAR observations on the binding preferences of TbAQP2 for pentamidine 1136 1137 and its analogues.

1139 **Table 1**. Pentamidine analogues with an aliphatic linker





Compound	R ₁	R ₂	R ₃	R4	x	Y	Z	Κ _i (μM)	δ(ΔG ⁰) PMD (kJ/mol)
Ethamidine	Am	Am	Н	Н	СН	СН	-O-(CH ₂) ₂ -O-	>100	>19.7
Propamidine	Am	Am	Н	Н	СН	СН	-O-(CH ₂) ₃ -O-	6.63 ± 1.40	13.0
Butamidine	Am	Am	Н	Н	СН	СН	-O-(CH ₂) ₄ -O-	3.87 ± 1.38	11.7
Pentamidine (PMD)	Am	Am	Н	Н	СН	СН	-O-(CH ₂) ₅ -O-	0.036 ± 0.0006	
Hexamidine	Am	Am	Н	Н	СН	СН	-O-(CH ₂) ₆ -O-	0.058 ± 0.011	1.3
Heptamidine	Am	Am	Н	Н	СН	СН	-O-(CH ₂) ₇ -O-	0.123 ± 0.010	3.1
Octamidine	Am	Am	Н	Н	СН	СН	-O-(CH ₂) ₈ -O-	2.16 ± 0.48	10.2
RT-48	Am	Am	Н	Н	СН	СН	-S-(CH ₂) ₅ -S-	2.01 ± 0.86	10.0
RT-50	Am	Am	Н	Н	СН	СН	-NH-(CH ₂)5- NH-	6.27 ± 1.30	12.9
RT-49	Am	Am	Н	Н	СН	СН	-SO ₂ -(CH ₂) ₅ - SO ₂ -	>150	>20.7

DD1(00					СН	СН	-O-(CH ₂) ₂ -O-	166+01	15.2
DB1699	Am	Am	н	Н			(CH2)2-O-	16.6 ± 2.1	15.3
RT-36	Am	NH ₂	Н	Н	СН	СН	-O-(CH ₂) ₅ -O-	0.43 ± 0.07	6.2
CHI/1/72/1	۸m	СН.	Ц	Ц	СН	СН	$O(CH_2) = O$	3.1 ± 0.7	10.7
CIII/1/72/1	AIII	C113	11	11	CII	CII	-0-(CII ₂)5-0-	3.1 ± 0.7	10.7
CHI/1/69/1	Am	Н	Н	Н	СН	СН	-O-(CH ₂) ₄ -O-	2.3 ± 0.5	10.4
DT 20				TT	CII	CH		NH 100	> 10.7
R1-38	Am	CA	н	Н	СН	СН	$-O-(CH_2)_3-O-$	NI, 100	>19.7
					СН	СН		2890 ±	
<i>meta</i> -PMD	Н	Н	Am	Н			-O-(CH ₂) ₅ -O-		28.1
								1050	
RT-32	Im	Im	Н	Н	СН	СН	-O-(CH ₂) ₅ -O-	0.40 ± 0.08	6.0
							(2)3		
RT-30	MeAm	MeAm	Н	Н	СН	СН	-O-(CH ₂) ₅ -O-	0.30 ± 0.07	5.3
Stilbamidine	Am	Am	Н	Н	СН	СН	-CH=CH-	54.8 ± 3.2	18.3
Suiteunie		1 1111			011	011		0 110 - 012	10.5
FR39	G1	G1	Н	Н	СН	СН	-(CH ₂) ₂ -	41.7 ±15.2	17.6
CPMI8	G2	G2	Ц	Ц	СН	СН	(CH ₂)	528 ± 127	18.1
CRIVIIO	02	02	11	11	CII	CII	-(C112)2-	J2.0 ±12.7	10.1
RT-43	Am	Am	Н	Cl	СН	СН	-O-(CH ₂) ₅ -O-	0.51 ± 0.15	6.6
				т	CII	CII		2.15 + 0.04	0.4
Iodo-PMD	Am	Am	H	I	СН	СН	-O-(CH ₂) ₅ -O-	2.15 ± 0.04	8.4
RT-46	Am	Am	Н	-C(O)NH ₂	СН	СН	-O-(CH ₂) ₅ -O-	>100	>19.7
RT-52	Am	Am	Н	Н	СН	Ν	-O-(CH ₂) ₅ -O-	8.84 ± 0.88	13.7
RT-53	Am	Δm	н	Н	N	СН	-O-(CH2)-O-	NI 250	>22
1(1-55		AIII	11	11	14		-0-(0112)5-0-	111, 250	- 22
141 Am, ar	nidine; N	MeAm, N	Aethyl	-amidine; Im	, imida	azole;	CA, carboxylic a	ncid; G1, 2-	

aminoimidazoline; G2, 1-methoxy-2-aminoimidazoline. PMD, pentamidine, NI, no inhibition at the indicated concentration in μ M. K_i is the inhibition constant for [³H]-pentamidine transport by TbAQP2/HAPT1. $\delta(\Delta G^0)$ PMD is the difference in Gibbs Free Energy of interaction of the substrate with TbAQP2 with the same value for pentamidine (PMD). All K_i values are the average and SEM of at least 3-4 independent experiments.

1148 Table 2. Selection of diamidine analogues with aromatic linkers





Compound	R ₁	R ₂	Ar	X	Y	K _i (µM)	δ(ΔG ⁰) PMD (kJ/mol)
DB75	Am	Am		СН	СН	38.2 ± 10.2	17.3
DB607	Am	OCH ₃		СН	СН	18.1 ± 1.9	15.5
DB960	Am	NMB ^a		СН	СН	16.6 ± 3.5	15.3
DB994	Am	Am		Ν	СН	167 ± 20	21.0
DB829	Am	Am		СН	N	39.9 ± 8.0	17.4
DB1061	EtAm	EtAm		СН	СН	32.3 ± 6.0	16.9
DB1062	2MeIm	2MeIm		СН	СН	59.6 ± 11.2	18.4
ER1004	Am	Am	S N	СН	СН	68.7 ± 16.0	18.8
DB320	Am	Am	N	СН	СН	71.3 ± 12.1	18.9

DB686	Gua	Gua	s	СН	СН	0.29 ± 0.11	5.2
DB1063	EtAm	EtAm	s	СН	СН	0.40 ± 0.10	6.0
DB1064	2MeIm	2MeIm	S	СН	СН	3.0 ± 0.82	11.0
DB1213	Am	Am	Se	СН	CH	0.72 ± 0.17	7.5
DB1077	Am	Am	S	СН	CH	13.8 ± 3.1	14.8
DB914	Am	Am		СН	СН	0.073 ± 0.013	1.8

Am, amidine; Im, imidazole; EtAm, ethylamidine; 2MeIm, 2-methylimidazoline; NMB, *N*methyl benzimidazole. ^aThis compound lacks the second benzene ring and features the terminal NMB moiety instead. K_i is the inhibition constant for [³H]-pentamidine transport by TbAQP2/HAPT1. $\delta(\Delta G^0)$ PMD is the difference in Gibbs Free Energy of interaction of the substrate with TbAQP2 with the same value for pentamidine (PMD). All K_i values are the average and SEM of at least 3-4 independent experiments.

Supplemental Materials with:

Pentamidine enters *Trypanosoma brucei* by passing through the pore of the aquaglyceroporin TbAQP2

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A.

Tco Tbr1 Tsu1 Tsu2 Tbr2 Tbr3	MTSPTVPNPMSTVPMTEMTE-ANGTTNPPIPDAGERTAVNFDTEQCK-TKEIL MSDEKINVHQYPSETDVRGLKARNGGACEVPFEENN-EPIPNRSANPQEKNENELV MSSEPVNVHRYTAEGDRSGLKDRHGKTCEVCVGDESAAAVPSAVYNPQEQSGDGPEVK MQNQPDAMTH-STAVQMV-NKNPEDGTGGADTERSDEMTAPTTRTGDAQK MQSQPDNVAY-PMELQAV-NKDGTVEVRVQGNVDNSSNERWDADVQK MQSQPDNVAY-PMELQAV-NKDGTVEVRVQGNDDSSNRK * . :	51 55 58 48 45 37
Tco Tbr1 Tsu1 Tsu2 Tbr2 Tbr3	-AGEGEAPHGPMDINYWPLRNLRMDFREYVGEFLGTFVLLFMGNGVVATTLLDNNLGFLS GDNADNEAHD <mark>AVDV</mark> NYWAPRQLRLDYRNYMGEFLGTFVLLFMGNGVVATTILDKDLGFLS AGGGEAEVQNAADVNNWAPRRLRLDYRDYMGEFLGTFVLLFMGNGVVATTMLDDGLGFLS CETTNTPKEG <mark>AGGI</mark> NYWAPRELRLKYRDYMGELLGTFVLLLMGNGVVATVVVDGKLGFLS HEVAEAQEKP <mark>VGGI</mark> NFWAPRELRLNYRDYVAEFLGNFVLIYIAKGAVITSLLVPDFGLLG HEVAEAQEEV <mark>PGGI</mark> NFWAPRELRLNYRDYMGELLGTFVLLFMGNGVVATVIIDGKLGFLS	110 115 118 108 105 97
Tco Tbr1 Tsu1 Tsu2 Tbr2 Tbr3	ITFGWGIAVTMGLYVSLGTSSGHLNPAVTVANAFFGGFPWKKVPGYIAMQMLGAFVGAAC ITLGWGIAVTMGLYISLGISCGHLNPAVTLANAVFGCFPWRRVPGYIAAQMLGAFVGAAC ITLGWGIAVTMGLYISLGTSCGHLNPAVTVANAVFGCFPWKKVAGYIAMQMLGAFVGAAC ITLGWGIAVTMALYISLGISSGHLNPAVTVGNAVFGDFPWRKVPGYIAAQMFGAFLGAAC LTIGIGVAVTMALYVSLGISGGHLNSAVTVGNAVFGDFPWRKVPGYIAAQMLGTFLGAAC ITLGWGIAVTMALYVSLGISSGHLNPAVTVGNAVFGDFPWRKVPGYIAAQMLGAFLGAAC :*:* *:****.*** * **** ***:.*** ***::* ****	170 175 178 168 165 157
Tco Tbr1 Tsu1 Tsu2 Tbr2 Tbr3	AYGVYADLLNKKVSDGEIEDYAGMFSTYPRDGNSLFSCIFGEFICTAMLTFCVCGI AYGVYADLLKQHSGG-LVGFGDKGFAGMFSTYPREGNRLFYCIFSEFICTAILLFCVGGI AYGVFADLLKQHSGG-LIPFGDKGFAGMFSTYPRDGNRLFYCIFGEFICTAMLLFCVSGI AYGVFADLLKEYCGGKLLAFGAKGIAGVFSTYPKEANSVFACVFGEFICTAILLFCVCGI AYGVFADLLKAHGGGELIAFGEKGIAWVFAMYPAEGNGIFYPIFAELISTAVLLCVCGI AYGVFADLLKAHGGGELIAFGEKGTAGVFSTYPRDSNGLFSCIFGEFICTAMLLFCVCGI ****:****: :. * :: ** :.* :* :* :* :* :* :* :* :*	226 234 237 228 225 217
Tco Tbr1 Tsu1 Tsu2 Tbr2 Tbr3	FDTHNAPATGHEPLAVGALVFAIGNNVGYATGYAINPARDFGPRVFSAILYGSTVFTRGD FDPNNSPAKGHEPLAVGALVFAIGNNIGYASGYAINPARDFGPRVFSAILFGSEVFTTGN FDANNSPAKGHEPLAVGALVFAIGNNIGYATGYAINPARDFGPRLFSAILFGSEVFTAGN FDPNNSPAKKHEPLAVGSLIFAIGNNIGYSTGYAMNPARDFAPRVFSALLLGGEVFSHGN FDPNNSPAKGYETVAIGALVFVMVNNFGLASPLAMNPSLDFGPRVFGAILLGGEVFSHAN FDPNNSPAKGHEPLAVGALVFAIGNNIGYSTGYAINPARDFGPRVFSSFLYGGKVFSHAN ** :*:**. :* :*::*:: **.* :: *:**: **.**:*:*:**.	286 294 297 288 285 277
Tco Tbr1 Tsu1 Tsu2 Tbr2 Tbr3	YYFWVPLFIPLLGGIFGIILYKYFVPH313YYFWVPLFIPFLGGIFGLFLYKYFVPY321YYFWVPLFIPFLGAIFGLFLYKYFVPH324YYFWVPLFIPFLGAIFGLFLYKYFVPH315YYFWVPLVVPFFGAILGLFLYKYFLPH312YYFWVPLVIPLFGGIFGLFLYKYFVPH304*******.:*::*:*:*:*:*:	

>Tco AQP1

MTSPTVPNPMSTVPMTEMTEANGTTNPPIPDAGERTAVNFDTEQCKTKEILAGEGEAPHGPMDINYWPLRNLRMD FREYVGEFLGTFVLLFMGNGVVATTLLDNNLGFLSITFGWGIAVTMGLYVSLGTSSGHLNPAVTVANAFFGGFPW KKVPGYIAMQMLGAFVGAACAYGVYADLLNKKVSDGEIEDYAGMFSTYPRDGNSLFSCIFGEFICTAMLTFCVCG IFDTHNAPATGHEPLAVGALVFAIGNNVGYATGYAINPARDFGPRVFSAILYGSTVFTRGDYYFWVPLFIPLLGG IFGIILYKYFVPH

>Tbr AQP1

MSDEKINVHQYPSETDVRGLKARNGGACEVPFEENNEPIPNRSANPQEKNENELVGDNADNEAHDAVDVNYWAPR QLRLDYRNYMGEFLGTFVLLFMGNGVVATTILDKDLGFLSITLGWGIAVTMGLYISLGISCGHLNPAVTLANAVF GCFPWRRVPGYIAAQMLGAFVGAACAYGVYADLLKQHSGGLVGFGDKGFAGMFSTYPREGNRLFYCIFSEFICTA ILLFCVGGIFDPNNSPAKGHEPLAVGALVFAIGNNIGYASGYAINPARDFGPRVFSAILFGSEVFTTGNYYFWVP LFIPFLGGIFGLFLYKYFVPY

>Tbr AQP2

MQSQPDNVAYPMELQAVNKDGTVEVRVQGNVDNSSNERWDADVQKHEVAEAQEKPVGGINFWAPRELRLNYRDYV AEFLGNFVLIYIAKGAVITSLLVPDFGLLGLTIGIGVAVTMALYVSLGISGGHLNSAVTVGNAVFGDFPWRKVPG YIAAQMLGTFLGAACAYGVFADLLKAHGGGELIAFGEKGIAWVFAMYPAEGNGIFYPIFAELISTAVLLLCVCGI FDPNNSPAKGYETVAIGALVFVMVNNFGLASPLAMNPSLDFGPRVFGAILLGGEVFSHANYYFWVPLVVPFFGAI LGLFLYKYFLPH

>Tbr AQP3

MQSQPDNVAYPMELQAVNKDGTVEVRVQGNDDSSNRKHEVAEAQEEVPGGINFWAPRELRLNYRDYMGELLGTFV LLFMGNGVVATVIIDGKLGFLSITLGWGIAVTMALYVSLGISSGHLNPAVTVGNAVFGDFPWRKVPGYIAAQMLG AFLGAACAYGVFADLLKAHGGGELIAFGEKGTAGVFSTYPRDSNGLFSCIFGEFICTAMLLFCVCGIFDPNNSPA KGHEPLAVGALVFAIGNNIGYSTGYAINPARDFGPRVFSSFLYGGKVFSHANYYFWVPLVIPLFGGIFGLFLYKY FVPH

>Tsu AQP1

MSSEPVNVHRYTAEGDRSGLKDRHGKTCEVCVGDESAAAVPSAVYNPQEQSGDGPEVKAGGGEAEVQNAADVNNW APRRLRLDYRDYMGEFLGTFVLLFMGNGVVATTMLDDGLGFLSITLGWGIAVTMGLYISLGTSCGHLNPAVTVAN AVFGCFPWKKVAGYIAMQMLGAFVGAACAYGVFADLLKQHSGGLIPFGDKGFAGMFSTYPRDGNRLFYCIFGEFI CTAMLLFCVSGIFDANNSPAKGHEPLAVGALVFAIGNNIGYATGYAINPARDFGPRLFSAILFGSEVFTAGNYYF WVPLFIPFLGGIFGLLLYKYFVPH

>Tsu AQP2

MQNQPDAMTHSTAVQMVNKNPEDGTGGADTERSDEMTAPTTRTGDAQKCETTNTPKEGAGGINYWAPRELRLKYR DYMGELLGTFVLLLMGNGVVATVVVDGKLGFLSITLGWGIAVTMALYISLGISSGHLNPAVTVGNAVFGDFPWRK VPGYIAAQMFGAFLGAACAYGVFADLLKEYCGGKLLAFGAKGIAGVFSTYPKEANSVFACVFGEFICTAILLFCV CGIFDPNNSPAKKHEPLAVGSLIFAIGNNIGYSTGYAMNPARDFAPRVFSALLLGGEVFSHGNYYFWVPLFIPFL GAIFGLFLYKYFVPH

В.



C.

T. brucei AQP1 v T. suis AQP1	0.21
T. brucei AQP3 v T. suis AQP3	0.30
T. brucei AQP2 v T. brucei AQP3	2.00

Figure S1.

A. Sequence alignment and individual sequences of the *T. congolense*, *T. b. brucei* and *T. suis*. The *T. brucei* and *T. congolense* sequences were obtained from tritrypDB, *T. suis* sequences (Kelly S, Gibson W and Carrington M. The genome of *Trypanosoma suis*. In preparation). The alignment was produced with Clustal Omega. The yellow highlighting indicates the N-terminus of the sequences used to determine non-synonymous v synonymous ratios.

dN/dS

B. Phylogenetic tree of these sequences. The tree is a Neighbour-joining tree produced in Clustal Omega with the lengths of the horizontals proportional to the differences.

C. The ratio of non-synonymous v synonymous (dN/dS) codon changes calculated for selected comparisons between *T. brucei* and *T. suis* AQPs. The ratios were calculated using a region of high confidence alignments from ~amino acid 60 (highlighted in Supplemental Figure 2A) to the C-terminus.



Fig. S2. Pentamidine transport analysis for TbAQP2^{1190T} and TbAQP2WT. (A) Transport of 30 nM [³H]-pentamidine by *tbaqp2/tbaqp3* null cells expressing TbAQP2^{1190T}, in the presence of unlabelled pentamidine at the indicated concentrations. Incubation time was 15 min, required to ensure sufficient radiolabel for accurate quantification, and uptake was linear and through zero over this period. The inhibition data were plotted to a double sigmoidal curve (Prism 7.0) with the bottom value fixed at 0. The high affinity component displayed an average an IC₅₀ of 30.9 ± 12.2 nM (n=3) and the lower affinity segment could be converted to a Michaelis-Menten plot for determination of K_m and V_{max} (inset), yielding an average K_m of $59.9 \pm 9.1 \mu$ M (n=3). The plot shown is one representative experiment in triplicate of three independent experiments. (B) Like (A) but with *tbaqp2/tbaqp3* null cells expressing TbAQP2WT. Incubation time was 20 s (linear phase). The high affinity phase had statistically identical EC₅₀ (41 ± 17 nM; P>0.05) as TbAQP2^{1190T}. The inset shows a zoom-in

on the low-affinity part of the curve, with the dotted line representing a theoretical sigmoid plot for 1 inhibitor, with the upper limit fixed at the value obtained for 10 μ M pentamidine. The low affinity component was also statistically identical in the two strains (TbAQP2-WT $K_m = 82.7 \pm 17.5 \ \mu$ M (n=3; P>0.05)). Note that the amount of [³H]-pentamidine taken up by the low affinity component is highly similar for the mutant (A) and control (B) cell lines, at approximately 0.0005 pmol(10⁷ cells)⁻¹s⁻¹. Both frames show one representative experiment of three repeats, each performed in triplicate. Error bars are SEM, when not shown, fall within the symbol.



Figure S3. EC₅₀ values for Cymelarsan, diminazene aceturate and phenylarsine oxide (PAO) against the *tbaqp2-tbaqp3* null cell line. AQP2-WT and various mutant versions thereof (indicated) were expressed in this cell line. EC50 values were determined using the alamar blue (resazurin) assay. Bars represent the average and SEM for at least three determinations. nd, not done.



Correlation transport vs EC50 for all cell lines

Strain	EC5	0	Transport rate		
	AVG	SEM	AVG	SEM	
2T1	5.52	1.65	21.86	2.34	
AQKO	100	0	1.05	0.16	
+AQP2-WT	0.46	0.27	100	0	
+L264R	126.1	10.2	0.91	0.28	
+L258Y	0.5	0.24	5.4	1.6	
+S131P/S263A	1.92	1.53	6.48	1.39	
+I190T/W192G	99.52	16.4	1.27	0.4	
+ I190T	6.89	2.28	2.728	0.744	
+AQP2 I110W	118.7	11.0	0.673	0.296	
+AQP3 WT	105.7	12.6	0.48	0.1	
+AQP3 W102I/R256L	83.83	7.7	0.7	0.18	
+AQP3 W102I/Y250L/R256L	31.94	4.25	1.13	0.1	
+L218W	4.4	0.45	6.94	1.58	
+L118W	46.3	3.36	1.83	0.32	
+L84W	31.98	2.08	2.18	0.52	
+L84W/L118W	59.1	3.58	0.97	0.76	
+L84M	0.43	0.09	48.84	4.68	
+L118M	34.24	5.03	9.33	0.63	

Figure S4. Correlation of the EC_{50} value with the rate of pentamidine transport for all 19 cell lines expressing a wild-type or mutant TbAQP2 in the aqp2/3 null *T. b. brucei line*. All EC_{50} values are expressed as percentage of the resistant control, aqp2/3 null transfected with an empty vector (no TbAQP2). 2T1 is the parental cell line of the aqp2/3 null. All values are the average of at least three independent determinations; the sensitive and resistant control cell lines were included in each independent experiment and the percentages taken are from the internal control rather than from the grand average over all experiments.



Figure S5. Growth Curve of CRK12 RNAi cells in full HMI-9 medium at 37 $^{\circ}C/5\%$ CO₂, in the presence or absence of 1 μ g/ml tetracycline (tet). Cell counts were performed with a haemocytometer and the average of duplicate determinations is shown.



Figure S6. Backbone RMSD of the protein inserted into a lipid bilayer showing convergence to \sim 3 Å in a simulation of 100 ns length.



	AQP2-KO RF	Ki value	Ki value HAPT1	
	AVG	AVG	SEM	
pentamidine	9.06	0.036	0.0006	
RT12	0.74	14.3	3.6	
RT14	1.08	0.061	0.015	
RT43	1.33	0.51	0.15	
RT52	5.79	8.84	0.88	
Diminazene	0.9	63	3	
DB75	0.4	38.25	10.2	
DB902	0.61	114	34	
DB1111	0.2	0.14	0.03	
DB1213	0.55	0.72	0.17	
DB1225	0.3	0.54	0.25	
CHI/1/72/1	0.8	2.6	0.68	
RT05	2.07	0.57	0.18	
RT13	3.17	16.1	4.9	
RT18	2.02	0.26	0.09	
RT24	1.31	0.12	0.03	
RT26	1.28	0.34	0.07	
RT30	1.35	0.3	0.07	
RT36	0.9	0.43	0.07	
RT44	1.1	16.78	3.93	
RT45	2.13	7.52	0.14	
RT01	0.97	49	11	
RT02	1.08	66	13	
RT08	1.41	75	14	
RT10	1.33	35.6	13.4	

RT32	1.23	0.4	0.008
RT48	6.97	2.01	0.86
RT50	17.47	6.27	1.3
Isometamidium	0.91	3.5	0.5
Ethidium	1.05	97	17

Figure S7. Correlation between the Resistance Factor (RF; $EC_{50}(aqp2/3 \text{ null})/EC_{50}(TbAQP2-WT)$) and the K_i value for inhibition of the High Affinity Pentamidine Transporter (HAPT1) encoded by TbAQP2. The pentamidine value (bold) is the K_m determined with radiolabeled pentamidine. The table lists the data points shown in the plot. The line was made by linear regression (Prism 6.0); correlation coefficient r² is 0.039. F-test: slope is not significantly different from zero (P = 0.29).

Supplemental Table S2: primers used for mutations in TbAQP2

Mutation	Primer	Sequence (altered base(s) underlined)	Generated	Original
			Plasmid	Template
S131P/S263A	S131P: HDK1062	CTCCGGTGGCCATCTCAAC <u>C</u> CTGCCGTCACCGTTGGCAA		ppoGFP-AQP2
	S131P: HDK1063	TTGCCAACGGTGACGGCAG <u>G</u> GTTGAGATGGCCACCGGAG		
	S263A: HDK1064	TCTCCCCTTGCGATGAATCCC <u>G</u> CACTTGATTTCGGTCCCAGGG		ркра
	S263A: HDK1065	CCCTGGGACCGAAATCAAGTG <u>C</u> GGGATTCATCGCAAGGGGAGA		
1110W	HDK607	CTCGGTCTTACGATTGGT <u>TGG</u> GGTGTGGCTGTCACGATG		pRPa ^{GFP-AQP2}
	HDK608	CATCGTGACAGCCACACC <u>CCA</u> ACCAATCGTAAGACCGAG	рник84	
L264R	HDK609	TCTCCCCTTGCGATGAATCCCTCAC <u>G</u> TGATTTCGGTCCCAGGGTCTTC		pRPa ^{GFP-AQP2}
	HDK610	GAAGACCCTGGGACCGAAATCA <u>C</u> GTGAGGGATTCATCGCAAGGGGAG A	pHDK167	
1110W/L264R	L264R: HDK609	TCTCCCCTTGCGATGAATCCCTCAC <u>G</u> TGATTTCGGTCCCAGGGTCTTC		pHDK84
	L264R: HDK610	GAAGACCCTGGGACCGAAATCA <u>C</u> GTGAGGGATTCATCGCAAGGGGAG A	ρησκ78	
1 25 8 2	HDK1109	CAACTTCGGCTTAGCGTCTCCC <u>TA</u> TGCGATGAATCCCTCACTTGAT		GFP-AQP2
LZJOT	HDK1110	ATCAAGTGAGGGATTCATCGCA <u>TA</u> GGGAGACGCTAAGCCGAAGTTG	рпритоо	μηγα
1100T	HDK1056	GCCTTCGGTGAAAAGGGGA <u>C</u> TGCGTGGGTGTTTGCCATG		pRPa ^{GFP-AQP2}
11901	HDK1057	CATGGCAAACACCCACGCA <u>G</u> TCCCCTTTTCACCGAAGGC	рпритоз	
W102C	HDK1058	CGGTGAAAAGGGGATTGCG <u>G</u> GGGTGTTTGCCATGTACCC		pRPa ^{GFP-AQP2}
W192G	HDK1059	GGGTACATGGCAAACACCC <u>C</u> CGCAATCCCCTTTTCACCG	рпокточ	
1190T/W192G	HDK1060	GCCTTCGGTGAAAAGGGGA <u>C</u> TGCG <u>G</u> GGGTGTTTGCCATG TACCC		pHDK163
	HDK1061	GGGTACATGGCAAACACCCCCCGCAGTCCCCTTTTCACCGA AGGC	DUDK102	
1.9.4\\\/	HDK1276	AAACTTCGTC <u>TGG</u> ATATATATCGCTAAGGG	nUDV210	pRPa ^{GFP-AQP2}
LO4VV	HDK1277	CCCAGAAATTCAGCCACG	pridk210	
L118W	HDK1274	CACCGCAGTG <u>TGG</u> CTGCTCTGTG	pUDV209	pRPa ^{GFP-AQP2}
	HDK1275	GAAATGAGTTCAGCAAAAATTG	μημκζυο	
L218W	HDK1272	CACGATGGCT <u>TGG</u> TATGTTTCACTG	pHDK200	pRPa ^{GFP-AQP2}
	HDK1273	ACAGCCACACCAATACCA	μηρικτορ	
19414/111914/	HDK1276	AAACTTCGTC <u>TGG</u> ATATATATCGCTAAGGG	דכנאסעמ	pHDK208
10400/111000	HDK1277	CCCAGAAATTCAGCCACG	μημκζζη	
L84M	HDK1364	AAACTTCGTCATGATATATATCGCTAAGG	pUDK224	pHDK210
	HDK1367	GAAATGAGTTCAGCAAAAATTGGATAAAATATAC	μημκζ34	
L118M	HDK1365	CACGATGGCT <u>ATG</u> TATGTTTCAC		pHDK208
	HDK1367	GAAATGAGTTCAGCAAAAATTGGATAAAATATAC	μηρικτορ	
L218M	HDK1366	CACCGCAGTG <u>ATG</u> CTGCTCTGTG	nHDK336	pHDK209
	HDK1367	GAAATGAGTTCAGCAAAAATTGGATAAAATATAC	PHDI230	
Supplemental Table S3: primers used for mutations in TbAQP3

Mutations	Primer	Sequence (altered base(s) underlined)	Plasmid	Template
W102I/R256L	W102I: HDK511	CTCAGCATTACGCTTGGT <u>ATC</u> GGCATTGCCGTCACGATG	pHDK71	pRPa ^{GFP-AQP3}
	W102I: HDK512	CATCGTGACGGCAATGCC <u>GAT</u> ACCAAGCGTAATGCTGAG		
	R256L: HDK513	TACGCAATAAATCCGGCTC <u>T</u> TGACTTCGGTCCCAGGGTC		
	R256L: HDK514	GACCCTGGGACCGAAGTCA <u>A</u> GAGCCGGATTTATTGCGTA		
W102I/R256L/Y 250L	Y250L HDK795	CATCGGTTACTCAACGGGT <u>CT</u> CGCAATAAATCCGGCTCTT	pHDK121	pHDK71
	Y250L HDK796	AAGAGCCGGATTTATTGCG <u>AG</u> ACCCGTTGAGTAACCGATG		

Chemistry of new compounds from Paul O'Neill laboratory.

1,2-Bis (4-cyanophenoxy) ethane (Ethamidine Precursor) (1a) ¹



Sodium (0.16 g, 6.96 mmol) was added portionwise to anhydrous EtOH (4.0 mL) under an atmosphere of nitrogen. After dissolution of the sodium pieces, a solution of 4-cyanophenol (0.75 g, 6.38 mmol) dissolved in anhydrous EtOH (4.0 mL) was added followed by dropwise addition of 1,2-dibromoethane (0.28 mL, 3.19 mmol). The reaction mixture was allowed to stir at reflux under a nitrogen atmosphere for 3 days after which the mixture was cooled, filtered, the solid washed with water and dried under vacuum. Purification by column chromatography eluting with DCM: hexane (8:2) gave the desired dinitrile ethamidine precursor (**1a**) as a white solid (1.42 g, 84%). Mp 211-212°C; ¹H NMR (CDCl₃, 400MHz) δ 7.61 (d, 4H, *J* = 9.0 Hz, ArH), 7.01 (d, 4H, *J* = 9.0 Hz, ArH), 4.39 (s, 4H, CH₂); ¹³C NMR (CDCl₃, 100MHz) δ 161.6, 134.1, 118.9, 115.3, 104.7, 66.4; v_{max} (NujOI) /cm⁻¹ 3326 (C-O-C), 3033 (ArH), 2898 (OH), 2223 (CN), 1602 (Ar), 1509 (Ar), 1247 (C-O-C); *m*/*z* (CI) 282 ([M+NH₄]⁺), found 282.12433, C₁₆H₁₆O₂N₃ requires 282.12424; anal. Found C 72.37, H 4.51, N 10.54, C₁₆H₁₂O₂N₂ requires C 72.71, H 4.57, N 10.60.

4,4'-(ethane-1,2-diylbis(oxy))dibenzimidamide dihydrochloride dihydrate (Ethamidine, Compound CHI/1/30/1) (**2**) ¹



(0.51 g, 1.92 mmol) of **1a** was dissolved in a mixture of anhydrous benzene (54 mL) and EtOH (2.90 mL), cooled to 0 °C and saturated with HCl gas. The mixture was sealed and allowed to stir at room temperature for 3 days after which anhydrous Et₂O (28 mL) was introduced and the mixture was allowed to stir for 10 minutes. The solids were filtered under nitrogen and dissolved in a mixture of anhydrous EtOH (36 mL) and EtOH.NH₃ (36 mL). The mixture was heated overnight (50 °C), cooled to room temperature and reduced by half *in vacuo*. Ether was added to precipitate the solid which was filtered, washed and dried under vacuum. Purification by recrystallisation (2N HCl) gave the desired compound **2** as fine white needles (0.54 g, 70%). Mp 333°C; ¹H NMR (MeOD, 400MHz) δ 7.85 (d, 4H, *J* = 9.0 Hz, ArH), 7.23 (d, 4H, *J* = 9.0 Hz, ArH), 4.52 (s, 4H, CH₂); ¹³C NMR (MeOD, 100MHz) δ 167.9, 165.3, 131.5, 121.7, 116.7, 68.5; v_{max} (Nujol) /cm⁻¹ 3362 (NH), 3037 (ArH), 2940 (C-H), 1658 (C=N-H), 1606 (Ar), 1505 (Ar), 1245 (C-0-C); *m*/z (ESP) 299 ([M-H]⁻); anal. Found C 47.54 H 5.62 N 14.30, C₁₆H₂₄N₄O₄Cl₂ requires C 47.18, H 5.94, N 13.76.

4, 4'-(propane-1,3-diylbis(oxy))dibenzonitrile (Propamidine Precursor) (1b)¹



Sodium (0.16 g, 6.96 mmol) was added portionwise to anhydrous EtOH (4.0 mL) with stirring under an atmosphere of nitrogen. After dissolution of Na, a solution of 4-cyanophenol (0.75 g, 6.38 mmol) dissolved in dry ethanol (4.0 mL) was added followed by dropwise addition of 1,3-dibromopropane (0.32 mL, 3.19 mmol). The reaction mixture was allowed to stir at reflux under a nitrogen atmosphere for 3 days after which the mixture was cooled, filtered, the solid washed with water and dried under vacuum. Purification by column chromatography eluting with DCM: hexane (8:2) gave the desired compound **1b** as a white solid (1.40 g, 79%). Mp 190-191°C; ¹H NMR (CDCl₃, 400MHz) δ 7.59 (d, 4H, *J* = 8.5 Hz, ArH), 6.96 (d, 4H, *J* = 8.5 Hz, ArH), 4.20 (t, 4H, *J* = 6.0 Hz, CH₂), 2.32 (m, 2H, CH₂); ¹³C NMR (CDCl₃, 100MHz) δ 161.9, 134.0, 119.1, 115.1, 104.2, 64.4, 28.8; v_{max} (Nujol) /cm⁻¹ 3104 (Ar-H), 2823 (C-H), 2221 (C=N), 1604 (Ar), 1509 (Ar), 1253 (C-O); *m/z* (CI) 296 ([M+NH₄]⁺), found 296.14037, C₁₇H₁₈N₃O₂ requires 296.13992; anal. Found C 73.22, H 5.13, N 10.03, C₁₇H₁₄N₂O₂ requires C 73.37, H 5.07, N 10.07.

4,4'-(propane-1,3-diylbis(oxy))dibenzimidamide

dihydrochloride d

dihydrate

(Propamidine, Compound CHI/1/25/5) (3)¹



(0.50 g, 1.79 mmol) of **1b** was dissolved in a mixture of anhydrous benzene (55 mL) and anhydrous ethanol (3.0 mL), cooled to 0 °C and saturated with HCl gas. The mixture was sealed and allowed to stir at room temperature for 3 days after which ether (30 mL) was added and the mixture was allowed to stir for 10 minutes. The solids were filtered under nitrogen and dissolved in a mixture of anhydrous EtOH (36 ml) and EtOH.NH₃ (36 mL). The mixture was heated overnight (50 °C), cooled to room temperature and reduced by half *in vacuo*. Et₂O (15 mL) was added to precipitate the solid which was filtered, washed and dried under vacuum. Purification by recrystallisation (2N HCl) gave the desired compound **3** as fine white needles (0.59 g, 78%). Mp 200°C; ¹H NMR (MeOD, 400MHz) δ 7.82 (d, 4H, *J* = 9.0 Hz, ArH), 7.19 (d, 4H, *J* = 9.0 Hz, ArH), 4.34 (t, 4H, *J* = 6.0 Hz, CH₂), 2.46 (m, 2H, CH₂); ¹³C NMR (MeOD, 100MHz) δ 167.9, 165.5, 131.5, 121.5, 121.4, 116.6, 66.4, 30.3; v_{max} (Nujol) /cm⁻¹ 3280 (N-H), 3038 (Ar-H), 2929 (C-H), 1504 (Ar), 1606 (Ar), 1240 (C-O-C); *m*/*z* (ESP) 313 ([M-H]⁻); anal. Found C 48.60, H 6.10, N 13.25, C₁₇H₂₆N₄O₄Cl₂ requires C 48.46, H 6.22, N 13.30.

4,4'-(Butane-1,4-diylbis(oxy))dibenzonitrile (Butamidine precursor) (1c)¹



Sodium (0.10 g, 4.35 mmol) was added portionwise to dry EtOH (4.0 mL) stirring under an atmosphere of nitrogen. After dissolution of sodium, a solution of 4-cyanophenol (0.47 g, 3.95 mmol) dissolved in dry ethanol (4.0 mL) was added followed by dropwise addition of 1,4-dibromobutane (0.24 mL, 1.98 mmol). The reaction mixture was allowed to stir at reflux under a nitrogen atmosphere for 3 days after which the mixture was cooled, filtered, the solid washed with water and dried under vacuum. Purification by column chromatography eluting with DCM: hexane (8:2) gave the desired compound **1c** as a white solid (1.03 g, 89%). Mp 174°C; 'H NMR (CDCl₃, 400MHz) δ 7.59 (d, 4H, *J* = 8.9 Hz, ArH), 6.93 (d, 4H, *J* = 8.9 Hz, ArH), 4.08 (m, 4H, CH₂), 2.01 (m, 4H, CH₂); ¹³C NMR (CDCl₃, 100MHz) δ 162,1, 134.0, 119.1, 115.1, 104.0, 67.7, 25.7; v_{max} (Nujol) /cm⁻¹ 3332 (C-O-C), 3033 (Ar-H). 2956 (C-H), 2219 (C≡N), 1604 (Ar). 1506 (Ar), 1251 (C-O-C); *m*/*z* (CI) 310 ([M+NH₄]⁺) found 310.15532, C₁₈H₂₀N₃O₂ requires 310.15555; anal. Found C 74.03, H 5.55, N 9.55, C₁₈H₁₆N₂O₂ requires 73.95, H 5.52, N 9.58.

4,4'-(Butane-I,**4-diylbis(oxy))dibenzimiamide dihydrochloride dihydrate** (Butamidine, Compound CHI/1/41/1) (**4**)¹



Compound **1c** (0.42 g, 1.44 mmol) was dissolved in a mixture of anhydrous benzene (46 mL) and anhydrous ethanol (2.50 mL), cooled to 0 °C and saturated with HCl gas. The mixture was sealed and allowed to stir at room temperature for 3 days after which anhydrous Et₂O (40 mL) was introduced and the mixture was allowed to stir for 10 minutes. The solids were filtered under nitrogen and dissolved in a mixture of anhydrous EtOH (34 mL) and EtOH.NH₃ (34 mL). The mixture was heated overnight (50 °C), cooled to room temperature and reduced by half *in vacuo*. Ether (15 mL) was added to precipitate the solid which was filtered, washed and dried under vacuum. Purification by recrystallisation (2N HCl) gave the desired compound **4** as fine white needles (0.46 g, 73%). Mp 286-287°C', 'H NMR (MeOD, 400MHz) δ , 7.80 (d, 4H, *J* = 9.0 Hz, ArH), 7.14 (d, 4H, *J* = 9.0Hz, ArH), 4.19 (m, 4H, CH₂), 2.02 (m, 4H, CH₂); ¹³ CNMR (MeOD, 100MHz) δ 165.7, 131.4, 121.2, 116.7, 69.7, 27.2; v_{max} (Nujol) /cm⁻¹ 3370 (N-H), 3129 (Ar-H), 2884 (C-H), 1650 (C≡N), 1606 (Ar), 1508 (Ar), 1257 (C-O); *m*/*z* (ESP) 327 ([M+H]⁺); anal. Found C 49.87, H 6.46, N 12.67, C₁₈H₂₈N₄O₄Cl₂ requires C 49.66, H 6.48, N 12.87.

4-(4-phenoxybutoxy) benzonitrile



Sodium (0.16g, 6.96 mmol) was added dropwise to dry ethanol (5 mL) and dissolved under a nitrogen atmosphere. To this a solution of 4-cyanophenol (0.53g, 4.47 mmol) dissolved in anhydrous ethanol (5ml) was added followed by addition of 1,4-dibromobutane (0.53 mL, 4.47 mmol). The reaction mixture was allowed to stir at reflux and monitored by TLC. After consumption of 4-cyanophenol, the reaction mixture was allowed to cool to room temperature. In a separate flask sodium (0 16 g, 6.96 mmol) was added portionwise to ethanol (5 mL) stirring under nitrogen. A solution of phenol (0.42 g, 4.47 mmol) in ethanol (5ml) was added and stirred for 10 minutes. This mixture was added dropwise to the cooled mixture and allowed to stir under reflux for 3 days after which the mixture was cooled, filtered, the solid washed with water and dried under vacuum. Purification by column chromatography eluting with DCM: hexane (8:2) gave the desired compound 4-(4phenoxybutoxy) benzonitrile as a white solid (0.98 g, 82%). Mp 130°C; ¹H NMR (CDCl₃, 400MHZ) δ 7.57 (d, 2H, J = 8.9 Hz, ArH), 7.28 (d, 1H, J = 7.5 Hz, ArH), 7.26 (d, 1H, J = 8.1 Hz, ArH), 6.92 (m, 5H, ArH), 4.08 (t, 2H, J = 5.9 Hz, CH₂), 4.03 (t, 2H, J = 5.9 Hz, CH₂), 1.99 (m, 4H, CH₂); ¹³C NMR (CDCl₃, 100MHz) δ 162.6, 159.2, 134.3, 129.8, 121.1, 115.5, 114.8, 104.3, 68.3, 67.5, 26.2; v_{max} (Nujol) /cm⁻¹ 3043 (Ar-H), 2884 (C-H), 2219 (C=N), 1602 (Ar), 1504 (Ar), 1247 (C-O); m/z (CI) 285 ([M+NH₄]⁺), found 285.16020, $C_{17}H_{21}N_2O_2$ requires 285.16031; anal. Found C 76.40, H 6.46, N 5.44, C₁₇H₁₇NO₂ requires C 76.38, H 6.40, N 5.24.

4-(4-Phenoxybutoxy)benzimidamide hydrochloride hydrate (Compound CHI/1/69/1)



4-(4-phenoxybutoxy) benzonitrile (0.27 g, 1.01 mmol) was dissolved in a mixture of anhydrous benzene (100 mL) and ethanol (1.60 mL), cooled to 0 °C and saturated with HCl gas. The mixture was sealed and allowed to stir at room temperature for 3 days after which anhydrous Et₂O (16 mL) was introduced and the mixture was allowed to stir for an additional 10 minutes. The solids were filtered under nitrogen and dissolved in a mixture of anhydrous EtOH (20 mL) and anhydrous EtOH.NH₃ (20 mL). The mixture was heated overnight (50 °C), cooled to room temperature and reduced by half *in vacuo*. Ether (30 mL) was added to precipitate the solid which was filtered, washed and dried under vacuum. Purification by recrystallisation (2N HCl) gave the desired compound 4-(4-Phenoxybutoxy)benzimidamide hydrochloride hydrate as fine white needles (0.28 g, 82%). Mp 134-135°C; ¹H NMR (DMSO, 400MHz) δ 9.28 (s, 2H, NH;), 9.08 (s, 2H, NHZ), 7.86 (d, 2H, *J* = 9.0 Hz, ArH), 7.29 (d, 1H, *J* = 7.0 HZ, ArH), 7.27 (d, 1H, *J* = 7.2 Hz, ArH), 7.16 (d, 2H, *J* = 9.0 Hz, ArH), 6.93 (d, 3H, *J* = 7.8 Hz, ArH), 4.16 (t, 2H, *J* = 5.9 Hz, CH₂), 4.03 (t, 2H, *J* = 5.9 Hz, CH₂), 1.89 (m, 4H, CH₂); ¹³C NMR (DMSO, 100MHZ) δ 165.0, 163.3, 158.9, 130.5, 129.8, 120.7, 119.6, 115.1, 114.7,

68.1, 67.2, 25.6, 25.5; v_{max} (Nujol) /cm⁻¹ 3288 (N-H), 1656 (C=N-H), 1604 (Ar), 1506 (Ar), 1234 (C-O-C); *m*/*z* (ESP) 285 ([M+H]⁺), found 285.1603, C₁₇H₂₄N₂O₂ requires 285.1599; anal. Found C 59.50, H 6.75, N 8.33, C₁₇H₂₃N₂O₃Cl requires C 60.26, H 6.84, N 8.27.

4-(5-(p-Tolyloxy)pentyloxy)benzonitrile



Sodium (0.12 g, 5.22 mmol) was added portionwise to anhydrous ethanol (4.0 mL) and dissolved under a nitrogen atmosphere. To this a solution of 4-cyanophenol (0.57 g, 4.79 mmol) dissolved in anhydrous ethanol (4.0 mL) was added followed by dropwise addition of 1,5-dibromopentane (0.65 mL, 4.79 mmol). The reaction mixture was allowed to stir at reflux and monitored by TLC. After consumption of 4-cyanophenol the reaction mixture was cooled to room temperature. In a separate flask, sodium (0.57 g, 4,79 mmol) was added portionwise to anhydrous EtOH (4.0 ml) with stirring under nitrogen. To this, a solution of p-cresol (0.5 ml, 4.79 mmol) in anhydrous EtOH (4.0 ml) was added and stirred for 10 minutes. This mixture was added dropwise to the cooled mixture and stirred under reflux for 3 days after which the mixture was cooled, filtered and the solid washed with water and dried under vacuum. Purification by column chromatography eluting with DCM: hexane (8:2) gave the desired compound as a white solid (1.02 g, 72%). Mp 133°C; ¹H NMR (CDCl₃, 400MHz) δ 7.57 (d, 2H, J = 9 Hz, ArH), 7.07 (d, 2H, J = 8.6 Hz, ArH), 6.93 (d, 2H, J = 9.0 Hz, ArH), 6.79 (d, 2H, J = 8.6 Hz, ArH) 4.02 (t, 2H, J = 6.4 Hz, CH₂) 3.96 (t, 2H, J = 6.4 Hz, CH₂), 2.28 (s, 3H, CH3), 1.86 (m, 4H, CH₂), 1.64 (m, 2H, CH₂); ¹³C NMR (CDCl₃, 100MHz) δ 162.7, 157.2, 134.3, 130.3, 130.2, 115.5, 114.7, 104.1, 68.5, 68.0, 29.4, 29.1, 23.0, 20.8; ν_{max} (Nujol) /cm⁻¹ 3322 (C-O-C), 3031 (Ar-H), 2921 (C-H), 2223 (C=N), 1602 (Ar), 1506 (Ar), 1234 (C-O-C); m/z (CI) 313 ([M+NH₄]⁺), found 313.19092, C₁₉H₂₅N₂O₂ requires 313.19162; anal. Found C 77.19, H 7.13, N 5.02, C₁₉H₂₁NO₂ requires C 77.26, H 7.17, N 4.74.

4-(5-(*p***-tolyloxy)pentyyloxy)benzimidamide hydrochloride hydrate** (Compound CHI/1/72/1)



4-(5-(*p*-Tolyloxy)pentyloxy)benzonitrile (0.27 g, 0.91 mmol) was dissolved in a mixture of anhydrous benzene (100 mL) and anhydrous ethanol (1.60 mL), cooled to 0 °C and saturated with HCl gas. The mixture was sealed and allowed to stir at room temperature for 3 days after which anhydrous Et₂O (20 mL) was introduced and the mixture was allowed to stir for 10 minutes. The solids were filtered under nitrogen and dissolved in a mixture of anhydrous EtOH (20 mL) and EtOH.NH₃ (20 mL). The mixture was heated overnight (50 °C), cooled to room temperature and reduced by half *in vacuo*. Ether (30 mL) was added to precipitate the solid which was filtered, washed and dried under vacuum. Purification by recrystallisation (2N HCl) gave the desired compound as fine white needles (0.25 g, 75%). Mp 132°C; ¹H NMR (DMSO, 400MHZ) δ 7.84 (d, 2H, *J* = 9.1 Hz, ArH), 7.15 (d, 2H, *J* = 9.1

Hz, ArH), 7.06 (d, 2H, J = 8.4 Hz, ArH), 6.80 (d, 2H, J = 8.4 Hz, ArH), 4.11 (t, 2H, J = 6.4 Hz, CH₂), 3.93 (t, 2H, J = 6.4 Hz, CH₂), 2.22 (s, 3H, CH₃), 1.78 (m, 4H, CH₂), 1.56 (m, 2H, CH₂); ¹³C NMR (DMSO, 100MHz) δ 165.0, 163.3, 158.9, 130.5, 130.1, 115.1, 114.5, 79.5, 79.3, 79.0, 20.4; v_{max} (Nujol) /cm⁻¹ 3430 (NH), 3309 (C-O-C), 3093 (Ar-H), 1658 (C=N-H), 1606 (Ar), 1508 (Ar), 1245 (C-O-C); m/z (ESP) 313 ([M+H]⁺), found 313.1916, C₁₉H₂₅N₂O₂ requires 313.1907; anal. Found C 62.19, H 7.42, N 7.66, C₁₉H₂₇ClN₂O₃ requires C 62.20, H 7.42, N 7.66.

Synthesis of ER 1004

4-bromobenzothioamide



Triethylamine (3.8 mL) was added to a solution of 4-bromobenzonitrile (5 g, 27.5 mmol) in pyridine (17 mL). The solution was cooled to 10 °C and H₂S (g) was bubbled through for 15 min. The resulting green solution was allowed to stir overnight (17 h). Nitrogen was bubbled through for 1 h to remove any excess H₂S. Water (27 mL) was added and the mixture was stirred for 10 min, a further portion of water (62 mL) was added and the pale yellow suspension left stirring overnight. The precipitate was filtered and rinsed with water to afford the title compound as bright yellow crystals (5.52 g, 93%). ¹H NMR (d₆-acetone, 400MHz) 9.07 (bs, 1H, NH), 8.92 (s, 1H, NH), 7.94 (dd, 2H, *J* = 2.0, 6.5 Hz, ArH), 7.62 (dd, 2H, *J* = 2.0, 6.5 Hz, ArH); ¹³C NMR (d₆-acetone, 100MHz) 201.9, 140.2, 132.3, 130.4, 126.5; *m/z* (CI) 216 (100%, [M⁺])

2,4-bis(4-bromophenyl)thiazole



2,4'-dibromoacetophenone (1 g, 3.60 mmol) was added to a solution of 4bromobenzothioamide (777 mg, 3.60 mmol) in EtOH (15 mL) and warmed to 45 °C for 1 h. The mixture was cooled to room temperature and left for 30 min before filtering. The precipitate was washed with EtOH: water (3:1, 10 mL) and dried to afford the thiazole as a pale solid (1.33 g, 94%). ¹H NMR (CDCl₃, 250MHz) 7.89 (d, 2H, J = 8.5 Hz, ArH), 7.85 (d, 2H, J = 8.5 Hz, ArH), 7.59 (d, 2H, J = 5.5 Hz, ArH), 7.56 (d, 2H, J = 5.5 Hz, ArH), 7.47 (s, 1H, CH); m/z (Cl) 396 (10%, [M+H]⁺).

4,4'-(thiazole-2,4-diyl)dibenzonitrile



A suspension of 2,4-bis(4-bromophenyl)thiazole (1 g, 2.53 mmol) and CuCN (906 mg, 10.12 mmol) in anhydrous DMF (15 mL) were heated to reflux for 21 h. On cooling, the reaction mixture was poured into aqueous NH₄OH (10%, 50 mL) and extracted with CHCl₃ (100 mL). Both layers were filtered to remove the dark precipitate. The organic layer was washed with water (2 x 50 mL), brine (50 mL) and dried MgSO₄. Removal of solvent gave a dark oily solid. Purification by column chromatography eluting with CHCl₃, afforded the title compound as a pale solid (361 mg, 50%). ¹H NMR (CDCl₃, 400MHz) 8.15 (d, 2H, *J* = 8.5 Hz, ArH), 8.11 (m, 2H, ArH), 7.77 (d, 2H, *J* = 8.5 Hz, ArH), 7.73 (t, 2H, *J* = 3.0 Hz, ArH), 7.26 (s, 1H, CH); ¹³C NMR (CDCl₃, 100MHz) 166.5, 137.4, 133.2, 130.1, 127.3, 119.1, 117.1, 114.2, 112.4; *m/z* (CI) 288 (100 %, [M+H]⁺).

4,4'-(thiazole-2,4-diyl)dibenzimidamide (ER1004)



The Garigipati Reaction is a little known reaction which effects the conversion of hindered nitriles to unsubstituted amidines in a mild and effective manner^{2, 3}. This is an efficient one step transformation involving direct nucleophilic addition of an amine to a nitrile, affording the corresponding amidine (Scheme 1).



Scheme 1.

The alkylchloroaluminium amides are effectively generated from trimethyl aluminium and ammonium chloride and the intermediate aluminium complex is easily hydrolyzed by water adsorbed on silica gel (Scheme 2).



Scheme 2.

The freshly prepared alkylchloroaluminum reagents (4 mL, 0.67 M, 2.7 mmol) were added to 4,4'-(thiazole-2,4-diyl)dibenzonitrile (78 mg, 0.27 mmol) in anhydrous toluene (1 mL) and heated to 80 °C overnight under nitrogen. On cooling, the aluminium complex was decomposed by pouring into a slurry of silica gel (2 g) in CHCl₃. The mixture was stirred for 5 min before filtering, the filter cake was washed with MeOH (20 mL). Removal of solvent gave the crude amidine as a pale solid in quantitative yield. The crude product (100 mg) was purified by reverse phase HPLC using a YMC-pack ODS-A column (250 x 20 mm I.D, 5 μ M) eluting with CH₃CN: Water 0.1% TFA (20-80 % gradient over 20 min). Removal of solvent afforded the desired compound as an off-white solid (38 mg, 22%). Mp 271-272 °C; ¹H NMR (DMSO, 400MHz) 9.32 (bs, 6H, NH & NH₂), 8.54 (s, 1H, CH), 8.25 (d, 2H, *J* = 8.5 Hz, ArH), 8.22 (d, 2H, *J* = 8.5 Hz, ArH), 7.92 (d, 2H, *J* = 8.5 Hz, ArH), 7.90 (d, 2H, *J* = 8.5 Hz, ArH); ¹³C NMR (DMSO, 100MHz) 165.5, 165.4, 154.4, 138.7, 137.4, 129.6, 129.2, 127.9, 126.9, 126.8, 119.6; *m/z* (ES) 322 (88 % [M+H]⁺); Found (ES) 322.1121 C₁₇H₁₆N₅S requires 322.1126; anal. Found C 46.05, H 3.10, N 12.44, S 5.77, C₂₁H₁₈N₅O₄F₆S requires C 45.91, H 3.12, N 12.74, S 5.83.

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Chemistry of new compounds from David Boykin laboratory.

General procedure for conversion of nitriles into amidine hydrochlorides (Method A).

To a cold and stirred suspension of the nitrile or dinitrile (0.001 mol) in 15 ml dry THF was added 6.0 ml, (0.006 mol) LiN(TMS)₂ (1 M in THF), stirred for 24 h, cooled, acidified carefully with saturated ethanolic-HCl, the precipitated white solid stirred for 2 h, solvent removed under reduced pressure, diluted with ether, filtered. The collected solid was added to 10 ml ice water, basified with 2M NaOH, the precipitate was filtered, washed with water and air dried. The solid was suspended in anhydrous ethanol (15 ml) and 5 ml saturated ethanolic-HCl and stirred for 6 h, ethanol was distilled off, triturated with dry ether and filtered. The solid was dried under reduced pressure at 80° C for 12 h to yield (70-75%) amidine hydrochloride.

General procedure for conversion of nitriles into amidine hydrochlorides (Method B).

A suspension of the nitrile or dinitile (0.001 mole) in 20 ml saturated ethanol-HCl was stirred for 4 days in a closed stoppered flask, followed by precipitation with anhydrous ether. The precipitated light yellow solid imidate ester dihydrochloride was filtered and dried under reduced pressure for 3 h to yield (65% -70%) amidine hydrochloride. The imidate ester dihydrochloride (0.0005 mole) in 20 ml anhydrous ethanol was saturated with ammonia(g) or 0.03 equivalents of ethylene diamine stirred in ethanol (at reflux) for 12 h, solvent removed, 20 ml ice water added, basified to pH 10 with aqueous 2N NaOH, filtered, and washed with water. The precipitated solid was dried in air, suspended in 10 ml of saturated ethanolic-HCl, and stirred for 2 h. The solvent removed, dry ether 20 ml added, filtered, washed with ether and dried under reduced pressure for 12 h. The product amidine hydrochloride was obtained as yellow solid 60-66% yield.

4-(5-(4-methoxyphenyl) furan-2-yl) benzimidamide hydrochloride (DB 607)

A mixture of 5-(4-cyanophenyl)-2-bromo furan¹ (1.23 g, 0.005 mole) and 4-methoxyphenyl boronic acid (0.93 g, 0.006 mole) in 75 ml dioxane under nitrogen was added K₂CO₃ (1.38 g, 0.01 mole, in 5 ml H₂O), followed by Pd(PPh₃)₄ 0.12 g (0.0001 mole) and the solution was heated under reflux for 12-24 h (tlc monitored). The solvent was removed under reduced pressure, solid filtered, washed with hexane and dried in air. The solid was suspended in DCM (100 ml), filtered through celite, concentrated under reduced pressure, triturated with ether: hexane (2:1), and filtered to yield 4-(5-(4-methoxyphenyl)furan-2-yl)benzonitrile as a yellow brown solid 0.76 g (74%) mp 250-2 °C dec ; ¹H NMR (DMSO-d₆): 7.94 (d, 2H, J= 10.4 Hz), 7.85 (d, 2H, J= 10.4 Hz), 7.78 (d, 2H, J= 10.8 Hz), 7.28 (d, 1H, J= 4.4 Hz), 7.02 (d, 2H, J= 10.8 Hz), 6.97 (d, 1H, J= 4.4 Hz), 3.81 (s, 3H); ¹³C NMR (DMSO-d₆): 159.3, 154.5, 150.0, 134.1, 132.9, 125.4, 123.5, 122.5, 119.0, 114.4, 111.8, 108.8, 106.9, 55.2; MS: HRMS-ESI-POS: Calcd. for C₁₈H₁₄NO₂ m/z 276.1024 (M⁺+1), found m/z 276.1021.

The amidine hydrochloride was obtained as yellow solid (Method A) 0.24 g (74%) ; mp >318°C dec ; ¹H NMR (DMSO-d₆): 9.41 (brs, 2H), 9.1 (brs, 2H), 8.01 (d, 2H, J= 8.4 Hz), 7.92 (d, 2H, J= 8.4 Hz), 7.82 (d, 2H, J= 8.4 Hz), 7.324 (d, 1H, J= 3.6 Hz), 7.04 (d, 2H), 7.02 (d, 1H, J= 3.6 Hz), 3.82(s, #H); ¹³C NMR (DMSO-d₆): 164.9, 159.2, 154.3, 150.2, 134.8, 128.8, 125.5, 125.4, 123.0, 122.5, 114.4, 111.4, 106.8, 55.2; ; MS: HRMS-ESI-POS.: Calcd. for C₁₈H₁₇N₂O₂ m/z 293.1289 (M⁺+1), found m/z 293.1274; Anal. calcd. for C₁₈H₁₆N₂O₂-HCl: C, 65.75; H, 5.21; N, 8.52; Found: C, 65.78; H, 5.23; N, 8.44.

4,4'-(thiophene-2,4-diyl) dibenzimidamide dihydrochloride (DB 1077)

A mixture of 2, 4-dibromothiophene 1.21 g (0.005 mole), 4-cyanophenylboronic acid 1.75 g (0.012 mole) following procedure for DB 607, yielded 2,4-(4-cyanophenyl)thiophene as a yellow solid, 0.86 g (72%) mp >220°C dec ; ¹H NMR (CDCl₃): 7.75-7.21 (m, 8H), 7.71 (d, 1H, J= 1.2 Hz), 7.64 (d, 1H, J= 1.2 Hz); ¹³C NMR (CDCl₃): 143.6, 141.6, 139.4, 138.0, 132.9, 132.8, 126.8, 126.1, 123.8, 123.7, 118.8, 118.6, 111.3, 111.1; MS: HRMS-ESI-POS: Calcd. for $C_{14}H_{11}N_2S m/z$ 239.0642 (M⁺+1), found m/z 239.0639.

The diamidine hydrochloride was obtained as yellow solid (Method A) 0.32 g (74%) mp>300°C dec ; ¹H NMR (DMSO-d₆): 9.55 (brs, 2H), 9.54 (brs, 2H), 9.31 (brs, 4H), 8.42 (s, 1H), 8.32 (s, 1H), 8.09 (d, 2H, J= 8.4 Hz), 8.03-7.97 (m, 6H); ¹³C NMR (DMSO-d₆): 164.9, 164.8, 142.5, 140.9, 139.5, 138.3, 129.1, 128.8, 126.6, 126.3, 126.2, 125.4, 125.0, 124.8; MS: HRMS-ESI-POS.: Calcd. for $C_{18}H_{18}N_4S$ *m/z* 161.0626 (M⁺+2)/2, found *m/z* 161.0621; Anal. calcd. for $C_{18}H_{16}N_2S$ -2HCl-2H₂O: C, 50.35; H, 5.16; N, 13.05; Found: C, 50.52; H, 5.23; N, 13.22.

3,3'-(furan-2,5-diyl bis (4,1-phenylene)) dipropanimidamide dihydrochloride (DB 1061) To a mixture of 4-bromophenyl propionitrile 0.63 g (0.003 mole) and 2, 5-bis (tributylstannyl) furan in 30 ml anhydrous dioxane under nitrogen was added Pd(PPh₃)₄ 0.14 g (0.00012 mole) and the solution was heated under reflux for 12 h (tlc monitored). The solvent was removed under reduced pressure, the solid was filtered, washed with hexane and dried in air. The solid was suspended in DCM (50 ml), stirred 2 h with 20 ml 10% KF (aqueous), the organic layer separated, filtered through celite dried over anhydrous MgSO₄, filtered, concentrated, triturated with hexane and the solid filtered was filtered to yield 0.34 g (70%) of 3,3'-(furan-2,5-diylbis(4,1-phenylene))dipropanenitrile as a yellow solid mp 120-2°C dec; ¹H NMR (CDCl₃): 7.73 (d, 4H, J= 8.4 Hz), 7.30 (d, 4H, J= 8.4 Hz), 7.28 (s, 2H), 3.0 (t, 4H, J= 7.6 Hz), 2.66 (t, 4H, J= 7.6 Hz); ¹³C NMR (CDCl₃): 153.2, 137.2, 130.0, 128.9, 124.4, 119.2, 107.2, 31.5, 19.5; MS: HRMS-ESI-POS: Calcd. for C₂₂H₁₈N₂ONa *m/z* 349.1317 (M⁺+Na), found *m/z* 349.1332.

The diamidine dihydrochloride was obtained using Method B: 0.14 g (60%) mp>300°C dec ; ¹H NMR (DMSO-d₆): 9.13 (brs, 4H), 8.74 (brs, 4H), 7.64 (d, 4H, J= 8.4 Hz), 7.49 (s, 2H), 7.33 (d, 4H, J=8.4 Hz), 3.0 (t, 4H, J= 7.2 Hz), 2.74 (d, 4H, J= 7.2 Hz); ¹³C NMR (DMSO-d₆): 170.6, 142.7, 139.4, 132.4, 129.6, 125.8, 125.2, 33.7, 32.03; MS: HRMS-ESI-POS: Calcd. for C₂₂H₂₆N₄O *m*/*z* 181.1053 (M⁺+2)/2, found *m*/*z* 181.1048; Anal. calc. for C₂₂H₂₄N₄O-2HCl-2H₂O: C, 56.29; H, 6.44; N, 11.93; Found: C, 56.35; H, 6.54; N, 11.86.

2,5-bis(4-(2-(4,5-dihydro-1H-imidazol-2-yl) ethyl) phenyl) furan dihydrochloride (DB 1062)

Similarly, 0.245 g (0.005 mole) of the above imidate ester in 20 ml of anhydrous ethanol was allowed to react under reflux (12 h) with 0.06 g (0.0015 mole) ethylene diamine. The solvent was removed under reduced pressure, diluted with water, solid was filtered, dried and converted to dihydrochloride using ethanolic-HCl to yield a yellow solid, 0.15 (62%), mp >325 °C, H NMR (DMSO-d₆): 8.29 (br, 4H), 8.74 (brs, 4H), 7.77 (d, 4H, J= 8.4 Hz), 7.34 (s, 2H), 7.05 (d, 4H, J=8.4 Hz), 3.0 (t, 4H, J= 7.2 Hz), 2.74 (d, 4H, J= 7.2 Hz); ¹³C NMR (DMSO-d₆): 170.2, 152.4, 138.5, 128.8, 128.6, 123.6, 108.0, 44.0, 30.5, 27.4; MS: HRMS-ESI-POS: Calcd. for C₂₆H₃₀N₄O *m*/*z* 207.1209 (M⁺+2)/2, found *m*/*z* 207.1203; Anal. calc. for C₂₆H₂₈N₄O-2HCl-2.75H₂O: C, 58.37; H, 6.68; N, 10.47; Found: C, 58.45; H, 6.54; N, 10.63.

3,3'-(thiophene-2,5-diylbis(4,1-phenylene)) dipropanimidamide dihydrochloride (DB 1063)

The dinitrile, 3,3'-(thiophene-2,5-diylbis(4,1-phenylene))dipropanenitrile, was prepared as described for DB1061 yielding a yellow solid 0.77 g (75%); mp 124-6^oC dec.;¹H NMR

(CDCl₃): 7.62 (d, 4H, J= 8.0 Hz), 7.29 (d, 4H, J= 8.0 Hz), 7.28 (s, 2H), 3.0 (t, 4H, J= 7.2 Hz), 2.66 (t, 4H, J= 7.2 Hz); ¹³C NMR (CDCl₃): 143.3, 137.5, 133.5, 129.1, 126.2, 124.2, 119.2, 31.4, 19.4; MS: HRMS-ESI-POS: Calc. for $C_{22}H_{18}N_2SNa$ *m/z* 365.1088 (M⁺+Na), found *m/z* 365.1089.

Similarly following the DB 1061 procedure the diamidine dihydrochloride was obtained as a yellow solid 0.16 g (66%), mp >280^oC dec; ¹H NMR (DMSO-d₆): 9.19 (brs, 4H), 8.77 (brs, 4H), 7.64 (d, 4H, J= 8.0 Hz), 7.51 (s, 2H), 7.33 (d, 4H, J=804 Hz), 2.99 (t, 4H, J= 8.4 Hz), 2.73 (d, 4H, J= 8.4 Hz); ¹³C NMR (DMSO-d₆):170.0, 142.2, 138.9, 131.9, 129.1, 125.3, 124.7, 33.2, 31.6; MS: HRMS-ESI-POS: Calcd. for $C_{22}H_{26}N_4S$ *m/z* 189.0939 (M⁺+2)/2, found *m/z* 189.0931; Anal. calcd. for $C_{22}H_{24}N_4S$ -2HCl-1.5H₂O: C, 55.45; H, 6.13; N, 11.76; Found: C, 55.52; H, 6.34; N, 11.63.

2, 5-bis(4-(2-(4,5-dihydro-1H-imidazol-2-yl) ethyl) phenyl) thiophene dihydrochloride (DB 1064)

Similarly following the procedure for DB1062 the diamidine dihydrochloride was obtained as a yellow solid, 0.17 g (62%); mp >225°C dec.; ¹H NMR (DMSO-d₆): 10.19 (s, 4H), 7.63 (d, 4H, J=7.6 Hz), 7.5 (s, 2H), 7.30 (d, 4H, J= 7.6 Hz), 3.78 (s, 8H),2.97 (t, 4H, J= 6.4 Hz), 2.81(t, 4H, J= 6.4 Hz); ¹³C NMR (DMSO-d6): 170.2, 140.2, 138.9, 131.9, 129.1, 125.4, 124.8, 44.1, 30.6, 27.4; MS: HRMS-ESI-POS: Calcd. for C₂₆H₂₉N₄S *m/z* 429.2113 (M⁺+1), found *m/z* 429.2109; Anal. calcd. for C₂₆H₂₈N₄S-2HCl-3.0H₂O: C, 56.21; H, 6.53; N, 10.08; Found: C, 56.34; H, 6.61; N, 10.24.

4-(5-(1-methyl-1H-benzo[d]imidazol-2-yl) furan-2-yl) benzimidamide dihydrochloride (DB 960)

To a stirred solution of 5-(4-cyanophenyl) furan-2-aldehyde² 1.23 g (0.005 mole), 1-amino-2-*N*-(methylamino) benzene 0.61 g (0.005 mol) in 20 ml dry DMF under N₂ was added sodium metabisulfite 0.95 (0.005 mol) and the mixture was heated at 130°C for 12 h (tlc monitored). The solvent was removed, the residue was triturated with cold water, separated solid was filtered, washed with water and air dried. The solid was stirred with 1:1 mixture of DCMether, filtered and dried in vac at 70°C for 4 h to give the nitrile, 4-(5-(1-methyl-1Hbenzo[d]imidazol-2-yl)furan-2-yl)benzonitrile, as a yellow brown solid, 1.1 g (72%), mp >290°C dec ; ¹H NMR (DMSO-d₆): 8.03 (d, 2H, J= 8.4 Hz), 7.92 (d, 2H, J= 8.4 Hz), 7.69-7.63 (m, 2H), 7.46 (d, 1H, J= 3.6 Hz), 7.40 (d, 1H, J= 3.6 Hz), 7.34-7.23 (m, 2H), 4.13 (s, 3H); ¹³C NMR (DMSO-d₆): 152.4, 145.9, 143.2, 142.4, 136.0, 133.2, 132.9, 124.2, 122.7, 122.2, 118.9, 118.5, 114.7, 111.1, 110.2, 109.9, 31.4; MS: HRMS-ESI-POS: Calcd. for C₁₉H₁₄N₃O *m/z* 300.1136 (M⁺+1), found *m/z* 300.1132.

The amidine hydrochloride was obtained as yellow solid (Method B) 0.3 g (78%); mp >300°C dec; ¹H NMR (DMSO-d₆): 8.09 (d, 2H, J= 8.7 Hz), 7.90 (d, 2H, J= 8.7 Hz), 7.78-7.70 (m, 2H), 7.61 (d, 1H, J= 3.9 Hz), 7.48-7.42 (m, 2H), 7.45 (d, 1H, J= 3.9 Hz), 4.13 (s, 1H); ¹³C NMR (DMSO-d₆): 165.5, 155.6, 141.5, 141.3, 135.3, 134.8, 133.9, 129.5, 128.0, 126.0, 125.8, 125.5, 119.6, 116.4, 112.4, 112.0, 32.9; MS: HRMS-ESI-POS: Calcd. for C₁₉H₁₈N₄O *m*/*z* 159.0740 (M⁺+2)/2, found *m*/*z* 159.0733; Anal. calcd. for C₁₉H₁₆N₄O-2HCl-1H₂O: C, 56.02; H, 4.94; N, 13.76; Found: C, 56.18; H, 4.91; N, 13.51.

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