

Chapter 6

Microalgae–methanotroph cocultures for carbon and nutrient recovery from wastewater

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ABSTRACT

Wastewater produced from municipal, agricultural, and industrial processes has caused detrimental impacts on local communities and environments. In addition, wastewater is the fifth largest anthropogenic source of methane emissions globally. Although anaerobic digestion is a proven waste management technology with many environmental benefits, its application is limited to large-scale water resource recovery facilities. This is due to the poor return-on-investment caused by the contaminants present in raw biogas and nutrient-rich liquid digestate that require further treatments, some of which are costly. In this chapter, we discuss our recent development of a microalgae–methanotroph coculture-based platform for integrated biogas valorization and nutrient recovery. Development of coculture-based biotechnology faces many technical challenges, including tracking the growth of individual species in the coculture over time, quantifying and understanding inter-species metabolic interactions, and developing kinetic models for the coculture. There are also many practical considerations when applying coculture-based biotechnology for wastewater treatment. We discuss our proposed solutions to address these technical challenges and practical concerns. We also offer our perspective on future directions.

Keywords: anaerobic digestion, biogas valorization, microalgae–methanotroph coculture, nutrient recovery, photobioreactor

6.1 BACKGROUND

Municipal, agricultural, and industrial processes produce significant amounts of wastewater that contain organic carbon and high content of nitrogen, phosphorus, and other pollutants. If not adequately treated before discharge into waterways, wastewater can have detrimental impacts on local communities and the environment. Wastewater is the fifth largest anthropogenic source of methane (CH₄) emissions globally, contributing to worsening the greenhouse effect. Moreover, the chemicals and pollutants found in untreated wastewater, such as chlorofluorocarbons and other ozone-depleting substances, can have a harmful impact on the ozone layer. When these substances are released into the atmosphere, they can react with and destroy ozone molecules in the stratosphere, harming the protective ozone layer. This, in turn, can increase the amount of harmful ultraviolet radiation that

reaches the Earth's surface, causing various health problems for humans and damaging ecosystems. Finally, wastewater also contributes to smog, acid rain, and drinking water contamination (Driscoll *et al.*, 2003; Galloway *et al.*, 2004).

Wastewater resource recovery facilities (WRRFs) typically use a combination of physical, chemical, and biological treatment techniques to remove pollutants and contaminants from wastewater. Among them, nitrification–denitrification is a biological process most commonly used in WRRFs to remove nitrogen from wastewater. It involves the conversion of ammonia (NH_3) to nitrate (NO_3^-) by nitrifying bacteria, followed by the conversion of nitrate to nitrogen gas (N_2) by denitrifying bacteria. A nitrification–denitrification process requires energy to operate and can be a significant source of operating costs. For example, nitrification requires oxygen (O_2), typically supplied using aeration systems that blow air into the wastewater. This is often the most energy-intensive process in the treatment plant, and the energy consumption can be in the range of 25–60% of the total energy use. Pumps and mixers are used to move and mix wastewater through the treatment process. These devices require energy to operate and can account for 10–25% of the total energy use (Siegrist *et al.*, 2008). A denitrification process often requires supplementing an organic carbon source (e.g., methanol) to support nitrate reduction (Tam *et al.*, 1992; Zhao *et al.*, 1999). Pumping air and supplying organic carbon sources are the primary contributors to high-operational costs for WRRFs (Drewnowski *et al.*, 2019).

On the contrary, the organic carbon stranded in wastewater is a valuable and often overlooked resource for producing fuels and chemicals. By transforming wastewater treatment processes to extract and utilize this carbon, it is possible to not only mitigate the negative environmental and societal consequences of wastewater, but also generate revenue to offset treatment costs and potentially create a profitable industry. This potential has spurred increasing research interest in waste-to-value (W2V) technologies, which include waste-to-energy, waste-to-fuel, waste-to-chemical, and other similar processes (Fei *et al.*, 2014; Haynes & Gonzalez, 2014; Henard *et al.*, 2016).

Anaerobic digestion (AD) is commonly regarded as the most effective waste management solution for wet organic waste and is currently the only widely commercialized W2V process at scale. In a controlled and contained manner, AD breaks down organic matter into biogas (consisting primarily of CH_4 and carbon dioxide (CO_2)), with trace amounts of other gases such as hydrogen sulfide (H_2S)), and nutrient-rich digestate (containing valuable elements such as N, P, and K). Furthermore, AD is highly effective at mitigating odors and pathogens (Angelidaki & Ellegaard, 2003; Nasir *et al.*, 2012; Topper *et al.*, 2006). Given these benefits, AD has gained significant traction for converting stranded organic carbon in wastewater to biogas and is widely adopted by large-scale WRRFs. In the USA, 48% of total wastewater flows are treated by AD, with at least 1,238 WRRFs (out of 14,780 WRRFs in the USA) processing solids through AD (Qi *et al.*, 2013).

Many W2V technologies could utilize AD as a key process. The three most prominent ones are: (1) combined heat and power (CHP), which utilizes biogas produced from AD as a fuel source to generate heat and/or electricity; (2) renewable natural gas or biomethane (BM) where biogas is upgraded to meet pipeline quality standards, allowing it to be used as a transportation fuel or injected into natural gas pipelines; and (3) microalgae cultivation (MC) utilizing AD-generated digestate (and CO_2) for biofuels or bioplastics production (Ahmed *et al.*, 2021; Angelidaki *et al.*, 2018; Kapoor *et al.*, 2020). Despite their wide recognition, there are challenges associated with these technologies. For example, both CHP and BM require significant capital expenditure and operating expenses due to the impurities present in biogas (e.g., H_2S and NH_3). In addition, both technologies produce low-value products (electricity, heat, or CH_4), further deteriorating their economic viability. As a result of poor return on investment (ROI), both CHP and BM have limited success in the USA. For example, among the WRRFs that have AD installed, 15% (~184) of them flare the produced biogas, 64% (~791) of them burn the biogas for digester and building heating, whereas only 21% (~263) of them use biogas for power generation or driving machinery (Qi *et al.*, 2013). Although European Union countries have built technologically and economically mature CHP and BM industries, this has primarily been

achieved through massive government subsidies that offset prices up to 10 times those of electricity and natural gas (Angelidaki *et al.*, 2018; Lombardi & Francini, 2020; Mishra *et al.*, 2021; Rosa, 2020; Scarlat *et al.*, 2018). As for integrated AD and microalgae cultivation (IADMC), the AD digestate provides the microalgae with nutrients, such as nitrogen and phosphorus, which are essential for their growth. The microalgae biomass can then be harvested and processed to produce biofuels, such as biodiesel and bioethanol, or other value-added products, such as nutraceuticals and cosmetics (Ho *et al.*, 2018; Rajagopal *et al.*, 2021; Vadiveloo *et al.*, 2021). IADMC achieves two goals simultaneously: AD provides a sustainable source of nutrients for MC, whereas MC helps reduce environmental impact of AD by recycling the nutrients produced. However, microalgae-based biofuels face several challenges that hinder their competitiveness with petroleum fuels, and sustainable production levels have not yet been achieved (Beckstrom *et al.*, 2020; Georgakopoulou, 2019; Richardson *et al.*, 2012; Sun *et al.*, 2011). These challenges include low biomass density, high costs associated with harvesting and downstream processing, and significant demands for water, energy, and land due to the limited light penetration in the liquid phase. Moreover, energetic and environmental viability of algae-based biofuels presents serious challenges that must be addressed (Choi *et al.*, 2021).

As mentioned earlier, wastewater is a significant contributor to anthropogenic CH₄ emissions, ranking as the fifth largest source globally and accounting for 7–9% of total global CH₄ emissions. The growing recognition of the prominent role of CH₄ emissions in climate change highlights the need for CH₄ remediation at WRRFs. This is especially crucial because CH₄ is 86 times more potent than CO₂ on a 20-year timescale and 34 times more potent on a 100-year timescale, making it a significant contributor to global warming. Multiple studies have demonstrated that mitigation of CH₄ can play an outsized role in limiting warming over the next few decades. It presents a potential for rapidly reducing climate warming, either in the near term to prevent temporary exceedance of the 1.5 or 2.0°C peak warming threshold, or later in the century to bring down temperatures after an overshoot to higher levels (Alvarez *et al.*, 2012; Collins *et al.*, 2018; Harmsen *et al.*, 2020; Ocko *et al.*, 2021; Rogelj *et al.*, 2015). Consequently, CH₄ remediation is a vital aspect of wastewater treatment and should be integrated into nutrient recovery.

6.2 OVERVIEW OF MICROALGAE–METHANOTROPH COCULTURES: A PROMISING W2V PLATFORM FOR WASTEWATER TREATMENT

In this chapter, we introduce a novel W2V biotechnology platform that employs a microalgae–methanotroph coculture for the integrated treatment of AD effluent (i.e., digestate) and conversion of biogas as shown in Figure 6.1. In microalgae–methanotroph coculture systems, a carefully selected microalgae–methanotroph coculture effectively converts both CO₂ and CH₄ in biogas to microbial biomass, while also recovering nutrients (e.g., N and P) from AD effluent to support microbial growth. We present some promising results from our laboratory experiments to demonstrate the potential of the microalgae–methanotroph coculture as a sustainable, profitable, and effective W2V platform.

The proposed microalgae–methanotroph coculture W2V platform is based on studies published in *Nature* (Raghoebarsing *et al.*, 2005) and *Nature Geoscience* (Kip *et al.*, 2010), which suggest that the

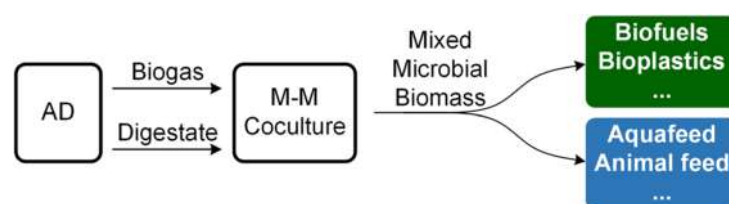


Figure 6.1 Overview of the proposed integrated AD and microalgae–methanotroph coculture W2V platform.

coupling of CH_4 oxidation (by methanotrophs) and oxygenic photosynthesis (by peat moss) is prevalent in wetlands to reduce CH_4 emissions and reuse CO_2 . Another report in *Nature ISME* (Milucka *et al.*, 2015) shows that true aerobic oxidation by methanotrophs, fueled by *in-situ* O_2 produced from photosynthetic algae, is responsible for CH_4 removal in anoxic waters. These discoveries suggest that a microalgae–methanotroph coculture could be a novel and effective approach to recycling biogas and nutrients.

The coupling of methanotrophs and microalgae offers several advantages as shown in Figure 6.2: (1) dissolved O_2 produced by microalgae will be consumed by methanotrophs for cell growth, which not only eliminates the safety risk of mixed CH_4 and O_2 , but also eliminates the O_2 inhibition to microalgae growth; (2) dissolved CO_2 produced by methanotrophs can further promote microalgae growth due to significantly reduced mass transfer resistance; and (3) microalgae growth can be significantly promoted by cofactors (e.g., biotin and thiamine), which are synthesized by methanotrophs (Cecchin *et al.*, 2018; Tandon *et al.*, 2017). By exploring the synergistic metabolic coupling of oxygenic photosynthesis and CH_4 oxidation, microalgae–methanotroph cocultures have demonstrated significantly increased biomass production and nutrient recovery. Compared to CHP, which only captures the energy contained in CH_4 while losing all carbon to CO_2 , the coculture can recover 100% of the carbon in CH_4 and CO_2 , resulting in zero emissions and close to 100% nutrient recovery (Roberts *et al.*, 2020).

To improve process economic feasibility, simple downstream processing that requires limited capital and operational cost is preferred. In addition, high-value products are needed to make the overall process profitable. With these considerations, coculture biomass produced from biogas and wastewater can serve as feedstock for animal feed supplements or bioplastics.

For wastewater produced from wineries and food-processing plants that are determined to be safe (e.g., low level of heavy metals and antibiotics), the coculture biomass could be used as single-cell protein (SCP) for aquafeed supplements. It is worth noting that both microalgae and methanotrophs have been studied extensively and tested as protein supplements for aquafeed. For example, trials in fish have shown that the protein meal derived from methanotrophs performs well as an alternative protein source to fish meal in feed formulations for Atlantic salmon (Aas *et al.*, 2006). Other studies on methanotroph-derived fish meal show improved growth performance and health benefits in aquatic and terrestrial animals (Øverland *et al.*, 2010; Romarheim *et al.*, 2010). For microalgae, positive testing results in fish and shrimp have suggested that a significantly higher dietary inclusion level of microalgal biomass in aquafeeds is expected (Becker, 2007; Gamboa-Delgado & Márquez-Reyes, 2018; Teimouri *et al.*, 2013). As a result, the coculture biomass of microalgae and methanotrophs could be a highly promising source for SCP.

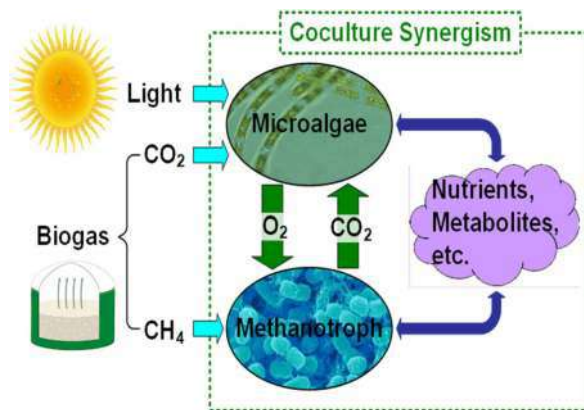


Figure 6.2 Synergistic interactions within microalgae–methanotroph cocultures.

For wastewater from municipal sewage and industrial processes, the produced coculture may not be safe for animal feed. In this case, they could be used as bioplastics feedstock (BPFS). Recent studies have shown that when simple and cost-effective processing of whole microalgal or bacterial cells is applied to produce bioplastics for packaging (which is in rapidly increasing demand), agricultural plastic products (e.g., horticultural containers), or apparel (e.g., footwear), these bioplastics products have been found to be immediately competitive without any government subsidy or incentive (Beckstrom *et al.*, 2020; Chia *et al.*, 2020; Choi *et al.*, 2021; Chong *et al.*, 2021; Coppola *et al.*, 2021; Karan *et al.*, 2019; Onen Cinar *et al.*, 2020; Tharani & Ananthasubramanian, 2020; Zeller *et al.*, 2013). This has been demonstrated by successful commercial applications at Algix – footwear brands, including Adidas and Merrell, who have launched commercial products containing bioplastics manufactured at Algix through thermo-mechanical molding of mixed whole-cell microbial biomass (Corcoran & Hunt, 2021). For microbial biomass to be used as BPFS, its protein content must be in the range of 35–60%; the protein content of the coculture biomass we produced from biogas and AD digestate was 43% ($\pm 4\%$) (Roberts *et al.*, 2020), ideal for BPFS.

In general, using mixed cultures as biocatalysts offers many advantages over conventional single or pure cultures. With mixed cultures, product yield and growth rate may be higher. In addition, mixed cultures usually enable better utilization of cheap, complex substrates. Mixed cultures often exhibit enhanced robustness and offer more protection against contamination. For example, studies have shown that, compared to a monoculture of microalgae, the cocultivation of microalgae and bacteria or fungi can deliver improved performance in terms of nutrient removal and biomass production. Zhang *et al.* (2020) systematically reviewed the recent progress in this area. However, using mixed cultures for bioconversion also presents many challenges. It is challenging to monitor and characterize a mixed culture in real time, which is the prerequisite for studying its growth kinetics. In addition, without real-time monitoring, it is almost impossible to maintain a mixed culture at its optimal state, such as controlling the optimal population of coculture to maximize substrate conversion. As a result, most existing research on mixed culture-based nutrient recovery from AD effluent mainly focused on the process performance (both biogas conversion and/or nutrient recovery), with few of them examining the inter-species interactions, and how such interactions would affect the overall performance of the process.

In this chapter, we present our recent progress on studying microalgae–methanotroph cocultures for biogas conversion and nutrient recovery, which includes the protocols for the coculture characterization in real time, kinetic modeling of the coculture, and many practical considerations for using microalgae–methanotroph cocultures for biogas conversion and wastewater treatment. The results presented in the following sections mainly utilized two-model coculture systems. One is *Arthrospira platensis*–*Methylobacterium buryatense* 5GB1, which is a cyanobacterium–type I methanotroph coculture that prefers high-pH and high-salt medium. Using this model coculture, we developed experimental and computational tools for studying microalgae–methanotroph cocultures. The other model coculture is *Chlorella sorokiniana*–*Methylococcus capsulatus* (Bath), which is a microalgae–type X methanotroph coculture that prefers neutral pH and low-salt medium. We demonstrated integrated wastewater treatment and biogas valorization using the latter coculture pair.

6.3 EXPERIMENTAL AND COMPUTATIONAL TOOLS FOR REAL-TIME CHARACTERIZATION OF THE MICROALGAE–METHANOTROPH COCULTURES

Multispecies associations are prevalent in nature, providing essential ecosystem services such as carbon, nutrient, and metal cycling. This can be explained by natural selection – mixed cultures can offer numerous benefits over monocultures, as discussed previously. Despite these advantages, the use of mixed cultures for biotechnological applications in bioenergy and related fields has been limited. This is mainly due to the fact that the successful commercialization of potential biotechnologies requires a comprehensive understanding of the fundamental biological conversion steps within microorganisms.

Acquiring this knowledge requires accurate characterization of cell growth dynamics, substrate conversion, and product excretion rates. However, there is a lack of effective tools to accurately characterize the mixed culture in real time. In addition, the involvement of gas substrates (both CH_4 and CO_2) makes the characterization of the coculture more challenging, as obtaining accurate measurements of gas component uptake or production rates can be tricky due to their high sensitivity to system pressure or volume changes. In the following section, we will first discuss the experimental protocols that can deliver accurate gas-phase measurements, then the experimental–computational (E–C) protocols to characterize the coculture in real-time are discussed.

6.3.1 Accurate measurement of gas component uptake and production rates in bioconversion

The challenges with measuring gas component consumption and production rates are not caused by the precision of analytical equipment (e.g., gas chromatography (GC)). Instead, they are rooted in the fact that the consumption and/or production of gases alter the system headspace pressure (for batch processes) or gas-phase flow rate (for continuous processes). For batch experiments conducted in closed systems with constant volume, such as vials, the system pressure often experiences significant reduction when gas substrate(s) are involved. This can be caused by the overall gas consumption exceeding gas production, as well as gas and/or liquid sampling. On the contrary, vials can become over-pressurized if gas production dominates. When a headspace gas sample is obtained from an under- or over-pressurized system with a gas-tight syringe, after the syringe is withdrawn from the system, air can enter the syringe if under-pressurized, or the gas sample can escape from the syringe if over-pressurized. Our study demonstrated that such pressure differences could cause significant errors in the measured gas composition. For continuous chemostat operation, the system pressure should be constant. However, due to the imbalance between gas-phase substrate consumption and gaseous product excretion, the off-gas flow rate can significantly differ from the feeding gas flow rate. For continuous operation, accurate off-gas flow rate measurements are critical for determining the gas consumption and production rates through the mass balance of the system.

In Stone *et al.* (2019), we reported two easy-to-implement experimental protocols and associated computation procedures to obtain accurate measurements of gas component consumption and production rates: one for batch experiments, the other for continuous operation. For depressurized (i.e., system pressure below 1 atm) batch cultures, we use N_2 (or other inert gases that do not interfere with the measurements of other gases) to repressurize the system to 1 atm before obtaining samples, whereas for pressurized systems, accurate measurements can be obtained by measuring the system pressure and scale-up the measured composition by multiplying the ratio between the system pressure and the atmospheric pressure. For continuous cultures, we use helium (or other inert gases that can be accurately measured by GC) as an internal tracer to accurately measure the off-gas flow rate. Two abiotic and two biotic systems were used to conduct several case studies to validate the effectiveness and accuracy of the protocols and associated computational procedures.

The effectiveness of the developed repressurization protocol for batch systems is demonstrated in Figure 6.3a. Figure 6.3a clearly shows that the repressurization protocol can significantly reduce the measurement error in O_2 concentrations for a batch system with known gas composition. For a continuous system that cultivates a type I methanotroph (*M. buryatense* 5GB1), the significant difference between the feed and off-gas flow rates due to cell growth is demonstrated in Figure 6.3b. This result confirms the necessity to accurately measure the off-gas flow rate. In addition, the accuracy of the tracer protocol for a continuous system is validated through the total carbon balance. Details can be found in Stone *et al.* (2019).

6.3.2 Quantitative characterization of microalgae–methanotroph cocultures

One major challenge associated with characterizing mixed cultures is how to accurately determine the individual biomass concentration for each microorganism during the dynamic growth of the mixed culture. Existing characterization approaches can be categorized into molecular biological,

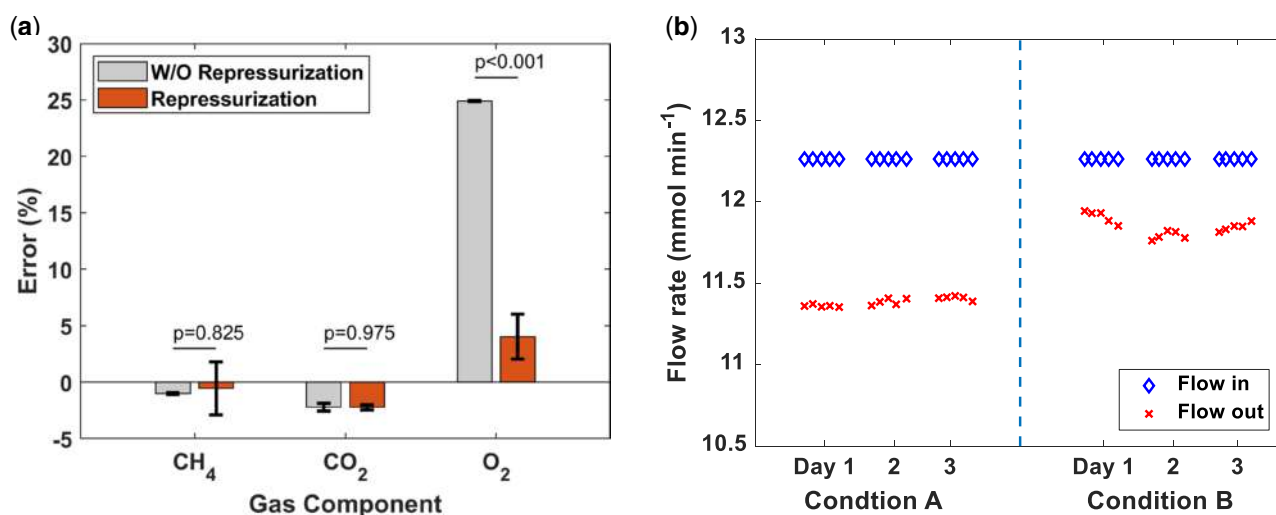


Figure 6.3 (a) Proposed experimental protocol significantly reduces measurement error in O₂ in a batch system; and (b) significant difference between the feed and off-gas flow rates due to cell growth in a continuous system.

biochemical, and microbiological methods (Sabra *et al.*, 2010; Spiegelman *et al.*, 2005). However, these methods require either expensive equipment, such as flow cytometry, community genome sequencing, or time-consuming and challenging techniques, such as ribonucleic acid (RNA)/deoxyribonucleic acid (DNA) extraction, isolation, or amplification. As a result, although these approaches deliver accurate offline characterization of a mixed culture, they are not suitable for frequent or real-time measurements desired for dynamic modeling of the mixed culture systems. As a result, among the published research on microalgae–methanotroph cocultures, only Hill *et al.* (2017) tracked the individual biomass concentration over time through cell counting using flow cytometry, whereas others only reported the total optical density (OD) of the coculture over time (Rasouli *et al.*, 2018; van der Ha *et al.*, 2012).

In addition to individual biomass concentration, the individual substrate consumption rates and product excretion rates of each organism over time are needed for the development of kinetic models. When there is cross-feeding in the coculture (i.e., any exchange of metabolite(s) between different organisms), the individual consumption/production rates cannot be measured directly. For the microalgae–methanotroph coculture, as shown in Figure 6.2, both O₂ and CO₂ are cross-feeding metabolites: O₂ is produced by microalgae while consumed by methanotrophs, whereas CO₂ is produced by methanotrophs and consumed by microalgae. The total consumption/production rates of O₂ and CO₂ by the coculture can be directly measured by GC, but accurate splitting of the total rates into two components is challenging. One could use labeled substrates, such as ¹³-CH₄, to determine the amount of cross-fed CO₂, but it requires expensive substrate and additional analysis, which is infeasible for real-time tracking.

To address these challenges and obtain real-time measurements for individual strains in the coculture, we have developed an E–C protocol to fully characterize the synthetic microalgae–methanotroph coculture that only requires commonly used analytical equipment. As shown in Figure 6.4, based on the measured total substrate consumption and production excretion rates, the E–C protocol computes the individual biomass concentration, individual substrate uptake rates, and product secretion rates based on the overall mass balance and growth stoichiometry of each organism. For microalgae–methanotroph cocultures, the substrate uptake rates include CH₄ and O₂ uptake rates for methanotrophs (can be computed or determined by Equations (6.1) and (6.2) in Figure 6.4, respectively), CO₂ uptake rate for microalgae (Equation (6.5)). The product secretion rates

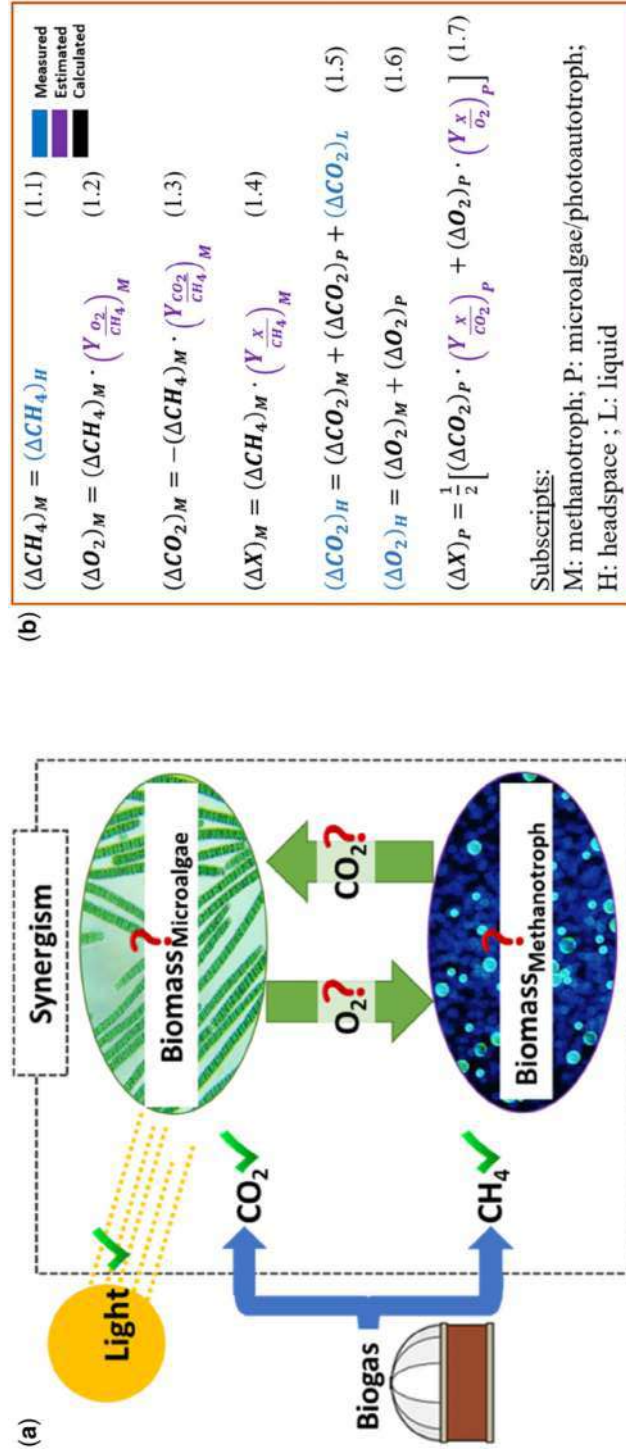


Figure 6.4 Illustration of the E-C protocol for microalgae-methanotroph coculture characterization: (a) schematic diagram of the relationships among known variables (with checkmarks) and unknown variables (with crosses); and (b) equations used to compute the unknown variables based on the known measurements (in blue font) and yield parameters (in purple font).

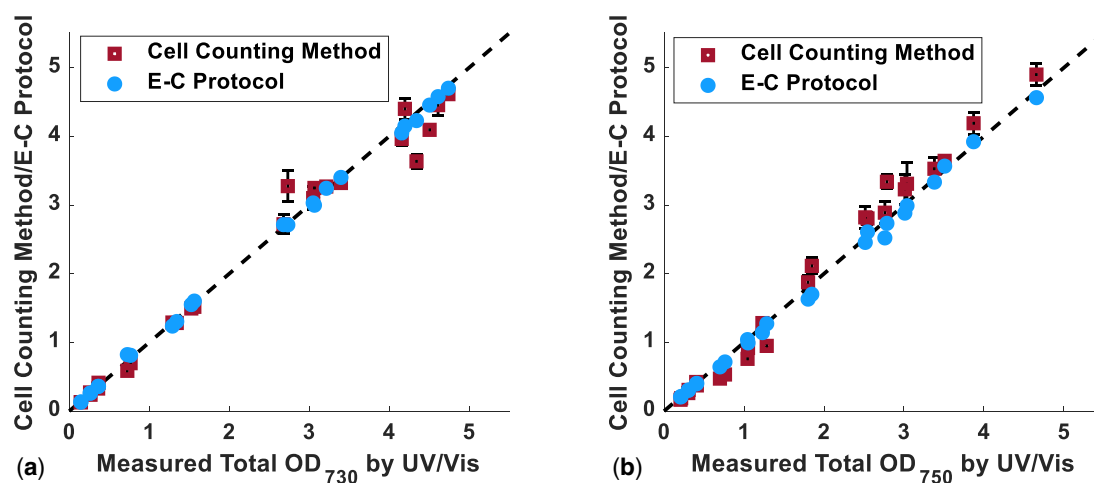


Figure 6.5 Comparison of the measured total OD vs. the total OD calculated using the individual biomass concentrations obtained through cell counting and the E–C protocol: (a) the salt water coculture pair *Synechococcus* sp. PCC7002–*M. alcaliphilum* 20ZR; and (b) the freshwater coculture pair *C. sorokiniana*–*M. capsulatus*. Points that lie closer to the diagonal dashed line have better accuracy than points that lie further away from the diagonal dashed line.

include CO_2 production rate by methanotrophs (Equation (6.3)), and O_2 production rate by microalgae (Equation (6.6)). Individual species growth rates include that of methanotrophs (Equation (6.4)) and microalgae (Equation (6.7)). More details can be found in [Badr *et al.* \(2021\)](#).

The E–C protocol was applied to characterize the growth of one cyanobacteria–methanotroph pair (*Synechococcus* sp. PCC7002–*Methylomicrobium alcaliphilum* 20ZR) and one microalgae–methanotroph pair (*C. sorokiniana*–*M. capsulatus*). The accuracy of the E–C protocol was validated by individual biomass concentrations measured through cell counting using flow cytometry. Moreover, we further showed that the E–C protocol provides better accuracy than the cell counting approach by comparing the predicted total OD from the individual biomass concentration by both approaches with the measured total OD and statistical testing. This is illustrated in [Figure 6.5](#). For both cases, the E–C protocol provides noticeably better estimates of total OD than those based on cell counting – points that lie closer to the diagonal dashed line have better accuracy than points that lie further away from the diagonal dashed line.

The E–C protocol only requires the commonly used analytical equipment, including GC, ultraviolet–visible spectrometry, and total carbon analyzer. It does not require any special sample preparation such as deoxyribonucleic acid (DNA) or ribonucleic acid (RNA) extraction or cell fixation. Therefore, it is suitable for real-time characterization of microalgae–methanotroph cocultures. In addition, it is shown that the E–C protocol is more accurate than cell counting such as flow cytometry. The significance of the E–C protocol is that it provides the real-time quantitative coculture characterizations that are required for the kinetic modeling of the coculture. We expect that the E–C protocol and its adaptations to other systems will become convenient and valuable tools for developing coculture or mixed culture-based biotechnologies.

6.4 SEMI-STRUCTURED KINETIC MODELING OF THE COCULTURE

A key enabler for the development of any biotechnology is a high-quality kinetic model that can predict the growth of biocatalysts under different conditions. Such a kinetic model can enable optimal design and scale up of the bioreactor. It also provides the foundation for model-based control of the bioreactor. For the microalgae–methanotroph coculture, there is a lack of effort in developing kinetic

models. This is largely due to the challenges in obtaining real-time characterization of microalgae–methanotroph cocultures and the lack of understanding of the inherently complex interactions in the coculture. For example, the time-series measurements of the coculture growth over time are prerequisite for the estimation of the kinetic parameters, which has not been available until recently. In addition, the *in-situ* exchange of O_2 and CO_2 between the microalgae and methanotroph, as well as additional unknown interspecies interactions make the kinetic modeling of the system highly challenging.

Enabled by the E–C protocol discussed in the previous section, we have published the very first kinetic model that captures the known cross-feeding mechanism within a microalgae–methanotroph coculture. Our kinetic model is a semi-structured model, as it explicitly models the exchange of *in-situ* produced O_2 and CO_2 between the two species. The other unknown interactions are captured in model parameters. For example, if the coculture promotes the growth of the methanotroph, we would expect the maximum methanotroph growth rate in the coculture model to have a greater value than that of the methanotroph in its single-culture model.

Figure 6.6 illustrates the flow chart and key equations for the semi-structured kinetic model. The model consists of four components: (1) microalgae growth; (2) methanotroph growth; (3) mass balance in the liquid phase; and (4) mass balance in the gas phase. Note that the growth of each organism in the coculture is coupled with the changes in the gas-phase composition through the mass balances in the liquid and gas phases. The equations that model each component are listed in Figure 6.6 as well, with the terms that coupling different components highlighted in dotted boxes. These terms include the O_2 produced by the microalgae, CO_2 produced by the methanotroph, and their contributions to the liquid phase mass balances.

Several sets of wet-lab experiments were conducted to test the accuracy of the semi-structured kinetic modeling using a cyanobacteria–methanotroph coculture (*A. platensis*–*M. buryatense* 5GB1). It was shown that the semi-structured kinetic model accurately predicted the individual growth

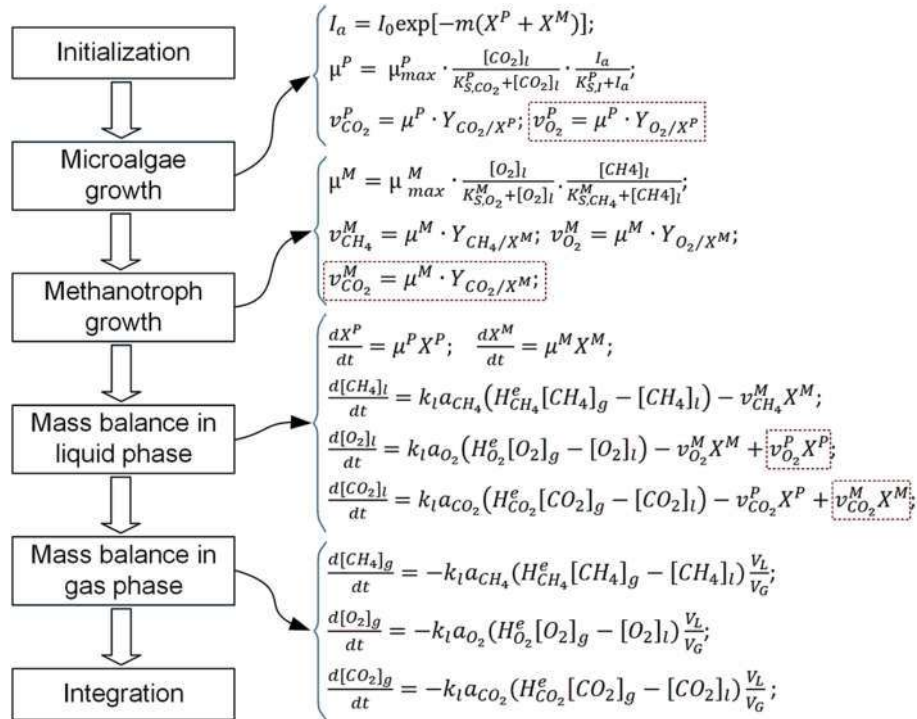


Figure 6.6 Flow chart of the semi-structured kinetic model and the associated model equations.

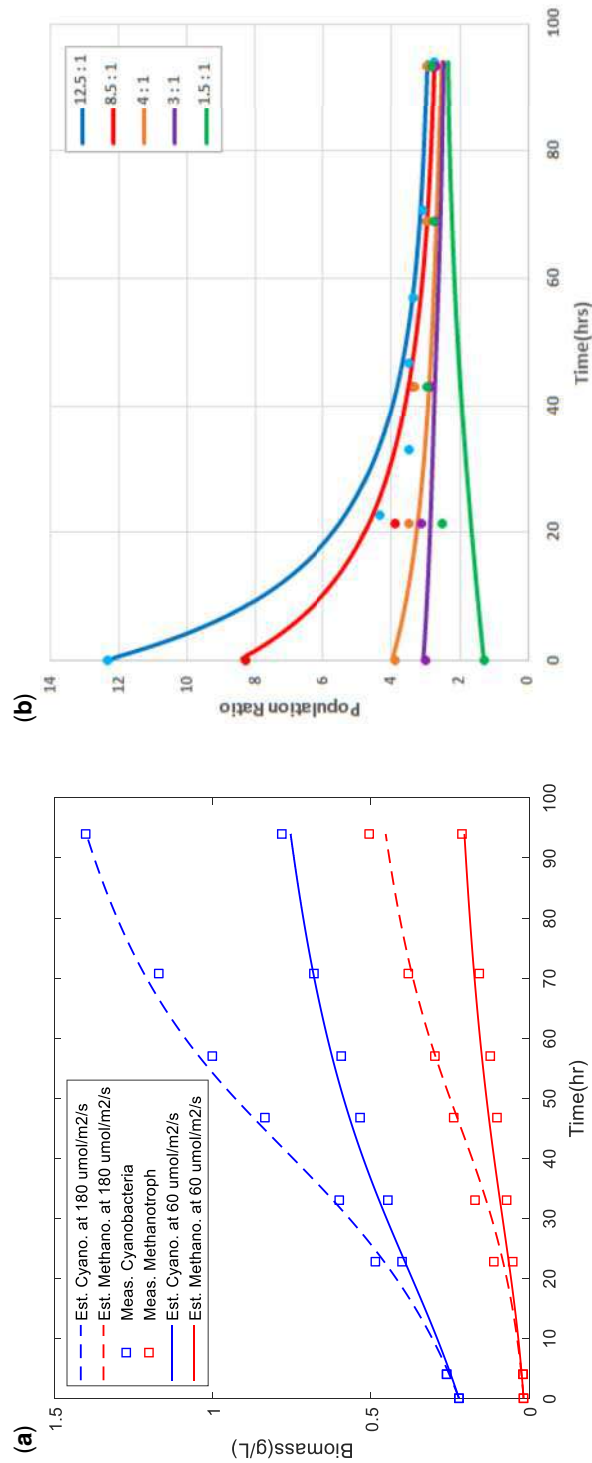


Figure 6.7 Experimental validation of the semi-structured kinetic model using a methanotroph–cyanobacteria coculture: (a) comparison of the biomass concentration in the coculture predicted by the model (lines and dashed lines) with measurements (squares) and (b) both model simulation (lines) and experimental measurements (dots) confirm that despite different inoculum ratios (cyanobacterium:methanotroph), after achieving steady state, the population ratio of the two strains converge to the same value.

rates, and the individual consumption/production rates of O_2 and CO_2 for the methanotroph and cyanobacteria in the coculture (Badr *et al.*, 2022). Specifically, experiments were conducted to examine the effect of several factors on the growth of the coculture, including light intensity, gas-phase composition, and inoculation ratio. The model predictions showed excellent agreement with the experimental data under all conditions examined, as illustrated in Figure 6.7.

With the details provided by the model that are often impossible to obtain through experiments, we revealed that for the cyanobacteria–methanotroph coculture cultivated on biogas without external O_2 supply, light limitation (due to self-shading effects) becomes the growth-limiting factor before mass transfer limitation (due to the small solubility of CH_4 and O_2). In addition, the semi-structured kinetic model captures the effect of other emergent metabolic interactions through the maximum growth rate. In fact, the maximum growth rates for the model coculture showed a 48 and 42% increase compared to their corresponding monocultures for cyanobacteria and methanotrophs, respectively. The significant increase in the maximum growth rates confirms the existence of the emergent metabolic interactions within the model coculture, although their identities are not yet known. More details can be found in Badr *et al.* (2022).

6.5 INTEGRATED NUTRIENT RECOVERY AND MITIGATION OF GREENHOUSE GAS EMISSIONS FROM WASTEWATER USING MICROALGAE–METHANOTROPH COCULTURES

As discussed in Section 6.1, due to many environmental benefits of AD and production of a valuable fuel (CH_4), it has been commonly deployed in WRRFs to convert organic wastes contained in wastewater into biogas, particularly in large-scale WRRFs. However, the installations of AD at mid- and small-scale WRRFs have been very limited. This is because the utilization of the AD-generated biogas has been limited to heating and electricity generation. The low value of heat/electricity is the main reason for poor ROI that prevents AD installations at mid- and small-scale WRRFs. Another drawback of AD is that its nutrient-rich liquid effluent is usually required to go through a nitrification–denitrification process prior to discharge, which is energy intensive and costly as discussed before.

The microalgae–bacteria or microalgae–fungi mixed culture has been studied for wastewater treatment and biogas upgrading (Kleerebezem & van Loosdrecht, 2007). In addition, inspired by how nature recycles biogas and nutrient, microalgae–methanotroph cocultures have been examined for integrated nutrient recovery and/or biogas valorization. For example, van der Ha *et al.* (2012) reported that a coculture of *Methylocystis parvus*–*Scenedesmus* sp. could completely convert a synthetic biogas (60% CH_4 , 40% CO_2) into microbial biomass without external O_2 supply. Hill *et al.* (2017) demonstrated that *M. alcaliphilum*–*Synechococcus* PCC 7002 could maintain stable growth on gas mixtures with a wide range of compositions, including raw biogas and synthetic biogas. Rasouli *et al.* (2018) demonstrated the application of using a microalgae–methanotroph (*C. sorokiniana*–*M. capsulatus*) coculture for nutrient recovery from a potato plant wastewater with synthetic biogas.

Most recently, through an ongoing collaboration with Columbus Water Works (CWW), our research (Roberts *et al.*, 2020) showed that *C. sorokiniana*–*M. capsulatus* can efficiently recover nutrients (N and P) in wastewater while converting biogas into microbial biomass. Located in Columbus (GA, USA), CWW is the fourth largest WRRF in Georgia and has been a leader among WRRFs in technology innovation. In 2012, CWW implemented and demonstrated a new technology that integrates green power and class-A biosolids production with wastewater treatment. The technology, known as the Columbus Biosolids Flow-Through Thermophilic Treatment and Cogeneration System, was the first thermophilic AD process in the USA that runs entirely off the heat derived from the AD-biogas-fueled power generation. Using *C. sorokiniana*–*M. capsulatus* (Bath) as the model coculture, we demonstrated that the coculture achieved 100% recovery of NH_3 (80% recovery of total nitrogen (TN)) and 100% recovery of orthophosphate (98% recovery of total phosphorous) when biogas supply is unlimited. In addition, the coculture achieved 100% CH_4 and CO_2 conversion into microbial biomass

when nutrient supply is unlimited. Also, biomass production, TN recovery, and total phosphorus (TP) recovery performance of the microalgae–methanotroph coculture were significantly better than those of the microalgae monoculture, achieving 120, 71, and 164% improvements, respectively, when the same amount of biogas was used.

The practical implementation of the microalgae–methanotroph coculture-based W2V technology requires consideration of several critical factors. In this section, we examine four major issues that could affect the performance of the microalgae–methanotroph coculture for nutrient recovery and biogas valorization, along with our preliminary results on how to address them. These factors include the selection of the biocatalyst, the tolerance of the coculture to contaminants in raw biogas, the freshwater consumption rate, and the pretreatment of AD effluent. Furthermore, in order to attract more attention to coculture or mixed culture W2V technology from researchers in the W2V field, and to achieve wide acceptance of the platform in the field, we need to provide convincing and experimentally verifiable results to demonstrate advantages of cocultures over single cultures or sequential single cultures.

6.5.1 Choice of a suitable biocatalyst

One key factor that has significant impact on the performance of the coculture W2V platform is the choice of biocatalyst. The ideal candidates should tolerate various inhibitors in the AD digestate and raw biogas, while delivering robust and stable growth under various disturbances, such as light intensity, light–dark cycle, and availability of macro- and micro-nutrients (e.g., N and P).

Our research has shown that not all microalgae–methanotroph pairs form a synergistic partnership. This is true even if their preferred growth condition matches, which include pH and salinity. In addition, different AD digestates contain different inhibitors and stressors depending on the organic waste fed to the AD, and the AD types and operations. Therefore, to optimize the performance of the coculture system in terms of carbon and nutrient recovery, it is important to screen different microalgae and methanotrophic strains to identify the best candidates.

The screening of different coculture pairs is a time- and labor-intensive process. Given the number of microalgae and methanotrophic species n_1 and n_2 , there are $n_1 n_2$ potential coculture candidates to be screened. To speed up the screening process, we have developed a specialized equipment, termed species screening station (S3), as shown in Figure 6.8. S3 consists of nine parallel-fed batch reactors that control temperature, pH, agitation rate, light intensity, gas composition, and feed rate. It is important to note that abiotic tests are necessary to verify that the equipment provides consistent or uniform growth conditions among different reactors, with the only difference being the biocatalysts in each reactor. Besides abiotic tests, biotic tests should be conducted by cultivating the same species in



Figure 6.8 Photograph of S3 that can run nine parallel-fed batch bioreactors simultaneously.

all nine reactors under the same conditions. The growth rates measured from different reactors should be comparable to confirm they are indeed nine replicates.

Another important issue to consider for strain screening is to provide a consistent adaptation process for all strains to be examined. Because of different inhibitors in the AD digestate, some strains may not adapt as fast as others. Without any adaptation, the fast-adaptive strains may exhibit better growth performance on the wastewater in short terms, whereas others that exhibit better growth performance in longer terms may appear to perform worse during the screening. To avoid missing out the slower adapters, we have developed a standard protocol to provide a consistent adaptation period as described below. Before the screening, each strain will first be introduced to a transition medium (5% AD digestate, 45% clarifier water, both from CWW, plus 50% defined medium). The strains will be harvested during their mid-exponential growth phase on the transition medium and used as inoculation for screening in the S3. With the added adaptation period, we can ensure that slower adapters would have sufficient time to adjust to the AD digestate. The strategy will reduce the risk of missing out any promising strains. Our results show that the S3 can greatly expedite the screening process. For example, we screened 12 monocultures and six cocultures in triplicate – a total of 54 individual runs in less than 3 months (Murphy *et al.*, 2022).

After the screening, we chose *C. sorokiniana*–*M. capsulatus* (Bath) as the model coculture pair for the following reasons: (1) the pair showed one of the best growth performances on raw biogas and diluted digestate provided by CWW; (2) both *M. capsulatus* and *C. sorokiniana* have long served as model organisms to understand CH_4 oxidation and phototrophy, respectively. And both strains have complete and expert-annotated genomic information, (draft) genome-scale metabolic models and genetic tools; (3) *C. sorokiniana* has been extensively studied for wastewater treatment, particularly for digestate treatment; and (4) *M. capsulatus* (Bath) is the only industrial methanotroph strain for commercial applications due to its robustness and stability. For example, it has been commercialized for the production of SCP as animal feed (e.g., FeedKind® from Calysta, Menlo Park, CA).

6.5.2 Coculture tolerance to contaminants in raw biogas

The major contaminants in raw biogas are NH_3 and H_2S . Although many microalgae–methanotroph coculture pairs (including the model coculture pair *C. sorokiniana*–*M. capsulatus*) prefer NH_3 as the nitrogen source, a high concentration of H_2S may inhibit the coculture growth. To confirm the tolerance of the model pair to H_2S in raw biogas, we have tested the pair by adding different concentrations of H_2S (1,000–5,000 ppm) to the synthetic feeding gas (70% CH_4 and 30% CO_2), as the raw biogas from CWW only contains ~1,000 ppm H_2S . The experiments were conducted in serum bottles and lasted for 8 days. Figure 6.9 shows the coculture OD over time, which suggests that 1,000 and 2,000 ppm H_2S had little inhibition to the coculture; whereas 3,000 ppm H_2S lengthened the lag phase, the growth rate was similar to that of 1,000 and 2,000 ppm after the lag phase. In addition, the

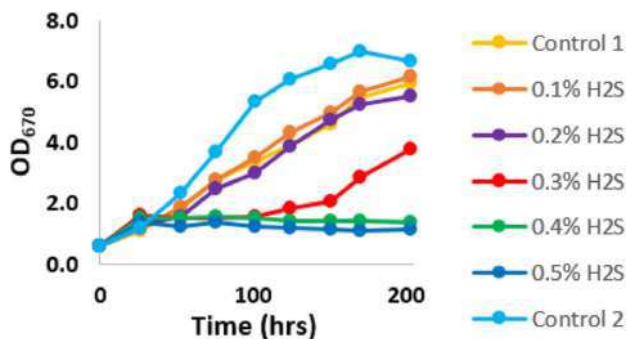


Figure 6.9 H_2S tolerance test shows healthy coculture growth up to 3,000 ppm (i.e., 0.3% H_2S).

measured individual biomass concentrations confirmed the steady growth of both strains. This result confirms that the model pair can tolerate biogas with 3,000 ppm H_2S , which is the upper limit of H_2S concentrations in most AD-generated biogases.

If the biogas feed contains higher H_2S concentrations that the coculture pair cannot tolerate, one could either add a pre-cleaning column to use chemical solutions (such as an alkaline solution) to remove part of the H_2S , or introduce a compatible sulfide-oxidizing bacterium to the coculture to achieve simultaneous H_2S removal. The coculture of sulfide-oxidizing bacteria with microalgae has been examined for biogas upgrading, which can effectively remove CO_2 , NH_3 , and H_2S from raw biogas, so that the upgraded biogas can be injected into existing natural gas pipelines (Muñoz *et al.*, 2015).

6.5.3 Freshwater consumption required by wastewater treatment

AD effluent often contains various inhibitors, including volatile fatty acids and antibiotics that may severely inhibit the growth of both microalgae and methanotrophs in the coculture. For microalgae-based wastewater treatment, the digestate is usually diluted 10 or 20 times to achieve sustained growth of microalgae and enable sufficient nutrient recovery rates (Wang *et al.*, 2018; Wen *et al.*, 2017; Xia & Murphy, 2016). However, using fresh water to dilute AD effluent is not practical because fresh water is a limited resource in most locations.

To eliminate or reduce freshwater usage, we have examined the possibility of using secondary clarifier water (the treated water at a WRRF before the final cleanup) as a diluent. In Roberts *et al.* (2020), we compared three diluents to examine their effects on coculture growth. The three diluents were: (1) tap water, (2) secondary clarifier water, and (3) a modified ammonium mineral salt medium (modified AMS), which is the standard AMS medium (Whittenbury *et al.*, 1970) without inorganic nitrogen ($\text{NH}_4^+\text{-N}$) and inorganic phosphorus ($\text{PO}_4^{3-}\text{-P}$).

The effect of different diluents on coculture growth was evaluated by biomass production, biogas utilization, and nutrient recovery. As reported in Roberts *et al.* (2020), three diluents had negligible effects on coculture growth and biomass productivity. A comparison of the coculture biomass growth and nutrient recovery of the coculture on AD effluent diluted with different diluents is presented in Figure 6.10. Clearly, the coculture performed similarly on the three diluents. This indicates that the secondary clarifier water can be used to replace fresh water to dilute AD effluent, thereby avoiding

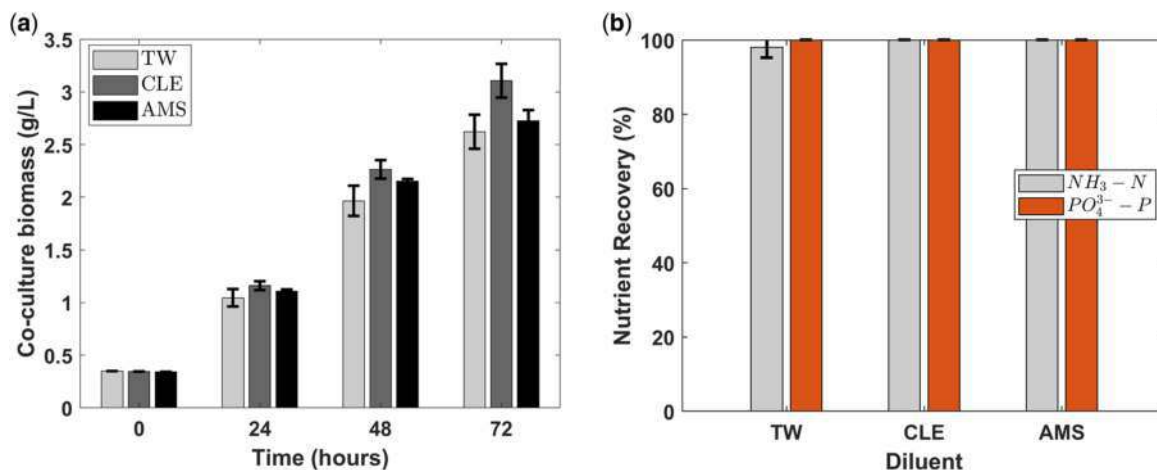


Figure 6.10 Comparison of (a) biomass growth and (b) nutrient recovery of coculture on AD effluent diluted with tap water (TW), secondary clarifier water (CLE), and modified AMS (AMS).

the need of fresh water for the proposed technology. In addition, the model coculture demonstrated effective nutrient recovery with nearly 100% recovery of ammonia nitrogen and orthophosphates.

6.5.4 Pretreatment of AD effluent

Besides macro- and micro-nutrients, AD effluent also contains solid contents, which require pretreatment before being fed to the coculture. Suspended solids often contain native microbial communities (e.g., methanogens) in the AD effluent, which could potentially compete with or cause contamination of the microalgae–methanotroph coculture. Liquid medium sterilization can completely eliminate potential competition or contamination. However, the process would incur significant cost, making the coculture-based W2V technology impractical for wastewater treatment from a financial perspective. To ensure the economic viability of the technology for wastewater treatment, it is necessary to lower or minimize the AD effluent pretreatment requirement. This is in fact possible because the native microbial communities in AD effluent thrive under anaerobic conditions. As a result, we argue that they do not compete well with the microalgae–methanotroph coculture due to the aerobic conditions in the bioreactor. This is supported by a study in which it was found that the fast-growing methanotrophs in AD effluent were significantly enriched and became the dominant species after prolonged cultivation on CH_4 and O_2 (Kim *et al.*, 2018). Specifically, in the enriched culture, *Methylosarcina fibrata* accounts for 94.1% of the methanotroph population, and 53.8% of the total microbial population. This result agrees with another recent study by Perera *et al.* (2022) where the synergistic interactions in a defined microalgae–bacteria consortium were not perturbed by the native heterotrophic bacteria in the wastewater, and a community shift occurred, which balanced the interactions and resulted in enhanced wastewater treatment (Perera *et al.*, 2022).

To further validate our argument and demonstrate the feasibility of using cocultures for wastewater treatment, we have conducted experiments to examine the effects of different AD effluent pretreatment methods on coculture growth. Three different pretreatment methods of AD effluent were examined: settling, filtering, and autoclaving. In addition, we tested coculture growth on a sterilized AMS medium, which served as the control. As reported in Roberts *et al.* (2020), there were no statistically significant differences among different pretreatment methods. In other words, the coculture grew similarly on the three differently treated AD effluents. Figure 6.11 shows the biomass productivity of individual strains in the coculture on differently pretreated AD effluents. This result further confirms the robustness of the model coculture, and that minimum treatment by settling (hence incurring minimum cost) would suffice the AD effluent pretreatment requirement.

6.5.5 Advantage of the coculture over sequential single cultures in carbon and nutrient recovery

There are many studies that have demonstrated advantages of mixed cultures over single cultures in a variety of applications. However, for the application of both microalgae and methanotrophs to wastewater treatment, it has been suggested that sequentially cultivating microalgae and methanotrophs is an alternative to their coculture and may achieve similar performance as the coculture (personal communication). We believe that this is highly unlikely due to the missing synergism when cultivated sequentially. Nevertheless, there was no experimental study to support one or the other. Therefore, we designed experiments to compare the two options. Specifically, to examine whether the coculture exhibits any advantages over the sequential single cultures, we conducted experiments to compare the coculture growth on biogas and diluted AD effluent, with the growth of sequential single culture, that is, *C. sorokiniana* followed by *M. capsulatus*. To achieve biogas conversion without external O_2 supply by sequential single cultures, we first cultivate microalgae on diluted AD using raw biogas as the carbon substrate, which fixes CO_2 and produces O_2 , then the spent gas is fed to the methanotrophs which convert CH_4 and CO_2 into microbial biomass. As nutrient limitation would limit the growth of both strains in the coculture, in these experiments 20 ml of undiluted AD effluent was added to the vessel 48 h after the inoculation to ensure there is unlimited supply of macronutrients.

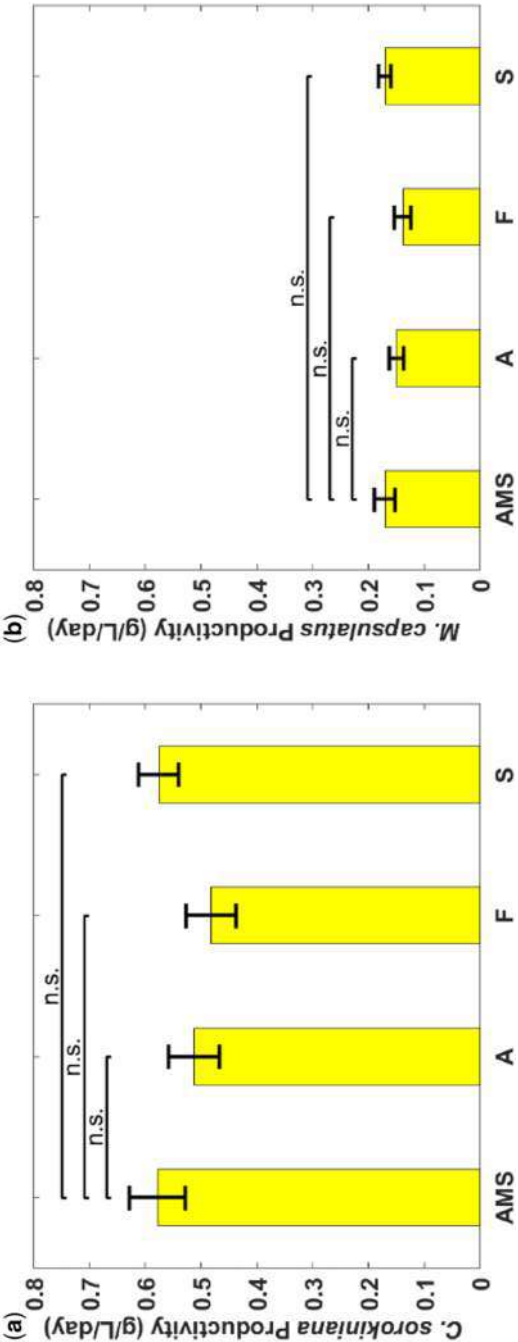


Figure 6.11 Biomass productivity of individual coculture strains on differently pretreated AD effluent diluted by CLE: (A) *C. sorokiniana* and (B) *M. capsulatus*. The Dunnett’s test, with AMS as the control, indicates that these differences are not statistically significant (n.s.) at the significance level $\alpha = 0.05$.

Our study (Roberts *et al.*, 2020) shows that both microalgae (*C. sorokiniana*) and methanotroph (*M. capsulatus*) in the coculture exhibited significantly improved growth compared to the sequential single cultures. With the same amount of biogas supply, the biomass production of *C. sorokiniana* and *M. capsulatus* in the coculture showed a 64 and 58% increase compared to the sequential single-culture counterparts. We believe this significantly improved biomass production can be attributed to the metabolic coupling of CH₄ oxidation and oxygenation through photosynthesis. Enabled by the metabolic coupling, *C. sorokiniana* in the coculture showed a significantly higher CO₂ fixation rate and O₂ evolution rate than the microalgae single culture; and *M. capsulatus* in the coculture showed a much higher CH₄ assimilation rate and CO₂ production rate than the methanotroph single culture. In other words, there was no negative impact such as inhibition on *C. sorokiniana* growth due to the extra CO₂ produced by *M. capsulatus*. In addition, the results showed that CH₄ had no effect on microalgae. This is likely due to the very limited solubility of CH₄ in the liquid medium. More details can be found in Roberts *et al.* (2020).

To determine whether the model coculture offers an improvement in nutrient recovery compared to the sequential single cultures, we conducted experiments similar to the growth experiment described above. The only difference was that no additional nutrients were added after 48 h. For all cultures, TN, NH₄⁺-N, TP, and PO₄³⁻-P were measured to assess the nutrient recovery by different cultures. These experiments confirmed that the coculture offered significantly enhanced nutrient recovery than the single cultures, which was mainly due to enhanced biomass production as the correlation between the biomass production and nutrient recovery for the coculture was almost the same as that of the single cultures (Roberts *et al.*, 2020).

6.6 NEXT-GENERATION PHOTOBIOREACTORS

To achieve broad adoption of the microalgae–methanotroph coculture-based W2V technology in real applications, we must address the relevant engineering challenges associated with microbial cell cultivation. Traditional suspended or planktonic MC has two critical bottlenecks that limit its commercial application (Georgakopoulou, 2019): (1) low footprint areal biomass productivity and (2) high-cost biomass recovery. These bottlenecks are caused by the light attenuation in culture broth and mass transfer resistance of gaseous substrate into the liquid broth, both of which severely limit the achievable cell density in the liquid medium and the scale-up of the biotechnology. The low cell density further results in a large footprint and high energy cost for biomass harvesting which drastically diminishes the economic feasibility. These challenges are the main reasons for the limited commercialization of microalgae-based waste-to-fuel technologies (Cheah *et al.*, 2016). The same challenges apply to microalgae–methanotroph cocultures, with additional challenge of low solubility of CH₄ and O₂ in aqueous solutions. Therefore, novel photobioreactors are needed for the microalgae–methanotroph coculture-based W2V technology.

In the past decade, biofilm-based MC has drawn increasing attention (Gross & Wen, 2014; Gross *et al.*, 2013, 2015). By cultivating microalgae in biofilms on a supporting substratum, biofilm-based cultivation offers many advantages: first, biomass harvest is made easy and energy-efficient – biomass can be simply scraped off the substratum. Second, biofilm can be cultivated on a moving belt to be in contact with the gas and liquid phases alternately, which offers additional benefits of reduced mass transfer resistance of gaseous substrate reaching microalgae cells. Exploring the idea of biofilm-based cultivation, we have developed a patented circulating coculture biofilm photobioreactor (CCBP) for the microalgae–methanotroph coculture (He *et al.*, 2022). The schematic diagram of the CCBP is shown in Figure 6.12, in which a conveyor belt (substratum) is stretched around shafts to form a zig-zag configuration that supports microalgae–methanotroph biofilm growth on it. The lower part of the configuration is submerged in the diluted AD effluent, while the upper part is exposed directly to biogas and sunlight. The movement of the conveyor belt is driven by connecting one shaft with a motor, enabling the attached biofilm to alternately access nutrients in the liquid phase and carbon

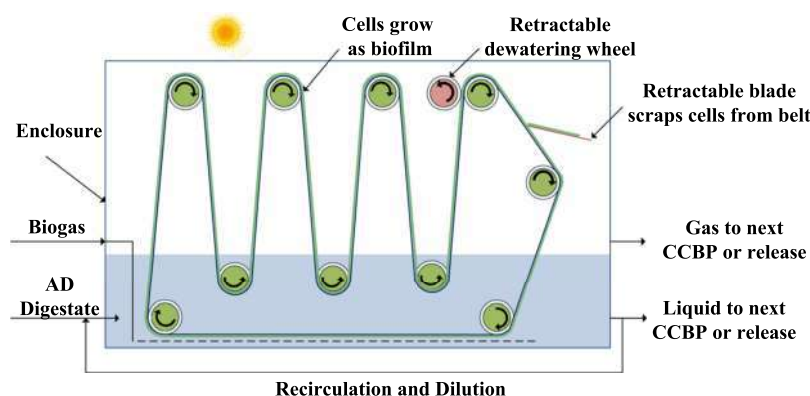


Figure 6.12 Schematic diagram of a patented CCBP (He *et al.*, 2022).

substrate (CH_4 and CO_2) in the gas phase. With the biocatalysts (coculture biofilm) directly exposed to the gas phase, and significantly increased surface area available for sunlight, the CCBP has achieved an aerial biomass productivity of 144.6 g Dry Cell Weight/ m^2/day .

6.7 OUTLOOK AND CONCLUSION

Wastewater, if not properly treated, causes acute economic and environmental problems. In addition, wastewater is the fifth largest anthropogenic source of CH_4 emission globally. Therefore, mitigation of CH_4 is an imperative task in wastewater treatment. AD converts wet organic wastes into biogas and inorganic nutrients in a controlled and contained fashion, thereby offering many environmental benefits while producing a valuable fuel (CH_4). However, the poor ROI, caused by the low value of biogas (due to contaminants) and nutrient-rich AD effluent that requires further treatment, limits the application of AD to large-scale WRRFs. To broaden the adoption of AD, it is necessary to convert biogas into high-value products and reduce the cost of nutrient recovery.

Recently, microalgae–bacterial cultures have drawn increasing research interest for their application in integrated biogas upgrading and wastewater treatment. In particular, microalgae–methanotroph cocultures have been demonstrated to provide a highly promising platform for integrated nutrient recovery from wastewater and biogas valorization. In this chapter, we showed that microalgae–methanotroph cocultures have the potential to play an important role in reducing greenhouse gas emissions and energy consumption from wastewater treatment processes and producing value-added products (e.g., animal feed or aquafeed). Existing research, including ours, has established the scientific foundations for using microalgae–methanotroph cocultures for practical applications. These foundations include demonstrated robustness in tolerating the contaminants/inhibitors in raw biogas, confirmed feasibility of minimum pretreatment of the AD effluent, demonstrated operation with treated wastewater instead of fresh water, and quantified high-protein content suitable for animal feed or bioplastics production.

To fully explore the significant potential of microalgae–methanotroph cocultures for biogas valorization and nutrient recovery, there are many research questions that remain to be answered. Currently, all existing research on microalgae–methanotroph cocultures utilizes wild strains, and the product is microbial biomass. This is due to the lack of understanding on the biological foundation for the synergistic effect within the coculture. The molecular mechanism of the inter-species interactions is still largely unknown. Understanding the molecular mechanism for the synergistic interactions between the microalgae and methanotroph will pave the road for many more applications, such as engineering the coculture for the production of desired biochemicals, instead of just producing biomass. Another unexplored area is to construct mixed cultures with other species, such as sulfide-oxidizing

bacteria, to handle challenging situations such as extremely high concentrations of H_2S or other contaminants. Such research will require the development of new analytical procedures to monitor the mixed culture in real time. Last but not least is the development of photobioreactors. Currently, there is little understanding on the growth of the microalgae and methanotrophs in biofilms. There has not been any research that examines the mass transfer of gaseous substrates and macronutrients in the biofilm, and whether the enhanced growth observed in the biofilm-based cultivation of the coculture is caused only by the increased surface area or if the biofilm (i.e., extracellular matrix) contributed in any way. There is a vast unknown space for scientists and engineers to explore.

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Part 3

Integration with Other Technologies

Chapter 7

Microalgae cultivation in bio-electrochemical systems

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ABSTRACT

Biofuels from algae have the potential to completely replace fossil-based fuels and provide energy security for the future. However, the cost of algae biofuels is still too high for commercial application. In this context, producing algae and electrical energy using photosynthetic microbial fuel cells (PMFCs) is an attractive option. PMFCs utilize the natural process of photosynthesis for algae generation or algae degradation at the anode. In the former system, the process of organic matter degradation complements the process of algae biomass production with concomitant power generation. Electrogenic bacteria oxidize organic matter at the anode anaerobically. The anode transfers the electrons released through oxidation to the cathode, where photosynthetic organisms produce oxygen (O₂) as a cathodic electron acceptor. The suitability of bio-electrochemical systems such as microbial fuel cells for algae cultivation can be assessed by comparing them with the conventional method of cultivation, namely open ponds and photobioreactors. PMFCs offer a process that can provide high carbon dioxide concentrations for algal growth, has a mechanism to prevent high inhibitory O₂ concentrations and can meet a fraction of the process electricity requirements. The algae biomass can go as high as 4–5 g/L in a PMFC and power output doubles due to activity of algae at the cathode compartment. This chapter discusses the algal growth in bio-electrochemical systems, the factors that influence them and directions for future research.

Keywords: bioelectricity, biofuel, carbon dioxide, dissolved oxygen, electrogenic bacteria, energy, microalgae, photobioreactor, PMFC, wastewater.

7.1 INTRODUCTION

The International Energy Agency (IEA) reported that biofuels have a high potential to meet a substantial fraction of the world's energy demands. As per a 2015 report, biofuel constitutes 10% of the world's cumulative energy source (IEA, 2015). Third-generation biofuels derived from algae are a prominent part of bioenergy initiatives. However, several bottlenecks hinder bioenergy generation from algae. This includes algae cultivation, harvesting, oil extraction and algae-to-fuel conversion efficiency (Reddy *et al.*, 2019). The processes that bring significant enhancement of efficiency in

any of these factors enable algae technology for bioenergy production. A sustainable technology for algae cultivation is the need of the hour. The definition of sustainability involves environmental, economic and social aspects. Algae cultivation is an environmental friendly process and an attractive greenhouse effect mitigation method.

Photosynthetic microbial fuel cells (PMFCs) offer a process that can provide high carbon dioxide (CO_2) concentrations for algal growth, has a mechanism to prevent high inhibitory oxygen (O_2) concentrations and can meet a fraction of the process electricity requirements. PMFCs are a modified form of microbial fuel cells (MFCs) and considered promising systems for wastewater treatment and power generation. In such systems, microorganisms convert organic matter into electricity (Khandelwal *et al.*, 2018). A conventional two-chamber MFC design contains an anodic chamber and a cathodic chamber separated by a proton exchange membrane (PEM) or an ion-selective membrane (Logan *et al.*, 2006). Microorganisms oxidize the organic matter in the anodic chamber to produce electrons, protons and CO_2 . The electrons are transferred onto the anode surface, from where they flow across an external circuit to the cathode, constituting the flow of current. The protons migrate to the cathode through the PEM. At the cathode, molecular O_2 reduction to water depletes electrons and protons. In PMFCs, the CO_2 produced at the anode is fed to the cathode for fixation by algae, which in turn produce O_2 as an electron acceptor to complete the MFC circuit (Wang *et al.*, 2010).

Wang *et al.* (2010) reported proof of using PMFCs for algae cultivation at the cathode with simultaneous anodic CO_2 fixation and generation of O_2 as an electron acceptor. The power output from the device was 5.6 W/m^3 , with algal growth in the cathode chamber linked directly to power generation. The cathode dissolved oxygen (DO) concentration was found to be 6.6 mg/L . The power generation from the MFC depends on the O_2 concentration at the cathode and a concentration level of 6.6 mg/L is considered suitable (Kang *et al.*, 2003). The device was demonstrated to be an efficient carbon capture device for algae biomass production. However, they used glucose as a carbon source at the anode. Pandit *et al.* (2012) used a similar technology with wastewater to generate cyanobacterial biomass at the cathode. A number of reports then published on either wastewater treatment using these devices (Zhang *et al.*, 2011) or comparing the efficiency of these devices with mechanically aerated devices (Juang *et al.*, 2012).

Schamphelaire and Verstraete (2009) reported an integrated system consisting of an algal biomass production unit, an anaerobic digester to convert algae biomass to biogas and an MFC to generate electricity while treating digester effluent. They estimated their system results in a $\sim 9 \text{ kW/ha}$ capacity power plant, with 23 kW/h prospects. Khandelwal *et al.* (2018) showed that the algae biomass degradation at the anode was coupled with algae biomass production at the cathode. The algae biomass harvested from the cathode was used for oil extraction and fed back into the system at the anode for degradation by a microbial consortium. The replacement of costly fuels such as glucose/acetate at the anode with algae biomass and capture of O_2 generated by the photosynthesis process via cathodic reduction to water was achieved, as high O_2 concentrations are inhibitory to the growth of algae, particularly in closed systems.

Considering the advantages of using PMFCs for algae cultivation, it is essential to look at the process with respect to some key points such as the feasibility of the process in dual-chambered systems/open pond reactors, the highest algae productivity obtainable in such systems, highest power/energy output, scalability and sustainability of the process. The points mentioned above require discussion on two different aspects of MFC operation: the first one being electrochemical and design parameters such as MFC configuration, electrodes, separators, electrode catalysts and power management systems (Figure 7.1). The second one is biological parameters such as the microbial community at the anode, choice of algae species at the cathode, interaction of algae cells with the cathode and optimal conditions for microbial growth. This chapter discusses algal growth with respect to these aspects. In addition, a brief comparison of algal growth in bio-electrochemical systems and conventional systems is presented.

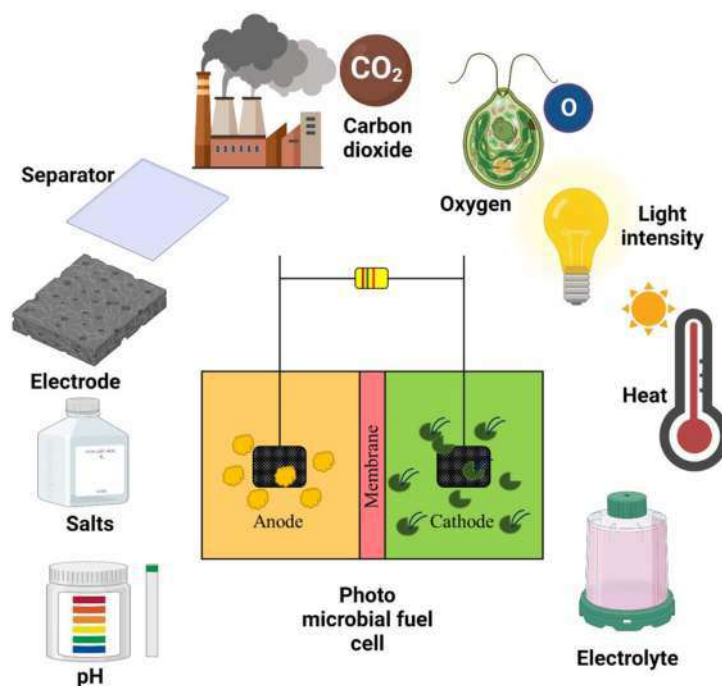


Figure 7.1 Overview of the factors affecting the efficiency of PMFCs.

7.2 USE OF ALGAE IN MFCS

7.2.1 Algae as primary producers

Microalgae are one of the primitive organisms on Earth that carry out oxygenic photosynthesis (Hopkins, 1999). The water splits with the help of light energy at the photosynthetic reaction center housed in the eukaryote's chloroplast and the cyanobacterial cell membrane. The splitting of water is accompanied by electron flow, adenosine triphosphate (ATP) and nicotinamide adenine dinucleotide phosphate hydrogen (NADPH) generation. The ATP and NADPH then reduce CO_2 through dark reactions (Hopkins, 1999). Algae are at the bottom of the aquatic food chain and are called autotrophs, which means they get their food from themselves via CO_2 fixation. Cyanobacteria, also known as blue-green algae (BGA), are considered living fossils of the planet. The algae class Euglenophyceae is thought to be the oldest lineage of algae that includes zooflagellates.

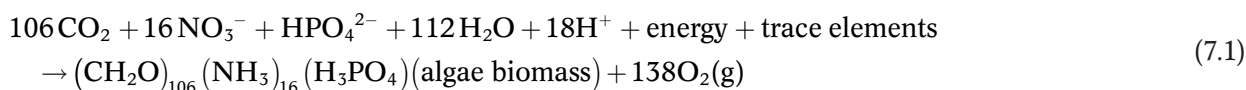
Algae, unlike other photosynthetic organisms, can grow in barren lands flooded with wastewater, exhibit much higher biomass productivity and effectively purify wastewater of nitrogen, phosphorus and chemical oxygen demand (COD). Microalgae (microscopic algae) are not only used as a source of bioenergy (Chapter 9) but also as food and dietary supplements (Chapter 10). Because of these tiny photosynthetic microorganisms known as phytoplankton, the ocean is a massive ecosystem that acts as a reservoir of fixed atmospheric CO_2 .

Only a small group of algal genera have been explored to date. About 30,000 species have been studied and classified (Enzing *et al.*, 2014). Algae are a diverse group of organisms that range from single-celled to filamentous, solitary to colonial, micro to macro and prokaryotes to eukaryotes, with shorter life cycles and the ability to survive extreme conditions. They are divided into distinct classes based on pigment composition, cell membrane (single-, bi- or multi-layered), mode of reproduction and food reserve. There are a total of 11 classes of algae classified by Fritsch (1945), namely Chlorophyceae, Xanthophyceae, Chrysophyceae, Bacillariophyceae, Cryptophyceae, Dinophyceae, Chloromonadineae, Euglenineae, Pheophyceae, Rhodophyceae and Myxophyceae (Cyanophyceae).

The most recent study by [Gololobova and Belyakova \(2022\)](#) shows how chloroplasts were crucial to the evolution of algae. They divided the algae into five monophyletic supergroups named mainly Archaeplastida (Glaucocystophyta, Rhodophyta, Prasinodermophyta, Chlorophyta and Charophyta), TSAR defined as telonemids, stramenopiles, alveolates and rhizaria (Ochrophyta, Dinophyta, Chlorarachniophyta and photosynthetic species of the genera *Chromera*, *Vetrella* and *Paulinella*), Haptista (Prymnesiophyta and Rappemonads), Cryptista (Cryptophyta) and Discoba (Euglenophyta).

7.2.2 Algae metabolism

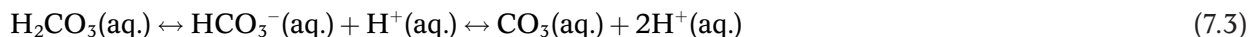
Algae can grow in various cultural media and environmental conditions ([Sharma et al., 2011](#)). Physicochemical factors such as sunlight, CO₂ concentration, pH, temperature, salinity and stress influence algal growth and biomass production ([Mata et al., 2010](#)). The photoperiod shows ascendancy by observing the growth of the algae. The biomass increases with an increase in photoperiod from 6 to 12 h ([Ip et al., 1982](#)). The following reaction represents the photosynthesis process of algae ([Wang et al., 2017](#)):



Temperature, light, dissolved CO₂, DO and pH affect each other, and their interplay determines the algal growth. Biomass tends to increase with a rise in temperature. The optimal temperature for algal growth is between 20 and 30°C ([Ip et al., 1982](#)). The temperature also influences the solubility of CO₂/O₂ and the mass transfer rate in the culture medium ([Vale et al., 2020](#)). Depending on the algal species, growth is affected by the concentration of CO₂. High levels of DO are detrimental to microalgae due to the formation of reactive oxygen species, which causes oxidative stress. As far as CO₂ is concerned, gaseous CO₂ has limited solubility in water and is not readily assimilated by algae. CO₂ first dissolves in the solution as carbonic acid (Equation (7.2)), which eventually dissociates to bicarbonate or carbonate (Equation (7.3)), depending on the pH. Carbonic acid formation is accelerated by carbonic anhydrase, an enzyme all algal species produce ([Shukla & Kumar, 2018](#)). Microalgae supplied with bicarbonates grow better than the ones provided with gaseous CO₂. The pH of the medium is another critical parameter that affects algal growth as it determines the predominant carbon species and nutrient bioavailability. An alkaline pH favors CO₂ dissolution in water and the formation of bicarbonates. The reaction between water and CO₂ dissolved in water can be represented by the below equation ([Hopkins, 1999](#)):



Depending on the pH of the solution, carbonic acid can dissociate to produce HCO₃⁻ and carbonate (CO₃²⁻). The below equation depicts the equilibrium between the three species ([Hopkins, 1999](#)):



At high pH values, the reaction proceeds in favor of carbonate formation. Carbonate is the predominant species in solution under highly alkaline conditions. Bicarbonate is the most common species when the pH is close to neutral. Microalgae prefer near neutral pH for optimal growth and readily assimilate bicarbonates from solution. The utilization of bicarbonates regenerates alkaline water that needs to be controlled by supplying CO₂.

An adequate concentration of CO₂ is mandatory for photosynthesis ([Creswell, 2010](#)). Algae exhibit low production rates at ambient CO₂ concentration (0.035%). A CO₂ concentration of 2–6% supports high photosynthetic activity and biomass growth ([Chinnasamy et al., 2009](#)). To supply a high CO₂ concentration, flue gas from power generation plants is an attractive source. A high flue gas temperature lowers the process efficiency and requires an additional step to cool the gas. Economically, algae cultivation is still

a costly affair. The mechanical stirring for mixing in photo-bioreactors/open ponds and degasification to remove high inhibitory O₂ concentrations in photo-bioreactors adds to the cost of the process. Nutrients such as nitrogen and phosphorus must be present in adequate quantities to support algal growth. A lack of nitrogen slows respiration rates and affects amino acids and protein synthesis. Limiting phosphorus concentrations affect protein synthesis, nucleic acid synthesis and cell metabolism. This restriction also impacts the photosynthetic energy conversion lowering microalgal growth.

7.2.3 Large-scale microalgae cultivation

There are several ways to grow microalgae, from batch cultures, which are simple and do not require any inputs or outputs, to continuous systems, in which fresh medium is added to the culture and the spent medium is withdrawn from the system at the same rate promoting steady biomass growth. Microalgae can be grown indoors or outdoors in closed or open systems. Open systems for growing microalgae are less expensive. Still, they are more likely to get contaminated or affected by environmental factors. This makes it harder to maintain microalgae growing for long periods. Regardless of the species used as an inoculum, several parameters, such as temperature, light intensity and pH, can significantly affect open pond growth. Also, outdoor cultures only work for a few species and are prone to crashes because the parameters cannot be controlled and sometimes differ from batch to batch. In contrast, closed cultivation systems such as photobioreactors (PBRs) are more expensive but enable strict control of the cultivation parameters that are suitable for microalgal growth. As closed systems, PBRs prevent direct gas exchange between the atmosphere and the algal culture. PBR tubes must be transparent to allow light to pass through and are typically made of glass or acrylic. These systems provide the cultures with a controlled environment (pH, temperature, light intensity and dissolved O₂ and CO₂) and prevent bacterial contamination. Algae productivity is higher in PBRs, but the productivities are not good enough to cover the operational costs of a PBR (Table 7.1).

Table 7.1 Comparison of the specific growth rate of microalgae in different operating systems.

Microorganism	Operational Condition	Specific Growth Rate (/day)	References
<i>Chlorella</i> sp.	Bubble column PBRs	0.3	Naira <i>et al.</i> (2019)
<i>Chlorella ellipsoidea</i>	Bubble column PBRs	Indoor; V = 2 L – 0.168 (±0.006) 20 L – 0.168 (±0.011) Outdoor; V = 200 L – 0.145 (±0.026)	Wang <i>et al.</i> (2014)
<i>Spirulina</i> sp.	Plastic rectangular tanks	Indoor; V = 50 L – 0.42 (±0.030) Outdoor; V = 500 L – 0.1 (±0.02)	Krishnamoorthy <i>et al.</i> (2019)
<i>Chlorella</i> sp.	Batch-scale PBRs for indoor Pilot-scale PBRs for outdoor	Indoor; V = 500 mL – 0.3225 (±0.0039) Outdoor; V = 40 L RDW 5% – 0.978 RDW 10% – 0.2012 RDW 25% – 0.1652	Lu <i>et al.</i> (2015)
<i>C. vulgaris</i>	PMFCs (indoor; V = 100 mL)	LEA – 0.275 (±0.02) FP – 0.208 (±0.015)	Khandelwal <i>et al.</i> (2018)
<i>C. vulgaris</i>	LDPE PBR PMFCs (outdoor; V = 10 L)	RPMFC 5% – 0.63 (±0.056) 10% – 0.59 (±0.049) CW-MFC – 0.54 (±0.035)	Khandelwal <i>et al.</i> (2020)

PBR, photobioreactor; RDW, raw dairy waste; RP-MFC, rock phosphate-microbial fuel cell; CW-MFC, clayware-microbial fuel cell; LDPE, low-density polythene; LEA, lipid extracted algae; FP, fruit pulp.

7.3 ROLE OF ALGAE IN PMFCs

Algae are O₂ producers at the PMFC cathode; the cathodic reactions require a continuous supply of electron acceptors, which are provided by the algae by generating O₂. In a PMFC, the need of O₂ supply is met by utilizing algae's ability to produce O₂ through photosynthesis. The below reactions describe the activity of microflora at the anode and cathode of PMFCs:

At the anode:

Organic matter → Acetate,



At the cathode:

(1) Photosynthetic reaction:



(2) Reduction reaction:



In PMFCs, *Chlorella vulgaris* microalgae produce 750 mmol/L of DO daily; the values exceed by a factor of 3 from the DO concentration of 250 mmol/L that was attained by bubbling air in the cathode (Commault *et al.*, 2017). Algae also fix CO₂, making it a carbon capture system. Continuous O₂ quenching via cathodic oxygen reduction further enhances CO₂ capture rates and algae biomass production. The rate of CO₂ fixation by algae is ~6.24 kg/m³/day through photosynthesis (Elmekawy *et al.*, 2014). Several researchers have explored algae cultivation in PMFCs with promising results obtained in terms of algal growth, water treatment and power generation.

7.3.1 Algal species tested in MFC cathode compartment

Different species of algae present different growth rates, O₂ production rates, carbon capture rates and nutrient assimilation rates. Algae differ in terms of their intracellular biomolecular content and composition as well. Depending on the intended application, a desirable species of algae can be grown in a PMFC. The role of algal species is to replace unsustainable chemical acceptors at the PMFC cathode with photosynthetic O₂. Algae species also impact the power output of a PMFC significantly. *C. vulgaris* is one of the many algal strains frequently used in the cathode compartment of a PMFC due to its high photosynthetic efficiency. The cathode of PMFCs was also tested with several other pure algal species, such as *Dunaliella salina*, *Chlamydomonas reinhardtii*, *Scenedesmus obliquus*, *Desmodesmus* sp. and *Microcystis aeruginosa* to capture solar energy for photosynthesis and generate bioelectricity (Table 7.2).

7.3.2 Mechanism of bioelectricity generation in PMFCs

An energy-generating redox reaction is separated into two chambers of a PMFC, with oxidation taking place at the anode and reduction at the cathode. The oxidation generates electrons that travel from the anode to the cathode constituting the current. The protons diffuse through the PEM to balance the moving charges. The electrons, protons and O₂ combine at the cathode surface to produce water. The main steps in PMFCs are: (1) photosynthesis by algae at the cathodic chamber, (2) oxidation of organic matter under anaerobic conditions by electrogenic bacteria at the anode chamber, (3) transfer of protons and electrons from the anode to the cathode chamber and (4) reduction of O₂ at the cathode.

Table 7.2 Performance assessment of PMFCs using different algal strains for power generation.

Class of Algae	Algae Used at the Cathode	Culture Medium	Light Intensity	Types of MFC	Electrode Used	MFC Material	Power Output	Biomass Produced	COD Removal Efficiency (%)	Reference
Cyanophyceae	BGA	Waste water		Single-chambered tubular	Graphite felts	Polyvinyl chloride	114 mW/m ²		78.9	Yuan <i>et al.</i> (2011)
Chlorophyceae	<i>Chlorella pyrenoidosa</i>	TAP	3,500 lux	Dual-chambered	Carbon paper	Glass	30.15 (± 0.02) mW/m ²	5.94×10^6 cells/mL		Xu <i>et al.</i> (2015)
Chlorophyceae	Mixed algae	Modified Hoagland		Dual-chambered	Graphite felts	Glass	14 mW/m ²			Strik <i>et al.</i> (2008)
Chlorophyceae	<i>C. vulgaris</i>	Modified Bold's	26 W	Dual-chambered	Carbon	Glass	0.6 mW/m ²	437–2,140 mg/L		Mitra and Hill (2012)
Chlorophyceae	<i>S. obliquus</i>	Bold basal	1,600 lux	Dual-chambered	Carbon		102 mW/m ²		74	Kondaveeti <i>et al.</i> (2014)
Chlorophyceae	<i>C. vulgaris</i>	Acetate-free phosphate-buffered solution		Dual-chambered	Carbon	Glass	187 mW/m ²			Liu <i>et al.</i> (2015)
Chlorophyceae	<i>C. vulgaris</i>	Modified Zehnder		Dual-chambered	Graphite plates	Polycarbonate	15.0 (± 0.1)/722 (± 62) mW/m ²		17.6	Lakaniemi <i>et al.</i> (2012)
Chlorophyceae	<i>Dunaliella tertiolecta</i>	Modified Zehnder		Dual-chambered	Graphite felts	Polycarbonate	5.3 (± 2.6)/277 (± 133) mW/m ²		7.7	Lakaniemi <i>et al.</i> (2012)
Pheophyceae	<i>Laminaria saccharina</i>	Minimal		Dual-chambered	Graphite felts	Glass with rubber septa	250 mW/m ²		60–85	Gadhamshetty <i>et al.</i> (2013)
Cyanophyceae	<i>Arthrospira maxima</i>	Basal		Dual-chambered	Graphite granules	Perspex frames, six rubber gaskets, stainless steel	5,800 mW/m ³		67	Inglesby and Fisher (2013)
Cyanophyceae	<i>A. maxima</i>	Zarrouk's		Dual-chambered	Graphite granules	Perspex frames, six rubber gaskets, stainless steel	10.2 (± 1.3) W/m ³		41–63	Inglesby and Fisher (2012)

(Continued)

Table 7.2 Performance assessment of PMFCs using different algal strains for power generation (Continued).

Class of Algae	Algae Used at the Cathode	Culture Medium	Light Intensity	Types of MFC	Electrode Used	MFC Material	Power Output	Biomass Produced	COD Removal Efficiency (%)	Reference
Chlorophyceae	<i>Scenedesmus</i> sp.	Open pond (fresh water)		Dual-chambered	Carbon brushes	Acrylic	1.78 W/m ²			Rashid <i>et al.</i> (2013)
Cyanophyceae	<i>M. aeruginosa</i>	Lake water		Dual-chambered	Graphite brushes	Polymethyl methacrylate	4.14 (±0.05) W/m ³		81 (±6) to 23 (±4)	Wang <i>et al.</i> (2012)
Chlorophyceae	<i>C. vulgaris</i>	Lake water		Dual-chambered	Graphite brushes	Polymethyl methacrylate	3.70 (±0.02) W/m ³		73 (±3) to 30 (±5)	Wang <i>et al.</i> (2012)
Cyanophyceae, Chlorophyceae, Bacillariophyceae	<i>Ankistrodesmus</i> , <i>Chlorella</i> , <i>Oscillatoria</i> , <i>Scenedesmus</i> , <i>Diatom</i> , <i>Cosmarium</i>	Synthetic wastewater	3,000 (±200) lux	Single-chambered	Graphite plates	Perspex	3.55 µW/m ²	2.87 g/L		Subhash <i>et al.</i> (2013)
Chlorophyceae	<i>Scenedesmus abundans</i>	BBM media	94.6 µmol/m ² /s	Dual-chambered	Graphite rods	Plexiglass	838.68 mW/m ²	0.94 (±0.01) g/L	97.24	Nayak and Ghosh (2019)
Cyanophyceae	<i>Synechococcus leopoliensis</i>	BG-11	40 (±5) µE/m ² /s	Dual-chambered	Carbon fiber veils	Acrylic	42.5 W/m ³	6 × 10 ⁵ cells/mL		Walter <i>et al.</i> (2015)
Cyanophyceae	<i>M. aeruginosa</i>	BG-11	7,000 lux	Dual-chambered	Carbon paper	Glass	58.4 mW/m ³			Cai <i>et al.</i> (2013)
Chlorophyceae	<i>Desmodesmus</i> sp.	BG-11	3,000 lux	Dual-chambered	Graphite felts	Plexiglass	99.09 mW/m ²			Wu <i>et al.</i> (2014)
Chlorophyceae	<i>C. vulgaris</i>	Bold's basal		Dual-chambered	Toray carbon cloths with 10% Teflon		13.5 mW/m ²	300 mg/dm ³	80	Campo <i>et al.</i> (2013)
Chlorophyceae	<i>C. reinhardtii</i> transformation F5	TAP	900 lux	Dual-chambered	Graphite electrodes	Poly-acrylic plastic	12.947 mW/m ²			Lan <i>et al.</i> (2013)
Chlorophyceae	<i>Scenedesmus quadricauda</i> SDEC-8	Domestic wastewater	135 µmol/m ³ /s	Single-chambered	Carbon cloth with titanium wire	Plexiglas cylinder	62.93 mW/m ²		62	Yang <i>et al.</i> (2018)
Chlorophyceae	<i>C. vulgaris</i> strain CS-42	SWW 4% and phosphate buffer	200 mmol/m ² /s	Dual-chambered	Graphite plates		34.2 (±10) mW/m ²			Commault <i>et al.</i> (2017)

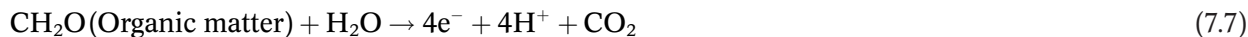
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Table 7.2 Performance assessment of PMFCs using different algal strains for power generation (Continued).

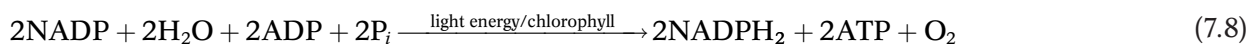
Class of Algae	Algae Used at the Cathode	Culture Medium	Light Intensity	Types of MFC	Electrode Used	MFC Material	Power Output	Biomass Produced	COD Removal Efficiency (%)	Reference
Chlorophyceae	<i>C. vulgaris</i>	Effluent water from a chocolate factory	75,000 lux	Dual-chambered	Graphite plates	Glass	23.17–327.67 mW/m ²	5.2 mg/mL	78.6	Huarachi-Oliviera <i>et al.</i> (2018)
Chlorophyceae	<i>Chlorella sorokiniana</i>	BG-11	2,000 lux	Dual-chambered	Carbon felts	Poly-acrylic	3.2 W/m ³		65.97 (±0.83)	Neethu <i>et al.</i> (2018)
Chlorophyceae	<i>C. vulgaris</i>	BG-11	Solar radiation	Dual-chambered	Graphite felts	RP-mixed CW	11.5318 kWh/m ³	0.307 kg/m ³ /day		Khandelwal <i>et al.</i> (2020)
Chlorophyceae	<i>C. vulgaris</i>	BG-11	PBRs	Dual-chambered	Graphite felts	Acrylic	2.7 W/m ³	0.028 kg/m ³ /day		Khandelwal <i>et al.</i> (2018)
Unspecified	Microalgae (unspecified)	Wastewater		Dual-chambered	Carbon veils	Terracotta cylinder	44 µW			Salar-García <i>et al.</i> (2016)
Chlorophyceae	<i>Dunaliella salina</i>	Modified Johnson's media	9800 lx	Dual-chambered	Graphite felt	Acrylic material	213.38 mW/m ²	4.02 ± 6 × 10 ⁶ cells/mL	59.32%	Mishra and Chhabra (2022)

BBM, Bold's basal medium; BG-11, Bluegreen-11; BGA, blue-green algae; TAP, Tris-acetate-phosphate; RP, rock phosphate; CW, clayware; SWW, Synthetic wastewater.

In the case of anodic electrogenic bacteria, oxidation of NADH to NAD⁺ during the mitochondrial respiration process plays a principal role in voltage generation (Logan *et al.*, 2006). Many dehydrogenase reactions are an integral part of the plasma membrane of algae and have an essential role in the electron transport chain (Shukla & Kumar, 2018):



In algae, photosystems (PSs) comprise several pigments, including chlorophyll. The pigments are arranged in a fashion that harvests maximum solar energy and transmits that to the reaction center. The carotenoids are at the outer surface and chlorophyll forms the reaction center. Algal photosynthesis is completed in two steps: light and dark reactions. The light mediates electron ejection from the two reaction centers of the P680 and P700 PSs during the light reaction. The oxidized P680 splits water into O₂, protons and electrons. The electrons fill the hole and stabilize the reaction center for the next cycle. The splitting molecule breaks water molecules into O₂, protons and electrons. Meanwhile, the ejected electron transports through the Z-scheme from PS II (P680) to PS I (P700), resulting in a proton gradient that generates ATP. In non-cyclic photophosphorylation, NADP⁺ accepts the electron generated through the oxidation of PS I, producing NADPH. In cyclic photophosphorylation, only ATP is generated as the electron flows in a cycle coming back to the same reaction center:



During the dark reaction, NADPH and ATP reduce CO₂ through the Calvin–Benson cycle (C₃ cycle). The first step of this cycle is catalyzed by ribulose-1,5-bisphosphate carboxylase/oxygenase (Rubisco) (Taiz *et al.*, 2015) as shown in the below equation:



The Rubisco enzyme shows a higher affinity for O₂, thereby reducing the rate of CO₂ fixation and algal growth (Hopkins, 1999). As O₂ inhibits the process of photosynthesis, continuous/periodic deoxygenation of the algal growth medium is essential to sustain healthy algal growth.

7.4 PMFC DESIGN PARAMETERS

7.4.1 Dual chambers vs sediment MFCs

PMFCs can be single- or dual-chambered. In both types, an anode and a cathode are immersed in two redox environments. In sediment MFCs, the top layers are suitable for cathode placement as these are exposed to light and support photosynthesis and O₂ generation (Shukla & Kumar, 2018). Anaerobiosis prevails in the bottom-most layers of the water column and sediment where the anode can be placed. In dual chambers, the electrodes can be on the same plane, but separated into different compartments via a selectively permeable membrane. Both systems can be operated in batch or continuous mode. The shape of the chambers can vary. Dual-chambered PMFCs are difficult for industrial scale-up due to their complex design and high-operational costs. Hence, single-chambered PMFCs are preferred over the dual-chambered design.

7.4.2 Construction materials, electrolytes, electrodes and separators

MFCs can be made of glass, concrete, acrylic, polyvinyl chloride, polycarbonate, perspex, stainless steel, polymethyl methacrylate, plastic and ceramics (Shukla & Kumar, 2018). In the case of PMFCs, the cathodic chamber has to be transparent for light penetration. Ceramics are the most sustainable for anodic compartments among the said materials as these are cost-effective, use renewable materials and do not create pollution. The most common ceramics are earthenware pots, terracotta pots and cylinders and goethite cylinders (Behera & Ghangrekar, 2011).

7.4.3 Electrode materials

An MFC electrode should have high conductivity, higher surface area, non-toxic, non-polluting and have a low charge transfer resistance. The latter often requires the modification of electrodes with electrode catalysts that accelerate the rate of charge transfer reactions. Researchers attempt to enhance power output of MFCs by changing the shape and material of electrodes. Carbon-based electrodes such as carbon cloth, carbon felts, carbon brushes, carbon paper, graphite plates, graphite rods and graphite foils are commonly used as base electrodes (Shukla & Kumar, 2018). Electrode catalysts can be platinum black or metal oxides as well as their nanoparticles coated on/dispersed in carbon materials. Similarly, carbon nanotubes, graphene and N-doped graphene/graphene oxide find applications as electrode materials. Other less commonly used electrode materials include gold, silver, copper, nickel, cobalt, aluminum, titanium and stainless steel (Zhao *et al.*, 2008). Baudler *et al.* (2015) found that carbon electrodes are better than other electrodes as they offer low internal resistance, are cheap, recyclable, non-corrosive and usable for a long time. The drawback of carbon electrodes, particularly for PMFC applications, is the dark and opaque appearance that blocks light or has a shading effect. In addition, these have low thermal conductivity, resulting in low power output. Therefore, transparent metal electrodes are suitable for PMFC applications (Baudler *et al.*, 2015).

The disadvantages of metal electrodes include low surface roughness that reduces charge transfer rates and the absence of well-defined pores that prevents proper microbial attachment (Logan *et al.*, 2007). Moreover, metal electrodes may corrode over time resulting in background currents and lower conductivity. Literature studies suggest that the nanoscale surface morphology of electrodes decides the conductivity and ability to interface with microbes (Bacakova *et al.*, 2011; Cao *et al.*, 2009; Legeay *et al.*, 2010; Sekar *et al.*, 2004). Even with electrodes made of similar carbon material, differences in conductivity, surface area, space and size of carbon microfiber can result in variable MFC performance (Sanchez *et al.*, 2015).

7.4.4 Separators

The electrolyte and separator should have high conductivity. The anions and cations flow in opposite directions in closed circuits with flow rates proportional to the magnitude of the current. The separator has two roles: (1) to prevent the mixing of anodic and cathodic substrates and (2) to allow the movement of ions across the chambers. Separators such as electrodes should be non-toxic, have high ionic conductivity and be impermeable to O_2 . The most commonly used separators are PEMs that wear negatively charged groups such as $-PO_3^{2-}$, $-PO_3^{-}$, $-COO^{-}$, $-C_6H_4O^{-}$ and $-SO_3^{-}$ that successively protonate and deprotonate (Rodenas *et al.*, 2015). The lower ion/proton conductivity increases the internal resistance, lowers conductivity and creates a pH gradient with low pH at the anode. Low pH disrupts microbial growth and lowers the power output (Winfield *et al.*, 2013). MFCs can be operated without any separator, but this decreases the Coulombic efficiency of the system due to the diffusion of O_2 to the anode and electron donor to the cathode (Ghangrekar & Shinde, 2007). The anion exchange membranes with positively charged groups, such as $-PR^{+}$, $-SR^{+}$ and $-NH_3^{+}$ help exchange negative ions creating an ionic balance (Zhuang *et al.*, 2012). Separators such as salt bridges, glass fibers, glass wool, clayware (CW) membranes and ceramic membranes can exchange cations and anions primarily due to their high water-holding capacity. Separators need periodic replacement/cleaning as they tend to clog in long-term operations lowering the MFC performance. Separators made of non-biodegradable materials are undesirable. Thus, ceramic separators such as CW hold promise for scale-up of MFC systems. Behera and Ghangrekar (2011) demonstrated the efficiency of the terracotta separator that has a wall thickness of 4 mm. Similarly, rock phosphate (RP)-blended CW can be used as a separator showing maximum power density, 5% RP blend yields 890 (± 95), 10% RP-blend yields 960 (± 120) and CW yields 1200 (± 152) mW/m³ of power density (Khandelwal *et al.*, 2018).

7.4.5 Effect of light intensity, temperature, DO, CO₂, pH and salts

Light is essential and the duration of the photoperiod determines the power output. High light intensity increases the temperature, enhancing the reaction rate and the substrate utilization rate up to a certain temperature (Shukla & Kumar, 2018). A light intensity of 3,500–100,000 lux with a light/dark regime of 18/6, 12/12 and 16/8 h was reported as optimum for PMFC operation (Reddy *et al.*, 2019). The optimum temperature for PMFC operation is 30°C but may vary from organism to organism. DO also increases with light intensity. However, as mentioned in the previous sections, high DO inhibits algal growth. The DO content also depends on the types of algal species used at the cathode. PMFCs can provide a DO content as high as 18–19 mg/L for cathodic reactions (Taskan & Taskan, 2022). A minimum DO of 2.2 mg/L is required for the continuous operation of PMFCs (Jang *et al.*, 2013). The DO content also decreases with salinity (Kim & Chung, 1984).

A neutral pH range of 6.8–7.5 is optimal for algal growth. CO₂ dissolves well in alkaline pH and algae absorb better bicarbonates (Reddy *et al.*, 2019). The CO₂ concentration at the cathode plays an essential role in MFC performance, with an optimum value of ~5% CO₂–air mixture (Reddy *et al.*, 2019). There are two methods for supplying CO₂ to the PMFCs: *in-situ* and *ex-situ*. In the *ex-situ* method, the catholyte is purged or supplied with pure CO₂ gas or bicarbonate generated elsewhere. In an *in-situ* way, CO₂ generated through anaerobic digestion of organic matter at the anode is used. There are benefits and drawbacks of both strategies. Microalgae grow faster with the *ex-situ* method, but the system requires additional units for the transport of CO₂. The *in-situ* method relies on the metabolic rate of microbes at the cathode, the partial pressure and pH of the catholyte. Generally, anodic off-gas supports effective algal growth (Khandelwal *et al.*, 2018). It directly affects how well the PMFC works, making the material more conductive.

7.5 ECONOMIC IMPORTANCE OF PMFCs

Algae accumulate a high oil content in their cells and the oil composition is suitable for biodiesel production. The residual algae biomass is rich in proteins and carbohydrates. In a biorefinery, the residual algae are often subjected to biogas generation (Figure 7.2). However, the same can be used as an anodic substrate in MFCs. Algae biomass is also bioconvertible to ethanol, biogas and biohydrogen (Figure 7.2). Algae harvest and drying prior to lipid extraction is challenging. Therefore, wet algal biomass transesterification has been tested. Direct or wet transesterification is the same as dry transesterification except that the extraction step is skipped and the whole biomass is used as the feedstock of the reaction (Shukla & Kumar, 2018). Microalgae only require a little pre-treatment before fermentation because their cell walls have only a thin cellulosic fence and lack lignin.

Conversely, macroalgae must be treated first to get the stored carbohydrates out, whereas microalgal fermentation produces ethanol, acetate, hydrogen and CO₂. Hydrogen production from algal biomass can be accomplished through photobiological and fermentative processes (Shaishav *et al.*, 2013). Hydrogenase is the main enzyme that speeds up reactions that lead to the production of biohydrogen. Some microbes can break down the organic compounds in algae without O₂ to generate methane and CO₂. Algal biomass can avoid the requirement of harvesting, dewatering, drying or oil extraction to produce biogas (Shukla & Kumar, 2018).

Algae biomass is rich in carotenoids, terpenoids, xanthophylls, chlorophylls, phycobilins, polyunsaturated fatty acids, polysaccharides, vitamins, sterols, tocopherols and phycocyanins (Ammar *et al.*, 2022). Algae are frequently utilized as dietary supplements because of their nutritional value. *Spirulina* and *Chlorella* are the two well-known algal species used as food sources. Algae aid in treating diseases such as diabetes, rheumatic disorders and high blood pressure in the arteries. Algae also benefit memory and concentration by providing omega 3 and omega 6 polyunsaturated fatty acids required for brain development. Additionally, they fight off bacterial, fungal and viral infections. Algae are also rich in natural antioxidants and antimicrobials that improve shelf life and circumvent artificial preservatives (Gonçalves, 2021).

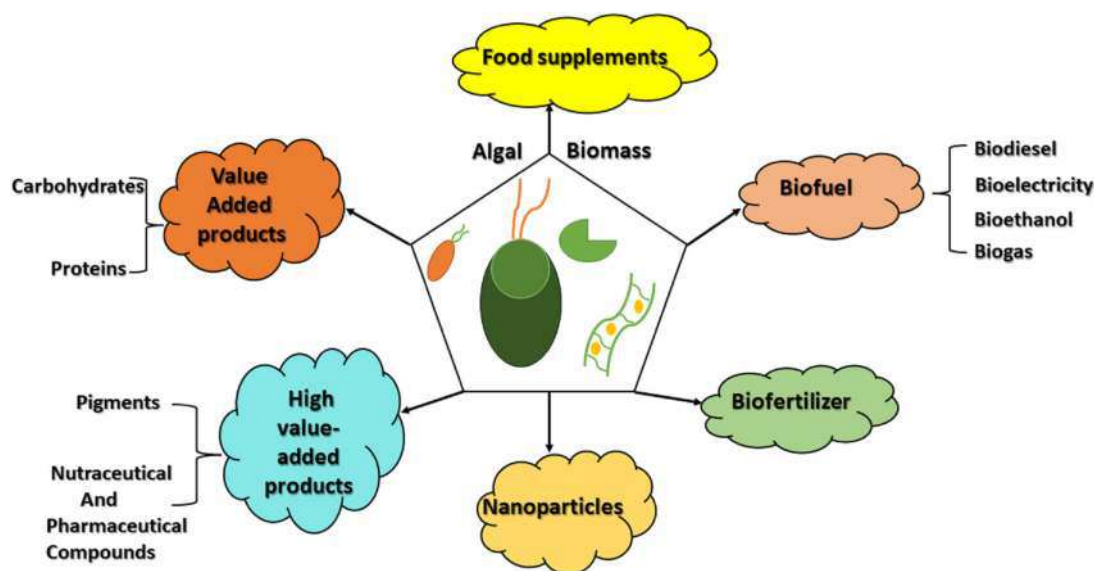


Figure 7.2 Cultivation, value-added product generation, power generation, biofuel generation and resource recovery of algae biomass.

Algae, which can be unicellular, multicellular, filamentous or saponaceous, are examples of organisms that produce their food through photosynthesis. With over 200,000 species, they are also the most prominent primary producers in the world. Microalgal production requires mass cultivation, biomass recovery and downstream operations to ensure a consistent yield for food, chemicals, feed, biofuel and biofertilizers, as shown in [Figure 7.3](#) (Balasubramaniam *et al.*, 2021). BGA (microalgae) can yield plant growth hormones, polysaccharides, chemicals that kill bacteria and other metabolites.

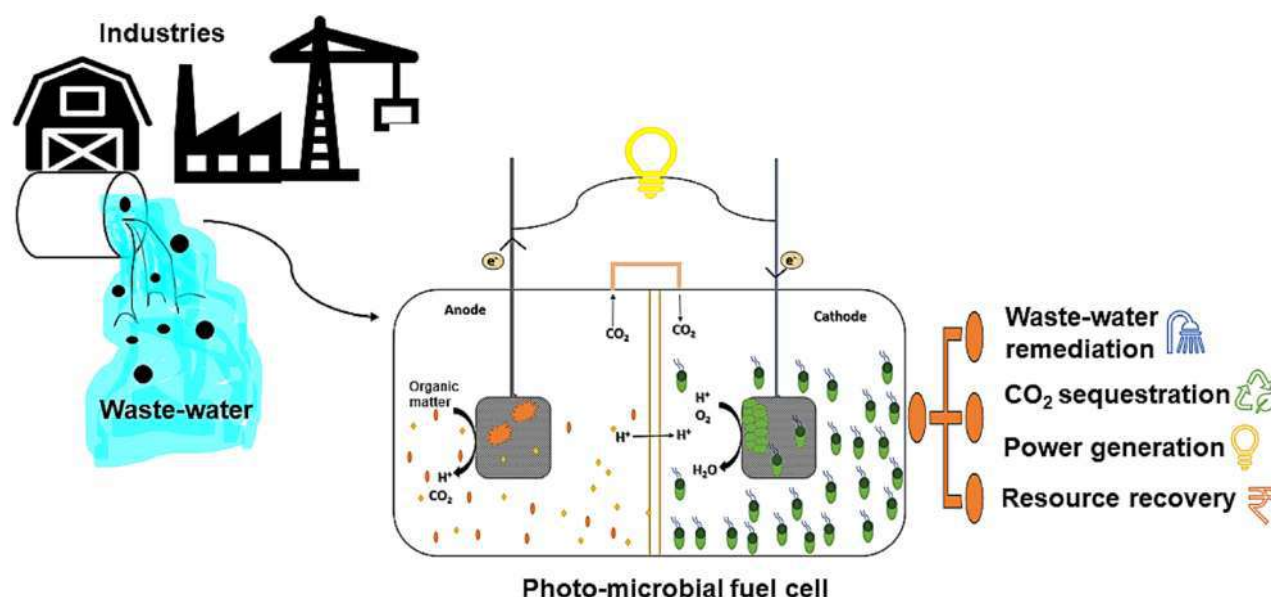


Figure 7.3 PMFCs for wastewater remediation, algae cultivation, biomass recovery, power generation and downstream operations to ensure a consistent yield for food, chemicals, feed and biofuel.

They can also improve the fertility and quality of the soil (Ronga *et al.*, 2019). According to Guo *et al.* (2020), the primary sources of organic matter of the agroecosystem are cyanobacteria and green microalgae. In the case of constructed wetland PMFCs, algae can make a big difference in the amount of organic carbon in the soil by taking up CO₂. Cyanobacterial heterocyst cells (e.g., *Nostoc*, *Anabaena* and *Aulosira*) may fix atmospheric nitrogen to satisfy the needs of the microbiota and plants of the soil (Fritsch, 1945). Several studies have shown that injecting cyanobacteria or groups of cyanobacteria into crops has made a big difference in the nitrogen content of the soil. Inoculating the soil with cyanobacteria can reduce the soil nitrogen fertilizer requirement by 25–40% (Ammar *et al.*, 2022).

7.6 FUTURE PERSPECTIVES

Microalgae use CO₂ as a carbon source in their metabolism, reducing the load of CO₂ in the environment. Because microalgae can grow independently while capturing CO₂, this process can be combined with other techniques already linked to microalgal growth, such as cleaning wastewater, generating biofuels and producing high-value products. Open ponds and PBRs are suitable techniques for algal growth in conventional methods, but compared with PMFCs, the latter showed better results for algal growth, where O₂ degassing is natural because it acts as the terminal electron acceptor.

Compared to heterotrophic MFCs or photovoltaic cells, PMFCs have some observable advantages. They can generate electricity solely from natural resources such as sunlight, water and CO₂. As a result, it is not necessary to load MFCs with organic compounds and the utilization of CO₂ also contributes to carbon sequestration, resulting in a clean environment. Regardless of day or night, PMFCs can produce power continuously. The procedure can turn the algal biomass into proteins, pigments and biofuels such as biodiesel, biogas and bioethanol. It can also be used with PBRs to supply O₂ to MFCs, thus allowing clean wastewater by MFCs. The performance of the system is influenced by light and DO, making it challenging to produce electricity continuously and sustainably. When assessing how sustainable these systems are, it is also essential to look at the relationship between the amount of electricity used and the removal of pollutants and how well the system is set up and run.

The challenge lies in the process scale-up. The scale-up studies of PMFCs have revealed higher capital costs ranging from \$735/m³ to \$36,000/m³ (Liang *et al.*, 2018; Wang *et al.*, 2020). Low-cost PMFC systems depend on CW separators, and low-cost PBRs have also been tested and the system worked well for power generation (Khandelwal *et al.*, 2020). Electrodes and membranes are the major causes for high cost of an MFC system. Sediment PMFCs are relatively cheaper and can be applied to real-world applications. The operating parameters, such as temperature, pH, organic loading, salinity, conductivity, start-up and hydraulic retention time have been optimized for a number of contaminated soils or artificial wetlands. Although promise associated with the process has been demonstrated, a need to study the process for large-scale *in-situ* bioremediation, bioaugmentation and algae cultivation remains. This involves a close understanding of anodic and cathodic microenvironments, mass transfer efficiencies, soil/sediment characteristics in the case of sediment MFCs and environmental conditions. The impact of these factors on algae biomass growth and composition is important for reproducible commercial applications of PMFCs.

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Chapter 8

Integrated anaerobic digestion and algae cultivation

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ABSTRACT

Anaerobic digestion is considered a versatile process that for years has been used to treat various types of waste. Besides being a low-cost technology applicable in rural and urban locations, anaerobic digestion produces multiple by-products that can be integrated into a biorefinery scenario. Similarly, microalgae biotechnology can adequately complement anaerobic digestion by improving resource recovery through a closed-loop process and contributing to a biobased green circular economy model. Therefore, this chapter aims to address current perspectives on the topic. It covers algae cultivation from anaerobic digestion residues as a post-treatment option and digestate as a potential medium for microalgae growth. Moreover, anaerobic digestion is presented as an energetic valorization route of algae biomass, including strategies to overcome main challenges, such as pre-treatment of microalgae biomass and anaerobic co-digestion. Biogas upgrading during algae cultivation is also discussed. Finally, it presents biorefinery models based on integrated microalgae and anaerobic digestion, reporting the technologies' sustainability and environmental impacts. Future perspectives on the subject are highlighted, encouraging further studies to improve microalgae biomass production, nutrient recovery, wastewater treatment, and biogas upgrading.

8.1 INTRODUCTION

Anaerobic digestion (AD) is one of the most diffused biotechnologies for converting organic biomass to bioenergy (Chen *et al.*, 2018). In this oxygen-deprived process, organic substrates are submitted to microbial conversion in reactors to produce biogas (Greene, 2019). In addition, the process also results in liquid or solid residues containing residual nutrients and microbiota (Zicari *et al.*, 2019). Moreover, organic waste streams' use as a substrate is highly attractive from the economic and environmental perspectives, with food wastes, agricultural, municipal solids, animal manure, poultry, and microalgae as reported substrates (Khan *et al.*, 2021).

Research on algal substrates for AD dates back to the late 1950s (Golueke *et al.*, 1957), based on its potential for biofuel production through biomass valorization. In addition, algal cultivation can be employed as photosynthetic biogas upgrading technology, recovering CO₂, and purifying CH₄

(Franco-Morgado *et al.*, 2021). The mobilization of nutrients (nitrogen (N) and phosphorus (P)) and the CO₂ availability made the AD process highly attractive for microalgae applications (Solé-Bundó *et al.*, 2019b). Anaerobically treated effluent or digestate can thus be useful as a microalgae culture medium, allowing effluent polishing and providing nutrients for algae growth (Barreiro-Vescovo *et al.*, 2020). In this context, several pathways can be employed to integrate AD technology with algae cultivation to promote resource recovery and waste treatment.

Different types of biomass can be obtained in microalgae biotechnology according to the wastewater used for growth medium or culture conditions. Multiple products can be produced due to the great versatility of microalgae biomass and AD process. In a circular economy biorefinery concept, AD and algal biotechnology integration occurs via nutrient recycling (Chen *et al.*, 2018), being an interesting multi-arrangement alternative for environmental, social, and economic development. In addition to compensating for each process's limitations, coupling two or more waste treatment technologies can act as an engine to improve resource recovery through a closed-loop process (Sikarwar *et al.*, 2021).

Bearing this context in mind, this chapter includes current perspectives on the topic, covering algae cultivation from AD residues, AD as an energetic valorization route of algae biomass, algae cultivation for biogas upgrading, and coupling technologies for sustainable biorefineries.

8.2 ALGAE CULTIVATION FROM AD RESIDUES

8.2.1 Liquid effluent

Considering biological systems for wastewater treatment, several technologies based on anaerobic treatment are available, such as high-rate anaerobic systems, namely up-flow anaerobic sludge blanket reactor (UASB), anaerobic contact process, anaerobic filter or fixed film reactors and fluidized-bed reactors (Khan *et al.*, 2011). Among them, the UASB reactor is widely used in countries with hot climates, such as Brazil, Colombia, and India (Chernicharo *et al.*, 2018). It offers economic and operating benefits and less area demand than conventional treatments like activated sludge and stabilization ponds (Vassalle *et al.*, 2020a).

Although these anaerobic reactors are used for organic matter removal, it is widely accepted that their performance needs to be improved to meet many wastewater discharge standards. It results in an effluent that may still contain high concentrations of organic matter, suspended solids, nutrients (Khan *et al.*, 2011), and pathogenic organisms, thus requiring post-treatment steps.

The chemical oxygen demand (COD) in domestic wastewater treated in different conventional anaerobic systems commonly presents concentrations ranging from 70 to 160 mg/L (Chernicharo, 2006). The typical COD removal efficiency in a UASB reactor varies around 55% and 70% (Chernicharo, 2006). Table 8.1 presents studies using anaerobic effluents for microalgae cultivation. Concentrations ranging from 141 to 45,875 mg COD/L can be present in anaerobic effluents, mainly explained by several factors, such as wastewater type, reactor scale, and treatment parameters. Notably, in some works, wastewater dilution was necessary to adapt the culture medium organic matter load to the ideal conditions for microalgae development (de Godos *et al.*, 2016; Kimura *et al.*, 2019; Xie *et al.*, 2018; Zhen *et al.*, 2022).

The greatest COD values (Table 8.1) do not meet the release standards of Brazil ($150 < \text{COD} < 225$ mg/L) (Morais & Santos, 2019) and the European Union ($\text{COD} = 125$ mg/L). Regarding nutrients, the Brazilian standard establishes a limit of 20 mg/L for NH₄⁺ (CONAMA, 2011), while the European Commission requires a more rigorous standard of 15 mg/L for total nitrogen. Thus, a post-treatment is needed not only at a secondary level but also at a tertiary level to maintain balanced discharge into ecosystems.

Some technologies have been suggested for UASB reactor effluent post-treatment. Among them are trickling filters, submerged aerated biofilters, rotating biological contactors, wetlands, sequencing batch reactors, chemically enhanced sedimentation, zeolite columns, and dissolved air flotation (Khan *et al.*, 2011). In recent decades, microalgae-based technologies, such as the high-rate algal pond (HRAP), have also been evaluated and shown to be promising (Assemany *et al.*, 2018; Benett

Table 8.1 Concentrations of organic matter and nutrients from effluents treated in an anaerobic system followed by microalgae-based treatment.

Type of Effluent	Anaerobic/Aerobic Treatment Unit	Raw Effluent (mg/L)	Microalgae	Treated Effluent (mg/L)	Biomass Productivity	References
Domestic sewage	UASB/Hybrid system composed of an HRAP and biofilm reactor	CODs = 116.0 N-NH ₄ ⁺ = 37.3 N-NO ₃ ⁻ = 1.6 Ps = 5.2	Mixed culture (autochthonous species)	CODs = 78.0 (33%) N-NH ₄ ⁺ = 6.1 (84%) N-NO ₃ ⁻ = 29.9 (-1.769%) Ps = 4.1 (21%)	6.79 g/m ² day	Assis <i>et al.</i> (2017)
Raw domestic wastewater (screened)	Anaerobic pond/HRAP	BOD ₅ = 94 N-NH ₄ ⁺ = 36 N-NO ₃ ⁻ = 0.1	Mixed culture (autochthonous species)	BOD ₅ = 52 N-NH ₄ ⁺ = 15.2 (33–76%) N-NO ₃ ⁻ = 0.2	1.281–4.112 mg/L (chlorophyll-a)	Sutherland <i>et al.</i> (2017)
Domestic sewage	Septic tank/HRAP fed with gas from the combustion of gasoline	CODs = 174.5 TKN = 87.8 Ps = 12.3	Mixed culture (autochthonous species)	CODs = 110.8 (30%) TKN = 36.2 (37.9%) Ps = 13.5 (-11.3%)	6.12 g/m ² day	Assis <i>et al.</i> (2019)
Anaerobically digested distillery (diluted)	UASB/Photobioreactor (rectangular tank with submerged mixer aerator)	COD = 45,875	<i>Spirulina</i> sp.	COD = (60–70%)	0.08–0.094 g dry biomass per L/d	Krishnamoorthy <i>et al.</i> (2019)
Domestic sewage	Septic tank/Hybrid system composed of a HRAP and biofilm reactor	COD = 329.2 N-NH ₄ ⁺ = 87.4 NO ₃ ⁻ = 1.1 TP = 9.1	Mixed culture (autochthonous species)	COD = 135.5 (58.8%) N-NH ₄ ⁺ = 19.9 (77.3%) NO ₃ ⁻ = 40.3 TP = 7.7 (16.2%)	6.13 g/m ² day	Assis <i>et al.</i> (2020)
Municipal wastewater	UASB/HRAP	COD = 232.69 (55%) TN = 54.33 N-NH ₄ ⁺ = 34.21	Mixed culture (autochthonous species)	COD = 146.08 (38%) TN = 24.31 (30%) N-NH ₄ ⁺ = 14.31 (44%)	1.01 g/L (volatile solids)	Vassalle <i>et al.</i> (2020b)
Domestic sewage	UASB/HRAP	COD = 141 ± 48 TKN = 41.1 ± 12.0 TP = 4.4 ± 0.8	Mixed culture	COD = 63.7 ± 11.3 TKN = 9 ± 4.2 TP = 3.4 ± 0.6	NA	Oss <i>et al.</i> (2022)

Note: Average (±standard deviation) final concentration values of water quality variables found in the literature were reported, and the values within parentheses refer to the removal efficiency. NA = not available. UASB: upflow anaerobic sludge blanket; HRAP: high-rate algal pond; BOD₅: biochemical oxygen demand; COD: chemical oxygen demand; CODs: soluble chemical oxygen demand; TKN: total Kjeldahl nitrogen; TN: total nitrogen; N-NH₄⁺: ammoniacal nitrogen; N-NO₃⁻: nitrate nitrogen; PO₄³⁻: phosphorus; Ps: soluble phosphorus; TP: total phosphorus.

et al., 2008; Couto *et al.*, 2020; Magalhães *et al.*, 2022; Santiago *et al.*, 2013; Vassalle *et al.*, 2020a, 2020b; Villar-Navarro *et al.*, 2018) (Figure 8.1). Systems that use algal–bacterial symbiosis represent a wastewater treatment technology (Zhen *et al.*, 2022) with the advantages of reduced energy consumption during aeration, efficient nitrogen and phosphorus removal, and effective biomass recycling (Xie *et al.*, 2018). Symbiotic interactions between microalgae, bacteria, and fungi have been used for wastewater treatment (Kabir *et al.*, 2022; Leng *et al.*, 2020; Leong & Chang, 2022; Zhang *et al.*, 2021). Some factors encouraging the UASB and HRAP integration are:

- (i) The anaerobic treatment partially removes turbidity and suspended solids from the wastewater, which contributes to the light incidence in the water column and, consequently, microalgae growth and photosynthetic activity (Couto *et al.*, 2021; de Godos *et al.*, 2016).
- (ii) The anaerobic effluent is rich in nutrients (NH_4^+ and PO_4^{3-}), which are more readily available and essential for microalgae growth. It is noteworthy that the main N form assimilated by microalgae is NH_4^+ .
- (iii) Bioremediation can be combined with simultaneous valuable bioproducts production, like pigments, biodiesel, bioCH_4 , and biofertilizer (Leong & Chang, 2022).

Microalgae are involved in O_2 production, CO_2 consumption, and nutrient removal via photosynthesis. At the same time, bacteria are responsible for fixing and regenerating inorganic nutrients (NH_4^+ , PO_4^{3-} , H_2S), consuming organic matter, and producing vitamins and siderophores (Lian *et al.*, 2018). Fungi and bacteria are involved in organic matter degradation in the anaerobic digestate, while microalgae can assimilate the CO_2 released during the degradation (Zhang *et al.*, 2022). This way, the produced biomass can remove N-NH_4^+ (volatilization or assimilation), PO_4^{3-} (precipitation or assimilation), and acetate along with metallic ions, for example calcium (Ca), magnesium (Mg), and iron (Fe) (Pacheco *et al.*, 2015). The biomass can also produce polypeptides, called chelating agents, capable of binding to heavy metals, for example mercury (Hg), cadmium (Cd), and lead (Pb) (Kabir *et al.*, 2022). Oliveira *et al.* (2021) observed that the presence of Cu and Zn, found in swine wastewater, altered the dynamics of HRAPs regarding nutrient removal, productivity, and biochemical composition of the biomass. Similarly, Oliveira *et al.* (2023) concluded that nutrient removal and biomass biochemical composition should be considered to combine the recovery of Zn and nutrients with the production of value-added biomass. Therefore, environmental parameters must also be considered, as they have effects on gene expression and can promote some biological pathways to the detriment of others, thus modifying the microbial structure and the inherent metabolism involved in the biotechnological process (Bose *et al.*, 2020; Lopatkin & Collins, 2020).

Vassalle *et al.* (2020b) investigated the combination of anaerobic (UASB) and aerobic (HRAP) treatment to treat municipal wastewater and reported that the HRAP was responsible for only 38% of COD removal, while the global mean removal efficiency of this variable was 72%. Still, according to the authors, HRAP was found efficient in removing estrogens (90–95%) and pharmaceuticals (64–70%).

Concerning nutrients, Oss *et al.* (2022), in a wastewater treatment plant (WWTP) composed of a UASB reactor followed by HRAP, produced activated carbon (C) from biomass and achieved removal rates for COD, N, and P similar to values already presented in the literature (Craggs *et al.*, 2012; Park & Craggs, 2010). Zkeri *et al.* (2021) compared two systems composed of a methanogenic moving-bed biofilm reactor (AnMBBR) followed by an aerobic MBBR (AeMBBR) and sequencing batch reactor (SBR) with *Chlorella sorokiniana*. The authors reported that the AnMBBR + AeMBBR combination removed COD, NH_4^+ , total Kjeldahl nitrogen, and PO_4^{3-} by 93 (± 4)%, 97 (± 3)%, 99 (± 1)%, and 49 (± 15)%, respectively, while the AnMBBR + SBR combination removed COD, but only partially the other pollutants.

Table 8.1 summarizes the biomass production using anaerobic effluent as a culture medium. Under Brazilian environmental conditions, studies have reported productivities (based on volatile solids (VS) value) around 6.5 g/m²/day (Assis *et al.*, 2019; Assis *et al.*, 2017), operating with autochthonous

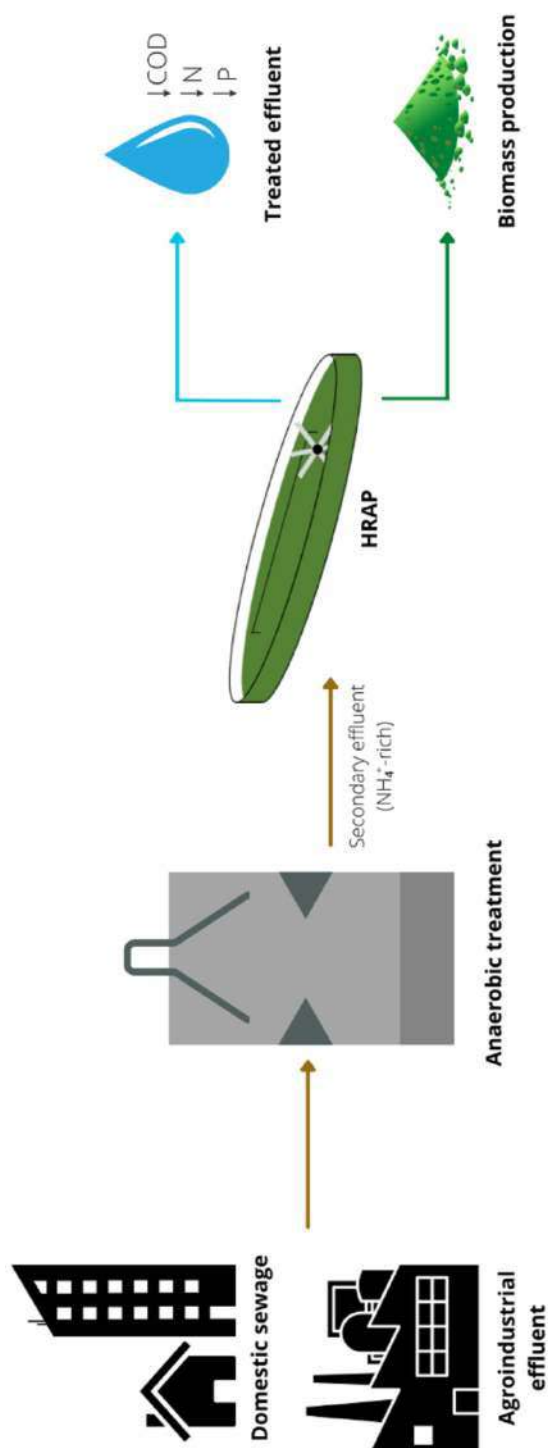


Figure 8.1 Schematic diagram of UASB and HRAP integration to increase pollutant removal and produce algal biomass. COD: chemical oxygen demand; N: nitrogen; P: phosphorus; HRAP: high-rate algal pond.

species adapted to the culture medium. This value can still be improved, and studies on biomass production optimization through operational strategies should be encouraged, considering the adversities of outdoor conditions and competition between microorganisms. Thus, a wide field of study can explore the treatability of both anaerobic systems and treatment system arrangements that allow maximum resource recovery.

8.2.2 Digestate

Organic waste AD produces a by-product named digestate. It contains many nutrients and other compounds that can cause undesired environmental impacts when discarded directly into the environment (Chen *et al.*, 2018). In this context, several alternatives have been investigated to value this nutrient-rich by-product. Recently, an emerging possibility is to couple microalgae cultivation with anaerobic digestate treatment (Barreiro-Vescovo *et al.*, 2020; Chen *et al.*, 2018; Patel *et al.*, 2021). Thus, digestate as a culture medium for producing microalgae biomass is an alternative to replace the demands for drinking water and fertilizers of conventional microalgae cultivation, reducing costs and environmental impacts (Al-Mallahi & Ishii, 2022). Given this, algal phycoremediation is a sustainable and efficient alternative to treat anaerobic digestate and allows simultaneous nutrient recycling (Leong & Chang, 2022).

Additionally, microalgae biomass is rich in lipids and proteins. Therefore, it may have several applications, such as biofuels, biofertilizers, and value-added products, such as biopolymers and pigments (Calijuri *et al.*, 2022). It creates an opportunity to develop the biorefinery concept and circular economy (Chen *et al.*, 2018). Microalgae cultivation using digestate has been studied in recent research. Among them, there is the AD of food (Barzee *et al.*, 2022; Patel *et al.*, 2021), animal (Lu *et al.*, 2022), and urban solid (Barreiro-Vescovo *et al.*, 2020) waste, as well as the combination of different residues (Chen *et al.*, 2018; Seelam *et al.*, 2022) (Figure 8.2).

Due to the remarkable ability of microalgae to adapt to extreme conditions and the possibility of nutrient recovery, microalgae cultivation in anaerobic digestate is a promising strategy. The digestate is rich in bioactive substances, such as monosaccharides, free amino acids, nucleic acids, and fulvic acid, stimulating microalgae development and providing greater tolerance to abiotic and biotic stress (Chong *et al.*, 2022). In addition, the processes that rule AD mineralize P and N into PO_4^{3-} and NH_4^+ , respectively, which are the preferred forms assimilated by microalgae (Al-Mallahi & Ishii, 2022). Still, the volatile organic acids (VOA) in the anaerobic digestate are promising compounds for microalgae production (Patel *et al.*, 2021).

There are, however, some challenges in microalgae cultivation in anaerobic digestate, mainly concerning their physical and chemical characteristics. Excess of suspended solids, turbidity, NH_4^+ , and metals in the digestate limit microalgae growth. Beyond that, a disbalanced nutrient proportion and presence of other competing organisms are other factors that can limit microalgae growth (Al-Mallahi & Ishii, 2022; Praveen *et al.*, 2018). Marcilhac *et al.* (2014) investigated the effect of light intensity and digestate color on nutrient removal and concluded that the initial optical density is inversely proportional to productivity and N assimilation. According to the authors, this fact is due to reduced light penetration and, consequently, reduced photosynthetic efficiency. To solve the problem of limiting light use due to suspended solids, Chen *et al.* (2018) proposed a membrane photobioreactor with a 0.1 μm pore size that resulted in removal efficiencies of 43.9% of NH_4^+ and 64.9% of PO_4^{3-} .

High NH_4^+ concentrations can also limit microalgae development, despite being the preferred assimilation form. Free ammonia (NH_3) is a toxic N form (Jiang *et al.*, 2021) that easily penetrates the cell membrane and accumulates in the cytoplasm, impairing photosynthetic processes (Uggetti *et al.*, 2014). Praveen *et al.* (2018) evaluated the effects of high N concentrations on growth (NH_4^+ between 20 and 120 mg/L) and concluded that microalgae growth was inhibited at concentrations exceeding 100 mg/L. However, the values can vary from 100 to 1,600 mg/L, depending on the microalgae species used and the cultivation conditions (Al-Mallahi & Ishii, 2022).

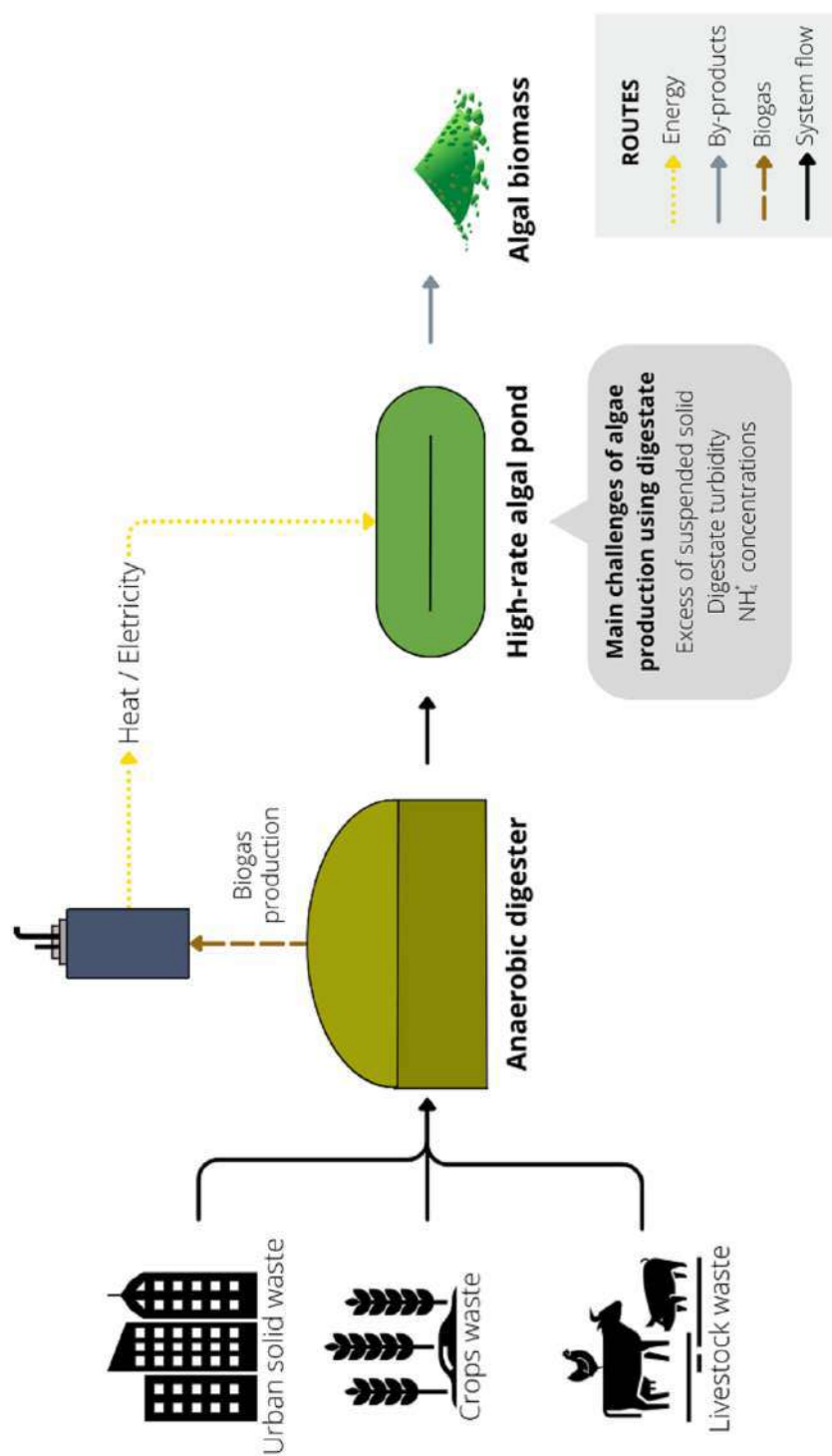


Figure 8.2 Schematic diagram of microalgae production using digestate from the anaerobic treatment of biomass.

Another limiting factor for using digestate as a culture medium is the nutrient proportion, given that an adequate C/N ratio is required for a synergistic microalgae and bacteria interaction (Fallahi *et al.*, 2021). Anaerobic digestate has a C/N ratio of 2:3 (Barzee *et al.*, 2022; Lu *et al.*, 2022). Therefore, it is considered a low ratio compared to the adequate C/N ratio for microalgae cultivation, which ranges between 6 and 8 (Dang *et al.*, 2022; Woertz *et al.*, 2009). The low C/N ratio of the anaerobic digestate is related to NH_4^+ accumulation and high pH value.

Some solutions can be implemented to improve the microalgae cultivation stage in anaerobic digestate, such as combining different raw materials during the AD stage (Chong *et al.*, 2022) or supplementing CO_2 from flue gases in the microalgae cultivation stage (Assis *et al.*, 2019). In addition, digestate pretreatment can be carried out. Pretreatment aims to facilitate the breakdown of complex organic compounds, which may reduce the suspended solids concentration, and mitigate possible toxicities due to high organic and inorganic matter concentrations, consequently reducing turbidity and promoting digestate sanitization (Chong *et al.*, 2022). For example, Praveen *et al.* (2018) investigated the microalgae–bacterial process performance through two stages: (1) digestate dilution with municipal wastewater, followed by (2) pretreatment in activated sludge, achieving COD, nitrate (NO_3^-), NH_4^+ and PO_4^{3-} removal efficiencies of 87%, 100%, 30% and 77%, respectively.

8.3 AD AS ENERGETIC VALORIZATION ROUTE OF ALGAE BIOMASS

8.3.1 AD of microalgae

Biogas production via AD of microalgae biomass obtained in wastewater treatment has been an energy recovery alternative since decades, with renewed research attention in recent years (Choudhary *et al.*, 2020). AD produces biogas in which CH_4 represents 55–70% of the composition, responsible for the process's energy potential due to its calorific value (37.27 MJ/m³) (Ganesh Saratale *et al.*, 2018). Some biogas valorization routes are generating heat or electricity, liquefaction into methanol, compression into fuel for automobiles, and fuel gas (Zabed *et al.*, 2020).

Increasing AD performance, with greater methane (CH_4) production, depends on several factors, for example operational and environmental conditions, as well as substrate composition. The organic loading rate is a key factor for the AD efficiency among the operational factors. It prevents VOA accumulation or shortage, which influences the reactor pH (ideal range 6.6–7.4), a critical factor for balancing acidogenic and methanogenic processes. A volumetric organic load of 1.6–4.8 kg VS/m³ is usually recommended for obtaining a high AD rate (Zabed *et al.*, 2020). Still, it may vary depending on the biomass and reactor types, the biomass biochemical composition, and the anaerobic microbial population. Other parameters are reflected in the production of VOA, including the C/N ratio, which, when low, causes high NH_4^+ concentrations in the digester, disturbing the microbial metabolism and consequently accumulating VOA. C/N ratios between 20 and 30 are considered adequate, and a C/N ratio equal to 25 usually gives better CH_4 yields (Zabed *et al.*, 2020).

Another variable that influences the AD quality is biomass moisture. A high solids content in the reactor decreases the available water, affecting the alkalinity availability, free NH_4^+ , and VOA concentrations (Zabed *et al.*, 2020). Higher rates of CH_4 production have been reported at 60–80% moisture (Kwietniewska & Tys, 2014). Also, the reactor hydraulic retention time (HRT) should not be too short (~16 days). This HRT value avoids washing out methanogenic archaea, causing a low CH_4 bioconversion efficiency. Contrary, it should also not be overly long (>50 days) to prevent the depletion of substrates and nutrients. For pilot and commercial plants, the optimal HRT varies between 30 and 50 days, whereas on a laboratory scale, it ranges between 15 and 30 days (Zabed *et al.*, 2020).

Regarding operational conditions, low temperatures generate high accumulation of VOA, which is reflected in the pH, affecting the methanogenic archaea metabolism. In contrast, elevated temperatures increase NH_4^+ toxicity in addition to foaming and odor formation. The operating range of mesophilic AD is 30–40°C and for thermophilic it is 50–60°C, with 35°C and 55°C being the ideal temperature, respectively (Zabed *et al.*, 2020). Figure 8.3 presents a schematic design of an anaerobic reactor,

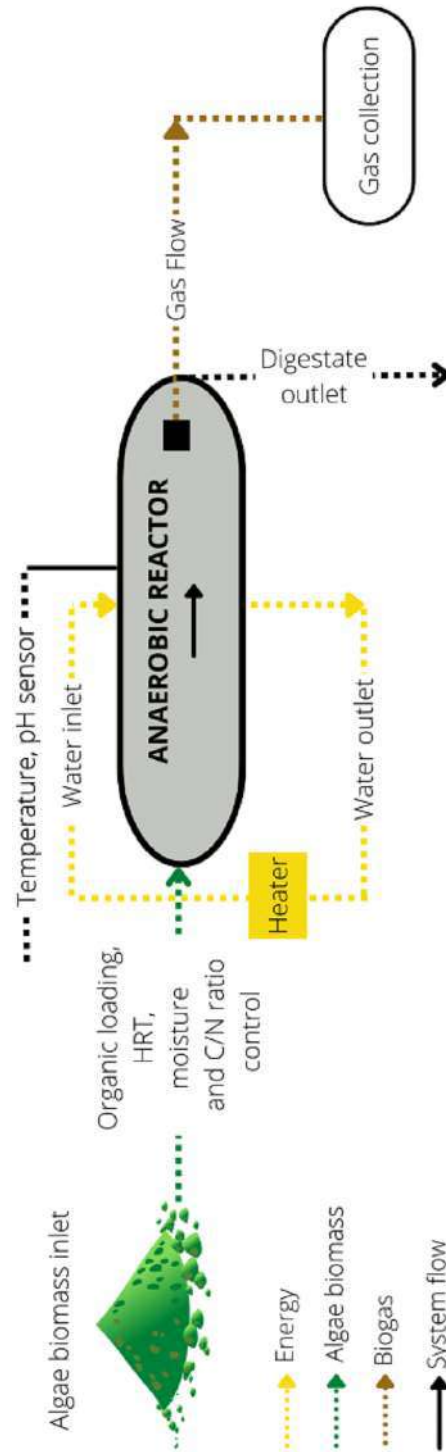


Figure 8.3 Schematic diagram of an anaerobic reactor with indications of the primary control parameters for enhancing biogas productivity of microalgae AD. HRT: hydraulic retention time; C/N ratio: carbon/nitrogen ratio.

highlighting the primary control parameters that can be utilized to enhance the biogas productivity of microalgae AD.

AD can be performed in different reactor types: fixed dome, floating drum, plastic, or textile reactors (Zabed *et al.*, 2020). As a rule, the reactor must create an oxygen-free environment. Furthermore, it must be protected against water, gas, and light leaks. In addition, it must contain protection mechanisms against corrosive chemicals and gases and avoid adverse weather conditions (Zabed *et al.*, 2020). The responses as a function of all these factors and parameters determine the CH₄ yield of algal biomass subjected to AD. Concerning the microalgae biomass produced in wastewater, several types of culture media and species have already been studied at bench scale (Choudhary *et al.*, 2020).

The main limitations of microalgae AD are: (1) low biomass biodegradability due to the microalgae cell walls resistance, causing low CH₄ potential (degradation extent) and low conversion rate (degradation speed); and (2) NH₄⁺ inhibition risk due to the biomass low C/N ratio (as mentioned in Section 8.2.2). Moreover, NH₄⁺ inhibition limits the maximum rate of organic discharge from the digesters and requires a longer HRT; therefore, a digester with a larger volume is required (Karupiah & Ebenezer Azariah, 2019; Solé-Bundó *et al.*, 2019b). The respective solutions to these issues are: (a) pretreatments to disturb the microalgae cell wall and make its intracellular content more available (de Oliveira *et al.*, 2022; Yakesh Kannah *et al.*, 2021); and (b) anaerobic co-digestion (AcoD) to increase alkalinity, provide a balanced macro and micronutrients composition, stabilizing the process at high organic loading rates and increasing CH₄ yields (de la Lama-Calvente *et al.*, 2022; Veerabadrhan *et al.*, 2021). Table 8.2 presents the CH₄ yield of microalgae AD, AcoD, and biomass pretreatments.

8.3.2 Pretreatment of microalgal biomass

Pretreatment methods can be divided into two groups: (a) energy-intensive, which are mechanical/physical (ultrasound, microwave, and milling), thermal or hydrothermal, and (b) energy-efficient, which are biological, enzymatic, chemical (acidic or alkaline) or combined (thermochemical) (de Oliveira *et al.*, 2022; Yakesh Kannah *et al.*, 2021). Therefore, strategies must be adopted for energy-intensive treatments to achieve better energy performance. Xiao *et al.* (2019a) proposed a hydrothermal pretreatment system for algal biomass using solar energy and obtained 348 mL CH₄/g VS. The CH₄ production was 57% higher than without pretreatment (221.70 mL CH₄/g VS). Biogas production with solar-powered hydrothermal pretreatment achieved a maximum exergy efficiency (40.85%) (Xiao *et al.*, 2019b). Biogas production with hydrothermal pretreatment with solar energy achieved a net energy ratio of 0.69, with emissions of −166.13 g CO₂_{eq} per kWh. Also, it achieved a leveled cost of 0.17 USD/m³, representing a better performance than biogas without solar energy pretreatment (Xiao *et al.*, 2020). With a thermo-acid pretreatment, Barros *et al.* (2022) estimated that biogas production with microalgae biomass AD produced at the tertiary level would result in an energy surplus of 2.8% in the WWTP. Fu *et al.* (2023) estimated that with thermo-alkaline pretreatment, also for energy efficiency, there would be an energy surplus in the system, increasing the CH₄ production.

8.3.3 Anaerobic co-digestion

In WWTPs, an AcoD option exists between sludge and microalgae or microalgae with other residues. Damtie *et al.* (2020) obtained a 36% increase in CH₄ production when studying AcoD from biologically pre-treated algal biomass and primary sludge. Solé-Bundó *et al.* (2019b) obtained a 65% increase in CH₄ production in the AcoD of microalgae with primary sludge and a generation of 4.5 times the energy consumed, whereas the microalgae mono-digestion AD generated 2.7 times the energy consumed. Solé-Bundó *et al.* (2019a) achieved a 60% increase in CH₄ yield applying a thermal pretreatment in the microalgae biomass and 15% after AcoD with WWTP residues (oil, grease, and fat). Zhang *et al.* (2020) added glycerol to the AcoD of microalgae and potato processing residues and found an increase of more than 50% in CH₄ production. Assemany *et al.* (2020) performed the

Table 8.2 AD performance when treating different types of microalgae biomass feedstocks.

CH ₄ Yield	Microalgae Species	Microalgae Culture Medium	AD Feedstock	Reference
348 L CH ₄ per kg VS	<i>Chlorella pyrenoidosa</i>	Freshwater	Solar-driven hydrothermal pre-treated microalgae biomass	Xiao <i>et al.</i> (2019a)
221.70 L CH ₄ per kg VS			Raw microalgae biomass mono-digestion	
382 mL CH ₄ per g VS	<i>Chlorella</i> and <i>Desmodesmus</i>	Freshwater	Co-digestion of microalgae biomass and rice straw	Srivastava <i>et al.</i> (2022)
252 mL/g VS	Microalgal consortium	Wastewater	Thermo-acid hydrolysis pre-treated microalgae biomass	Barros <i>et al.</i> (2022)
308 mL/g VS	<i>Chlorella vulgaris</i>	Freshwater	Aerobic digestion as a pre-treatment for co-digestion of microalgae biomass and sludge	Damtie <i>et al.</i> (2020)
188 mL/g VS			Raw microalgae biomass mono-digestion	
0.33 m ³ CH ₄ per kg VS	Microalgal consortium	Wastewater	Co-digestion of microalgae biomass and primary sludge	Solé-Bundó <i>et al.</i> (2019a)
0.20 m ³ CH ₄ per kg VS			Raw microalgae biomass mono-digestion	
0.73 (±0.07) L CH ₄ per g COD	<i>Chlorella vulgaris</i>	Freshwater	Raw microalgae biomass co-digested with 1% v/v glycerol	Zhang <i>et al.</i> (2020)
0.30 (±0.04) L CH ₄ per g COD			Raw microalgae biomass mono-digestion	
207.35 mL CH ₄ per g VS	<i>Chlorella</i> sp.	Freshwater	Co-digestion of algal residues after lipid extraction with mixed enzymes pretreatment and energy grass	Zhang <i>et al.</i> (2018)
128.75 mL CH ₄ per g VS			Algal residue after lipid extraction from untreated microalgae mono-digestion	
128.80 mL CH ₄ per g VS			Algal residue after lipid extraction from mixed enzymes pretreatment microalgae mono-digestion	
368.94 mL CH ₄ per g VS	<i>Chlorella pyrenoidosa</i>	Wastewater	Co-digestion ratio of 1:2 of thermo-alkaline pretreated microalgae biomass and secondary sludge	Fu <i>et al.</i> (2023)
328.43 mL CH ₄ per g VS			Co-digestion ratio of 1:1 of thermo-alkaline pretreated microalgae biomass and secondary sludge	
293.39 mL CH ₄ per g VS			Co-digestion ratio of 2:1 of thermo-alkaline pretreated microalgae biomass and secondary sludge	
0.10 m ³ CH ₄ per kg VS	Microalgal consortium	Wastewater	Co-digestion of municipal wastewater growth microalgae biomass and 10% olive mill wastewater	Assemany <i>et al.</i> (2020)
0.062 m ³ CH ₄ per kg VS			Mono-digestion of municipal wastewater growth microalgae biomass	
0.13 m ³ CH ₄ per kg VS			Co-digestion of brewing industry wastewater growth microalgae biomass and 10% olive mill wastewater	
0.16 m ³ CH ₄ per kg VS			Mono-digestion of brewing industry wastewater growth microalgae biomass	

Note: AD: anaerobic digestion; VS: volatile solids; COD: chemical oxygen demand; v/v: volume/volume.

AcoD of microalgae biomass grown in brewing industry wastewater and olive mill wastewater. They obtained 61% more CH_4 compared to the microalgae biomass mono-digestion. As presented in Section 8.4, in WWTPs, microalgae can also capture CO_2 from biogas (Nguyen *et al.*, 2021), a form of biogas purification (Miyawaki *et al.*, 2021).

The generation of multiple by-products can be an approach to make microalgae biogas even more attractive. For example, Zhang *et al.* (2018), through the lipid extraction of *Chlorella* sp. with pretreatment by mixed enzymes (cellulase, xylanase, and pectinase), achieved 169% more energy in the combined biodiesel and CH_4 production than with biodiesel alone or with AcoD of the residual biomass and C-rich material (energy grasses). Srivastava *et al.* (2022) extracted lipids for biodiesel production and performed the AcoD of the remaining biomass with rice straw. They obtained a 382 mL CH_4 /g VS yield, almost 50% higher than the control. Another opportunity is co-production of CH_4 and hydrogen gas (H_2) through AD in two stages. The acidogenic and methanogenic processes are carried out separately, allowing recovery of the H_2 generated in the first phase (Zabed *et al.*, 2020).

8.4 ALGAE CULTIVATION FOR BIOGAS UPGRADING

The biogas composition should be at least 95% CH_4 before feeding into the natural gas grid (Khan *et al.*, 2021). However, biogas is usually composed of 45–70% CH_4 , 20–55% CO_2 , and other gases, namely, nitrogen gas (N_2) (0–3%), O_2 (0–1%), water vapor (1–10%), H_2S (0–10,000 ppm), NH_3 (0–100 ppm), and traces of hydrocarbons, siloxanes, and chlorine (Bose *et al.*, 2019). These gases, except for the CH_4 , decrease the heating value of the biogas or can produce environmental pollutants (Angelidaki *et al.*, 2018). Beyond that, they can corrode metal components of boilers, internal combustion engines, and gas pipelines (Khan *et al.*, 2021). Thus, the biogas can be cleaned by removing these compounds, and the heating value can be increased through a process named ‘biogas upgrading’.

Many conventional biogas upgrading technologies can be used, such as pressure swing adsorption, chemical scrubbing, water scrubbing, organic solvent scrubbing, and membrane separation (Nguyen *et al.*, 2021). However, emerging biogas upgrading systems are being investigated as economic and environmental alternatives, such as adsorption by biochar, cryogenic upgrading, and biological upgrading. Among the biological upgrading systems, microalgae have attracted research interest (Miyawaki *et al.*, 2021; Thi Nguyen *et al.*, 2019; Toro-Huertas *et al.*, 2019; Xie *et al.*, 2023). It is noteworthy that selecting the appropriate technology for upgrading raw biogas depends on its final use, the economics involved, and the efficiency of the upgrading process (Khan *et al.*, 2021).

When using microalgae for biogas upgrading, CO_2 can be assimilated as a C source to produce chemical energy through photosynthesis (Thi Nguyen *et al.*, 2019). Microalgae can remove CO_2 from biogas using open or closed systems (Figure 8.4). The most common open system is the HRAP, also named raceway pond. It can be interconnected to an absorption bubble column (ABC). An ABC is fed with raw biogas, and the liquid containing microalgae produced in the HRAP is recirculated, allowing microalgae to capture the CO_2 from the biogas (Toro-Huertas *et al.*, 2019). Zabed *et al.* (2020) stated that although cultivation in open systems is techno-economically more convenient than in closed systems, open systems pose a higher risk of contamination with relatively lower biogas purification. The main drawback of the closed system is the higher energy requirements for light penetration and high capital costs. The potential CH_4 recovery by the photoautotrophic biogas upgrading process is 97%, with H_2S removal achieved simultaneously (Khan *et al.*, 2021).

Khan *et al.* (2021) reported that CO_2 solubility, mass transfer to microalgae, difficulty in biogas harvesting, and CH_4 solubility in microalgae media are the main challenges of open or closed systems. These limitations can be overcome by using indirect biogas upgrading systems (Figure 8.4). As Nguyen *et al.* (2021) stated, indirect methods can overcome the limitations of direct biogas upgrading. In this approach, CO_2 can be captured in a carbonate solution such as potassium carbonate (K_2CO_3).

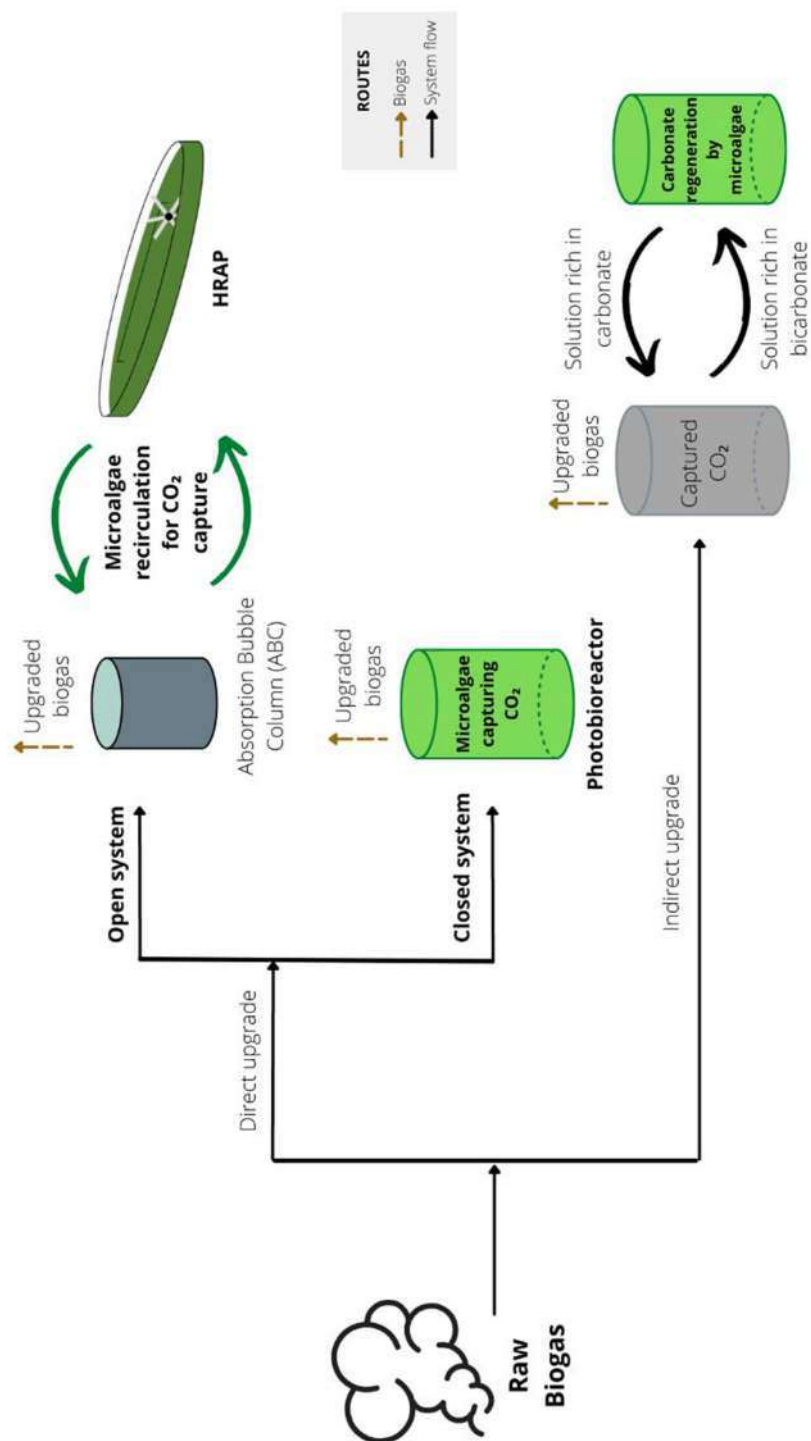


Figure 8.4 Biogas upgrading routes through algal technology. HRAP: high-rate algal ponds.

Notably, these systems are limited to only specific microalgal species that can tolerate an environment with strong ion concentrations and high alkali levels.

In a recent study removing H_2S and NH_4^+ by microalgae was also approached (Xie *et al.*, 2023). The authors could upgrade biogas while recovering N and P using microalgae treatment. They used *Chlorella vulgaris* in closed photobioreactors containing synthetic swine manure digestate. Different biogas-to-digestate liquid feed ratios were investigated to achieve a ratio that would maximize both CH_4 production and nutrient recovery. The authors achieved simultaneous biogas decarbonization and desulfurization with a 1:1 to 40:1 biogas-to-digestate ratio range, and nearly all CH_4 remained in the upgraded biogas. NH_4^+ was removed at higher biogas-to-digestate ratios. This finding demonstrates that the proposed system is suitable for treating high H_2S and NH_4^+ concentrations, both common contaminants from biomass processing units.

8.5 COUPLING TECHNOLOGIES FOR SUSTAINABLE BIOREFINERIES

Sustainable biomass conversion into a wide range of biobased products (food, feed, chemicals, and materials) and bioenergy (biofuels, power, and heat) is known as biorefinery (de Jong *et al.*, 2012). A biorefinery is usually associated with products with high environmental, social, and economic sustainability levels. According to Trivedi *et al.* (2015), a new biorefinery approach must integrate industry and the environment, improving resource use and minimizing the ecological footprint of the entire system.

Biorefineries integrate different processes into the same installation (physical, chemical, thermochemical, or biotechnological) to obtain a wide product range. There are many biomass types and possible combinations between platforms and end products, with the flexibility of a biorefinery being a key feature in incorporating new processes into existing facilities (Pascual *et al.*, 2015). As mentioned in Sections 8.2, 8.3, and 8.4, several studies have proposed coupling microalgae biotechnology with AD in many different ways. A biorefinery could be established if those multiple pathways are integrated (Figure 8.5). However, environmental, energy, and economic sustainability aspects must be better understood, requiring more effort in future research.

8.5.1 Biorefinery based on integrated microalgae and AD technologies

As stated in Section 8.2, UASB reactors are commonly used during wastewater treatment and can be integrated with microalgae cultivation for domestic sewage polishing. In this scenario, the microalgae biomass produced in the HRAP can be used as an anaerobic substrate in the UASB reactor, consisting of an AcoD between domestic wastewater and algal biomass. For a population of 20,000 inhabitants, this configuration provided an energy surplus between 0.15 and 0.32 KWh/ m^3 , and revenue between 10,321.89 and 21,822.60 USD/year, indicating the UASB reactor energy sustainability associated with HRAP (Gonçalves *et al.*, 2020). In addition, the energy production ranged from 70 to 180% more than consumed and could be applied in the WWTP or the neighboring community (Vassalle *et al.*, 2020b).

In the agroindustrial context, wastewater treatment based on microalgae tertiary treatment can also be interconnected to an anaerobic digester for bioenergy and biofertilizer production from sludge and microalgae AcoD. Avila *et al.* (2022) evaluating a circular bioeconomy model for nutrient and energy recovery from winery wastewater, highlighted the secondary sludge and algal biomass AcoD as a strategy to increase CH_4 yield and the importance of using other bioproducts from this route to reduce fertilizer costs. A virtual model proposed by Siqueira *et al.* (2022) to anaerobically treat vinasse integrated with microalgae biotechnology presented an electricity surplus of +14.49 MJ/ m^3 of vinasse and a positive net energy ratio equal to 2, establishing a better integration of WWTPs and biorefineries.

8.5.2 Environmental impacts of integrated microalgae and AD technologies

Regarding environmental sustainability, the life-cycle assessment (LCA) is a powerful tool for measuring new processes, technology, or product impacts on the environment (Marangon *et al.*,

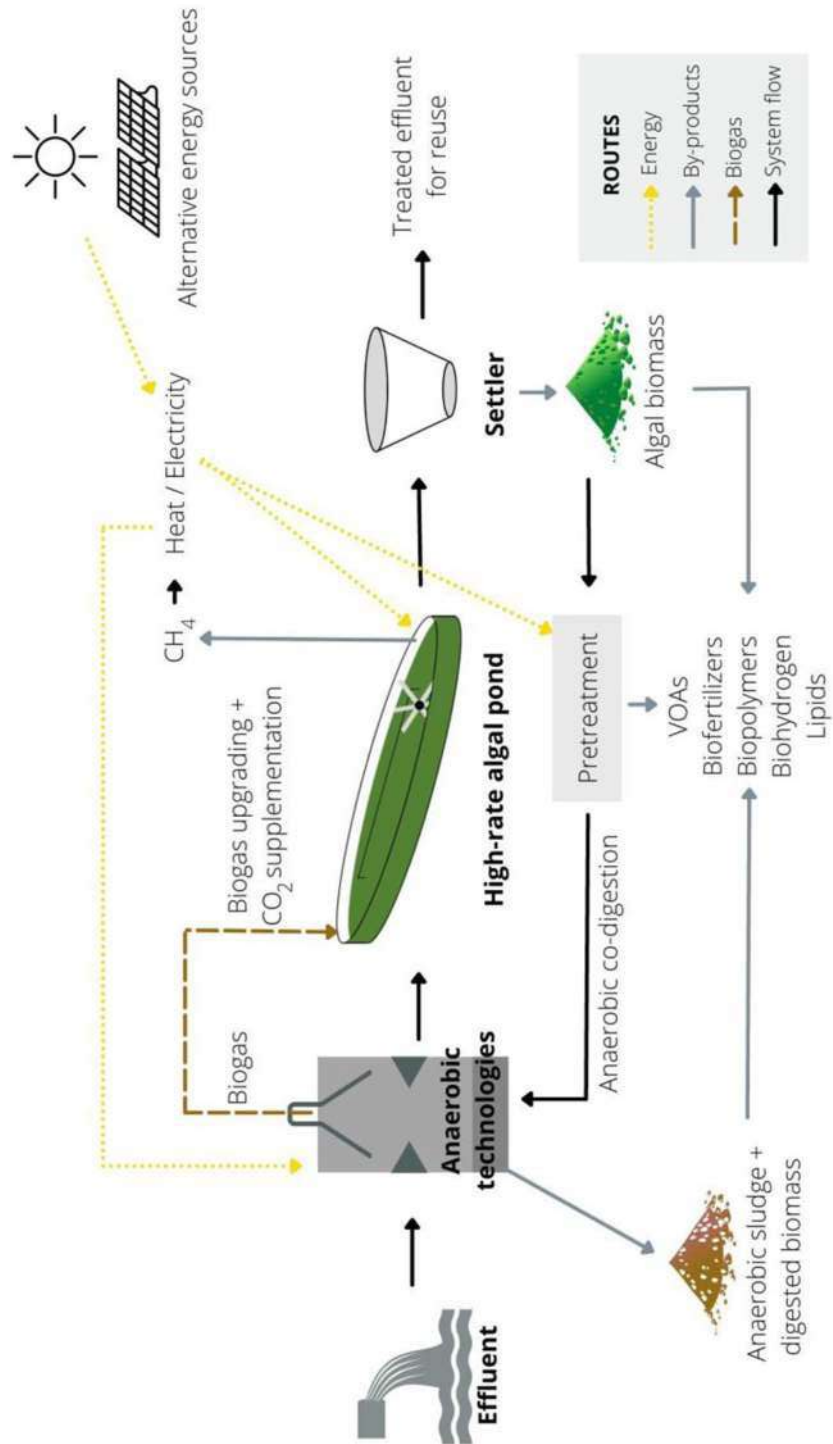


Figure 8.5 Sustainable biorefinery proposal by integrating microalgae and AD biotechnologies. VOA: volatile organic acids.

2022). Arashiro *et al.* (2018) performed an LCA of different systems, including a scenario with a HRAP followed by an energy recovery system through AcoD of the algal biomass with primary sludge. The authors emphasized the importance of AcoD in electric energy generation and the positive impact of this system on the environment. However, considering the eutrophication potential, it was the biggest polluter, mainly due to the lower nutrient removal efficiency. The lower energy consumption compared to the activated sludge scenario presented lower environmental impacts in climate change, ozone depletion, fresh and marine water eutrophication, photochemical oxidants formation, and fossil fuels depletion. However, the HRAP construction stage, demanding more material, harmed abiotic resource depletion. Also, the HRAP was responsible for greater environmental damage due to atmospheric emissions of nitrogenous compounds. Comparing all scenarios, the authors concluded that biomass valorization as biofertilizer instead of energy valorization via AD was the most economical alternative. Although presenting the most expensive operation, this option had a lower area requirement and a greater commercialization potential.

Arashiro *et al.* (2022) analyzed the integration of microalgae and AD in two different scenarios by comparing microalgae-based systems for wastewater treatment and bioproducts generation. The first was a HRAP followed by a closed photobioreactor treating domestic wastewater with biogas recovery after microalgae biomass AD. The second was characterized by an UASB reactor followed by a HRAP treating food industry wastewater with biogas recovery from the UASB reactor. Both scenarios included the recovery of other bioproducts, that is natural pigments and biofertilizer. The authors concluded that the wastewater type was the most decisive parameter in the LCA, as the second scenario presented lower environmental impacts in 8 of 10 categories. Compared to domestic wastewater, industrial wastewater resulted in lower air emissions due to lower NH_4^+ concentrations, higher biogas production, and lower heavy metal concentrations in the digestate.

Tua *et al.* (2021) investigated improvements in an existing municipal WWTP integrating a microalgal cultivation unit with the AD of the produced biomass. Microalgae were cultivated in the centrate from sludge dewatering and with CO_2 supplementation from flue gas of the combined heat and power unit. The biomass was separated in a settler and sent to AD for extra biogas production. Among the environmental indicators, the proposed system improved 7 of 15 indicators, mainly due to the electric energy generation. However, the system had negative environmental impacts, mainly due to nitrogenous compounds released into the environment, impacting particulate matter, terrestrial and marine acidification, and eutrophication categories. Another effect of the new proposed system was related to human toxicity, linked to residual biomass that can generate an environmental burden after co-incineration and subsequent disposal in landfills (carcinogenic toxicity). The non-carcinogenic toxicity was linked to the zinc (Zn) contribution to soil pollution when using biomass for agricultural purposes.

Alternatively, microalgae biotechnology can be used within the source-separated nutrient approach. Li *et al.* (2022) proposed a scenario that municipal wastewater and human urine were placed in different modules for microalgal cultivation coupled to struvite and biofuel production (heat, electricity, bio-oil, biogas, and biochar). The authors concluded that separating nutrients by urine precipitation was essential for the system's environmental sustainability, regardless of cost.

8.5.3 Insights for improving the sustainability performance of integrated microalgae and AD technologies

After considering different proposals from the literature, the main sustainability aspects of integrated microalgae and AD technologies can be highlighted (Figure 8.5). The beneficial use of digestate and residual biomass after AD as a nutrient source in microalgae cultivation and a valuable by-product is essential (Bussa *et al.*, 2020). Otherwise, it will be considered an emission to the environment, causing pollution. In that way, when valorizing the digestate as a biofertilizer, heavy metal recovery before soil application should be considered (Arashiro *et al.*, 2022). Emissions during microalgae cultivation are another point of interest, especially N emissions. Thus, pH control in the HRAP and CO_2 supplementation may be an alternative to minimize the negative impacts of ammonia volatilization (Tua *et al.*, 2021). The

CO₂ source for C supplementation during microalgae growth is also critical (Bussa *et al.*, 2020). In that way, the biogas upgrading through CO₂ bio-assimilation in the cultivation reactor may represent an environmental and economic benefit (see Section 8.4). Biogas production improvement is highly appreciated to increase the system energy yield. Besides AcoD, biomass pretreatment (see Section 8.3) before AD can be a good option to improve CH₄ production and the system's energy feasibility (Xiao *et al.*, 2020). In addition, renewable energy sources integration, such as solar energy, should be considered to reduce impacts related to electricity consumption (Arashiro *et al.*, 2022).

Lastly, AD will become an important technology for future biorefinery development. The process is already used as an auxiliary technology to recover waste streams. However, its use as a leading technology should be promoted. The challenge is to rethink existing biogas plants and expand their range of final products, going much further than selling electricity (Pascual *et al.*, 2015). For example, other bioproducts can be obtained: (1) biopolymers, bioalcohol, and medium-chain fatty acids through the VOAs platform and (2) biofertilizers, such as struvite and NH₄⁺ salts via the digestate platform.

Rajendran and Murthy (2019) stated that acquisition of raw materials, plant operation aspects, and modernization costs are the major uncertainties regarding LCA and economic assessment for biogas production. Also, operational capacity and energy efficiency are the ones that most impact the system's economic performance (Aui *et al.*, 2019). Thus, proposals for biorefineries and technology integration and their sustainability will vary depending on local characteristics. Regional specificities must be taken into account to propose routes that are more favorable within each context. It is highly appreciated that regional aptitude (mainly in economic terms) is explored, considering the market cost, public acceptance, and by-products applicability, minimizing transport and logistic costs. So, there will not be an optimal biorefinery system applicable to any case, and the various local factors involved should be considered. For example, Bussa *et al.* (2020) concluded the high potential of integrated microalgal cultivation with AD in rural regions with cattle farming and in areas with a higher degree of urbanization where large municipal WWTPs were in operation. By doing a geospatial analysis, the authors stated that low potential areas require larger transportation distances for substrates or digestates, reducing the environmental benefits while increasing the economic burden.

8.6 CHALLENGES AND FUTURE PERSPECTIVES

The management of anaerobic digestate and microalgae cultivation is a sustainable strategy from environmental and economic points of view. In this way, proper treatment is provided for this nutrient-rich by-product with a high organic load, and, at the end of the treatment, a microalgae biomass is obtained with several applications that can guarantee the overall viability of the process. Coupling these two technologies on a large scale is a possibility that has already been studied. However, it needs further research to solve some limitations due to the presence of high suspended solids concentrations and ammonia toxicity, among other factors. The challenge, especially for high-strength wastewater treatment, is the need to dilute the anaerobic effluent so that the microalgae can withstand the organic load. Considering full-scale wastewater treatment, this would be disadvantageous due to the consumption of water and inputs and the need for larger units to hold the diluted effluent. Thus, future research can focus on strategies to overcome this bottleneck, for example, by combining two complementary effluents. In addition, there are still opportunities to evaluate the performance of microalgae technologies to treat micropollutants recently attracting attention, such as pharmaceuticals, endocrine disruptors and microplastics.

Regarding AD's technical limitations, pretreatment methods are recommended in further studies, especially those classified as energy efficient or associated with renewable energy sources. In addition, the co-production of CH₄, H₂, and biodiesel, together with other valuable by-products and the AcoD of microalgae and other biomass types deserve continuous efforts. Furthermore, these techniques can be performed concurrently and applied to WWTPs, making the microalgae AD energy recovery more attractive. Despite all the technological advances, upgrading biogas through microalgae needs further

research studies to make it feasible on a larger scale. Finally, coupling AD and microalgae technologies could be an affordable way to encourage a biobased green circular economy model, able to improve microalgae biomass production, nutrient recovery, wastewater treatment, and biogas upgrading.

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Chapter 9

Algae for wastewater treatment and biofuel production

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ABSTRACT

Biofuels can be used for the provision of electricity, heating, and transport. Interest in biofuels has been sparked by their suitability to decrease carbon emissions and fossil fuel dependency without major modifications to our existing energy infrastructure. Microalgae grown in wastewater are a suitable feedstock to produce two of the most utilized types of biofuels: bioethanol and biodiesel. Biodiesel is obtained as fatty acid methyl esters from microalgae via a chemical reaction known as transesterification. Bioethanol is produced from biomass by microbial fermentation. So far, microalgae growing in wastewater has been characterized for containing a mixture of lipids, carbohydrates, and proteins. Hence, another area of interest is the use of wastewater-derived microalgae for the sequential production of bioethanol, biodiesel, and protein compounds. A biorefinery concept emerges for the generation of multiple co-products from the wastewater-derived microalgae that can maximize the use of unit operations and the valorization of microalgal biomass. In this chapter, concepts for biodiesel and bioethanol production are evaluated and a biocircular economy prospected.

Keywords: bioethanol, biodiesel, sewage, pre-treatment, fermentation, transesterification, microalgae

9.1 INTRODUCTION

Microalgae for wastewater treatment have been studied for more than 70 years. Initial observations of microalgae species were in facultative lagoons. Even if microalgae were able to subsist in lagoon treatment systems, cultivations presented challenges such as: the requirement of a large surface area (Lavoie & de la Noüe, 1985); the needed maintenance of introduced microalgae, as per replacement and succession of species (Gantar *et al.*, 1991); and the difficulty to harvest diluted algal biomass (Tredici *et al.*, 1992). Current research in microalgal growth in wastewater aims to address these challenges, and solutions have gradually been elucidated such as the growth of microalgae using high-rate algal ponds or bioreactors. The main advantage of using microalgae emerges as currently nitrogen and phosphorus are not completely removed in wastewater treatment plants and microalgae are species that can remove them once most of the carbon has been depleted. Additionally, microalgae can

co-exist with other microorganisms such as bacteria and yeast allowing the treatment of wastewater with high carbon loads (Hernández-García *et al.*, 2019).

Microalgae can assimilate inorganic and organic compounds from wastewater and at the same time accumulate biomolecules of interest, for example lipids, carbohydrates, or proteins, under unfavorable environmental conditions. The generated algal biomass can be used to produce different bioproduct such as biofertilizers, biohydrogen, biodiesel, bioethanol, bio-oil; or biomaterials like biofertilizers, biopolymers or biofilms (Jebali *et al.*, 2018; Salama *et al.*, 2017). In addition, biomass can be thermally processed to produce heat and electricity (Gouveia *et al.*, 2016; Romero Villegas *et al.*, 2017). For this reason, microalgae are considered a biomass source with good bioenergetic potential.

In the bioenergy context, microalgae are classified as a third-generation feedstock offering several advantages with respect to first- and second-generation terrestrial plants. Microalgae characteristics supersede terrestrial plants, as per the high growth rate and productivity; dual photosynthetic or heterotrophic growth; high microbial carbon dioxide fixation under autotrophic growth, no dependence on fertile soil, and no compromise in food production for human consumption. Microalgae growth in wastewater fits well with its later use in bioenergy products such as biodiesel and bioethanol. The quality of the produced biodiesel and bioethanol using microalgae grown in wastewater is comparable to other biomass feedstocks or microalgae grown in synthetic medium. Additionally, the risks of using biodiesel and bioethanol are lower than their fossil fuel-derived competitors.

This chapter provides an insight in the development of wastewater-grown microalgae for biodiesel and bioethanol production. Our laboratory has been researching the optimization of microalgae growth and their transformation to energy products through process understanding, integration and intensification (Velasquez-Orta *et al.*, 2022). The conversion of microalgae into biofuels requires the selection of operations that are economically viable and environmentally friendly. Biodiesel and bioethanol are the primary biofuel products globally produced. Previous communications have reviewed the conversion of microalgae to biodiesel or bioethanol, but few take as a basis the growth of microalgae in wastewater. Processing will play a significant role in the economic feasibility of biofuel production given their low-cost commodity.

9.2 CHARACTERIZATION OF MICROALGAE GROWN IN WASTEWATER FOR BIOFUEL PRODUCTION

Microalgae growth using wastewater has gained vast attention in the last three decades. According to Science Direct publication numbers, there were six times more publications in 2022, than a decade ago. Most studies utilize a consortium of microalgae for inoculation in wastewater. Commonly inoculated strains are *Chlorella* sp., cyanobacteria, *Desmodesmus* sp. or *Scenedesmus* sp. as they have positively prevailed in wastewater. Given the non-sterile nature of wastewater, cultivations end-up being a mixture of microbial strains where usually a desired microalgae dominates a consortium. As a result, the biochemical composition of microalgal biomass segregates into different fractions rather than a high fraction of a specific compound. The fractions can be generally divided into proteins, carbohydrates, and lipids. Carbohydrates are of interest for bioethanol production, whilst neutral lipids can be transformed into biodiesel. Velasquez-Orta *et al.* (2014) reported a total of 0.3 mg lipids/mg of biomass from mixed microalgae cultures growing in a wastewater treatment lagoon. Oliveira *et al.* (2018) inoculated *Scenedesmus* sp. in wastewater, after 16 days of growth, obtaining a biomass composition of 0.2–0.3 mg lipids/mg of dry biomass, 0.2–0.3 mg carbohydrates/mg of dry biomass and 0.4 mg protein/mg of dry biomass. Hernández-García *et al.* (2019) indicated 0.4 mg lipids/mg of biomass and 0.5 mg carbohydrates/mg of biomass after microalgal cultivation under nutrient limitation conditions. As can be seen, the composition of microalgae cultivations in wastewater varies and should be monitored in

wastewater treatment systems. Algae cultivations produce usable fractions for both bioethanol and biodiesel production.

9.3 BIODIESEL PRODUCTION FROM MICROALGAE GROWN IN WASTEWATER

9.3.1 Biodiesel production process

Biodiesel is defined as a mixture that contains at least 96.5%, by weight, of fatty acid methyl esters (FAME), in accordance with the EN 14214:2003 standard. FAME is derived from the conversion of the neutral lipid fraction of microalgae, known as triglycerides (TAG). The transesterification of microalgal lipids involves a chemical reaction that converts the extracted lipids into FAME and glycerol as shown in Figure 9.1. Apart from the chemical reaction, a series of unit operations are required prior and after, to obtain biodiesel from microalgae. The starting stages involve microalgae cultivation, harvesting, cell disruption and drying. Figure 9.2 provides an overview of the refining stages needed after the transesterification reaction to ensure that the FAME mixture is classified as biodiesel. Figures 9.1 and 9.2 show the production of glycerol as a by-product from the transesterification reaction. Glycerol production amounts to around 15% of the total volume and can be refined to chemical, edible or cosmetic applications. Glycerol biodegradability enables its application as feedstock for biological transformation. Elahinik *et al.* (2022) proposed the use of glycerol effluents emanating from biodiesel and epoxy resin industrial plants. The glycerol-rich wastewater was used to obtain propionate via aerobic granular sludge fermentation. Glycerol has also been converted to bioelectricity using stackable microbial fuel cells (Zhao *et al.*, 2017). Some considerations on the use of glycerol should be its price volatility as per its by-product nature and regulations on the trading of glycerol emanating from a waste fraction.

The global share of biodiesel production was 23% in 2009 and rose to 37% in 2020 (BP, 2021). Production of biodiesel from microalgae grown in wastewater has been shown to be feasible since our first publication (Komolafe *et al.*, 2014). Combining these two systems provides the benefit of wastewater bioremediation with fuel biorefining. Once microalgal biomass is available, the amount of unit operations needed for biodiesel production can be reduced through intensification (Velasquez-Orta *et al.*, 2022). For example, *in situ* transesterification can potentially combine the stages involving total drying, cell disruption, lipid extraction, and transesterification. Hence, *in situ* transesterification offers a one-step approach for cell disruption, lipid extraction, and conversion. The chemical transesterification reaction requires significantly higher amounts of alcohol ($\times 100$ times the usual

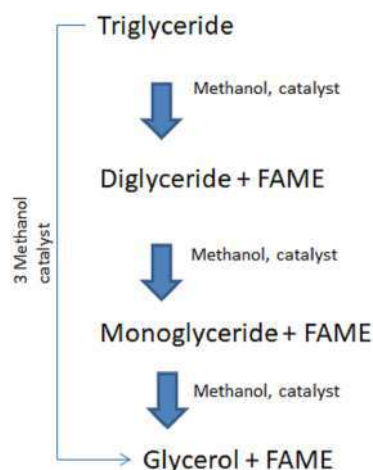


Figure 9.1 Overview of transesterification reaction to convert microalgal lipids into FAMES.

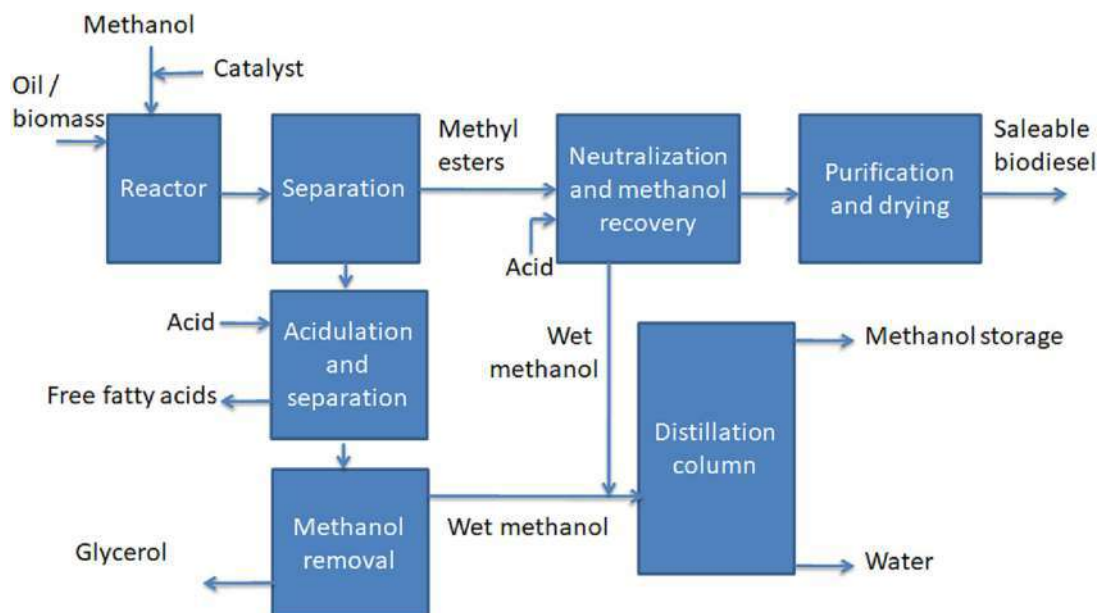


Figure 9.2 Overview of unit operations following the transesterification reaction for biodiesel production.

value), as per its dual act as both solvent lipid extraction and reactant. However, the alcohol can be recovered and reutilized. In contrast, the conventional route requires organic solvent mixtures for lipid extraction (e.g., chloroform, methanol, ethanol, hexane, or mixtures) which can be difficult to recover.

9.3.2 Types of microalgae grown in wastewater for biodiesel production

Different types of microalgae have been studied for FAME harvesting based on their lipid production. The conventional FAME microalgal fingerprint is showcased in [Figure 9.3](#), generated from information taken from [Komolafe et al. \(2014\)](#), demonstrated the conversion of *Desmodesmus* sp. grown in wastewater into FAME via *in situ* transesterification. The highest recovery was reported as 77.6 (± 2.3) wt% of FAME at a reaction time of 75 min, equivalent to 0.2 mg/mg of microalgae biomass, using a catalyst/lipid (NaOH) molar ratio of 0.15:1 and a methanol/lipid molar ratio of 600:1. [Vasistha et al. \(2023\)](#) obtained approximately 0.3 mg FAMES per mg of *Coelastrella* sp. KJ-04 grown in distillery wastewater.

[Figure 9.3](#) shows the common fingerprint of microalgae-derived biodiesel, the highest fractions correspond to oleic (C18:19c) and γ -linoleic (C18:3n6) methyl esters (20–27%), followed by steric (C18:0) and palmitoleic methyl esters (C16:1n9t). These C16 and C18 carbon chains make-up 60% of the overall FAMES. [Vasistha et al. \(2023\)](#) also reported carbon chains C16–C18 with no more than 2 degrees of unsaturation ($16-18 < 3$), which seems to be a deterministic factor in FAMES obtained from green microalgae.

9.4 BIOETHANOL PRODUCTION FROM MICROALGAE GROWN IN WASTEWATER

9.4.1 Bioethanol production process

Bioethanol is a type of biofuel with the formula: C_2H_5OH produced from fermentation of plant material with high sugar/carbohydrate content. Its overall fermentation reaction is provided in [Figure 9.4](#) and can be simplified into two main reaction mechanisms. Reaction 1 showcases the conversion of glucose to ethanol. Reaction 2 demonstrates the glucose consumed for microbial (yeast) growth. Bioethanol is currently the highest globally consumed biofuel having an 82% share of all biofuels commercialized. Its production has attracted extensive biomass studies, lately including the use of lignocellulosic

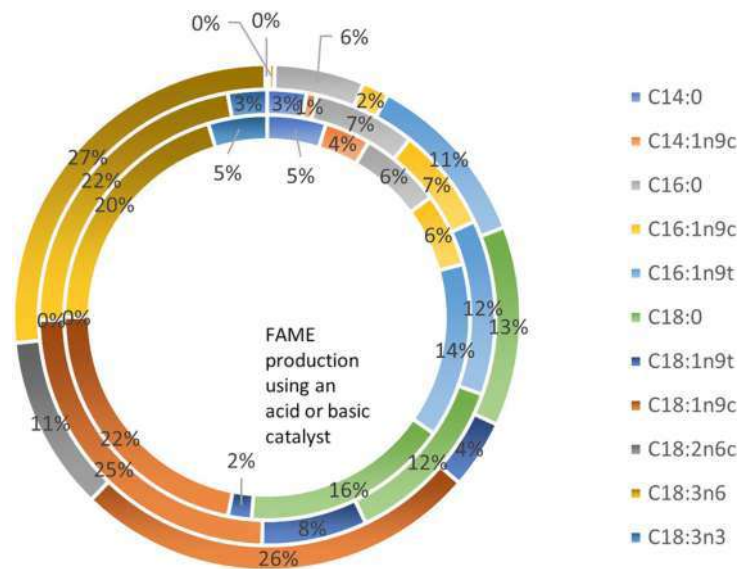


Figure 9.3 Fingerprint of FAMES contained in microalgal biodiesel production. Percentages by weight of the types of methyl esters found. Inner circle corresponds to the lipid extraction of algae crude oil, middle circle showcases FAME production using an acid catalyst, outer circle shows FAME composition using an alkaline catalyst. (Source: Biodiesel composition obtained from [Komolafe et al., 2014](#)).

biomass and fermentations via synthetic developed strains. The major producers of bioethanol are in America. The United States and Brazil had an annual increase from 34.4 to 59.7 billion litres between 2019 and 2020. The demand for bioethanol continues to grow and it is expected to increase by 9.7% in 2026. Bioethanol is a building block in the production of other chemicals and solvents. These products include drugs, plastics, lacquers, polishes, plasticizers, and cosmetics. Hence, ethanol is an essential commodity and organic chemical needed in large volumes for consumer products and industry.

Microalgae including *Chlorella*, *Dunaliella*, *Chlamydomonas*, *Scenedesmus*, and *Spirulina* have a carbohydrate content up to about 50% (w/w), which make them good candidates for bioethanol production ([Chen et al., 2013](#); [Dragone et al., 2011](#)). Cultivation strategies, such as nutrient starvation, can help promote the accumulation of energy-rich compounds: carbohydrates and lipids ([Hernández-García et al., 2019](#)). Most studies in the literature report bioethanol production from microalgae cultivated in synthetic medium, however, microalgae *Scenedesmus obliquus* was shown to be able to grow in wastewater more than 70 years ago ([Gotaas et al., 1954](#)). *Scenedesmus* sp. has been one of the most studied species because of its ability to remove a high percentage of phosphorus (85–99%) and nitrogen (88–99%) as well as its microalgal biomass productivity between 0.073 and 0.15 g/L/d

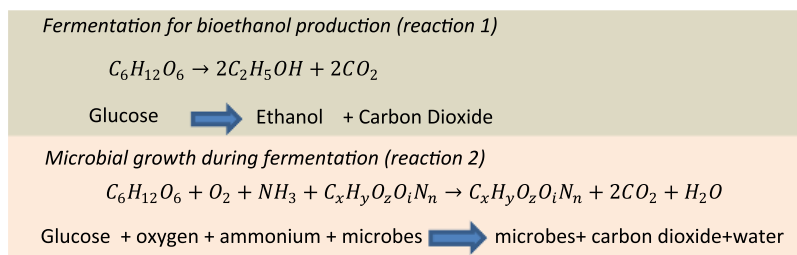


Figure 9.4 Overview of the main fermentation and growth reactions during microalgae sugars conversion to bioethanol.

(Ji *et al.*, 2015; Ruiz *et al.*, 2014; Zhang *et al.*, 2014). Hernández-García *et al.* (2019) observed that cultures of *Desmodesmus* sp. accumulated up to 41% carbohydrate by weight (and 20%w of lipid) after nutrient-limiting conditions.

Microalgal biomass is first harvested and hydrolysed to obtain fermentable sugars, which can be transformed into ethanol. Once simple sugars are obtained, a conventional fermentation process is conducted, continued by the separation and refining of the bioethanol produced (Figure 9.5). Usually, batch fermentations are conducted between 8 and 12 hours. The fermented products are then separated using a centrifuge. The wine/beer (liquid fraction) output from centrifugation is then distilled to achieve a mixture of 95% bioethanol and 5% water (Figure 9.5a). Following this, ethanol is further refined using processes such as azeotropic distillation (Figure 9.5b) or pressure swing adsorption (Figure 9.5c). These two last processes can increase the ethanol purity to 99.6% (w/w).

9.4.2 Hydrolysis

Hydrolysis or saccharification of the harvested biomass is a crucial step to release fermentable sugars. Miranda *et al.* (2012) compared different methods of cell disruption and extraction of sugar from *S. obliquus*, including physical (homogenization, sonication at 120°C temperature) and physicochemical (acid or alkaline hydrolysis), concluding that the best method was acid hydrolysis using sulphuric acid. Acid hydrolysis provides a high efficiency in converting cellulosic materials into fermentable sugars (Harun & Danquah, 2011; Phwan *et al.*, 2018). Figure 9.6 showcases the hydrolysis reaction of starch to produce simpler carbohydrate molecules. Romero-Frasca *et al.* (2021) conducted an acid hydrolysis of microalgae using 0.1M sulphuric acid, a temperature of 85–90°C, and constant stirring for 120 min. The reaction was then neutralized using a 5 M sodium hydroxide solution.

9.4.3 Fermentation

The hydrolysate obtained is then fermented into bioethanol as shown in Figure 9.4. Fermentation releases carbon dioxide which can be recovered and incorporated into the system for microalgae cultivation. Bioethanol has a high-octane number and high heat of vaporization. Hence it is an adequate gasoline replacement or blend in concentrations between 10 and 80% (v/v) following minor engine modifications (e.g. the intake manifold needs to be redesigned as per bioethanol's high heat of vaporization).

Initial studies mainly reported yields on the bioethanol production from microalgal biomass grown in synthetic media. Ho *et al.* (2013) obtained bioethanol from the acid hydrolysate of *S. obliquus* (51.8%, carbohydrate content) using *Zymomonas mobilis* for the fermentation process. After 4 h of fermentation, an ethanol concentration of 8.6 g/L was obtained with a yield of 0.22 g of ethanol/g of biomass. In this study, acid hydrolysis (2% H₂SO₄) was used to saccharify the wet biomass of microalgae, achieving a glucose yield of 96–98% and a transformation to ethanol of 99.8%, respectively. Reyimu and Özçimen (2017) reported bioethanol with yields of 0.04 g of ethanol/g biomass using *Tetraselmis suecica* cultivated in treated municipal wastewater. On the contrary, Tighiri and Erkurt (2016) reported an ethanol yield of 0.05 (g ethanol/g biomass), using biomass from a mixed culture of microalgae, also cultivated in wastewater.

Our laboratory has recently identified *Candida* sp. as growing species during wastewater treatment (Romero-Frasca *et al.*, 2021; Walls *et al.*, 2019). It was first noted that the species were able to produce bioethanol during wastewater treatment at low quantities as per previous literature (Reyimu & Özçimen, 2017). The *Candida* strains were then isolated and utilized for the transformation of acid hydrolysed microalgae to bioethanol. *Candida* sp. were able to convert 75% of glucose to bioethanol, whilst *S. cerevisiae* achieved an 87% conversion at 28°C, pH 6.5. Relatively similar ethanol yields were determined for both species, achieving 0.45 (±0.05) and 0.46 (±0.05) g ethanol per g glucose for *S. cerevisiae* and *Candida* sp., respectively (Romero-Frasca *et al.*, 2021). This indicated that the wild-type species of *Candida* have the potential to conduct fermentations using wastewater as growth medium. Additionally it also demonstrated acid hydrolysis as a viable method for producing bioethanol from microalgae, without significant inhibition in alcoholic fermentation due to possible toxic compounds.

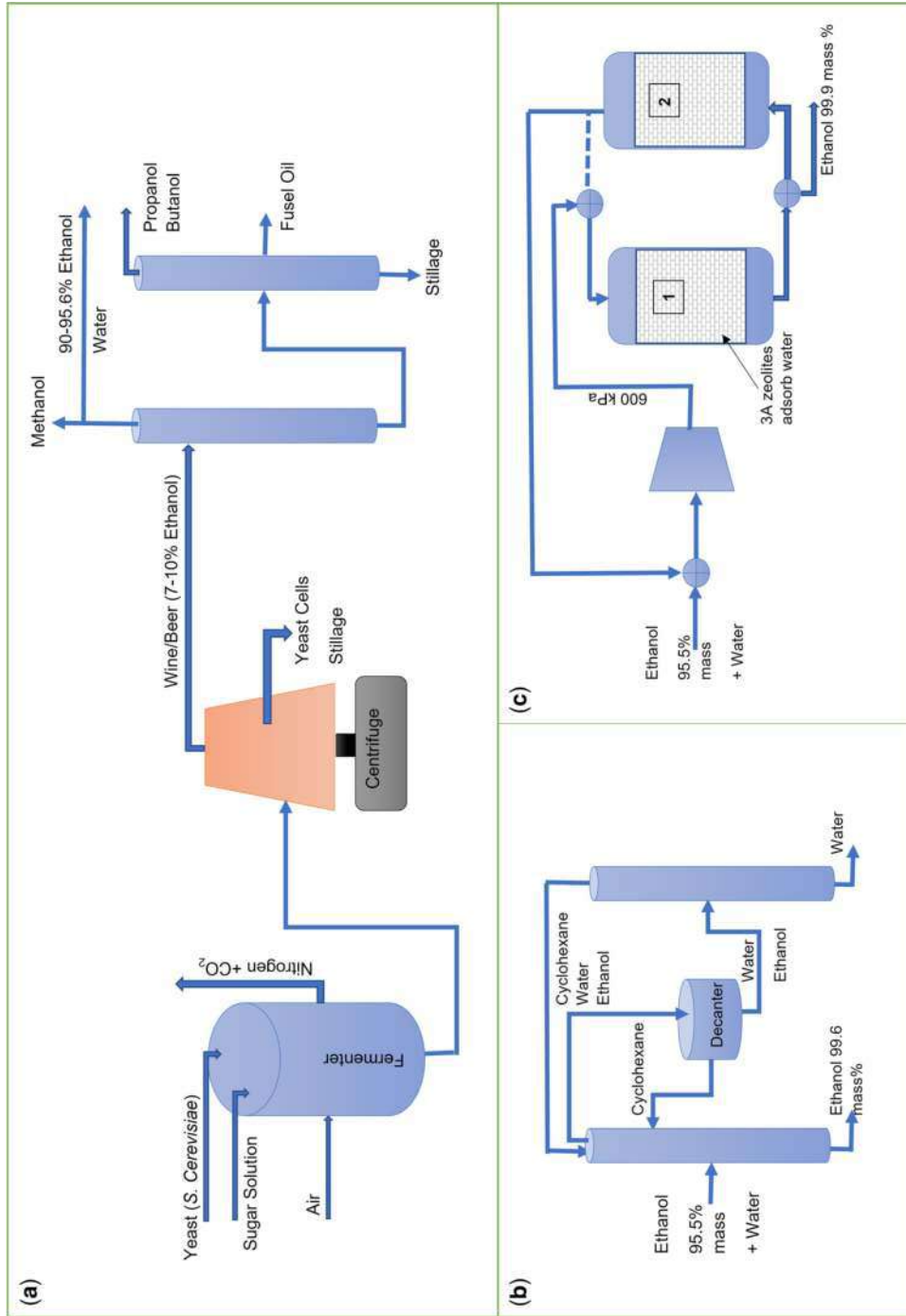


Figure 9.5 Overview of unit operations following the fermentation reaction for bioethanol production. (a) First separation and distillation unit operations to achieve 95% bioethanol, (b) azeotropic distillation and (c) pressure swing adsorption where bioethanol is dehydrated. The system consists of two identical adsorption columns (1,2) that switch on/off in sequential mode. This is done to maintain the pressure from one column to another and save energy.

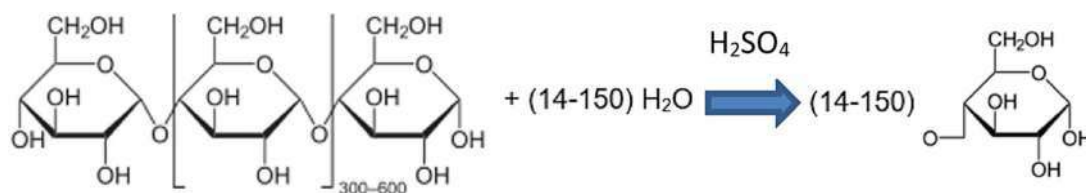


Figure 9.6 Example of an acid hydrolysis reaction.

9.5 CONCLUSIONS AND PERSPECTIVES

A circular bioeconomy framework can be established by using microalgae for the dual purpose of wastewater treatment and biofuel recovery. One of the main initial detriments of biofuel production from microalgae is the high microbial cultivation costs using synthetic medium, hence by using wastewater this cost is alleviated. However, there are still other challenges that need to be resolved on microalgal biofuel production systems. For example, currently biomass pre-treatment and refining require energy-intensive unit operations. Hence, costs could be reduced through investigating other process integration routes similar to this dual bioremediation and biorefining approach. Another approach is the process intensification of microalgae growth and processing. In this last one, *in situ* transesterification has shown advantages over separate extraction and transesterification processes. Further research should elucidate other mechanisms for integration and intensification. One example is the concept of a biorefinery system where high-value compounds are produced using similar unit operations to obtain biofuels.

The production of biodiesel and bioethanol from microalgae grown in wastewater has been shown at the laboratory- and pilot scales. Wild microalgae can contain up to 40%w lipids and 50%w carbohydrates after nutrient limitation conditions. Microalgal biodiesel is produced via a transesterification reaction. The reaction will only convert neutral lipids to FAME, with reported conversions between 80 and 99% using either alkaline or acid catalysts. However, a biodiesel production route involves a series of refining processes, apart from the main reaction, that require special consideration. Microalgal biodiesel have a specific fingerprint with carbon chains of C16 and C18. Regarding bioethanol production, the complex structure of microalgae biomass needs pre-treatment via lysis and hydrolysis before fermentation. In hydrolysis, using an acid has been shown a straight forward mechanism, however, costs and associated risks at a large scale hinder its industrial economic use. It is interesting that bioethanol was also found to be produced during wastewater treatment, giving the possibility for future explorations using a dual fermentation and wastewater treatment process.

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Part 4

Algal Biotechnology

Chapter 10

Advanced value-added bioproducts from microalgae

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ABSTRACT

Currently, the world is facing major issues of degradation of land by natural as well as anthropogenic activities such as desertification, salinization, industrialization, pollution and population growth. The limited resources and the expanding global population require alternative resources to meet the demands in the future. Microalgae are contemplated as a favorable resource for high-value products, including carotenoids, phycobilin, astaxanthin, docosahexaenoic acid, eicosahexaenoic acid and omega-3/6 polyunsaturated fatty acids. Although the use of algae is not new, the idea of developing high value-added products concerning sustainability, economic viability, nutrition enrichment and environmental friendliness is attracting researchers to explore more about the potential of microalgal flora. Microalgae not only thrive under extremophilic conditions but also do not compete with plants for land resources. Having a short generation period, diverse biochemical composition, low-cost nutritional needs and fixation of CO₂ are also significant reasons to promote their products. Also, the biorefinery concept and sustainable cultivation possibilities can substantially add to enabling sustainable production of high-value biomolecules, while proposing opportunities for increasing sustainable food and fuel supplies. However, a few challenges like inadequate domestic demand, constant maintenance of ideal conditions for cultivation and food regulations still need to be overcome.

Keywords: bio-stimulant, microalgae, polyunsaturated fatty acids, carotenoids, pigments

10.1 INTRODUCTION

The worldwide population is expected to reach nearly 10 billion by the end of 2050. The rising population will certainly increase the demand for food, beverages, supplements, pharmaceuticals and personal care products (Rahman 2020). Now the world is looking for alternatives to fulfil the demand and sustainability criteria for the future and among the alternatives microalgae can be considered as a promising resource

(Caporgno & Mathys, 2018). The estimated microalgae-based industry was US\$3.4 billion in 2020 which is expected to become USD 4.6 billion in 2027 (Zhuang *et al.*, 2022). To develop the microalgae industry to their full ability, further research is necessary in terms of improving product yields and lowering overall costs. The utilization of microalgae for industrial purposes is nevertheless facing issues like low yield, energy consumption, maintenance of the cultures and products. The ongoing technological advancements will surely pave the way for these issues, for instance, selection of heterotrophic strains are in preference to limit the impact on natural conditions. As these strains grow rapidly using organic compounds, they knock out the limitation of environmental conditions and help in utilizing waste resources like lignocellulosic materials. However, with the advancement of genetic and metabolic engineering, the progress of culture and screening procedures as well as incorporation of nanotechnology, microalgae will become the most reliable resource of products and applications (Harun *et al.*, 2011).

Microalgae having universal presence and thriving survival under extreme conditions could be one of the major advantages for the industry. Microalgae are abundantly present organisms: 0.2–0.8 million species are present within the algae kingdom, yet underexplored in different sectors. Although their use is not new to mankind, still their utilization for the maximum extent with sustainability brings a new horizon for industrial research. Further, microalgae are an efficient fixer of atmospheric CO₂, which could substantially lead to the decrease of the greenhouse effect and will empower environmental health (Liyanaarachchi *et al.*, 2021; Mironiuk & Chojnacka, 2018). Therefore, development of microalgal-based products by industries not only provides benefits to the human health but it also supports the improvement of environmental health. Moreover, nutritional needs of the microalgae are limited, under the presence of sunlight rapid generation of microalgae can be easily achieved, which could be cost-effective from the industrial point of view. In addition, microalgae will produce high biomass concentrations in comparison to the terrestrial plants without engaging hectares of land (Russell *et al.*, 2022). Also, algae being a primitive plant ensure easy extraction and purification of the bioactive metabolites/compounds for further application in comparison to the complex procedures required in higher plants (Mironiuk & Chojnacka, 2018). Microalgae-based products besides being organic, possess the nutrition values higher than the usual supplements and are more potent to human health (Korzeniowska *et al.*, 2018). Also, the use of microalgae-based pigments for cosmeceuticals and nutraceuticals is attracting a lot more attention (Saxena *et al.*, 2020; Zhuang *et al.*, 2022). This chapter is an overview of the recent research conducted on microalgae for the detection, extraction and commercialization of their biomolecules in various sectors, including their presence in the market and concerning challenges of industries.

10.2 MARKET VALUE OF ALGAE-BASED HIGH-VALUE COMPOUNDS

The market value assessment of algal products is based on their nutritional composition, formulation, level of purity and usage (Vieira, 2016). Also, it is important to understand the regulatory law framework, technical and economic aspects and risk management for the development of microalgae-based market products. The crucial challenges with the market are expensive operational cost, requirement of infrastructure and maintenance, optimization of commercial scale harvesting quantity and optimization of market financial affairs regarding microalgae-based products. Besides these difficulties, it is estimated that microalgae-based product markets will reach up to US\$ 53.43 billion by 2026 (<https://www.credenceresearch.com/report/algae-productsmarket>)

According to Khattar *et al.* (2009), the global microalgae-based product astaxanthin market was assessed around US\$555.4 million in 2016. Microalgae is the natural resource for this pigment which is widely in use for nutraceuticals and aquaculture industry due to its antioxidant properties and fortification. Its market value is way higher than its synthetic version in the market (Li *et al.*, 2011; Pérez-López *et al.*, 2014). Also due to its potential application in neuro and cardio-related diseases, their market values have been influenced remarkably (Wu *et al.*, 2015). For 130 metric tons of annual production currently more than \$200 million have been utilized. The average market prices are

estimated to be between 1000 and 2000 USD per kg depending upon the purity level. Due to the high cost till now only 1% of the market is covered by astaxanthin produced by microalgae (Shah *et al.*, 2016). Similarly, beta-carotene another biomolecule extracted from algae was having a 3.5% compound annual growth rate and was US\$224 million in 2018 (Transparency Market Research, 2018) and the largest shareholder is Europe (Market Watch, 2018). For multiple applications in personal care, food and pharmaceuticals have raised its demands, also in the Asia-Pacific regions.

Furthermore, with the raising consciousness about health in mankind, industries are witnessing accelerated demand of Omega-3 (Market Research Future, 2019). Omega-3 is an essential fatty acid which is not produced in the human body. The market value of microalgae-based omega-3 is increasing by 13.5% per year. Currently its market is expanding in US, Europe and Asia-Pacific due to numerous health benefits. Likewise, the market size of eicosapentaenoic acid (EPA) and docosahexaenoic acid (DHA) are 300 million and 1.5 billion USD, respectively, and the price is 0.2–0.5 USD/g and 18–22 USD/g, respectively. Over 75% of the manufacture volume of microalgae was used in the health food marketplace as nutritional enhancements (Chacon-Lee and Gonzalez-Marino, 2010). The algae-based valued food additives and ingredients, for example DHA, represent a rising market. Martek's (now DSM) algae-derived DHA is found in 99% of all baby foods in the USA (Eckelberry, 2011).

High-value products that are extracted from microalgae thus improve the economics in a biorefinery approach and have market scope and opportunities (Figure 10.1). However, it needs to be

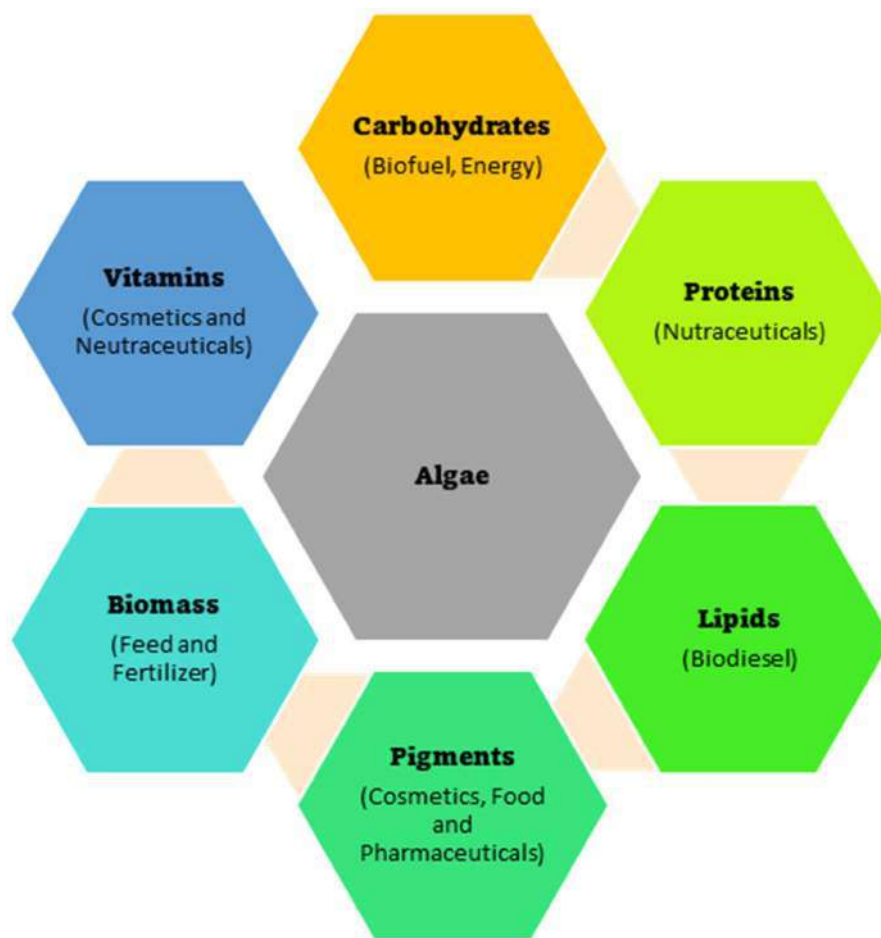


Figure 10.1 Various application of algae.

understood whether it is market driven or technology driven. Production economics such as the cost effectiveness of the food needs to be offset against the market opportunities and high-value products from microalgae: the technology also needs to be robust and reliable for its market flow.

10.3 HIGH-VALUE PRODUCTS USED IN DIFFERENT SECTORS

10.3.1 Cosmetics

Cosmetics are the products used globally by people to protect their skin and this industry is growing very fast due to modern lifestyle. The daily used cosmetics contain different synthetic chemicals which may cause adverse side effects on continuous exposure, due to these reasons, there is need to replace synthetic chemicals with environmentally sustainable products (Ariede *et al.*, 2017). Cosmeceutics is a nonofficial term and defined as cosmetic products with biologically active ingredients having medical or drug-like properties (Kligman, 2000; Martin & Glaser, 2011). Nowadays, natural ingredients from algae gained tremendous attention as an alternative for safe and high-quality products (Table 10.1). Microalgae contain natural pigments/metabolites having biological activities such as antioxidants, anti-bacterial, anti-aging, anti-inflammatory, anticancer and antiviral which makes them useful organisms in cosmetics industry for skin care products, anti-aging creams, sun protection products, thereby increasing the growth of this market (Fernando *et al.*, 2016; Talero *et al.*, 2015; Thomas & Kim, 2013; Wang *et al.*, 2017). There are hundreds of bioactive metabolites predicted from cyanobacteria and thousands more are predicted from eukaryotic microalgae.

Algae produce various secondary metabolites/antioxidants during their adaptation to stress and for survival in harsh conditions (Sansone & Brunet, 2019; Wang *et al.*, 2015). The secondary metabolite mycosporine-like amino acids (MAAs) received much attention these days due to their various applications in daily use materials such as fabrics, plastics, varnishes and paints to protect them against ultra-violet radiations (UVR) (Chrapusta *et al.*, 2017). Hence, MAAs are promising in many cosmetical and pharmaceutical industries (Kageyama & Waditee-Sirisattha, 2019). MAAs are present in some microalgae such as *Anabaena* sp., *C. vulgaris*, *D. salina*, *Eutreptiella* sp., *Scenedesmus* sp., *S. platensis* and *Leptolyngbya* sp. (Martínez-Ruiz *et al.*, 2022a). Another pigment, scytonemin which is present in the mucilaginous sheath around numerous marine cyanobacterial cells producing extracellular polysaccharides imparts yellowish-brown color to the cells (Martínez-Ruiz *et al.*, 2022a). Scytonemin is used in sunscreen for UV protection because it absorbs the light spectrum around 315–400 nm and is mainly extracted from *Scytonema* and *Nostoc* sp. majorly *N. punctiforme* (Sen & Mallick, 2022). *Nostoc* sp. can survive high levels of radiation. Natural antioxidants, such as

Table 10.1 Major microalgal products involved in cosmetic preparations.

Cosmeceutical Compound	Name of Microalgae	References
Scytonemin	<i>Nostoc punctiforme</i> , <i>Scytonema</i> sp., <i>Nostoc commune</i> , <i>Calothrix</i> sp., <i>Lyngbya</i> sp., <i>Leptolynbya mycodia</i> ,	Stolz and Obermayer (2005), Nowruzi <i>et al.</i> (2020), Rastogi <i>et al.</i> (2020), Santiesteban-Romero <i>et al.</i> (2022), Sheibani and Naeimpoor (2023)
Sporopollenin	<i>Dunaliella salina</i> , <i>Chlorella fusca</i> , <i>Scenedesmus</i> sp.	Priyadarshani and Rath (2012), Pallela (2014), He <i>et al.</i> (2016), Gupta <i>et al.</i> (2023)
Mycosporine	<i>Isochrysis</i> sp., <i>Chlorella minutissima</i> , <i>Dunaliella tertiolecta</i> , <i>Chlorella sorokiniana</i> , <i>Thalassiosira weissflogii</i> , <i>Lyngbya purpurem</i> , <i>Oscillatoria</i> sp. <i>Dunaliella tertiolecta</i> , <i>Chlorella sorokiniana</i> ,	Stolz and Obermayer (2005), Kim and Chojnacka (2015), Chandra <i>et al.</i> (2020), Geraldine <i>et al.</i> (2020), Rosic (2021), Zaytseva <i>et al.</i> (2021) Tossavainen <i>et al.</i> (2019), Fawcett <i>et al.</i> (2022)

astaxanthin, carotenoids and lycopene protect the skin from oxidative stress and damage caused due to the production of free radicals through exposure of ultra-violet (UV) and further prevents skin aging (Gao *et al.*, 2021; Mourelle *et al.*, 2017). Lutein, a compound from different microalgae majorly *C. protothecoides*, prevents skin damage caused by ultraviolet-C (UVC) (Saha *et al.*, 2020). *Dunaliella tertiolecta* and *Tetraselmis suecica* produce high concentrations of α -tocopherol and vitamin E, which are widely used in cosmetic formulations (Arora & Philippidis, 2023). *Dunaliella salina* and *Spirulina platensis* sp. are rich in β -carotene and *Porphyridium* is rich in sulphated polysaccharides which prevent the formation of reactive oxygen species (ROS), inhibit lipid peroxidation and ultimately prevent oxidative damage to skin cells and produce hyaluronic acid, a glycosaminoglycan which helps in skin hydration (Gupta *et al.*, 2023). The secondary metabolites of brown algae, *Macrocystis pyrifera* (i.e. phlorotannins) and *Turbinaria conoides* (i.e. laminarin, fucoidan and alginates) have antioxidant activity, thus preventing the formation of free radicals and protect skin from aging (Peng *et al.*, 2011). β -1,3-Glucan polysaccharide, rich in *Chlorella* sp. and *Skeletonema* diatom, as well as *Porphyridium* and *Nostoc flegelliforme*, acts as a free-radical collector and active immunostimulator which makes them potential candidates for preventing aging (Hamed, 2016; Shao *et al.*, 2013). The main carotenoids present in microalgae are β -carotene, lycopene, astaxanthin, zeaxanthin, violaxanthin and lutein and the most common microalgae commercially used for pigment production are *Dunaliella salina*, *Haematococcus pluvialis*, *Chlorella* sp., *Scenedesmus* sp., *Muriellopsis* sp., *Spirulina* sp. and *Porphyridium* sp. (Sathasivam & Ki, 2018).

The natural pigments present in microalgae and cyanobacteria are chlorophylls, carotenoids (carotenes and xanthophylls) and phycobilins and used in cosmetics such as in lipstick, eye shadows and eyeliners as a natural colorant (Begum *et al.*, 2016; Morocho-Jácome *et al.*, 2020). For example, β -carotenes from *Dunaliella salina*; astaxanthin from *Haematococcus pluvialis* (red color), phycocyanobilins (blue pigment) from *Spirulina* and phycoerythrobilins from rhodophyte *Porphyridium* are natural dyes (Hamed, 2016). Other pigments such as chlorophyll are easily extracted and used in deodorants, due to their ability to mask odors, as well as in toothpastes and hygiene products (Mourelle *et al.*, 2018). Canthaxanthin pigment from *Nannochloropsis* sp. is commercially used in tanning pills (Rebello *et al.*, 2020). Different type of lipids such as triacylglycerides, waxes, fatty acids, ceramides, glycerophospholipids, sterols, hydrogenated, esterified and oxidized lipids are used in cosmetics as dermatological delivery and moisturizing agents (De Luca *et al.*, 2021). The extracts from algae *Arthrospira platensis*, *H. pluvialis* and *T. suecica* proteins and polysaccharides are used in gels as thickeners and moisturizers (Martínez-Ruiz *et al.*, 2022b). Various marine strains secrete extracellular polysaccharides which act as physical barriers protecting the cells from external stimuli. Color and fragrance are two important characteristics for cosmetic products. The coloring is mostly done through pigments and essential oils provide aroma. Some commercially available products produced by *D. salina*, its extract known as blue retinol, help in growth and proliferation of skin cells (Mourelle *et al.*, 2017). Another product, silidine from the purple-red alga *Porphyridium cruentum* improves the skin texture and decreases redness. GoldenChlorella and AlgaPür Algae Oils from exopolysaccharides are beneficial for skin and hair treatments. Some companies are using extracts from algae. Recently, lipid extract from *Phaeodactylum tricornutum* is used as an anti-aging agent because it stimulates cell detoxification from oxidized proteins through proteasome, thus preventing aging by inhibiting the accumulation of harmful proteins (Vasilopoulou *et al.*, 2021). *Chlorella vulgaris* extract is also used for collagen repair and supporting tissue regeneration, thus reducing wrinkle (Ariede *et al.*, 2017; Wang *et al.*, 2015). A protein-rich extract from *Arthrospira* repairs the aging, tightens the skin and prevents stria formation (Bilal *et al.*, 2017).

10.3.2 Pharmaceuticals

The naturally derived valuable compounds from algae are gaining attention due to their useful biomedical properties such as anticancer, antidiabetic, antiviral and antimicrobial compounds. These compounds can be primary and/or secondary metabolites and used as a sustainable and cheap

source for various pharmaceutical products such as antibodies, recombinant proteins, vaccines and drug delivery in the pharmaceutical sector (Table 10.2). The high-value compounds from microalgae and cyanobacteria are screened for anti-diabetic properties having specific enzymes with catalytic activities (Abo-Shady *et al.*, 2023). Some examples of enzymes are α -amylase, α -glucosidase, *N*-acetyl-glucosaminidase, aldose reductase, hexokinase, glucose-6-phosphatase, dipeptidyl peptidase IV, glucose transporter 4 and glycogen synthase kinase-3 β from *Chlorella* sp. *Nitzschia laevis*, *Isochrysis galbana*, *Chaetoceros furcellatus*, *Skeletonema marinoi* and *Porosira glacidis* species (Lauritano & Ianora, 2016; Mutanda *et al.*, 2020).

Pharmaceutical industries showed much interest in lipids such as polyunsaturated fatty acids (PUFAs), phytosterols and carotenoids and used in prevention and treatment of cardiovascular

Table 10.2 Multiple applications of pigments extracted from different microalgal strains.

Name of the Pigment	Name of the Alga	Applications	References
Astaxanthin	<i>Chlorella zofingiensis</i> , <i>Haematococcus pluvialis</i> , <i>Monoraphidium Chlorococcum</i> sp., <i>Scenedesmus</i> sp., <i>Chlamydomonas nivalis</i> , <i>Nannochloropsis</i> sp., <i>Chlamydocapsa</i> sp., <i>Chlorella vulgaris</i> , <i>Eremosphaera viridis</i> , <i>Neochloris wimmeri</i> and <i>Coelastrella striolata</i> , <i>Chromochloris zofingiensis</i>	Nutritional food, cosmetics, Aquaculture, poultry and food	Borowitzka (2013), Allen <i>et al.</i> (2018), Mao <i>et al.</i> (2020), Perozeni <i>et al.</i> (2020), Zhang <i>et al.</i> (2021), Ritu <i>et al.</i> (2023)
Canthaxanthin	<i>Chlorella</i> sp. <i>Asterarcys quadricellulare</i> , <i>Coelastrum</i> sp. <i>Tetraspora</i> sp. <i>Coelastrella</i> sp., <i>Chlorococcal</i> sp.	Aquaculture, poultry and food	Nasrabadi and Razavi (2010), Singh <i>et al.</i> (2019), Rebelo <i>et al.</i> (2020), Maswanna <i>et al.</i> (2022), Janchot <i>et al.</i> (2019), Corato <i>et al.</i> (2022)
β -Carotene	<i>Dunaliella salina</i> . <i>Tetrademus obliquus</i> , <i>Scenedesmus</i> sp. <i>Chlamydomonas reinhardtii</i>	Biomedical Research	Borowitzka (2013), Tran <i>et al.</i> (2019), Singh <i>et al.</i> (2020), Harvey and Ben-Amotz (2020), Goswami <i>et al.</i> (2022)
Zeaxanthin	<i>Chlorella ellipsoidea</i> , <i>Dunaliella salina</i> , <i>Synechococcus</i> sp., <i>Synechocystis</i> sp., <i>Rhodorus</i> sp., <i>Chromochloris zofingiensis</i> , <i>Nannochloropsis oculata</i>	Antioxidant, food pigment	Koo <i>et al.</i> (2012), Bourdon <i>et al.</i> (2021), Chen <i>et al.</i> (2022), Victor and Camarena-Bernard (2023)
Lutein	<i>Scenedesmus</i> sp., <i>Muriellopsis</i> sp., <i>Chlorella sorokiniana</i> , <i>Scenedesmus almeriensis</i>	Antioxidant	Fernández-Sevilla <i>et al.</i> (2010), Xie <i>et al.</i> (2019), Molino <i>et al.</i> (2019), Patel <i>et al.</i> (2022)
Echinenone	<i>Botryococcus braunii</i>	Antioxidant	Borowitzka, 2013; Indrayani <i>et al.</i> 2022
Phycoerythrin	<i>Spirulina</i> sp. <i>Rhodomonas</i> sp., <i>Porphyridium purpureum</i> ,	Pharmaceuticals	Allen <i>et al.</i> (2018), Sosa-Hernández <i>et al.</i> (2019), Rodas-Zuluaga <i>et al.</i> (2021), Ji <i>et al.</i> (2022), Derbel <i>et al.</i> (2022)

diseases and blood coagulation (Xia *et al.*, 2021). Most studied lipids from microalgae are PUFAs and its derivatives such as DHA, EPA, α -linolenic acid, arachidonic acid (ARA) and docosapentaenoic acid are used for the treatment of diabetes, inflammatory bowel disorders, skin and respiratory disorders (Khan *et al.*, 2015). DHA and EPA also act as anti-inflammatory agents and used to reduce hypertension, stroke and arthritis. Additionally, DHA compounds are also used for the proper function and development of the nervous system (Jha *et al.*, 2017). ARA and DHA are essential for the development of eyes and brain in infant and are used as fortifications to infant formula (Mimouni *et al.*, 2012). Some examples of microalgae producing lipids are *Phaeodactylum tricornutum* producing EPA, *Nitzschia conspicua* producing arachidonic acid and *Schizochytrium* sp. accumulating DHA, EPA and palmitic acid (Ramos-Vega *et al.*, 2018; Xia *et al.*, 2021). To produce high amounts of PUFA, different extraction methods and systems need to be evaluated.

Phytosterols inhibit cholesterol absorption in the intestine and thus reduce cholesterol in humans (Le Goff *et al.*, 2019). Carotenoids from *Chlorella ellipsoidea* and *Chlorella vulgaris* have anti-proliferative effect on a human colon carcinoma cell line thus promoting cell death particularly in colon cancer (Cha *et al.*, 2008). β -Carotene from *Dunaliella salina* has good anti-inflammatory and immunomodulatory effects (Hyršlova *et al.*, 2022).

Cyanobacteria (*Spirulina*) and microalgae such as *Chlorella vulgaris*, *Scenedesmus quadricauda* and *Dunaliella* sp. produce sulphated polysaccharides which have a wide range of bioactivities such as antiviral, anticancer and anti-inflammatory (de Moraes *et al.*, 2015; Kiran & Venkata Mohan, 2021). The polysaccharides prevent the attachment of viruses with the target molecule. Microalga *Gyrodinium impudium* produce p-KG103, which inhibits the growth of tumor cells by stimulating cytokine production (Guo *et al.*, 2017). The polysaccharide from *Chlorella pyrenoidosa* showed promising improvement in hyperlipidemia disorder in rats (Wan *et al.*, 2020). Various polysaccharides from microalgae, such as alginate, carrageenan, laminarin and fucoidan, are used in drug delivery via nanoparticles (Yang *et al.*, 2022). The secondary metabolites fucoxanthin, sargaquinoic acid, sargahydroquinoic acid and sargaquinal produced by *Sargassum heterophyllum* show anti-malarial properties (Mutanda *et al.*, 2020). The polyphenols, phycobiliproteins and vitamins have antioxidant properties which help in preventing the oxidative damage caused by free radicals inhibit the growth of cancer cells and help to fight against various diseases such as chronic disorders, cardiovascular diseases and inflammations (Coulombier *et al.*, 2021).

Production of recombinant proteins using algal expression systems is also gaining consideration as it has many advantages like rapid growth rate, post-translational modifications like mammalian cells, cost effective and easy scale-up for bioprocessing and purification. *C. reinhardtii* was used to express malaria antigens from *Plasmodium falciparum* (Shamriz & Ofoghi, 2019). A chimeric gene having hemagglutinin-neuraminidase and fusion epitopes of Newcastle disease virus was successfully expressed in *C. reinhardtii* through an agrobacterium-mediated genetic transformation system (Shahriari *et al.*, 2019). *Phaeodactylum tricornutum* diatom and *D. salina* microalgae were successfully engineered to produce human IgG antibodies against the Hepatitis B virus surface protein (Geng *et al.*, 2003; Vanier *et al.*, 2015). Still some hurdles need to be addressed such as safety evaluation of transgenic strains, downstream processing, purification, cost and clinical trials.

10.3.3 Food supplements

10.3.3.1 Protein content of algae

An increase in the global population and their high demand for meat and dairy products has created pressure on protein supply. To meet those needs, it is highly essential to find out alternative protein sources. In this context, available protein in microalgae offers an excellent nutritional substitute by delivering all essential amino acids (Hariskos and Posten, 2014; Bhagia *et al.*, 2022). Protein is an integral component of the structure and metabolism of microalgal cells. Many microalgae contain very high amounts of protein, ranging from 42% to 70% (Barkia *et al.*, 2019; Milovanovic *et al.*, 2012; Plaza *et al.*, 2009). Microalgae can produce 2.5–7.5 tons of proteins annually per hectare (Bleakley & Hayes,

2007). *Chlorella* sp. has more than 70% of protein content, which has been commercialized recently (Eilam *et al.*, 2023). Similarly, *Spirulina* sp. and *Arthrospira* sp. are two well-known protein-rich microalgal strains. However, some other cyanobacteria like *Lyngbya majuscula*, *Nostoc* sp., *Anabaena* sp. and *Porphyridium* sp. are observed due to the production of microcolin-A (immunosuppressive agent), cyanovirin (antiviral agent against HIV) and enzyme superoxide dismutase (antioxidant), respectively (Arya and Gupta, 2001). *Microcystis aeruginosa* produces amino acids like serine, glycine, proline and valine. A carbonic enzyme anhydrase is produced by *Isochrysis galbana* that converts carbon dioxide into carbonic acid and bicarbonate (Khan *et al.*, 2018).

10.3.3.2 Single-cell protein

Single-cell protein (SCP) refers to a conventional or substituent for a protein found either from pure or mixed cultures of microalgae (also extracted from fungi, bacteria, or yeast) mostly used for animal and human consumption. These macromolecules with various chemical structures lead to various important functions (morphological, technological and physiological). Those protein components can be used as individual protein concentrates and can be integrated into processed food products. In this regard, microalgae are considered one of the most reliable protein sources and most of the algal domain is involved in the food sector. They also possess higher protein contents than conventional plant and animal protein sources. For example, according to Moorhead *et al.* (2011), *Chlorella* sp. is considered a human diet and used in mariculture due to the presence of immune-active substances, for example, 3-glucan β -1. Similarly, the protein content in *Spirulina platensis* is 65% which is 36%, 37%, 22%, 24%, 26% and 24% greater than dried skimmed milk, chicken, soy flour, beef, fish and peanuts, respectively. Some other microalgal strains like *Aphanizomenon* sp., *Nostoc* sp., *Dunaliella* sp., *Porphyridium* sp., *Arthrospira* sp., *Scenedesmus* sp., *Anabaena* sp. and *Tetraselmis* sp. are involved in SCP production.

10.3.3.3 Carbohydrates

Algal cells are an important food source as they have an excellent content of carbohydrates. They may be monosaccharides, oligosaccharides, or polysaccharides that perform structural and metabolic activities. They can attach to proteins or lipids as glycoproteins or glycolipids (Arad & Levy-Ontman, 2010). Moreover, the microalgae can also generate carbohydrates such as glucose and starch through photosynthesis which are the basic constituents of the cell wall. Some species have a high carbohydrate content (Barkia *et al.*, 2019; Harun *et al.*, 2011), that is *Spirogyra* sp. and *Porphyridium cruentum* contain 35–65% and 40–60% carbohydrate, respectively. Microalgal polysaccharides, another form of carbohydrates, play a vital role in manufacturing pharmaceuticals such as antiviral, antibacterial, antioxidant and anticancer compounds. The microalgal polysaccharides are also involved in synthesizing cosmetics, nutritional components, anti-herpes drugs and pharmacological compounds in the business market. These are produced in different forms depending on the microalgae species. More specifically, several cyanophytes synthesize glycogen, some accumulate semi-amylopectin (Nakamura *et al.*, 2005) and various species of chlorophyta can synthesize starch in the shape of 2 glucose polymers, namely amylose and amylopectin (Busi *et al.*, 2014). Similarly, diatoms are well known for synthesizing floridian starch and chrysolaminarin (Gugi *et al.*, 2015). The microalgal polysaccharides benefit the cosmetic industry, acting as hygroscopic agents and antioxidants for topical creams and lotions (Gujar *et al.*, 2019).

10.3.3.4 Lipids

Microalgal lipids have attracted much attention for their commercialization due to biodiesel production. Moreover, poly-unsaturated fatty acids, such as omega fatty acids, have noticeably high trade values in infant formulations and nutraceuticals (Qu *et al.*, 2013). This provides structural support to plasma membranes and acts as energy reservoirs. The lipid percentage of microalgae is a major portion of

neutral (acylglycerols, carotenoids and free fatty acids) and polar (phospholipids and galactolipids) lipids. Most microalgae are well-off in polar lipids in their exponential growth phase and commonly pile triacylglycerols in their stationary phase under unfavorable conditions (Rodolfi *et al.*, 2009). Fatty acids in microalgae are normally categorized as unsaturated and saturated fatty acids. These saturated–unsaturated fatty acids are mostly associated with neutral and polar lipids.

The lipid content of algal biomass ranges from 20% to 50% of dry cell weight (w/w). The production of different lipids depends upon the types of microalgal species and different cultivation conditions like salinity, temperature, growth phase and availability of nutrients, light intensity and pH (Guschina & Harwood, 2006). It was also reported that the lipid content increases considerably by limiting the nitrogen amount during their stationary phase. Microalgal lipids are given the most attention as healthy food supplements and vegan alternatives to fish oil and can be utilized as a foundation for industrial chemicals like cosmetic industry waxes and polymer lubricants (Mendes *et al.*, 2003).

10.3.3.5 Vitamins

Microalgae are recommended for their high content of different vitamins. They produce a wide variety of cost-effective and commercially important products. Vitamins from microalgae are easily available and their production is highly dependent upon nitrogen availability in the biomass and culture medium (Bonnet *et al.*, 2010). The different vitamin composition of microalgae is observed mostly during both the log and stationary phases of growth (Chew *et al.*, 2017). The microalga *Dunaliella salina* is well known for synthesizing pyridoxine, vitamins E and A (tocopherols), nicotinic acid, biotin, thiamine and riboflavin (Santhosh *et al.*, 2016). Another microalga, *Haslea osteria*, is rich in vitamin E. High quantities of vitamin A, E, C and β -carotene are synthesized by *Porphyridium cruentum* (Sheih *et al.*, 2009). The algal vitamins are highly nutritious for animals and humans (Borowitzka, 2013). The diatom *Navicula* sp. releases a blue-colored pigment called marennine, which is rich in tocopherols (Gastineau *et al.*, 2018).

10.3.3.6 Minerals

Microalgae are also well known for the accumulation of trace metals; however, few reports are available on the mineral content of microalgae. Minerals in the microalgal biomass include phosphorus, zinc, fluorine, potassium, iron, calcium, magnesium, sodium, sulphur, chlorine, manganese, copper, iodine, cobalt, selenium and molybdenum (Alsenani *et al.*, 2015). They are present either in elemental form or incorporated as compound forms in microalgal biomass and carry out several important functions. According to Tokusoglu and Ünal (2003), a significant number of elements like zinc, manganese, phosphorus, iron, magnesium, potassium, calcium and sodium are present in *Chlorella stigmatophora*, *C. vulgaris*, *Isochrysis galbana*, *D. tertiolecta*, *Tetraselmis suecica* and *S. platensis* (Tokusoglu & Ünal, 2003).

10.3.4 Agricultural products

Algal extracts can be applied in agriculture in the form of bio-stimulants, biofertilizers, or bioregulators of plant growth. Plant growth regulators can alter cell division, root and shoot elongation, initiation of flowering and other metabolic functions, whereas fertilizers are the supplements needed for normal growth of the plant (Figure 10.2). Microalgae can be exploited as natural soil conditioners and biofertilizers to significantly improve the soil characteristics. Recent studies indicate that algae contain several phytohormones as well as high amounts of micronutrients and macronutrients that are essential for plant growth, health and development with better growth and crop yield (Guo *et al.*, 2020; Renuka *et al.*, 2018). Moreover, microalgae can also be utilized to improve soil health and to reduce erosion by crust formation; to treat wastewater for irrigation by removing agrochemical, fertilizers and pesticides as well as for metal removal and nitrogen recovery (Castro *et al.*, 2020).

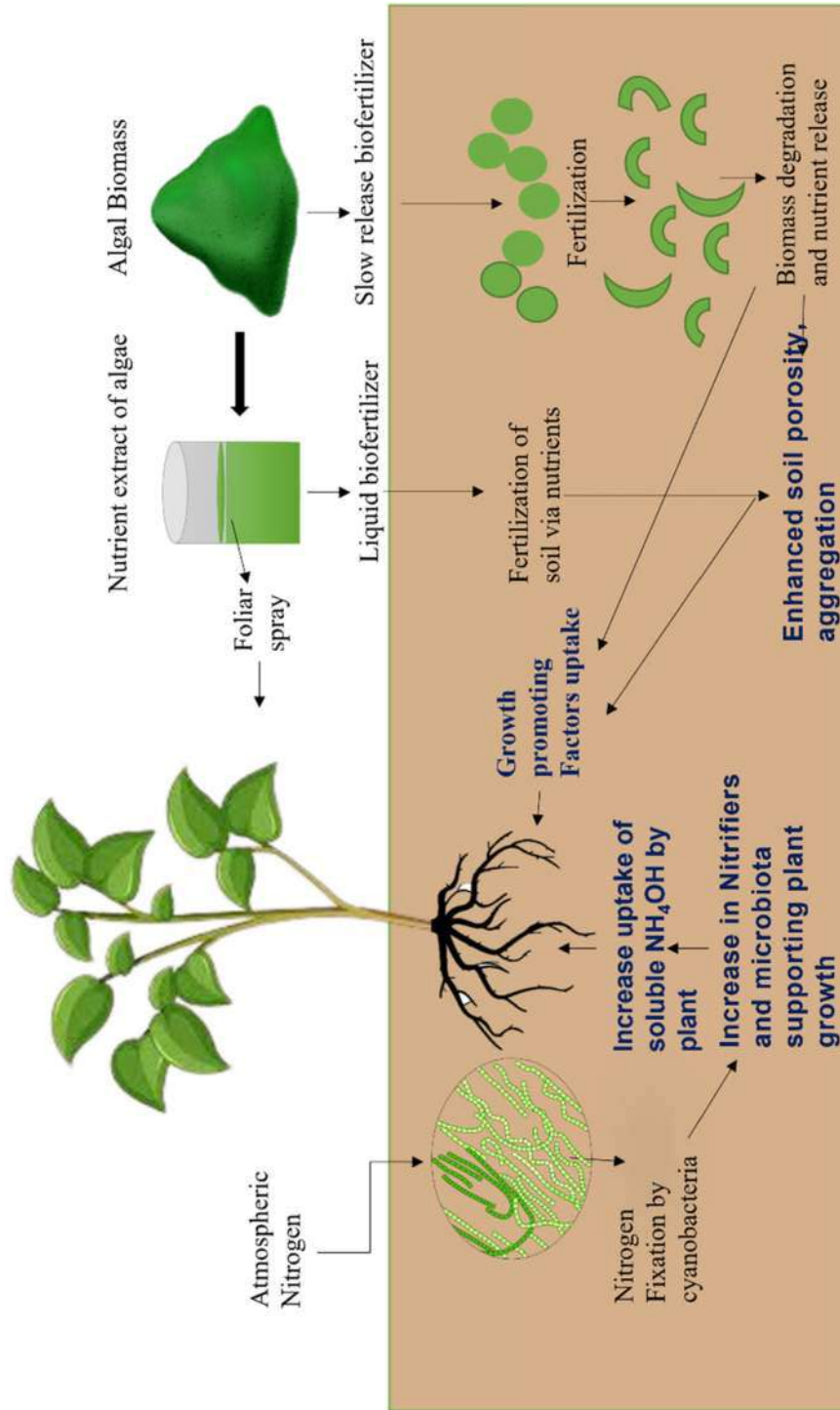


Figure 10.2 Use and role of algae products for enhancing plant growth and development.

10.3.4.1 Biofertilizer/biostimulants

Nitrogen, phosphorus, potassium, carbon and some trace elements are essential for plants for better growth, development and productivity and their deficiency can be corrected by applying ample biofertilizers. Because long-term and excessive usage of chemical or synthetic fertilizers leads to deposition of heavy metals in the cultivation land and eventually cause ecosystem imbalance (Ritika & Utpal, 2014). Biofertilizers comprise of living or dormant microbes alone or in combination, which improve the chemical and biological traits of soils, refurbish the soil fertility and enhance plant growth (Ammar *et al.*, 2022). The leftover crude of defatted or residual biomass after the extraction of value-added products can be used as biofertilizers and hence, reduce the production cost (Guo *et al.*, 2020). Microalgal extract and their biomass (such as *Spirulina* sp., *Chlorella* sp. and Cyanobacteria) itself can be used as suitable biofertilizer source (Balasubramaniam *et al.*, 2021). They are being regarded as the major organic matter sources in the agro-ecosystem as they can produce several polysaccharides, plant growth hormones, antibacterial chemicals and other metabolites required for plant growth (Guo *et al.*, 2020; Ronga *et al.*, 2019). Algae are capable of photosynthesis and carbon dioxide sequestration; hence, they can add enough organic carbon to the soil. Similarly, cyanobacteria contain heterocysts in their cell, thus enabling atmospheric nitrogen fixation. In this regard, some mutant strains of cyanobacteria can also be employed to enhance their resistance towards harsh, extreme, or unfavorable conditions or their efficiency for stimulating growth of different plants. A biofertilizer obtained from blue green algae (BGA) in the brand name 'Algalization' is commercialized having great economic viability in paddy cultivation. This helps to fix nitrogen under anaerobic conditions and deposits about 25–30 kg N/ha/season which enhances the crop yield by 10–15% (Mehta *et al.*, 2018).

According to Bilal *et al.* (2017), it is worth noting that microalgae could supply a set of plant-protecting biological substances which can enhance germination percentage, stem and leaf growth and flowering. They can also be used as plant biostimulants for seed germination (Stirk & van Staden, 2020). A few algal extracts are available in the market as commercial biofertilizers for plants in the name of Acadian (Canada), Seamac Ultra Plus Liquid and Turfcomplex (UK), Ekologik R (Chile), Maxicrop (UK), Kelpak 66 (South Africa), Seasol (Australia), Göemill (France), Algamino Plant (Poland), SeaCrop16 (USA) and Actiwave R (Italy).

10.3.4.2 Plant growth-promoting substances/hormones

After extraction of high-value products, some of the nutrients remain in the processed/residual biomass that can be used as biofertilizer for the growth of plants. Some algal extracts can be obtained by the extraction in water simply by boiling, autoclaving and homogenization and it has great application in modern agriculture. They can be used to promote health, growth and crop yield of many cereals and vegetable plants due to the availability of numerous biological components in them. Those extracts can be applied on both soil and plants as well as used as hydroponic solutions or foliar applications (Ali *et al.*, 2022).

Algae are also considered as rich sources of plant growth promoting hormones or substances, that is, cytokinins, auxins, gibberellins, abscisic acid, ethylene, betaine and polyamines (Ammar *et al.*, 2022). Extracts from some specific algal strains can be used effectively and commercially as growth stimulants and as amendments in agricultural crops and crop production systems (Ronga *et al.*, 2019). It has been documented that few microalgae are a rich source of phytosterols belonging to the steroid group having specific biological activities (Fernandes & Cordeiro, 2021; Luo *et al.*, 2015). A study has been done by Plaza *et al.* (2018) regarding the phytohormone content in *Scenedesmus* sp. and *Arthrospira* sp., where they have found that *Scenedesmus* sp. showed higher concentrations of phytohormones as compared to *Arthrospira* sp. Another study has shown the impacts of *Aulosira fertilissima* on the growth of rice (*Oryza sativa* L.) and reported the occurrence of root-promoting hormones (auxins, cytokinins and gibberellic acid) that induced increased growth of rice seedlings (Ronga *et al.*, 2019). Another study showed the enhanced effects of *Chlorella thermophilla* biomass on rice plant (*Oryza sativa* L.) seedlings, grown on chromium-enriched soil (Majhi & Samantaray, 2021).

Similarly, a herbicide-resistant mutated strain of *Anabaena variabilis* was designed that showed plant growth promoting activity in rice plants (Singh & Datta, 2007).

Cyanobacterial strains like *Tolypothrix* sp., *Anabaena* sp., *Aulosira* sp. and *Nostoc* sp. can maintain soil fertility, physico-chemical properties and fix atmospheric nitrogen with some positive effects on plants and soil (Song *et al.*, 2005). Symbiotically, *Azolla* and *Anabaena* sp. provide various growth-promoting components such as indole-3-acetic acid, 3-methyl indole, indole-3-propionic acid and vitamin B₁₂ and it adds about 60 kg/ha of nitrogen to the soil. It has been reported that dry microalgal biomass possesses around 1% phosphorus and 7% nitrogen (Wijffels & Barbosa, 2010). Moreover, the pyrolysis of algal biomass results in the formation of biochar, which can be a promising source of agricultural biofertilizer, bioenergy production and CO₂ sequestration (Mona *et al.*, 2021).

10.3.4.3 Biopesticides

Pesticides include insecticides, herbicides and fungicides and are mostly applied in agriculture to control pests and pathogens to get high crop yields (Abu-Ghosh *et al.*, 2021). Vigorous application of synthetic pesticides leads to several environmental problems, ground water contamination toxicity to humans and animals and induce harmful transformation on non-target pests (Rani *et al.*, 2020; Yadav & Sharma, 2019). Biopesticides are well known for their activity against plant pathogens which typically possess antioxidant, antimicrobial, antiviral and antifungal properties as well as encourage crop development (Gonçalves, 2021). In this regard, some strains of green algae and cyanobacteria are regarded as the most effective biocontrol agents against fungal pathogens and several soil-borne diseases and can increase the defence mechanisms in plants (Renuka *et al.*, 2018). Chlorellins, from *Chlorella* sp., is an algal isolated bioactive compound having pesticidal effects against pathogenic bacteria (Gupta *et al.*, 2013). Cryptophycin 1 from *Nostoc* sp. (ATCC 53789) is another biocontrol agent found to be most active against fungi and yeasts due to antimitotic and antiproliferative activities (Abu-Ghosh *et al.*, 2021).

10.3.5 Construction sector

The total energy consumption by the building sector is about 40% and it annually contributes up to 30% of the global GHGs emissions. Furthermore, it is also expected that GHG emissions from buildings become double over the next 20 years. Therefore, the vindication of GHG emissions from buildings is one of the utmost requirements of every national climate change strategy which needs holistic approaches to recover building energy performance (Elrayies, 2018; UNEP, 2009). The application of algae in architecture has so many benefits in terms of reducing carbon dioxide emissions, energy saving, oxygen generation, biofuel production, wastewater treatment at micro and macro level using building facades and creating urban space (Ilvitskaya & Chistyakova, 2020). Implementation of algae can reduce the overuse of agricultural land and transportation cost.

New techniques are now introduced to design unique photobioreactors to convert natural resources into energy. Holistic urban spaces, building façade and individual small architectural buildings, integrated with vertical flat panel, helical tubes and tube panel photobioreactors are the major contributions of microalgae-based photobioreactor systems (Ilvitskaya & Lobkova, 2018; Pruvost *et al.*, 2016). The most famous photobioreactor integrated building blocks are Process Zero Concept Building (Los Angeles, California, USA), B.I.Q House (Hamburg, Germany) and Photo. Synth.Etica (Dublin, Ireland). Similarly, the formation of integral urban spaces involves Alga Energety City (Istanbul, Turkey), Carbon T.A.P. (Tunnel Algae Park) (Philadelphia, USA) and Culture Urbaine (Geneva, Switzerland). Moreover, Urban Algae Canopy, EcoLogicStudio, living beings (by Jacob Dunias and Ethan Frier) and Street lamp (by Pierre Callech) are the most known small microalgal architectural forms. Another unique architectural and spatial construction is the Algae Dome culturing *Spirulina* sp. inside it, a four-meter-high bioreactor presented at the CHART art fair in Copenhagen (Denmark). There are some major factors that should be given importance during this kind of construction (Figure 10.3). They are day light performance, potential visibility, capital cost, environmental viability, thermal performance and acoustical performance (Elrayies, 2018).

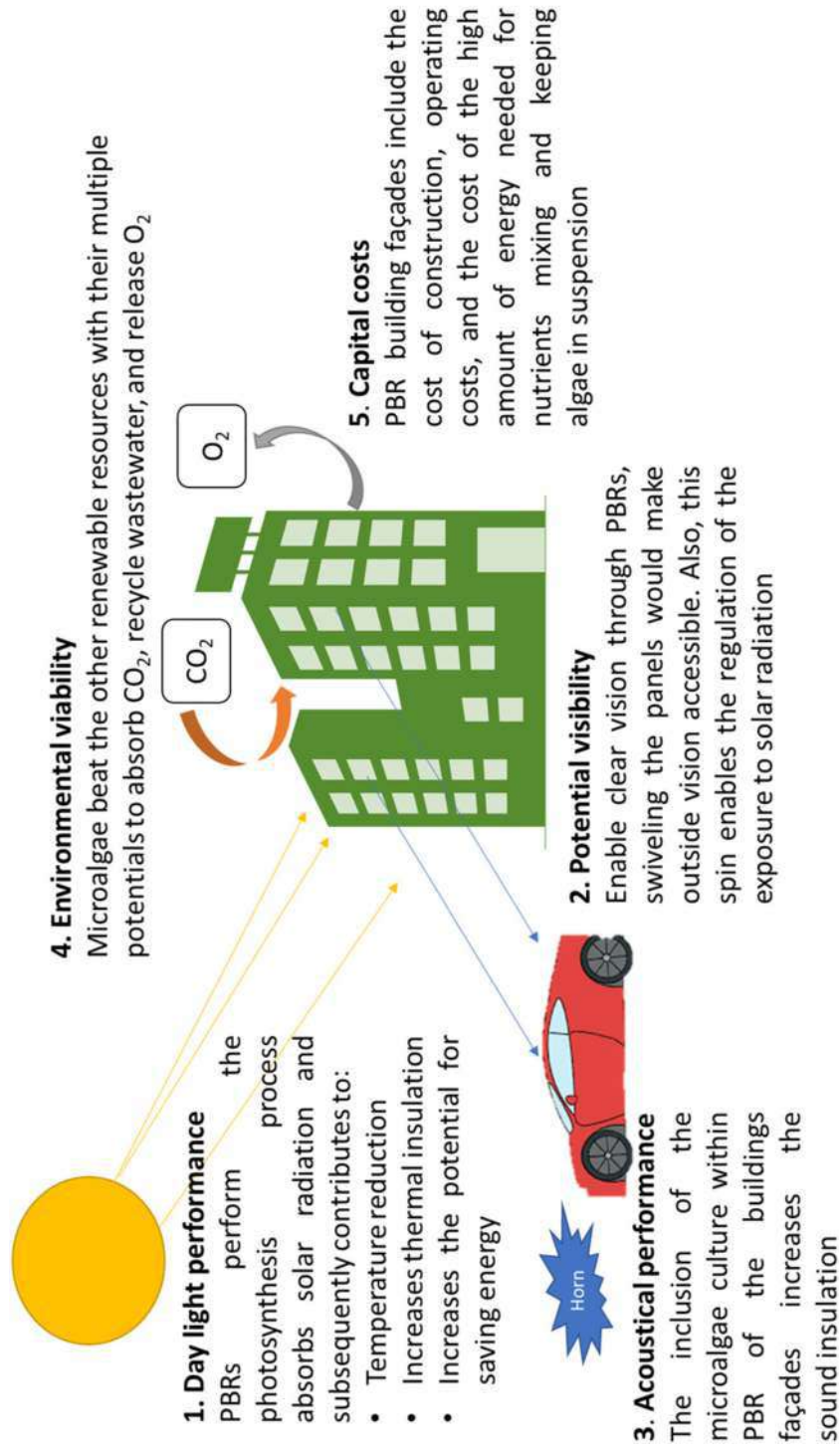


Figure 10.3 Potential factors taken into consideration while using algal photobioreactors as construction material.

10.4 CONSTRAINTS OF ALGAL BIOMASS PRODUCTION AND APPLICATION

Microalgae are a potential source of fuel, fodder and food yield (Table 10.3). The nutrient media required for cultivation and the energy-demanding methods of harvesting microalgal culture have a high recurring rate which is the major obstacle to the improvement of algal technologies. From the biotechnological point of view, microalgae need significant investigation. Hence, extensive innovation is required in various sectors of algal biotechnologies. This lacuna leads to the failure of algal research, although more than a thousand algal species are available worldwide. Among the 10 000 identified species, only a few are investigated to date for their potential chemical composition and very few are cultivated on an industrial scale. Similarly, genetic modification of microalgae along with their cultivation mode is another important goal in the field of algal research. The past few decades have accepted the use of microalgal biomass and their biomolecules in the improvement of many innovative food and other commercial products.

Table 10.3 Global production of nutraceutical products from different algal strains.

Microalgae Genus	Main Producers	Products
<i>Spirulina</i> sp.	Myanmar <i>Spirulina</i> Factory (Myanmar) https://www.spirulinasource.com/slideshows/myanmar-spirulina/ Cyanotech Corp. (USA) https://www.cyanotech.com/ Earthrise Nutritionals (USA) https://www.earthrise.com/ Pondicherry Spirulina Farms (India) http://www.pyfarms.com/	Tablets, pasta, chips and liquid extract Tablets, beverages, powders, extracts Tablets, powders, extracts Powder, capsules
<i>Chlorella</i> sp.	Hainan Simai Pharmacy Co. (China) https://www.chinafirm.biz/company-simai-pharmaceutical-haikou-35358 Taiwan <i>Chlorella</i> Manufacturing Co. (Taiwan) https://www.taiwanchlorella.com/ Algomed, Klotze (Germany) https://www.algomed.de/en/homepage/	Powders, extracts Nectar, tablets, powders, noodles Powders
<i>Haematococcus pluvalis</i>	Parry Nutraceuticals Ltd. (India) https://www.parrynutraceuticals.com/ Britannia Health Products Ltd. (U.K.) https://www.britannia-pharm.co.uk/ Nutrex Hawaii (USA) https://www.nutrex-hawaii.com/	Soft gel, oleoresins and beadlets Capsule Soft gel
<i>Dunaliella bardawil</i>	AquaCarotene Ltd. (Australia) http://www.aquacarotene.com/ Betatene® (Australia) https://www.apfoodonline.com/industry/betatene-australias-own-natural-beta-carotene/ Cyanotech Corporation (USA) https://www.cyanotech.com/	Whole-dried biomass Tablet, soft gel, powders, capsule Capsule, soft gel, oil
<i>Aphanizomenon flos-aquae</i>	Vision (USA) Blue Green Foods (USA) https://bluegreenfoods.com/	Powder, capsules, crystals Capsules, crystals

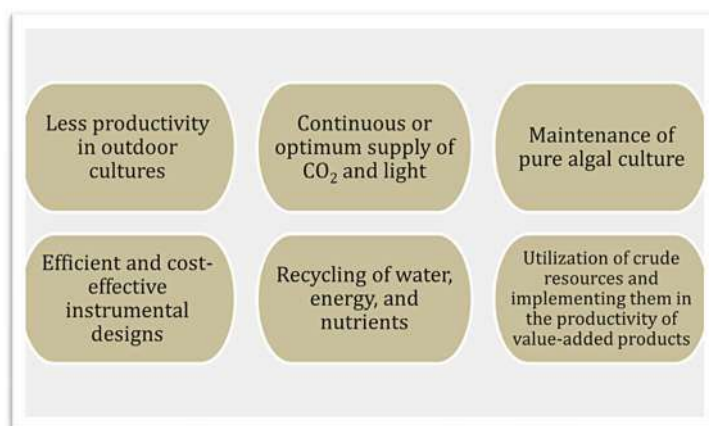


Figure 10.4 Major challenges in growing microalgae at large scale.

Although the microalgae-based product market is steadily expanding now, it is still not profitable as its substantial growth is troubled by the manufacturing techniques employed. Cost-effective production and optimized recovery operation are the two major challenges of this time (Figure 10.4). For example, the harvested biomass's wet slurry contains nearly 75% water which needs to be extracted using electrical or mechanical energy. Then, the dried biomass undergoes an extraction process to harvest the desired products. To date, there is not a single extraction technique that is commercially feasible. Moreover, the yield and nature of the desired product are also influenced by cell disruption techniques, which further require specific optimization steps. The biorefinery concept will be economically beneficial only when the extracts of biomass and the biomass itself can be utilized to produce commercially attractive value-added products.

10.5 CONCLUSION

Microalgae are sustainable and precious resources that have an ideal role in biofuel production, wastewater treatment and applications in agriculture, nutrition, pharmacy and the construction sector. They have high productivity properties and can expand even in wastelands. The growing population has created an opportunity for finding more suitable sustaining solutions. Changing the way of life needs development to create alternative options to provide both nutritional and health security in an eco-friendly and economical manner. Though microalgae were used many years ago to nourish their culture and harvest, they are growing rapidly now. If this continues to expand, then only the revolutionary changes in the pharmaceutical, cosmetics, energy and food industries can be performed in the coming years. The unique chemical contents in microalgae provide various functional ingredients, leading to the synthesis of high-value-added products. Additionally, there are some algal toxins, heavy metals and undefined compounds present in algal biomass which need profound research regarding them to address their deleterious effects on their consumption. Hence, this chapter provides specific data based on the available microalgal resources, which should further focus on the economical and nutritional improvement of microalgal applications.

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Chapter 11

Production of biopolymers from microalgae and cyanobacteria

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ABSTRACT

Over the past few decades, plastic-derived pollution has been recognized as a major environmental issue because the use of conventional plastics results in vast amounts of waste as well as in fossil-fuel depletion. Biodegradable and biobased polymers are a promising alternative to conventional plastics. In this context, polyhydroxyalkanoates (PHAs) are bioplastics with similar mechanical and thermal properties to petroleum-based plastics which can be used in a wide range of applications. Several studies have reported the accumulation of PHAs in the biomass of microalgae and cyanobacteria. Under optimal conditions for PHA accumulation, that is, nutrient limitation, and optimal light intensity, PHA content can significantly increase, achieving 85% of dry biomass weight. Downstream recovery of PHAs is also a critical step that affects the properties and the yield of PHAs. Bioplastic production from microalgae and cyanobacteria on a commercial scale is still limited due to its high cost, with the cultivation medium accounting for up to 50% of the total production cost. The use of wastewater as a growth medium can improve the economic feasibility and sustainability of PHA production from microalgae and cyanobacteria and contribute to a more circular economy.

Keywords: biodegradable bioplastics, bioplastic recovery, biorefinery, cyanobacteria, downstream processing, microalgae, PHA blends, polyhydroxyalkanoates, sustainability, upstream processing, wastewater.

11.1 INTRODUCTION

Plastic has made our lives more convenient. This increased convenience has led to an increase in demand, which, in turn, caused an exponential increase in the production of plastics since the beginning of their industrial production, resulting in ~8 billion tons of plastic generated from 1950 onward (European Environmental Agency, 2020). The annual production of plastics steadily increases at a

yearly average of 4%, from 279 million tons in 2011 to 391 million tons in 2021 (Statista, 2023). Plastics are polymeric substances, the properties of which depend on the structure of individual monomers and range from flexible to stiff, from permeable to impermeable, from hydrophilic to hydrophobic. Conventional plastics are derived from fossil-based chemicals, and are a cheap solution for strong and durable materials. Commonly used plastics include polyethylene (PE), polyethylene terephthalate (PET), polypropylene (PP), polyvinyl chloride, polystyrene (PS), and polyamides (nylons), and are constituents of a wide variety of products, including medical equipment, agricultural tools, electronic devices, and packaging (Leal Filho *et al.*, 2019).

The unblemished optimism regarding plastics changed around the 1970s. Due to their short usable life and their non-biodegradable nature, the accumulation of plastic in the environment became hard to miss, thereby damaging their reputation (Carpenter & Smith, 1972). From then on, the view on plastics has drastically changed. It is now well-known that due to their long lifetime, for instance up to 800 years as the average reported lifetime for PET (Ward & Reddy, 2020), plastics accumulate in the environment if not properly handled. In more tangible terms, estimations show that the primary plastic waste generation amounted to ~7,500 million tons in 2020, whereas over 200,000 tons of plastic waste enter the Mediterranean Sea every year (European Environmental Agency, 2020), with an economic cost ranging between \$3,300 and \$33,000 per ton marine plastics per year (Beaumont *et al.*, 2019). When improperly disposed of in the environment, plastics break into small insoluble pieces, referred to as microplastics (with diameters between 1 μm and 5 mm), the size of which makes them difficult to track, trace, and remove (Tirkey & Upadhyay, 2021). Microplastics have thereby entered the food chain (especially via marine animals) and are even found in women's placentas (Ragusa *et al.*, 2021). In addition to this repulsive fact, plastic production consumes ~6% of the global crude oil supply and is responsible for the generation of 2% of the global carbon dioxide (CO₂) emissions (Rosenboom *et al.*, 2022). Therefore, apart from the global health, it is crucial to pursue alternative solutions that also tackle the environmental impact. All these facts call for drastic changes regarding the generation, use, and disposal of plastics.

Biobased bioplastics, polymers derived from biological sources, can be a more sustainable alternative to conventional plastics (European Bioplastics, 2018). They can be divided into two categories, namely (1) biodegradable, for example, polylactic acid (PLA) derived from lactic acid, polyhydroxyalkanoates (PHAs), cellulose, and starch-based bioplastics, and (2) non-biodegradable, such as organic PE and PET (Rosenboom *et al.*, 2022). Advantages of bioplastics over conventional plastics include improved circularity due to the use of renewable resources, lower environmental footprint, biodegradation, and improved properties, which depend on the specific bioplastic type (Rosenboom *et al.*, 2022). Life-cycle assessments show that the substitution of conventional plastics with bioplastics, even from first-generation biofuels, requires 86% less non-renewable energy (Singh *et al.*, 2022). The production of fully biobased bioplastics is currently estimated at ~2 million tons per year (Chen, 2019), and they are expected to play a key role in future circular economy (Cheng & Gross, 2020). In this context, the biomass of microorganisms is increasingly gaining interest as a raw material for biobased products such as bioplastics. Microalgae and cyanobacteria are two microbial groups that have gained a significant share of the attention for this application, due to their potential bioplastic production from recovered resources such as nutrients and organics from wastewater, or CO₂ from off-gasses as well as their high content in targeted biopolymer precursors (Mastropetros *et al.*, 2022).

11.2 STRUCTURE AND PROPERTIES OF BIODEGRADABLE BIOPLASTICS

Biodegradable bioplastics include a range of materials derived from biological processes such as agriculture-derived polysaccharides (e.g., starch- and cellulose-based bioplastics) (Abe *et al.*, 2021), microbial fermentation products (e.g., lactic acid for PLA), and intracellular microbial components (e.g., PHAs), while the feasibility of converting the whole microbial biomass into bioplastic composites has recently been shown as well (Singha *et al.*, 2021). Starch- and cellulose-based bioplastics are

interesting due to their abundance, affordability, durability, strength, and biodegradability (Abe *et al.*, 2021; Nanda *et al.*, 2022). Even though cellulose-based biopolymers are water-sensitive and lack interfacial adhesion and thermal stability, research shows that pretreatment can overcome these challenges and increase the popularity of these polymers (Polman *et al.*, 2021). Applications of cellulose-based polymers include packaging films, frames for eyeglasses, and food packaging (Nanda *et al.*, 2022). Starch-based polymers are considered to be promising to produce edible films and have similar mechanical properties and transparency to conventional polymers (Shahabi-Ghahfarrokhi *et al.*, 2019). Similar to cellulose-based polymers, starch-based polymers are also sensitive to moisture, do not have optimal mechanical properties and thermal stability, are gas permeable, and have odor issues (Nanda *et al.*, 2022; Toh *et al.*, 2008). However, combination with other polymers, essential oils, fibers, or plasticizers improves their properties (Syafiq *et al.*, 2020), and enables applications in food-packaging and pharmaceutical fields. Lactic acid monomers are further polymerized to yield PLA, a non-toxic, biocompatible polymer with mechanical properties similar to PET and PS (Karamanlioglu *et al.*, 2017). Owing to its stiffness, mechanical strength, flexibility, thermal stability, lower temperature heat sealing ability, aroma, and flavor resistance, PLA finds applications, among others in food packaging, agriculture, transportation, furniture, electronic appliances, and fabrics (Jamshidian *et al.*, 2010). Finally, the versatility and durability of PHAs has placed them in the spotlight, and their market is increasing, with projections showing an increase from 81 million USD in 2022 to 167 million USD in 2027 (Markets and Markets, 2022).

Microalgae and cyanobacteria produce various types of PHAs, including polyhydroxybutyrate (PHB), poly-3-hydroxybutyrate (P(3HB)), and co-polymers such as poly(3-hydroxybutyrate-co-3-hydroxyvalerate) (PHBV) (Mastropetros *et al.*, 2022). These PHAs have properties comparable to conventional plastics such as PP and PE and find applications in the food and bulk-packaging sectors. Furthermore, due to their high biocompatibility and complete biodegradability, they can have high-value applications in the biomedical sector (Costa *et al.*, 2019; Koller, 2018; Paulraj *et al.*, 2018). PHB, the most prevalent PHAs, has a higher melting point and a comparable tensile strength compared to PP and PS (Khanna & Srivastava, 2005). Nevertheless, the low flexibility (i.e., elongation at break), and high brittleness and crystallinity limit the potential applications, excluding their conversion to durable materials (Muneer *et al.*, 2020). Especially regarding crystallinity, levels above 50% yield brittle polymers and are therefore undesirable (Laycock *et al.*, 2013), with microalgal and cyanobacterial PHAs approaching this range (Table 11.1). Additionally, the temperature at which PHB undergoes thermal degradation is very close to its melting point, which causes failures in many applications (Aydemir & Gardner, 2020). Therefore, medium-chain PHA (6–14 carbon atoms) or co-polymers are preferred because they present improved properties (Table 11.1). These properties are correlated with the molecular weight and structure of the monomers (Bugnicourt *et al.*, 2014), the composition of which is determined by the genetic potential of the microorganisms to produce them. Nevertheless, common chemical modification methods have been shown to improve the properties of these microalgal and cyanobacterial PHAs and are recommended to improve the properties and increase the number of applications.

Microalgae and cyanobacteria that are able to produce PHAs have been recently reviewed and summarized by Mastropetros *et al.* (2022), and species with the highest content (up to 78%) belong to the genera *Arthrospira* sp., *Synechocystis* sp., *Synechococcus* sp., *Nostoc* sp., and *Anabaena* sp. Importantly, microalgal and cyanobacterial PHAs can be produced on side-streams, further increasing their sustainability. Despite their good prospects, currently there are only a limited number of studies that show the feasibility of this concept and test the properties of microalgae- and cyanobacterial-derived PHAs, which will be discussed in the following sections.

11.3 EMPLOYING MICROALGAE AND CYANOBACTERIA FOR BIOPLASTIC PRODUCTION

Among the different types of bioplastics currently considered as more sustainable alternatives to conventional plastics, biodegradable and biobased PHAs are considered to be a promising solution

Table 11.1 Average physical properties of conventional, fossil-based polymers, and biopolymers that are produced by microalgae and cyanobacteria.

Polymer	Crystallinity (%)	Elongation at Break (%)	Tensile Strength (MPa)	Melting Point (°C)	Glass Transition Temperature (°C)	References
PP	60	400	38	176	−10	Balaji <i>et al.</i> (2013); Hazer and Steinbüchel (2007); Verlinden <i>et al.</i> (2007)
HDPE	70	12	—	129	—	Costa <i>et al.</i> (2019)
PHB	57	6.2	31	173	1.6	Balaji <i>et al.</i> (2013); Verlinden <i>et al.</i> (2007); Koller and Rodríguez-Contreras (2015); Garcia-Garcia <i>et al.</i> (2016); Simonazzi <i>et al.</i> (2021); Bhati and Mallick (2012)
PHBV	53	70	23	153	−2.9	Balaji <i>et al.</i> (2013); Hazer and Steinbüchel (2007); Verlinden <i>et al.</i> (2007); Bhati and Mallick (2012); Samantaray and Mallick (2014)
PHB/PCL (75/25)	58	11	21	169	—	Garcia-Garcia <i>et al.</i> (2017)
PHB/PCL (25/75)	—	125	11	154	—	Przybysz <i>et al.</i> (2018)

Source: Adapted from Mastropetros *et al.* (2022).

PP, polypropylene; HDPE, high-density polyethylene; PHB, polyhydroxybutyrate; PHBV, poly(3-hydroxybutyrate-co-3-hydroxyvalerate); PCL: polycaprolactone. —, not reported.

with similar thermal and mechanical properties to petroleum-based plastics (Bhatia *et al.*, 2021; Medeiros Garcia Alcântara *et al.*, 2020). The first reported microbial production of PHAs dates back to 1926, from the bacterium *Bacillus megaterium* (Możejko-Ciesielska & Kiewisz, 2016). Even though bacteria have been reported to accumulate up to 90% of cell dry mass in PHAs (Obruča *et al.*, 2022), the high demand for organic carbon results in increased costs that pose a challenge in its widespread application. Microalgae and cyanobacteria can be promising alternative ways to produce PHAs. As photosynthetic microorganisms, they can utilize solar energy and CO₂ for their biomass growth while they have low-nutrient requirements (Costa *et al.*, 2019).

11.3.1 Cultivation conditions

Around 100 strains of microalgae and cyanobacteria have been reported to produce PHAs during their growth. Microalgae and cyanobacteria naturally accumulate these biopolymers as a source of carbon and energy. However, the production of PHAs is a complex metabolic process. The biomass productivity as well as the percentage and the type of PHAs that are produced depend on various parameters such as the selected carbon source and the availability of nutrients and light (Bagatella *et al.*, 2022; Cassuriaga *et al.*, 2018).

11.3.1.1 Photoautotrophic, heterotrophic, or mixotrophic operational mode

In response to shifting environmental conditions, microalgae and cyanobacteria employ different metabolic pathways. In cyanobacteria and microalgae, there are three delineated growth mechanisms. During photoautotrophic metabolism, the cells use CO₂ as a source of carbon and light as a source of energy. Under heterotrophic growth, microalgae and cyanobacteria meet their carbon and energy needs by consuming organic substances. Mixotrophic conditions combine both photoautotrophic

and heterotrophic metabolic functions: energy and carbon needs can be covered by light or organic substances and organic or inorganic carbon sources, respectively. The significance of the different metabolic pathways in the cultivation of microalgae lies in their impact on the substrate that is being utilized, the amount of biomass produced, the growth rate, and the macromolecular composition of the cells.

Table 11.2 summarizes PHA production from microalgae and cyanobacteria during their growth by employing natural metabolic pathways. Photoautotrophic microalgae and cyanobacteria can accumulate PHAs (Phalanisong *et al.*, 2021). Apart from the production of these valuable compounds, these photosynthetic microorganisms can capture and utilize atmospheric CO₂ which contributes to carbon fixation (Phalanisong *et al.*, 2021). It has been reported that *Chlorella vulgaris*, *Scenedesmus obliquus*, and *Spirulina* sp. can remove CO₂ with efficiencies of 80, 28, and 53%, respectively (de Moraes & Costa, 2007; Sadeghizadeh *et al.*, 2017). Few studies have reported the production of PHAs from eukaryotic microalgae in a photoautotrophic environment. More specifically, during their cultivation, *Botryococcus braunii*, *Chlorella pyrenoidosa*, and *Chlorella fusca* accumulated PHB at a concentration of 16, 27, and 5.5%, respectively (Cassuriaga *et al.*, 2018; Das *et al.*, 2018; Kavitha *et al.*, 2016b). Unlike microalgae, substantial amounts of PHAs are found in many cyanobacteria grown photoautotrophically. Several cyanobacterial strains such as *Nostoc*, *Synechocystis*, *Synechococcus*, and *Spirulina* naturally synthesize PHB at a content lower than 10% (Sirohi *et al.*, 2021).

Perceptibly, the addition of organic substances (e.g., acetic acid, xylose, glucose, and sucrose) increases the yields of PHB in cyanobacteria and microalgae (Price *et al.*, 2020). With the addition

Table 11.2 PHA production from microalgae and cyanobacteria during their cultivation under photoautotrophic, heterotrophic, or mixotrophic conditions.

Microbial Species	Mineral Medium	Carbon Source	Condition	Type of PHA	PHA Content (%)	References
<i>B. braunii</i>	CHU-13	—	p	PHB	16	Kavitha <i>et al.</i> (2016b)
<i>S. salina</i>	BG-11	—	p	P(3HB)	6.6	Kovalcik <i>et al.</i> (2017)
<i>C. pyrenoidosa</i>	Fogg's	—	p	PHB	27	Das <i>et al.</i> (2018)
<i>Spirulina</i> sp.	Zarrouk	—	p	PHB	21	Martins <i>et al.</i> (2017)
<i>Nostoc ellipsosporum</i>	BG-11	—	p	PHB	19	Martins <i>et al.</i> (2017)
<i>C. fusca</i>	BG-11	—	p	PHB	0.5–5.5	Cassuriaga <i>et al.</i> (2018)
<i>N. muscorum</i>	BG-11	—	p	PHB	8.5	Sharma and Mallick (2005)
<i>N. muscorum</i>	NO ₃ -free BG-11	0.11% acetate + 0.08% propionate	m	PHBV	31	Mallick <i>et al.</i> (2007)
<i>C. fusca</i>	BG-11	0.002% xylose	m	PHB	17	Cassuriaga <i>et al.</i> (2018)
<i>Aulosira fertilissima</i>	BG-11	1% fructose	m	PHB	16	Samantaray and Mallick (2012)
<i>A. fertilissima</i>	BG-11	0.3% acetate	m	PHB	27	Samantaray and Mallick (2012)
<i>Synechocystis</i> sp.	BG-11	0.4% fructose + 0.4% acetate	h	PHB	38	Panda and Mallick (2007)
<i>Chlorogloeopsis fritschii</i>	BG-11	0.06% acetate	m	P(3HB)	15	Zhang and Bryant (2015)

p, photoautotrophic; m, mixotrophic; h, heterotrophic.—, not reported.

of 20 mg/L xylose, the PHB content in *C. fusca* LEB 111 increased to 17% from the 5.3% that was observed in the photoautotrophic culture under the same conditions (Cassuriaga *et al.*, 2018). A significant increase was observed in the PHB content of *Nostoc muscorum* by adding different sources of organic carbon (Sharma & Mallick, 2005).

11.3.1.2 Nutrient availability

Nitrogen is a key component of proteins, nucleic acids, and chlorophyll which are necessary for the structure and function of cells (Zarrinmehr *et al.*, 2020), and therefore is an important macronutrient that affects growth. In microalgae and cyanobacteria cultivation, the availability of nitrogen must be carefully monitored to achieve optimal growth and productivity. Nitrogen can be obtained from various sources including inorganic nitrogen compounds such as nitrate, nitrite, and ammonium, and organic compounds such as urea. Nitrate is the most commonly used source of nitrogen in microalgae (Yaakob *et al.*, 2021).

Phosphorus is another important macronutrient, where nucleic acids, cell membranes, and energy storage molecules such as adenosine triphosphate are among the many cellular structures that depend on it. To promote their growth, microalgae and cyanobacteria can absorb phosphorus in the form of polyphosphate or orthophosphate, with preference for the latter due to easier assimilation (Yaakob *et al.*, 2021).

Table 11.3 presents the PHA content from different microalgae and cyanobacteria strains during their cultivation under nitrogen and phosphorus deficiency. Nitrogen and phosphorus limitation affect both biomass growth and productivity. When these nutrients are limited, the cells redirect the excess carbon toward the biosynthesis of storage compounds such as PHAs, which can be used as an energy and carbon sources under adverse conditions (Costa *et al.*, 2019). Several studies have shown an increase in PHA content in many species under nitrogen and phosphorus starvation, regardless of the cultivation mode (photoautotrophic, heterotrophic, or mixotrophic) (Dang *et al.*, 2022; Troschl *et al.*, 2017; Yashavanth *et al.*, 2021). Kaewbai-Ngam *et al.* (2016) tested 137 cyanobacterial strains for their ability to accumulate PHB. Under nitrogen limitation conditions, PHB yield increased more than 50% of

Table 11.3 PHA production from microalgae and cyanobacteria under nutrient limitation.

Species	Culture Conditions	Nutrient Limitation	Type of PHA	PHA Content (%)	References
<i>Synechocystis</i> sp.	Photoautotrophic	N-deficiency, P-deficiency	PHB	16	Kamravamanesh <i>et al.</i> (2017)
<i>N. muscorum</i>	0.28% acetate, 0.38% glucose, 0.30% valerate	N-deficiency	PHBV	78	Bhati and Mallick (2015)
<i>Scenedesmus</i> sp.	Glucose	P-deficiency	PHB	30	García <i>et al.</i> (2021)
<i>Synechococcus</i> sp.	Photoautotrophic	P-deficiency	PHB	55	Nishioka <i>et al.</i> (2001)
<i>N. muscorum</i>	Photoautotrophic	N-deficiency, P-deficiency	PHB	23	Panda <i>et al.</i> (2005)
<i>N. muscorum</i>	0.20% acetate	N-deficiency, P-deficiency	PHB	35	Sharma and Mallick (2005)
<i>A. fertilissima</i>	0.50% acetate	P-deficiency	PHB	77	Samantaray and Mallick (2012)
<i>Spirulina platensis</i>	0.50% sodium acetate	N-deficiency	P(3HB)	10	Toh <i>et al.</i> (2008)
<i>A. fertilissima</i>	0.26% citrate, 0.28% acetate	P-deficiency	PHB	85	Samantaray and Mallick (2012)

the screened cyanobacterial strains. *Synechococcus* sp., a thermophilic cyanobacterium, accumulated 55% PHB when it was grown photoautotrophically and under phosphate limitation (Nishioka *et al.*, 2001). The PHB content in *N. muscorum* grown photoautotrophically and heterotrophically (0.2% acetate) achieved 22.7 and 35% under nitrogen and phosphate deficiency, respectively (Panda *et al.*, 2005; Sharma & Mallick, 2005).

11.3.1.3 Light

The growth of microalgae and cyanobacteria is in most cases significantly affected by light. Under photoautotrophic and mixotrophic conditions, light is essential for photosynthesis, which produces the required energy for cell growth. Under low-light intensities, the growth of microalgae and cyanobacteria can be limited due to limited photosynthetic activity. Excessively high-light intensities can also have negative effects such as photoinhibition and cell damage. Consequently, the intensity and availability of light affect biomass production as well as the accumulation of valuable compounds such as PHAs. However, the optimal light intensity and periodicity vary based on several factors, including the selected strain and the turbidity of the cultivation medium.

Several studies have investigated the impact of light intensity and alternation of light-to-dark cycles on the productivity of PHAs by microalgae and cyanobacteria (Costa *et al.*, 2019; Price *et al.*, 2020). In the study of Ansari and Fatma (2016), *N. muscorum* was cultivated at a light intensity of 25 $\mu\text{mol}/\text{m}^2/\text{s}$ under three different photoperiods. At 0.4% glucose and in 14:10, 12:12, and 10:14 h light/dark periods, its PHB content was 18, 21, and 24%, respectively. In another study, the PHB content in *C. fusca* increased from 5.3 to 17.4% when light intensity decreased from 58 to 28 $\mu\text{mol}/\text{m}^2/\text{s}$ under a 6:18 h light/dark period, whereas at the same light intensities, the PHB content was 5.5 and 2.7% under a 12:12 h light/dark period (Cassuriaga *et al.*, 2018). Optimizing the light intensity and photoperiod based on the specific strain being cultivated and culture conditions can be an effective strategy to enhance PHA production.

11.3.1.4 Wastewater as a feedstock for microalgae and cyanobacteria cultivation

Wastewater instead of a potential environmental hazard can be seen as a potential source of nitrogen and phosphorus for the growth of microalgae and cyanobacteria and be upgraded to valuable products (Sakarika *et al.*, 2022). The cultivation of these photosynthetic microorganisms in wastewater is a promising alternative to wastewater treatment as high nutrient removal and high biomass productivity can be achieved (Rizwan *et al.*, 2018). Cultivation of *S. obliquus* in soybean wastewater removed 72% of chemical oxygen demand, 95% total nitrogen, and 54% total phosphorus (Shen *et al.*, 2020). Similarly, *C. vulgaris* cultivated in meat wastewater removed 89% of chemical oxygen demand, 52% of total nitrogen, and 70% of total phosphorus (Hu *et al.*, 2019).

The cost of PHA production from microalgae and cyanobacteria is high compared to the conventional plastics industry. High feedstock and water requirements account for more than 50% of the production cost (Medeiros Garcia Alcântara *et al.*, 2020). To enable cost-effectiveness and feasibility on a larger scale, scientific interest has focused on the utilization of wastewater as a substrate for the cultivation of microalgae and cyanobacteria. Apart from the reduction of the upstream cost of the process and the bioremediation of wastewater, using wastewater as feedstock does not compete with raw materials such as sugars, which can also be used for PHA production (Medeiros Garcia Alcântara *et al.*, 2020). Studies have demonstrated that during the cultivation of microalgae and cyanobacteria in wastewater it is feasible to produce substantial amounts of PHAs with similar properties to conventional plastics. A PHB yield of 247 mg/L was reported by *B. braunii* grown in sewage wastewater at a concentration of 60% (Kavitha *et al.*, 2016a). In another study, *Synechocystis* sp. grown on shrimp wastewater accumulated PHB at a concentration of 33% while the removal efficiency of phosphate was 97% (Krasaesueb *et al.*, 2019).

Table 11.4 summarizes the production of PHAs from microalgae and cyanobacteria cultivated in different types of wastewater. Among the different wastewater types, anaerobic digestion effluents are

Table 11.4 PHA production from microalgae and cyanobacteria cultivated in wastewater.

Species	Type of Wastewater	Temperature (°C)	pH	Type of PHA	PHA Content (%)	References
<i>Synechocystis</i> sp.	Shrimp wastewater	27–30	7.0–9.0	PHB	34	Krasaesueb <i>et al.</i> (2019)
<i>N. muscorum</i>	Poultry litter	25 ± 2	7.0	PHB	23	Bhati and Mallick (2016)
<i>N. muscorum</i>	Poultry litter + 10% CO ₂	25 ± 2	7.0–8.0	PHBV	65	Bhati and Mallick (2016)
<i>S. salina</i>	Digestate supernatant	25 ± 1	—	PHB	6.3	Meixner <i>et al.</i> (2016)
<i>Synechocystis</i> sp.	30% palm oil mill effluent + BG-11 medium	28	8.2	PHB	15	Nur <i>et al.</i> (2023)
<i>Synechococcus</i> sp.	30% palm oil mill effluent + BG-11 medium	28	8.2	PHB	15	Nur <i>et al.</i> (2023)
<i>B. braunii</i>	50% palm oil mill effluent + glycerol + Fe-EDTA*	30	7.5	PHB	33	Nur (2022)

*EDTA, ethylenediaminetetraacetic acid; —, not reported.

generated in high volumes urging the need to implement a more sustainable disposal way than the current use as fertilizer. Digestates can be upgraded to higher value products when used as a substrate for the cultivation of microalgae and cyanobacteria and the production of value-added compounds as they contain high organic matter and are rich in nutrients such as ammonium-nitrogen and phosphorus (Kaur *et al.*, 2020; Koutra *et al.*, 2018). Only a few studies have investigated the production of PHAs from microalgae and cyanobacteria in digestates. For instance, *Synechocystis salina* was cultivated in diluted digestate and accumulated PHB at a concentration of 6.3% (Meixner *et al.*, 2016).

Overall, the cultivation of microalgae and cyanobacteria in wastewater seems to be an environmentally friendly and promising alternative for sustainable wastewater treatment and production of PHAs. However, fluctuations in the composition of the produced wastewater and the presence of potentially hazardous components can affect the entire process and even inhibit the growth of microalgae and cyanobacteria. Further studies to address these challenges are necessary before the implementation of the process at a larger scale (Mastropetros *et al.*, 2022; Medeiros Garcia Alcântara *et al.*, 2020).

11.3.2 Advantages of PHA production from microalgae and cyanobacteria compared to bacteria

PHAs are naturally produced by various microorganisms, with bacteria in the genera *Pseudomonas*, *Ralstonia*, *Bacillus*, and *Aeromonas* accumulating PHAs at high content. However, PHA production using bacteria is expensive and not feasible for large-scale applications due to the prohibitive cost of the organic carbon sources and oxygen requirements (Mozejko-Ciesielska & Kiewisz, 2016; Samantaray & Mallick, 2015).

On the contrary, microalgae and cyanobacteria seem to be promising microorganisms for PHA production, utilizing atmospheric CO₂ and generating energy through photosynthesis. These photosynthetic microorganisms do not need exogenous organic sources for their biomass growth reducing the overall cost of the process by up to 50% (Medeiros Garcia Alcântara *et al.*, 2020; Phalanisong *et al.*, 2021). Microalgae and cyanobacteria can utilize CO₂ that is present in flue gases for PHA production, providing a sustainable solution to greenhouse gas emissions and making the process economically feasible. For instance, *S. salina* and *Synechococcus elongatus* directly utilizing

industrial flue gases accumulated PHB at a content of 6.6 and 11%, respectively (Roh *et al.*, 2021; Troschl *et al.*, 2017).

Moreover, microalgae and cyanobacteria can produce more than one bioproduct. For instance, they can accumulate high amounts of lipids, proteins, polyunsaturated fatty acids, and pigments that can be utilized as raw materials for producing bioenergy and other valuable products used in a variety of sectors including food, cosmetics, nutraceutical, and pharmaceutical industries (Kumar *et al.*, 2020). The implementation of a biorefinery concept is a complex procedure, where both upstream and downstream processing can significantly affect the entire process. The type and yield of the produced compounds strongly depend on the selected strains and the cultivation conditions while developing effective methods for extracting and purifying the various compounds from the microbial biomass is challenging and requires a considerable amount of energy (Siddiki *et al.*, 2022). There are only a few experimental data available for the simultaneous production of PHAs and other valuable compounds in a biorefinery concept. *Arthrospira platensis* cultivated in palm oil mill effluent was investigated for the co-production of PHB and C-phycocyanin. Results showed that the productivities of PHB and C-phycocyanin using 50% palm oil mill effluent were 7 and 16 mg/L/day, respectively (Nur, 2022). Another study demonstrated the possibility of cultivating *Synechocystis* sp. in secondary effluent to produce PHB and lipids (Senatore *et al.*, 2023). The ability of microalgae and cyanobacteria to utilize flue gases and wastewater to produce PHAs as well as other valuable compounds in a biorefinery concept can render the microalgae cultivation technology at a large scale economically feasible and environmentally friendly.

11.3.3 PHA blends

PHAs present several environmental benefits as bio-based and biodegradable polymers. However, their industrial application is still limited due to their high production cost. Additionally, for PHA production to become competitive with the petroleum-based plastic industry, the produced biopolymer must have similar properties to conventional plastics. To overcome these limitations, blending PHAs with raw materials and other biodegradable polymers has emerged as a promising and simple approach. The type and properties of the produced polymer blends depend on the choice of the starting constituents and their blending ratio. PHA blending aims to improve the mechanical properties, such as increased tensile strength, elongation at break, and impact resistance, that can be used in a wide range of applications, enhance the biodegradability of the material, reduce the cost, and improve the overall performance (Kumar *et al.*, 2021).

11.3.3.1 PHA blends with raw materials

Starch is considered a highly promising natural polymer because it is biodegradable and widely available in large quantities. PHA/starch blends have improved mechanical properties compared to pure PHAs. In the study of Asl *et al.* (2021), an electrospinning method was used to blend PHB with different concentrations of starch (5–15 wt%). By adding starch at a concentration of up to 10%, the tensile strength of the PHB/starch scaffolds increased from 3 to 16 MPa. The presence of starch also enhanced the thermal stability and degradation rate. The results of this study suggest that electrospun scaffolds produced from PHB/starch could be used in bone tissue engineering applications. In another study, the blend of PHB with modified corn starch was investigated. When the starch concentration increased, an increase in glass transition temperature from 2 to 37°C was observed (Lai *et al.*, 2015). The mechanical and thermal properties of PHA/starch blends can be significantly improved with the addition of cross-linking agents such as citric and adipic acids (Sun *et al.*, 2018).

Lignin is a complex organic polymer and the second most abundant renewable natural polymer on the Earth. PHA blends with lignin offer a promising approach to the development of new materials with improved mechanical properties and biodegradability compared to either material alone. Therefore, by combining lignin with PHAs, it is possible to create new materials with a range of desirable properties that can be used in various applications (Kumar *et al.*, 2021). Lugoloobi *et al.* (2020) reported that

PHB/lignin blends showed higher glass transition temperatures, improved ultraviolet resistance and tensile performance, and higher melt viscosity making them suitable for packaging applications.

Cellulose derivatives are becoming increasingly popular as components that can be blended with PHAs due to their compatibility and their ability to accelerate the degradation of PHAs. Cellulose, acetate, butyrate, ethyl cellulose, and cellulose propionate are cellulose derivatives that are commonly used as drug carriers, blood coagulants, and coatings for pharmaceutical tablets (Sharma *et al.*, 2021). Cellulose-based microfibers (MFs) can be used to enhance the properties of PHA films. According to Mármol *et al.* (2020), the addition of MFs made the PHA film 23% more durable, as both the tensile strength and Young's modulus increased. Overall, PHA blends with raw materials result in new biopolymers with improved properties when compared to either of the individual components.

11.3.3.2 PHA blends with biodegradable polymers

PLA is a biodegradable polymer derived from renewable resources. Blending PHAs with PLA is the most studied approach as it can result in material with improved mechanical properties and biodegradability, with the specific properties depending on the PHA to PLA ratio. The PHA/PLA blend has been used in three-dimensional printing, where the printed materials exhibited favorable mechanical properties and thermal stability (Ausejo *et al.*, 2018). In another study, the PHA/PLA blend was demonstrated to have the capacity to absorb oil from water, which is similar to that of currently utilized absorbents (Iordanskii *et al.*, 2019).

Polycaprolactone (PCL) is a synthetic biodegradable polymer with a low melting point. Blending PCL with other biopolymers, such as PHAs, can improve biodegradability and decrease the production cost. The degradation rate of the blend as well as the mechanical properties can be controlled by adjusting the ratio of PCL to PHA. For instance, higher maximum stress was exhibited in blends rich in PHB, whereas blends rich in PCL led to greater strain at break. The PCL/PHA blend can be utilized in various biomedical applications, especially in tissue engineering (Kumar *et al.*, 2021; Li *et al.*, 2016).

Blending PHAs with poly(butylene adipate-co-terephthalate) (PBAT) is another approach to developing biodegradable polymer blends with improved properties and increasing the field of their application (Tian & Wang, 2020). Similar to PCL, PBAT is a synthetic biodegradable polymer. The blend can be processed using common techniques such as injection molding and extrusion. During injection molding of PHBV with PBAT, as the PBAT content increased the toughness and strain at break increased, while the specific modulus and strength decreased (Javadi *et al.*, 2010). In another study, the addition of PBAT increased the shear storage modulus of the PHB/PBAT blends and decreased the tensile storage modulus (Larsson *et al.*, 2016). Overall, blending PHAs with other biodegradable polymers is a promising alternative to reduce production costs and develop new materials with improved properties and biodegradability, making them attractive for a wide range of applications.

11.4 DOWNSTREAM PROCESSING OF BIOPLASTIC RECOVERY FROM MICROALGAE AND CYANOBACTERIA

During the past few years, increased research efforts aim at developing more efficient harvesting, pretreatment, and extraction techniques, with the hope of lowering the cost of microalgal bioplastics production. These efforts include exploring and evaluating various methods and technologies that can be used to optimize the downstream process. By reducing the costs associated with these production stages, the development of sustainable bioplastics derived from microalgae and cyanobacteria can become economically viable and contribute to a more sustainable future. In addition, this can also lead to the development of new and more efficient approaches to produce other valuable products from these microorganisms toward a biorefinery concept.

11.4.1 Harvesting

The relatively low final biomass concentrations in microalgal and cyanobacterial cultures (with values from 0.5 g/L in open-pond systems to 5 g/L in photobioreactors), resulting from light restriction due to shading from cell growth, lead to the urge for separation of the biomass from a large water volume (Pahl *et al.*, 2013; Vandamme *et al.*, 2013). To achieve the desired solid–liquid separation during harvesting, various mechanical-, chemical-, biological-, or electrical-based techniques can be used through one or more steps (Mata *et al.*, 2010; Morais Junior *et al.*, 2020). The selection of an appropriate harvesting method depends on several factors, such as the microalgal cell morphology (e.g., filamentous, spherical, or elongated), the biomass concentration in the culture medium, the specific gravity, and size (typically microalgae will be in the range of 0.5–200 μm) of cells (Caroppo & Pagliara, 2022; Gerardo *et al.*, 2015; Roy & Mohanty, 2019). Additionally, the surface charge of microalgae and cyanobacteria, which is estimated by their zeta potential, plays a key role in downstream processing by preventing cells from clumping together and leading to a stable cell suspension (Krishnan *et al.*, 2022). This potential can fluctuate significantly, depending on factors such as cell age and culture conditions (e.g., salinity and pH), and ranges from -5 to -80 mV (Greenwell *et al.*, 2010; Yang *et al.*, 2022; Zhang *et al.*, 2013). Considering the above, harvesting biomass is one of the main obstacles in downstream processing as it requires large amounts of energy and it has been stated that the cost of collecting and drying the biomass from wet cultures is ~ 20 – 30% of the total operational cost of biomass production (Molina Grima *et al.*, 2003; Price *et al.*, 2022).

Highly efficient and minimally damaging methods for separating the biomass from the culture medium are essential during harvesting. There is a plethora of available methods for harvesting microalgae and cyanobacteria (Vasistha *et al.*, 2021), where the appropriate method depends on the characteristics of the microorganism, the properties of the culture medium, and the intended application of the harvested biomass. Furthermore, it is a widespread practice to combine two or more methods to achieve a higher separation efficiency while reducing the costs involved (Barros *et al.*, 2015). Next, we discuss various separation techniques for microalgal and cyanobacterial biomass harvesting.

11.4.1.1 Centrifugation

Centrifugation methods use force to separate particles based on the different densities between the particles. This allows microbial cells, that are denser than the culture medium, to settle (Pahl *et al.*, 2013). Centrifugation is one of the most common harvesting methods on lab scale and can be applied to most microalgae and cyanobacteria. Compared to gravity sedimentation, centrifugal force accelerates sedimentation, leading to a higher biomass recovery efficiency. Additionally, centrifugation eliminates the need for chemicals (e.g., flocculants), which could decrease the quality of the biomass. However, the high energy requirements (up to 8 kWh/m³) limit its large-scale application to high-value products (Barros *et al.*, 2015; Laamanen *et al.*, 2016; Pahl *et al.*, 2013).

11.4.1.2 Filtration

Membrane filtration is a commonly used technique for biomass separation and can be considered a viable harvesting option. During this process, the liquid fraction of the culture is allowed to pass through a porous membrane, usually by applying pressure or a vacuum to the system, while the cells are retained. The ability of solute or solid to pass through a particular porous membrane is dependent on its dimensions, electrical charge, and morphology. Additionally, factors such as the viscosity and mixing rate of the suspension can impact this process (Mathimani & Mallick, 2018). Due to the relatively low-energy requirements (0.2–0.88 kWh/m³) and cost, combined with the ease of scalability, this method is highly advantageous (Pahl *et al.*, 2013). Also, similar to centrifugation, no chemicals are needed, thereby avoiding the qualitative degradation of the recovered biomass. However, the accumulation of microalgal deposits on the filter, leading to fouling (or clogging) of the membrane is the primary limitation of these methods, and it raises their operational costs. Membrane fouling is

primarily caused by extracellular polymeric substances (EPSs), which are organic compounds secreted by microalgae during their growth or released upon cell lysis (Singh & Patidar, 2018).

11.4.1.3 Flocculation and coagulation

Flocculation/coagulation is an economical method for harvesting microalgae and cyanobacterial biomass due to large culture volumes and the need for a universal process that can be applied to various species. Flocculation/coagulation involves the use of inorganic (e.g., $\text{Al}_2(\text{SO}_4)_3$, FeCl_3 , and $\text{Fe}_2(\text{SO}_4)_3$) or organic (e.g., poly(diallyldimethylammonium chloride), PDADMAC) salts, which work by neutralizing the negative charges of cells, resulting in clustering of particles, allowing the suspension to concentrate up to 100 times (Mubarak *et al.*, 2019; Singh & Patidar, 2018; Vandamme *et al.*, 2013). Combining this technique with gravity sedimentation reduces the energy demand of the overall operation, leading to an economically viable harvesting process (Barros *et al.*, 2015). However, a major disadvantage of using aluminum or iron salts as flocculants is that any remaining chemicals can be a potential environmental and health hazard. Also, the use of organic flocculants appears to negatively affect the levels of unsaturated fatty acids in the recovered biomass (Laamanen *et al.*, 2016). In recent years, several studies have been conducted on bioflocculation, in which microalgae cluster together with various microorganisms, including bacteria, fungi, or other microalgae (Kumar *et al.*, 2023). The above procedure can be carried out with the use of bioflocculants, which are usually EPSs produced by several microorganisms (Moreira *et al.*, 2022).

11.4.1.4 Gravity sedimentation

One of the simplest methods for liquid–solid separation is gravity sedimentation. Although this form requires low operating and designing costs, the fluctuating densities and consequently the low sedimentation rates (0.1–2.6 cm/h) of most microalgae, make the process relatively time-consuming, with the risk of degrading the collected biomass (Barros *et al.*, 2015; Greenwell *et al.*, 2010). Therefore, in most cases, gravity sedimentation takes place after a flocculation/coagulation step (Chatsungnoen & Chisti, 2016). Finally, the high self-sedimentation property of some species, such as cyanobacteria *Chlorogloea fritschii*, *Phormidium* sp., and microalga *Golenkinia* sp., eliminates the need for additional energy and reduces the cost and time required to harvest biomass (Hotos *et al.*, 2023; Monshupanee *et al.*, 2016; Nie *et al.*, 2018).

11.4.1.5 Flotation

Flotation is another separation technique based on air or gas bubbles that adhere to the surface of the particles, achieving their transport to the surface and facilitating the separation (Pahl *et al.*, 2013). Furthermore, some cyanobacteria float on their own, as they possess intracellular gas vesicles (aerotopes) (Duval *et al.*, 2021). Flotation is often combined with flocculation/coagulation techniques for optimal harvesting results. Flotation cells are typically supplied with air via dispersed air, dissolved air, or electrolytic mechanisms. Currently, the most common flotation methods are dissolved air flotation (with bubble diameters less than 100 μm), dispersed air flotation or foam flotation (bubble diameters between 100 and 1,000 μm), electrolytic flotation, and dispersed ozone flotation (Barros *et al.*, 2015). The efficiency of smaller sized gas bubbles increased, compared to larger bubbles, as they possess a larger surface area per unit volume. The larger the surface area, the greater is the chance of collision between air bubbles and particles (Pahl *et al.*, 2013). Qi *et al.* (2022) achieved a harvesting efficiency of 96% for the microalga *Tribonema* sp. using flotation, with a significantly lower amount of energy (0.19 kWh/kg biomass) compared to other harvesting methods.

11.4.2 Drying

The extraction techniques for most bioplastics (especially PHAs) from microalgae and cyanobacteria presupposes the drying of the biomass, as the residual water can have a significant effect on their efficiency. Thus, a reliable drying method such as freeze drying, convective drying, spray drying, or

solar drying is necessary (Levett *et al.*, 2016). It is estimated that biomass drying can account for up to 20% of the overall cost of producing PHAs from cyanobacteria, posing a barrier to upscaling commercial production (Costa *et al.*, 2019). Solar drying is an inexpensive dehydration technique but requires extended drying periods, due to the low temperature, and a large land area. In addition, the slow dehydration rate can promote bacterial growth and consequently degradation of the microalgal biomass (Chen *et al.*, 2015). However, in closed solar systems an increase in the drying rate can be achieved, leading to drying of the biomass in 3–5 h at a temperature of 60°C (Prakash *et al.*, 1997). Lyophilization (freeze drying) and spray drying are techniques commonly used to remove water from microalgal biomass. Unlike convective drying, these methods preserve all cellular components without damaging the cell wall (Chen *et al.*, 2015). Spray drying is generally considered more advantageous than lyophilization due to its faster drying speed, ability for continuous operation, and lower cost, but there is a greater possibility of oxidation of carotenoids (Zhang *et al.*, 2022). Nevertheless, until now there is no evidence on how each drying method affects the structure and physicochemical characteristics of the recovered PHAs from microalgal/cyanobacterial biomass.

11.4.3 Extraction

Recent research efforts have focused on developing extraction techniques that reduce the overall cost of producing bioplastics. The commonly used extraction methods are based on organic solvents, usually halogenated. Chloroform and dichloromethane are commonly used solvents as they dissolve bioplastics but no other biological products (Levett *et al.*, 2016). After biomass dehydration, the disruption of the cell membrane takes place, which can be achieved using organic solvents or physical stress, so that the solvent can come into contact with the PHA granules, which are trapped intracellularly (Mastropetros *et al.*, 2022). Additionally, a pretreatment step could be applied prior to extraction to enhance the recovery, usually using sodium hypochlorite (Kosseva & Rusbandi, 2018). Following the extraction of bioplastics from the dry biomass, a suitable solvent such as methanol is used for the recovery, and partial purification of the product, a method known as liquid antisolvent precipitation. Although organic solvents are effective in creating a product with minimal reduction in the molecular weight of the polymer, they are costly and are an environmental hazard (Kosseva & Rusbandi, 2018). Therefore, new environmental-friendly, sustainable, and profitable technologies are needed to scale up and commercialize bioplastic production.

Biomass hydrolysis could be a potential method for recovering biopolymers. By using an acid or base solution, the cells are hydrolyzed, leaving the bioplastic granules undissolved (López-Abelairas *et al.*, 2015). However, some chemical compounds used for this process seem to have a negative effect on the molecular weight and characteristics of the recovered bioplastics (Mastropetros *et al.*, 2022). To prevent such issues, the use of enzymes (e.g., trypsin, bromelain and lysozyme) has been proposed, because they can denature the cell wall during biomass treatment without degrading PHAs. Enzymatic methods for PHA extraction typically involve a heat pretreatment and enzymatic hydrolysis (Kapritchkoff *et al.*, 2006).

Supercritical fluids, such as supercritical CO₂, have been suggested as substitutes for organic solvents for extracting and purifying PHAs. More specifically, supercritical CO₂ can extract up to 90% of the PHA content at purity levels ranging from 86 to 99% and can be used as a secondary step to remove oily biomass residues and refine the bioplastics (Kosseva & Rusbandi, 2018; Mastropetros *et al.*, 2022). However, the high operational costs associated with supercritical fluid extraction and purification processes have impeded their widespread implementation. Nevertheless, the non-hazardous, non-combustible, and low-reactivity nature of supercritical fluids makes them an attractive alternative to organic solvent extraction methods (Mastropetros *et al.*, 2022).

There are various biodegradable, eco-friendly, and recyclable solvents that can be used for the extraction and purification of PHAs, including alcohols, acetone, ketones, and ethylene carbonate. Dimethyl carbonate is another green solvent that shows good performance and does not cause degradation of PHAs such as halogenated solvents. Ethylene carbonate is also used to recover a higher

quantity of PHAs without causing degradation (Kurian & Das, 2021). A recent study compared various solvents and found that dimethyl carbonate is a more environmentally friendly and less hazardous choice for PHA extraction from biomass (Koller, 2020). In addition, ionic liquids are being increasingly favored as a solvent for extraction, and they have the potential to replace traditional organic solvents, as they behave similarly because of their electrically charged ions (Mastropetros *et al.*, 2022). It has been noted that the use of ionic liquids as solvents for extraction offers the benefit of being able to recover the ionic liquid, thereby increasing the viability of the process (Dubey *et al.*, 2018).

11.5 CHALLENGES AND FUTURE PERSPECTIVES

One of the main bottlenecks in the widespread adoption of PHAs from microalgae and cyanobacteria is their accumulation at low percentages on a dry weight basis. Therefore, strategies to enhance the productivity of PHAs are necessary. Process optimization by controlling the cultivation conditions, such as light intensity, pH, and temperature can improve microalgal growth and PHA accumulation. Supplementation of organic carbon sources (e.g., simple sugars) and nutrient limitation (e.g., nitrogen or/and phosphorus starvation) have also been reported to increase PHA accumulation by microalgae and cyanobacteria (Costa *et al.*, 2019). Strain improvement via genetic engineering could be another option to enhance the production of PHAs. For instance, genetic modification of *Synechocystis* sp. enhanced PHB production up to 35% in dry cell weight (Sirohi *et al.*, 2021). However, there are concerns related to the safety and ethical implications of using genetically modified microorganisms (Chia *et al.*, 2020; Sirohi *et al.*, 2021).

The industrial application of PHA production is still limited due to its high cost. Despite the technological advances, PHA costs 5€/kg compared to the production cost of synthetic plastics, which ranges from 0.8€ to 1.5€/kg. One way to reduce the production cost is to reduce the cultivation cost. The feedstock used for the cultivation of microalgae and cyanobacteria represents more than half of the production cost. The utilization of wastewater streams as raw materials seems to be a promising alternative as they are widely available and enriched in organic carbon and nutrients that microalgae and cyanobacteria need for their growth and the production of valuable compounds. This approach will not only diminish the cost of PHA production but also contribute to the bioremediation of wastewaters. However, several issues need to be addressed. The feedstock composition strongly affects the type and yield of PHA produced. The combination of different streams of wastewater or their dilution with water can assure its constant characteristics and decrease the turbidity caused by suspended particles. Furthermore, cultivation in wastewater can affect the end-life of the produced PHA as it may contain impurities that could potentially compromise the biocompatibility of the resulting plastics (Khatami *et al.*, 2021; Medeiros Garcia Alcântara *et al.*, 2020). As discussed in Section 11.3.1.1, microalgae and cyanobacteria can photoautotrophically accumulate PHAs using CO₂ as the sole carbon source. The capture of flue gases, which are rich in CO₂, for PHA synthesis can reduce the production costs while promoting CO₂ mitigation and the reduction of greenhouse gases with several environmental benefits (Sirohi *et al.*, 2021).

The downstream processing is also a critical and costly step in the production of PHAs from microalgae and cyanobacteria. The properties, purity, and yield of the produced PHAs, apart from the potential for specific microalgae to produce them, also depend on the extraction methods used. The most common strategy is the extraction of PHAs with organic solvents such as chloroform and acetone. However, this method creates waste and need extra costs. Therefore, it is necessary to investigate alternative extraction methods, such as enzymatic ones, and the use of different solvents that are recyclable to establish downstream processes that are both cost-effective and environmentally sustainable without affecting the efficiency of the process (Kurian & Das, 2021).

The implementation of a biorefinery concept is a sustainable and economically viable method for PHA production. Cultivation of microalgae and cyanobacteria using carbon flue gases and wastewater for the production of PHAs and other value-added compounds has several environmental and

economic benefits. The downstream processing in this approach is still challenging as the separation of the various compounds is difficult and a high amount of energy is required. In conclusion, PHA production from microalgae has the potential to be a more sustainable and environmentally friendly alternative to petroleum-based plastics. However, several challenges need to be addressed to enable cost-effective and scalable PHA production technologies.

11.6 CONCLUSION

To enable a more sustainable future, the transition toward the utilization of biodegradable bioplastic materials derived from renewable sources is necessary. Microalgae and cyanobacteria are promising candidates for PHA production which can have similar thermal and mechanical properties to conventional plastics. Considering that the downstream processing significantly affects the yield and the properties of the produced PHAs, further research is required to optimize the extraction methods as well as to decrease the dependence on organic solvents. However, the industrial application of bioplastics is still limited due to their high cost. Ongoing research has focused on enhancing the PHA productivity and reducing the cost of the process. PHA production is possible during the cultivation of microalgae and cyanobacteria in various types of wastewaters and side-streams, which could increase the sustainability of the process. Valorization of wastewater and CO₂ from flue gases in the cultivation of microalgae and cyanobacteria to produce PHAs and other valuable co-products such as biofuels and pigments can be the key to the application of bioplastics on an industrial scale.

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Chapter 12

Processes and biorefinery approach for enhanced algal bioproduct recovery in the form of lipid and UV protectant

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ABSTRACT

This chapter discusses various methodologies for lipid extraction, including solvent extraction, enzymatic treatment, ultrasonic aided extraction, and supercritical carbon dioxide extraction, underscoring the need for further research and optimization for large-scale applications. The chapter further explores the potential symbiotic relationship between algal fuel production and waste treatment. This strategy effectively utilizes microalgae's natural ability to thrive in adverse conditions and sequester CO₂ and other pollutants. This approach can simultaneously reduce the environmental footprint while generating valuable biomass for biodiesel production. Another noteworthy point the chapter brings forward is the ability of microalgae to produce valuable compounds under environmental stress, particularly UV radiation. The UV-absorbing compounds such as mycosporine-like amino acids (MAAs) and scytonemin, present substantial potential for use in the cosmetic and pharmaceutical sectors due to their potent UV absorption and photoprotective properties.

12.1 INTRODUCTION

The idea of a 'biorefinery' has arisen as a set of integrated processes for turning microalgal biomass into fuel and other high-value products (Cherubini, 2010; Thomassen *et al.*, 2018). A more sustainable and cost-effective strategy that just concentrates on fuel production is made possible by the diverse and complementary outputs (Salama *et al.*, 2018). Based on current capital costs per unit of fuel production, the generation of biofuel from microalgae is not economically viable. As a result, producing high-value co-products is necessary to increase a microalgae biorefinery's profitability.

Microalgae are microbial factories that can produce a variety of substances besides lipids for biodiesel, having a lipid (7–23%), carbohydrate (5–23%), and protein (6–52%) composition (Chandra *et al.*, 2014). Microalgae can be an excellent source of raw materials for commercially significant value-added products utilized in the food, nutraceutical, cosmetic, and pharmaceutical industries (Haznedaroglu *et al.*, 2016). An integrated biorefinery can maximize product outputs from a single biological source, capturing the value of numerous components (Oh *et al.*, 2018).

The concept of a biorefinery was inspired by petroleum refineries, which provide fuels, oils, and other materials used in the chemical industry (Roux *et al.*, 2017). A biorefinery uses a series of interconnected processes to utilize all the components of the raw materials without causing any loss or harm to the finished goods. In an algae-based biorefinery, there are major hurdles to the sustainable extraction of these chemicals when taking green chemistry principles into account (Yellapu *et al.*, 2018). Maximizing microalgae biomass utilization requires a lot of energy, while utilizing the least amount of energy is still the key goal (Bakonyi *et al.*, 2018). For instance, the Department of Energy's (DOE) primary goal, as stated in the outlook provided in the U.S. multi-year program plan, is cost reduction to produce algal biofuel (Barry *et al.*, 2016).

In this chapter, the most recent research on how to effectively use algal biomass in a sustainable way by using biochemical processes and a bio-refinery technique is discussed. In addition, the framework enables an algal bio-refinery to effectively create value-added products like oil and UV protectant. This chapter also provides a thorough overview of current advancements in the processing of algal biomass utilizing various sustainable methods in an integrated biorefinery.

12.2 FERMENTATION

12.2.1 Selective fermentation

Despite its benefits, microalgal biofuel has not been commercially successful in part because of technical and financial difficulties with algae harvesting and lipid extraction. Pretreatment techniques including pulsed electric fields (PEFs), ultrasound, and acid/alkaline hydrolysis can be effective but are typically too energy-intensive and therefore expensive (Lai *et al.*, 2014; Laurens *et al.*, 2015; Sheng *et al.*, 2011a; Zbinden *et al.*, 2013). To lessen risks to the environment and workers, the present 'gold standards' for lipid extraction, Folch (1:1 chloroform:methanol) and Bligh & Dyer (1:1:0.5 chloroform:methanol:water), must be replaced by non-toxic 'green' solvents. Hexane and isopropanol mixed 1:1 (v/v) is an illustration of a non-toxic solvent (Lai *et al.*, 2014, 2016a, 2016b).

A revolutionary biological strategy for simplifying and improving the economics of lipid extraction is called selective fermentation (SF) (Lai *et al.*, 2016a, 2016b). SF takes advantage of the fact that, under anaerobic conditions, lipids typically biodegrade more slowly than do carbohydrates and proteins. Because they grow slowly, lipid-fermenting bacteria (Christ *et al.*, 2000) can be removed from a reactor with a short solids retention time (SRT) by their washout (Lai *et al.*, 2016a, 2016b). As a result, SF permits the fermentation of carbohydrates and protein in microalgae cells while leaving lipids unaltered. Yet this results in a condition that is much easier to extract because the 'protection' provided by the carbohydrates and proteins has been removed (Lai *et al.*, 2016a, 2016b). Protein fermentation may be a bottleneck in SF since it proceeds more slowly than carbohydrate fermentation (Lu *et al.*, 2012).

The process of biohydrogenation, which transforms long-chain fatty acids (LCFAs) into saturated forms like C18:0, C16:0, and C14:0, is another advantage of SF. Because they have a higher energy content, a higher-octane number for improved combustion efficiency, and a stronger resistance to oxidation, saturated fatty acids are advantageous for the production of transportation fuel (Knothe, 2011). There are two main ways that biohydrogenation can take place. One method is the direct conversion of unsaturated bonds to saturated bonds. In this process, H₂ serves as the electron donor and the LCFA molecule's carbon content remains constant (Lai *et al.*, 2016a, 2016b). An example of direct biohydrogenation is the reduction of C18:1 to C18:0.

Low H₂ concentrations can thermodynamically limit direct biohydrogenation, whereas high H₂ concentrations can accelerate direct biohydrogenation (Cavaleiro *et al.*, 2016; Lai *et al.*, 2016a, 2016b). Strains of the genera *Butyrivibrio* and *Pseudobutyrvibrio* (both in the order Clostridiales) can directly biohydrogenate C18:2 n-6 and C18:3 n-3 to C18:0 (John Wallace *et al.*, 2006; Van De Vossenberg & Joblin, 2003). The family Porphyromonadaceae (Order Bacteroides) and Ruminococcaceae (Order Clostridiales) are involved in direct biohydrogenation in ruminants, according to in vivo research (Castro-Carrera *et al.*, 2014; Huws *et al.*, 2011).

The second method involves the beta-oxidation process, which converts an unsaturated LCFA into a saturated LCFA with the loss of two C atoms as acetate (Cavaleiro *et al.*, 2016). An example of the second route is transformation of C18:1 to C16:0.

As H₂ is produced during beta-oxidation, this pathway does not require an external source of H₂. In theory, processes that utilize H₂, acetate, or both might give this method of biohydrogenation a thermodynamic boost (Cavaleiro *et al.*, 2016).

Lipid conservation is valuable and varies with the biohydrogenation route. β -Oxidation reduces saturated LCFA chain length by two C atoms per step, and the loss is more substantial if multiple steps of beta-oxidation occur. An example is the transformation from C16:0 to C14:0, which produces 2 moles of H₂ and 1 mole of acetate per 1 mole of C14:0 produced.

12.2.2 Electrofermentation

Anode respiring bacteria (ARB) establish a biofilm on the anode in microbial electrolysis cells (MECs), oxidize short-chain fatty acids (SCFAs), and then extracellularly transmit the extracted electrons to the anode (EET) (Reguera *et al.*, 2005; Torres *et al.*, 2009b, 2009a; Yang *et al.*, 2012). Through the external circuit, electrons move to the cathode, where they are absorbed by water molecules to create H₂, which emerges from the cathode as a gas. The MEC is a potential technique for accelerating protein biodegradation in substrates made of complex organic molecules in case of microalgae (Lu *et al.*, 2012).

Due to the need for pretreatment and the use of harmful solvents, extracting lipids from microalgae has been shown to be both commercially and environmentally unfeasible. By selectively biodegrading proteins and carbohydrates while preserving lipids, SF aids in the resolution of these issues. Electro-selective fermentation (ESF) enhances the fermentation performance through anode respiration in a microbial electrolysis cell (MEC) (Liu *et al.*, 2019). ESF was assessed and compared to SF using biomass from *Scenedesmus acutus*. Even though anode respiration only accounted for 1% of the total electrons supplied, ESF boosted protein breakdown three times more than SF did. Although ESF increased the total lipid loss, it tripled the effectiveness of lipid wet extraction with a non-toxic solvent.

The long-chain fatty acid (LCFA) profile changed from C18:1 to C16:0 and C14:0 as a result of lipid loss caused by beta-oxidation associated with biohydrogenation. Anode-respiring bacteria (ARB) on the ESF anode and protein-degrading bacteria and biohydrogenators in the ESF suspension were highlighted by microbial community analysis. Overall, ESF enhanced the quality of biofuel and lipid extractability.

12.2.3 Coupling SF and electrofermentation

A combination ESF is created with the aid of the MEC and SF. It aids in enhancing protein and carbohydrate conversions while protecting lipids for extraction. A strong biofilm of ARB oxidizes SCFAs quickly in the ESF (Ki *et al.*, 2015; Torres *et al.*, 2007), leading to a low concentration of SCFAs in the anode liquid. By reducing a thermodynamic barrier, a lower concentration of SCFAs should encourage upstream fermentation reactions (Fukuzaki *et al.*, 1990; Jones *et al.*, 2015; Pratt *et al.*, 2012). Unfortunately, this method could potentially speed up beta-oxidation as well, which would lead to a loss of all LCFAs.

H₂ is also an ARB substrate, either directly or indirectly through its homo-acetogenic conversion to acetate (Parameswaran *et al.*, 2009). A well-known strategy for overcoming thermodynamic obstacles to fermentation is scavenging H₂ (Cavaleiro *et al.*, 2016; Parameswaran *et al.*, 2010). ESF may therefore hasten the fermentation of proteins and carbohydrates. The loss of protein could cause the cell membrane to rupture and release intracellular lipids for easy extraction by interfering with the hydrogen bonding between membrane proteins and lipids (Cooney *et al.*, 2009; Sheng *et al.*, 2011b).

12.3 BIODIESEL EXTRACTION FROM MICROALGAE

12.3.1 Pretreatment

Several techniques can be used to algae biomass in order to extract intracellular substances. There are various conversion techniques, but the mechanical-based techniques are among the most significant (Cherubini *et al.*, 2009). The biomass is concentrated once microalgae cultures reach the stationary growth phase, and the desired products can be recovered using either dry or wet biomass. To break down the cellular walls and encourage the release of microalgae components that are not released outside the cell, the initial biomass can be dewatered by centrifugation, which is then followed by a cell disruption technique. The approaches employed typically involve a disturbance, break, or breakdown (Dong *et al.*, 2016). The biomass is then thermally dried to obtain a dried form after the dewatering process, which typically results in a paste-like biomass with a dry weight above 85% (Xu *et al.*, 2011).

12.3.2 Extraction

12.3.2.1 Principle of solvent extraction

One of the primary methods for recovering valuable compounds from microalgae is organic solvent extraction. Based on the polarity of the target chemicals, solvents should be selected. Because TAGs, the primary lipid target for the manufacture of biodiesel, are non-polar molecules, a non-polar solvent is an appropriate option for extraction. The majority of solvent-based extraction methods used to extract lipids from microalgae are based on conventional procedures for extracting plant oils, including organic solvent extraction, the Folch method, and the Soxhlet method. Organic solvents penetrate the cell wall of the microalgae, where they promote swelling and cell rupture, releasing the contents of the cell for further separation steps (Grima *et al.*, 2003). When selecting a solvent to extract lipids from microalgae, the primary factors to take into account are polarity or extractability, lipid solubility, water miscibility (ability to operate in two-phase systems), and low toxicity (Bensalem *et al.*, 2018).

12.3.2.2 Solvent extraction methods

12.3.2.2.1 Folch method

The Folch method, which is the foundation of many solvent extraction techniques currently in use, uses a 2:1 chloroform–methanol mixture to extract intracellular lipids. A cell homogenate is first stirred to equilibrate with 25% volume of saline solution. The lipids are allowed to settle on the top layer of this mixture until biphasic separation (Ranjith Kumar *et al.*, 2015). This procedure requires the breaking of microalgae cell walls as a preliminary step. It was initially intended for animal cells and tissues (Grima *et al.*, 2003).

12.3.2.2.2 Soxhlet extraction

In the Soxhlet extraction (SE) procedure, components of a solid sample that are only partially soluble are transported to a liquid phase (solvent) using a Soxhlet extractor. This method uses hexane and other non-polar solvents to produce neutral lipids. The extraction process involves inserting the solid sample into the Soxhlet apparatus's main chamber in a filter paper thimble. The solvent is then heated to reflux and forced into the main chamber, where the less soluble chemicals are recovered. Due to the recovery of complex lipids and pigments, a greater extraction yield from microalgae can be attained when the extraction solvent polarity increases (Baumgardt *et al.*, 2016). This is a crucial factor to take into account because whole lipid extracts using polar solvents are complicated and contain other metabolites besides lipids. The characteristics of a Soxhlet extraction are the solvent of choice, sample particle size, and extraction time (Sharif *et al.*, 2014). SE is typically done on a small scale in the lab and requires a lot of solvent and a long extraction period.

12.3.2.3 Bligh and Dyer method

The Bligh and Dyer method involves partitioning and extracting lipids simultaneously, with protein precipitation occurring at the interface of two liquid phases. This extraction method is comparable to

the Folch method, but with a different solvent combination composition and ratio. A cell homogenate's lipids are first extracted with a 1:2 solution of chloroform and methanol, and the chloroform phase – which is rich in lipids – is then recovered. Lipids from microalgae are removed and quantified using gravimetry. Both pilot and large-scale operations use this approach (Ranjith Kumar *et al.*, 2015). Instead of using water, this approach can be improved by adding 1 M NaCl to prevent the binding of acidic lipids to denatured lipids. The addition of 0.2 M phosphoric acid and HCl has resulted in shorter separation times. By adding 0.5% acetic acid (v/v), acidic phospholipid recovery has been improved (Ranjith Kumar *et al.*, 2015).

12.3.3 Mechanical methods

Solid shear, cavitation and collapse, PEFs, chemical hydrolysis, enzymatic digestion, subcritical water extraction, high-pressure homogenization, and bead milling are a few techniques used to destroy cells and thus release their content.

12.3.3.1 Milling

Bead milling is the process of breaking down the walls of microalgae cells by agitating and grinding the cells over a surface made of glass beads (Ghasemi Naghdi *et al.*, 2016). A disruption needs beads that are between 0.3 and 0.5 mm in size. Typically, zirconia-silica or zirconium oxide can be used to create the beads. The temperature, biomass concentration, flow rate, agitator movement type, and speed all affect how effectively the process works.

Milling can be carried out using agitated beads or shaken vessels. In the shaking vessel method, a vibrating platform is used to shake the culture vessel, which causes the microalgae cells to migrate and crash into one another. When Ryckeboesch *et al.* (2012) used this technique, they were able to recover 40% of the lipids from a culture of *Phaeodactylum tricornutum*, which was the highest lipid recovery achieved. On the other hand, Zheng *et al.* (2012) used a bead milling vessel to extract 11% of lipids from a culture of *Chlorella vulgaris*. According to Lee *et al.* (2010) agitated beads use a method in which the beads and the culture are stirred around by a rotating agitator inside the culture vessel while also being heated to aid in the disruption process. Using cultures of *Botryococcus* sp., *Chlorella vulgaris*, and *Scenedesmus* sp., the authors employed this methodology and obtained an oil yield between 7.9 and 8.1 g/L.

12.3.3.2 Pressing

One of the traditional techniques for obtaining value-added goods from a variety of sources is the use of presses (Kumar *et al.*, 2020a, 2020b). The mechanical crushing of materials with a very low moisture content is the foundation of this technique. Dried biomass is first put under intense mechanical pressure to shatter and crush the cells, and then it is squeezed to extract the oil. Variations in the pressure force, algae strain, and press and piston arrangement can all increase the extraction efficiency (Kumar *et al.*, 2020a, 2020b). With the gel-press approach, algae are first rinsed before employing diluted alkali to extract the carbohydrates. Centrifugation is used to separate the residues, then they are filtered through porous silica, and finally concentrated using evaporation. The recovered material is extruded into a cold potassium chloride solution using spinnerets, and the threads that have gelled are then compressed to remove water (Amin, 2009).

High pressures are used by shear-based machines like the French press and Hughes press to push a biomass solution through a tiny aperture. The average oil recovery is between 70% and 75% (Kumar *et al.*, 2020a, 2020b). Mechanical crushing is occasionally employed in addition to chemical procedures for better oil recovery. The primary limitations of this technology are that it requires expensive maintenance and is less effective than other mechanical extraction methods (Ranjith Kumar *et al.*, 2015).

12.3.3.3 Freeze-thaw method

Since the loss of volatile lipids owing to evaporation is reduced to a minimum with the freeze-thaw process, lipid extraction from microalgae biomass is favored. By freezing the wet biomass at a temperature of -80°C , the intracellular water crystallizes in this process. The samples are then

thawed, causing the ice crystals to expand and lyse the frozen cells. To maximize yield efficiency, this process is typically used in conjunction with another technique, such as ultrasonication, microwave-assisted extraction (MAE), or bead milling (Esquivel-Hernández *et al.*, 2017; Parfati *et al.*, 2018). Cycles of freezing and thawing must be carefully controlled, though. Unfrozen samples showed a 10% decrease in reproducibility after the first cycle and a further 7% decrease after the second, according to a study of the metabolic profile of marine microalgae after freeze-thawing under standard freeze-storage temperatures (-20°C and -78°C) for 1 and 2 cycles of 7 days each (Chr. Eilertsen *et al.*, 2014).

12.3.4 Enzymatic methods

A mixture of enzymes is used in enzymatic extraction procedures to dissolve the algal cell wall, expel lipid bodies from the cell, and separate the lipid fraction from the lipid/protein matrix. An alternative to mechanical cell destruction is enzymatic lysis. Due to the presence of polysaccharides like cellulose and hemicellulose in algal cell walls and lipids, packed in a sac surrounded by phospholipids, in algal cell walls, the lytic enzymes must be specific for the microalgae species, with cellulase and lipase being the most prevalent (Parfati *et al.*, 2018).

Microalgae lipids can be extracted using the cell disruption method known as aqueous enzymatic aided extraction (AEAE). High selectivity, gentle reaction conditions (neutral pH, incubation from 25°C to 37°C), and the lack of labor-intensive drying processes are noteworthy characteristics (Sierra *et al.*, 2017). The best extraction parameters were determined to be 37°C , pH 5.0, 1.3% cellulase, liquid/solid ratio 15 mL/g, and 5 h. An improved approach for enzymatic lysis combined with thermal treatment for extracting lipids from *N. oceanica*. Up to 28.8% of lipids were produced under these circumstances (Amin, 2009).

Biomass collection, enzyme conditioning and addition, stirring incubation to break down algal cell walls, solvent addition (if necessary), centrifugation, and lipid fraction recovery are the primary steps in the enzymatic extraction of lipids from microalgae (Lee *et al.*, 2010). Moreover, after the removal of lipids, the carbohydrate biomass can be saccharified via enzymatic digestion to produce bioethanol (Parfati *et al.*, 2018).

12.3.5 Physical extraction methods

12.3.5.1 Supercritical fluid extraction

By exerting pressure and temperature above a compound's or mixture's critical point, supercritical fluid extraction (SFE) makes use of a supercritical fluid's solvating capability. Solvent, temperature, pressure, solvent flow rate, extraction time, sample size, usage of a modifier, and particle size are some of the adjustable parameters to take into account for SFE (Sharif *et al.*, 2014).

To avoid using hazardous solvents, supercritical fluid extraction with carbon dioxide (SFE- CO_2) has been used as an alternative green extraction technique (Hernández *et al.*, 2014). SFE- CO_2 has several benefits, including being generally recognized as safe (GRAS) by the Food and Drug Administration (FDA), having a low critical point of CO_2 at near room temperature and relatively low pressure (30.9°C and 73.9 bar), and being ecologically benign (Reverchon & de Marco, 2006). Moreover, CO_2 is converted to gas after depressurization, which allows it to be removed from the sample without leaving any traces of solvent behind. This allows it to be recycled for additional extraction cycles, which has both financial and environmental advantages. Supercritical CO_2 , which is especially helpful for the extraction of biodiesel, is very selective for non-polar lipids like triglycerides and does not solubilize phospholipids (Hernández *et al.*, 2014). Hydrocarbons (hexane, pentane, and butane), nitrous oxide, sulfur hexafluoride, and fluorinated hydrocarbons are some of the additional solvents employed in SFE (Reverchon & de Marco, 2006).

12.3.5.2 Microwave-assisted extraction

MAE depends on the interaction of a dielectric polar substance (such as water) and a rapidly oscillating electric field created by microwaves (Esquivel-Hernández *et al.*, 2017; Moretto *et al.*,

2022). This electric field generates heat as a result of the friction created by the movement of the molecules within and between it. The cell begins to produce water vapor as a result of the heat, which finally ruptures the cell and leads to increased intracellular component leakage and release, driven by the electroporation action (Ghasemi Naghdi *et al.*, 2016). As a result, MAE is recognized as a quick, easy, safe, efficient, and affordable approach for the extraction of lipids that does not necessitate sample dewatering beforehand (Bensalem *et al.*, 2018). Moreover, microalgae processed with microwaves have numerous microfissures in the cell wall, which increases the amount of bio-oil recovered (Šoštarič *et al.*, 2012).

In addition to oil extraction, microwaves can be used to transesterify oils into biodiesel, which is a desirable alternative due to its quick reaction time (15–20 min), low operating costs, and effective extraction of algal oils. The substantial maintenance costs associated with using this technology on a commercial scale are a significant downside (Kumar *et al.*, 2015). The primary factors to be considered for MAE are extraction time, temperature, the process mixture's dielectric characteristics, the solid/liquid ratio, and the kind and concentration of the solvent (Ghasemi Naghdi *et al.*, 2016).

12.3.5.3 Ultrasound-assisted extraction

Using cavitation, ultrasonic-assisted extraction (UAE) can recover oils from microalgae cells (Harun *et al.*, 2010). Little vacuum bubbles with a high intensity are produced in the liquid during the low-pressure cycle. A high-pressure cycle occurs when the bubbles violently collapse once they reach a particular size. Locally, extremely high pressures and fast-moving liquid jets are created during the implosion, and the ensuing shear stresses cause the mechanical breakdown of the cell structure. The extraction of lipids from algae is supported by this outcome (Wei *et al.*, 2008). Solvent diffusion into the cell structure is supported by the high-pressure cycles of the ultrasonic waves. Lipids are more easily transferred from the cell into the solvent when using ultrasound because it mechanically breaches the cell membrane through cavitation shear pressures (Cravotto *et al.*, 2008).

By extending the exposure period and combining polar and non-polar solvents, lipid recovery can be improved. UAE also supports mass transfer and solvent penetration inside the cell to release the contents of the cells into the solvent. UAE may be carried out at low temperatures, which is ideal when dealing with the extraction of compounds that are thermally sensitive (Ghasemi Naghdi *et al.*, 2016).

12.3.5.4 Pressurized liquid extraction

Wet algal biomass is used in the wet lipid extraction procedure along with a corresponding amount of solvent (Al-Jabri *et al.*, 2022; Sathish & Sims, 2012). Although it differs depending on the biomass type, this technique is similar to the wet solvent extraction procedure (see Section 3.2). Biomass is transformed into liquid biocrude through the process of hydrothermal liquefaction in hot, compressed water (Biller *et al.*, 2012; Zhang *et al.*, 2022). Because the water must remain in the subcritical area to prevent the latent heat of vaporization, processing temperatures vary from 200°C to 350°C with pressures of around 15–20 MPa (Biller *et al.*, 2012). Complex molecules are disassembled and repolymerized to oily substances under these circumstances. This process eliminates the need to dry the feedstock, making it suitable for converting high-moisture biomass like microalgae.

12.3.5.5 Osmotic pressure

A quick shift in the solute concentration around a cell results in a rapid alteration in the transport of water across the cell membrane, which is known as osmotic shock or osmotic stress (Fajardo *et al.*, 2007). The microalgae's cellular contents are released as a result of this shock. The technique works better with strains grown in maritime conditions (e.g. *Nannochloropsis* sp.). To release cellular components for biochemical examination, osmotic stress is also generated (Alami *et al.*, 2021; Shen *et al.*, 2010). *Halorubrum* sp. isolated from saltern ponds can likewise be treated using this technique. Increased lipid productivities and different lipid compositions were observed (Lopalco *et al.*, 2004).

12.3.5.6 PEF technologies

A technique called pulsed electric field (PEF) processing uses brief bursts of a powerful electric field to process cells (Chittapun *et al.*, 2020). Between two electrodes, algae biomass is positioned, and a PEF is applied (Käferböck *et al.*, 2020). The holes in cell membranes are made larger and release their contents when exposed to an electric field (Wang *et al.*, 2023).

12.4 BIOREFINERY

12.4.1 Cyanobacterial biorefineries

The most basic type of photosynthetic microorganisms, cyanobacteria, have a significant potential for the generation of bioenergy as well as high-value food and pharmaceutical items (Thajuddin & Subramanian, 2005). Since it remains a difficult task, extensive research is being done to turn cyanobacterial lipid into a significant industrial process (Patnaik & Mallick, 2015). In recent years, the process of scaling up and commercializing cyanobacterial or microalgal products has begun, but cautious and methodical development is required to make it a sustainable industrial process.

At an industrial scale, cyanobacteria have enormous promise for recovering value-added products and biofuels. They are rich in lipids, susceptible to metabolic engineering, and contain value-added components like antioxidants (Esquivel-Hernández *et al.*, 2017), phycobiliproteins (Chandra *et al.*, 2017), UV protectants (Rastogi & Incharoensakdi, 2014), and vitamins (Esquivel-Hernández *et al.*, 2017). This makes it a potential feed stock for biorefinery (Sheng *et al.*, 2011b; Vermaas, 1996). Despite the fact that cyanobacteria are ideally suited for biorefining due to the diverse composition of their biomass, recovering co-products from cyanobacteria remains a difficult problem. Therefore, it is necessary to investigate moderate and sequential extraction techniques that maintain the usefulness of different cell compounds such as UV protectant, protein, vitamins, lipid, and carbohydrates.

12.4.2 Cyanobacterial biorefinery products

12.4.2.1 Biodiesel

A possible renewable feedstock for the manufacture of value-added goods and ethanol appears to be cyanobacteria. A comprehensive assessment of multiple product recovery is required to guarantee the economic and sustainability of biofuel production. Chandra *et al.* (2019, 2020) provided strong evidence in favor of producing mycosporine-like amino acids (MAAs) and high-quality biodiesel in succession. A procedure in which *Lyngbya* biomass was sequentially collected from all experimental variations after UV irradiation and treated with 100% HPLC-grade methanol for 12 h at 4°C before being centrifuged at 4000 rpm for 15 min at 4°C. The supernatant was gathered, dried at 38°C, and combined with the pellet for lipid extraction. It was determined that the dried methanolic residue is MAAs. To remove the photosynthetic pigment, this MAA was washed with chloroform and water. Together with the residue from the previous stage, the aqueous phase used to collect MAAs and the chloroform phase were both treated for lipid recovery. According to this method, recovering UV protectant after UV exposure with biodiesel is a more sustainable solution for high fuel productivity. The manufacture of algae biodiesel gains value as a result of this procedure. The problem of heterotrophic bacterial contamination is lessened by UV exposure, and the lipids content and saturation index of biodiesel are increased. This results in a biorefinery that is both affordable and sustainable (Chandra *et al.*, 2020).

12.4.2.2 UV protectant

Due to its great market demand and value, UV protectant could be a significant product of an algal biorefinery. For instance, it is predicted that by 2024, the global demand for this kind of goods will increase to \$13.2 billion from more than \$7.6 billion in 2012 (Oilgae, 2014). By doing so, it is possible to enhance the value of the process, the number of products with added value, and the environmental impact all at once. Because of their special makeup, microalgae can contain a variety of pigments, such as UV filters and carotenoids, astaxanthin, lutein, zeaxanthin, phycocyanin, and phycoerythrin.

Oil accumulation in microalgae is known to be significantly affected by exposing cells to environmental stress (ultraviolet radiation) (Arakaki *et al.*, 2017), nutrient depletion (Pancha *et al.*, 2014), salinity (BenMoussa-Dahmen *et al.*, 2016), oxidative stress (Yilancioglu *et al.*, 2014), temperature or pH changes. In order to reduce the stress condition, these parameters also have an impact on the other cell components and trigger the production of new molecules. Ultraviolet radiations (UVR) are a key method in this regard to generate UV inhibitors like mycosporin-like amino acids (MAAs) and investigate their impact on lipid productivity (Chandra *et al.*, 2019). By manufacturing MAAs and scytonemin, cyanobacteria are known to defend themselves against photochemical harm (Chandra *et al.*, 2020). They reside in the sheath of cyanobacteria and are lipophilic. Because the growth medium is not changed and heterotrophic contamination may be minimized, UVR provides various benefits for oil production.

A range of defense mechanisms are used by the cyanobacteria and marine algae group to endure and thrive under high UV fluxes. These organisms synthesize UV-absorbing substances like mycosporine-like amino acids (MAAs) and scytonemin as a mitigation mechanism (Chandra *et al.*, 2019, 2020). By scavenging significant amounts of reactive oxygen produced by supersaturated oxygen in deep, light-exposed water, MAAs display substantial antioxidant activity. The 3-dehydroquinate and 4-deoxygadusol precursors of the shikimate pathway are the sources of the main MAA, mycosporine-glycine (Chandra *et al.*, 2019). Mycosporine-glycine is converted into secondary MAAs via the addition or subtraction of amino acids as well as metabolic processes. However, it has been discovered that the cyanobacterium *A. variabilis* has an MAA biosynthetic gene cluster that transforms sedoheptulose-7-phosphate to shinorine via 4-deoxygadusol and switches the precursor 3-dehydroquinite (Balskus & Walsh, 2010). Tryptophan and tyrosine, two aromatic amino acids that are byproducts of the shikimate pathway, are thought to be the source of scytonemin. Moreover, a cluster of genes for scytonemin production has been found, and UV-A activation of these genes has been demonstrated (Rastogi *et al.*, 2015). MAAs and scytonemin can be used as active ingredients in the cosmetic and pharmaceutical sectors due to their potent UV absorption and photoprotective qualities.

12.5 CONCLUSION

Establishing a sustainable algal biomass-based biorefinery requires multidisciplinary research on biorefinery methodologies. A multi-product, integrated, and sustainable approach is crucial for efficient product recovery and process development. Microalgae biomass serves as a flexible feedstock for biodiesel production, utilizing techniques such as photobioreactors, fermenters, and open raceway ponds. Various lipid extraction techniques have been explored, including solvent extraction, enzymatic treatment, ultrasonic-aided extraction, and supercritical carbon dioxide extraction. Optimization work is needed for large-scale applications, particularly for supercritical carbon dioxide extraction (SFE-CO₂), PEFs, osmotic shock, hydrothermal liquefaction, and wet lipid extraction. Integrating algal fuel production with wastewater and waste treatment enhances economic viability. UV exposure stimulates UV defense synthesis and lipid productivity in *Lyngbya*, with UVB favoring fuel qualities and UVA benefiting food properties. The recovery of lipids and UV protectants from the same feedstock promotes cost-effective and environmentally responsible options in a sustainable biorefinery.

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